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Experimental Assessment of the Relationships Between Electrical Resistivity, Crustal Melting and Strain Localization beneath the Himalayan-Tibetan Belt

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Abstract: Seismic and magnetotelluric field campaigns carried out across the Himalaya and the Tibetan Plateau show mid-crustal low resistivity and low velocity zones. Interpretation of these anomalous observations, either saline fluids or partial melts, is still largely debated partly because experimental data simulating crustal melting under relevant pressure, temperature and water content conditions have not been provided. We report laboratory measurements constraining the resistivity as a function of temperature and the viscosity at 800°C of natural metapelite during partial melting. Dehydration-melting of muscovite is simulated using a Paterson press at 300 MPa in the temperature range 600-850°C. Electrical resistivity has been measured from the solid state (<650°C) to 25 vol. % of partial melt (>800°C) and viscosity has been determined at 800°C. Our results together with recent experimental constraints on seismic properties of partially molten rocks strongly suggest that the electrical and seismic anomalies measured underneath the Himalayan-Tibetan collisional orogen are best explained by partially molten rocks or local accumulation of pure melt bodies. This is also remarkably corroborated by the temperature-depth conditions of crustal partial melting and melt ponding expected from petrological surveys in the Himalayan range. However, our data suggest much higher melt fraction than previously thought and this implies regions in the middle crust having viscosities several orders of magnitude lower than previously assumed. High degree partial melting in the middle crust of the Himalayan-Tibetan orogenic system suggests revision of conceptual models on the development of mountain belts or that geophysical models addressing electrical resistivity at depth must be re-evaluated.

Keywords: Electrical resistivity; viscosity; partial melting

1. Introduction

The construction of mountain belts resulting from continental collisions involves crustal thickening, regional deformation and high-grade metamorphism. As observed in many

ancient orogenic terrains, elevated temperatures locally induce melting and strain weakening, which may profoundly affect the rheology of the continental crust, its deformation regime and hence the development of these orogens (Jamieson et al., 2011; Sawyer et al., 2011).

The Himalaya-Tibetan system is an active collisional belt allowing us to probe the three-dimensional thermo-mechanical distribution of an archetypal continent-continent orogen. The Himalaya-Tibetan orogenic system was initiated 70-50 My ago by the Indo-Asian collision resulting in crustal thickening and uplifting of the Tibetan Plateau (Royden et al., 2008; Yin and Harrison, 2000). This vast plateau is separated from the Himalayan mountains by the Indus-Tsangpo suture. Two major faults are located south of this suture: the Main Central Thrust (MCT), which is a low angle high strain crustal shear zone, and the South Tibetan Detachment system (STDs), a low angle normal fault (Fig. 1). The MCT and the STDs bound the Greater Himalayan Sequence (GHS), composed of medium to high-grade metamorphic rocks, migmatites (i.e. once partially molten rocks) and km-thick leucogranites pods (Law et al., 2004; Searle et al., 2006). These High Himalayan Leucogranites (HHL, Fig. 1), late Oligocene to Miocene in age (25-13 Ma), have been shown to be products of partial melting (750-800°C) of underlying metapelites (Law et al., 2004; Patiño Douce and Harris, 1998; Scaillet et al., 1995; Searle et al., 2006). However, north of the Indus-Tsangpo suture young granitoids (< 15 Ma) are typically biotitegranodiorites with ³He anomalies indicating a mantle contribution (Harrison, 2006; Hoke et al., 2000; Yokoyama et al., 1999).

Several magnetotelluric (MT) and seismic field campaigns (INDEPTH, Hi-CLIMB and HIMPROBE projects) have identified low resistivity layers with well-constrained tops at a depth of 10-15 km in southern Tibet (3 Ω.m) and 20-25 km in northwestern Himalaya (10 Ω.m), coincident with low seismic velocity zones (Arora et al., 2007; Brown et al., 1996; Caldwell et al., 2009; Hetényi et al., 2011; Li et al., 2003; Makovsky and Klemperer, 1999;

Nábělek et al., 2009; Nelson et al., 1996; Unsworth et al., 2005; Wei et al., 2001; Zhao et al., 1993). Such electrical anomalies have been interpreted as evidence of high fluid concentrations and two main hypotheses are still largely debated today: either aqueous fluids (Hetényi et al., 2011; Li et al., 2003; Makovsky and Klemperer, 1999) or melts (Arora et al., 2007; Brown et al., 1996; Caldwell et al., 2009; Gaillard et al., 2004; Li et al., 2003; Nábělek et al., 2009; Nelson et al., 1996; Unsworth et al., 2005; Wei et al., 2001). However, both hypotheses remain poorly demonstrated (Harrison, 2006; Hetényi et al., 2011), each having considerable impact on the thermal structure of the thickened crust in the Himalayas. The melt hypothesis suggests low melt percentages, between 2 and 14 vol.%, from electrical constraints (Arora et al., 2007; Unsworth et al., 2005), which would be in agreement with seismic observations (Caldwell et al., 2009) (less than 10 vol.% for $V_S = 2.9\text{-}3.3 \text{ km.s}^{-1}$). Recent experimental shear wave velocity measurements (Caricchi et al., 2008), however, indicate that such velocities better match 20-25 vol.% partial melting. Experimental studies on the electrical resistivity of partially molten rocks have been carried out on dry granulite samples at atmospheric pressures (Roberts and Tyburczy, 1999; Schilling and Partzsch, 2001). Under these conditions, partial melting starts at temperatures higher than 1000°C, which is well above expected crustal temperatures in continent-collisional settings. Melts produced by crustal melting are hydrous (Scaillet et al., 1995), which implies lower melting temperatures and lower electrical resistivity of rocks and melts (Gaillard et al., 2004). Notwithstanding, resistivity values on hydrous partially molten metapelites are still not well constrained from electrical modelling such as SIGMELTS (Pommier and Le Trong, 2011).

In this experimental study, we have performed electrical resistivity measurements as a function of temperature and determined the viscosity of metapelites undergoing partial melting at 800°C. These experiments were conducted using a Paterson-type deformation apparatus on partially molten metapelites. By using these types of rocks in this study, we have

simulated the appropriate geological processes happening at depth according to previous petrological studies in the Himalayan-Tibetan range (Law et al., 2004; Patiño Douce and Harris, 1998; Scaillet et al., 1995; Searle et al., 2006). We therefore show here that the geophysical anomalies underneath this orogenic system must pinpoint high degree crustal melting and local melt accumulation, forming low-viscous regions, which are several orders of magnitude weaker than previously assumed (Beaumont et al., 2001, 2004, 2006; Jamieson et al., 2004; Medvedev and Beaumont, 2006). These results suggest revision of conceptual models on the development of mountain belts.

2. Materials and Methods

2.1. Starting Material

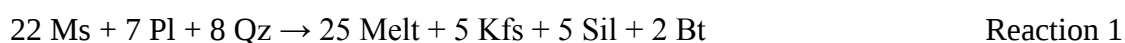
The starting material is a natural metapelitic rock from the Sioule metamorphic series (two-micas paragneiss formation, Supplementary Fig. 1), which belongs to the central metamorphic domain of the European Hercynian Belt (Faure et al., 2002). The Hercynian chain resulted from the collision of Gondwana and Laurussia in the Late Devonian-Early Carboniferous. The Sioule metamorphic series show strong evidence of partial melting and leucogranite emplacements similar to the Himalayan systems (Faure et al., 2002).

The composition of the Sioule sample we used (Table 1) is essentially similar to the Himalayan metapelite samples used by Patiño Douce and Harris (1998). The starting material contains Na-plagioclase, muscovite, biotite, quartz and K-feldspar in order of abundances (Table 1). Its composition and texture indicate no involvement in partial melting events and did not suffer any associated chemical depletions (in alkalis and water). Small amount of graphite (<0.5 vol.%) has been observed in thin sections. The bulk composition, i.e. major elements and water content, was obtained from electron microprobe (EMP) and Fourier transformed infrared (FTIR) analyses, respectively, on a glass that has been rapidly quenched after total melting of this rock at high-pressure high-temperature (4 kb, 1170°C, 21 hours).

This melting experiment was conducted in an internally heated pressure vessel at Institut des Sciences de la Terre d'Orléans (ISTO) (see Iacono-Marziano et al., 2012). The total water content of the glass analysed by FTIR is 1.75 wt. % and a large amount of dissolved CO₂ has been observed (1000 ppm CO₂). This CO₂ most likely derives from graphite oxidation during high temperature melting.

Oxygen fugacity of all the experiments presented in this paper was buffered by the coexistence of graphite and CO₂ in the sample (observed after the runs). This corresponds to redox conditions close to FMQ-2 and mimics natural conditions of crustal melting that also occurs at graphite saturation (Pichavant et al., 1996).

Published experiments (Patiño Douce and Harris, 1998) and our results show that the dehydration-melting of muscovite starts at 650°C at 300 MPa. Previous experimental work established the following reaction (Patiño Douce and Harris, 1998):



with Ms standing for muscovite, Pl for plagioclase, Qz for quartz, Kfs for K-feldspar, Sil for sillimanite and Bt for biotite.

Reaction 1 indicates that water, liberated upon muscovite breakdown, induces partial melting and incorporates into the melt, with about 1.14 volume of melt produced for 1 volume of muscovite. As our sample contains 23 vol. % Ms, completion of reaction 1 induces 26.2 vol. % melt in the rock.

2.2. Electrical resistivity measurements

The conventional Paterson assembly was modified in order to measure the resistance of the samples (Fig. 2). The sample-electrode geometry is similar to the configuration used by Caricchi et al. (2011). For resistivity measurements, an inner Ni electrode was inserted in the centre of the sample (perpendicularly to the layering configuration) and an outer Pt foil was

placed around the samples. The resistance of the samples was measured along their radii, between the inner Ni electrode and the outer Pt foil. The compositional difference between the two electrodes had no effect on the collected resistance values. The electrical resistivity, ρ ($\Omega\cdot\text{m}$), was obtained from the resistance, R (Ω), using (Caricchi et al., 2011; Gaillard, 2004; Pommier et al., 2008):

$$\rho = (2 \cdot \pi \cdot L \cdot R) / (\ln(d_{\text{ext}} / d_{\text{int}})) \quad (1)$$

with L the length of the sample (m), d_{ext} the outer diameter of the sample (m) and d_{int} the diameter of the inner electrode (m).

Electrical resistivity measurements were collected solely under static conditions (i.e. without applied deformation) in the temperature range of 500 to 850°C at 300 MPa and at frequencies between 1 Hz and 1 MHz. Error on the calculated resistivity values is estimated to less than 7% except for run PP216 where the maximum error is of about 28% for temperatures < 690°C (Table 2). These estimations are based on the geometrical constraints of the samples (width, length, inner diameter) and on repeated electrical measurements at the same experimental conditions.

The electrical response of the samples is plotted at each frequency in the Nyquist plane ($-Z'$, Z'') since the complex impedance (Z) can be written as the sum of real (Z') and imaginary (Z'') components (where $Z = Z' + jZ''$ and $j^2 = -1$). In this plane, the electrical response is an arc centred on the real axis where the first part of this arc (high frequencies) represents the electrical response of the sample whereas the second low-frequency arc is related to the sample/electrode interactions (Huebner and Dillenburg, 1995; Supplementary Fig. 2). The impedance arcs from this study can be fitted by a parallel circuit of a resistance and a capacitance (Huebner and Dillenburg, 1995; Roberts and Tyburczy, 1991).

2.3. Rheological measurements

Deformation experiments were performed at a pressure of 300 MPa and at a temperature of 800°C. During dynamic experiments, strain rates, $\dot{\gamma}$, range from $1 \cdot 10^{-5}$ to $2.5 \cdot 10^{-4} \text{ s}^{-1}$ and bulk finite strain, γ , reaches a maximum value of 0.91 (Table 2). Strain rates were controlled by applying a continuous torque during the experiments. Stepping strain rate experiments were also conducted in order to calculate the stress exponent, n , enabling us to calculate the stress τ (MPa) from the applied torque (Nm). Stepping strain rate consists of increasing strain rate in a stepwise manner once steady state stress conditions were achieved for a given strain rate (Paterson and Olgaard, 2000). Apparent viscosities, η_a (Pa.s), of our samples were calculated for a given strain rate by using the following equation (Paterson and Olgaard, 2000):

$$\eta_a = \tau / \dot{\gamma} \quad (2)$$

with τ the shear stress (Pa) and $\dot{\gamma}$, the strain rate (s^{-1}).

2.4. Analytical techniques and imaging

After the static and dynamic experiments, the samples were cut along longitudinal axial sections (Fig. 7 in Paterson and Olgaard (2000)) and then mounted in epoxy. A scanning electron microscope (SEM) was used for microscopic observations at back-scattered electron mode at ISTO (see Supplementary Figs. 3, 4 and 5). Melt distribution and volumetric percentages were determined by manually selecting the melt phase (consisting of glass and crystals formed from reaction 1) on SEM images of the radial faces of the samples (Supplementary Figs. 3 and 4). Binary images were created with solely the melt appearing in black. Volumetric percentages were then calculated using a ratio of pixels of melt over the total pixels of the studied binary image (Table 2).

Chemical compositions of bulk rock and mineral phases as well as the melt produced by dehydration-melting of muscovite crystals were obtained by EMP analyses. The water

content (in the glass) of the bulk rock was analysed by FTIR following Iacono-Marziano et al. (2012).

3. Experimental Strategy

To better constrain electrical resistivity values during our Paterson experiments, we have conducted partial melting experiments in internally heated pressure vessels at ISTO (PM experiments, Supplementary Materials). These partial melting experiments have enabled us to constrain the onset of partial melting in the temperature range 650-700°C (Supplementary Table 1; Supplementary Fig. 5), which is in agreement with published phase equilibria experiments on similar mineralogical assemblages (Patiño Douce and Harris, 1998). In the latter study, the associated melt production, via reaction 1, is completed at $T = 820^{\circ}\text{C}$.

Partial melting of metapelites has also been shown to be an extremely slow process. Previous studies reveal that run durations of one week are not sufficient to achieve equilibrium melt percentages at temperatures lower than 800°C (Rubie and Brearly, 1990). Indeed, Rubie and Brearly (1990) have shown that the melting kinetics of a reaction involving muscovite, quartz and H_2O is controlled by diffusion of Si, Al and K through the melt for temperatures $< 760^{\circ}\text{C}$. The width of the reaction zone between muscovite and quartz is a square root function of time and is modelled by a parabolic law, which indicates that at least two weeks are required to fully dissolve small muscovite grains. Textural and crystal-melt chemical equilibrations at $T < 800^{\circ}\text{C}$ thus operate at rates that are not compatible with the duration of experiments in Paterson apparatus.

Our time-series resistivity measurements performed at temperatures ranging from 650 to 825°C (PP187, PP202, PP205; Table 2) didn't show any clear electrical stabilization. We have interpreted this behaviour as a record of slow chemical and textural evolution of the samples during partial melting (Supplementary Fig. 6). To counteract this effect and obtain chemical and textural equilibrium, the temperature was increased to 850°C (PP213, PP214,

PP216; Table 2), enhancing reaction kinetics and leading to nearly electrical steady state within ~ 300 min (Supplementary Fig. 7). Such high temperature ensured the rapid and total melting of muscovite. For the reason explained below (see section 5.1), the results used in this study for interpreting MT anomalies beneath the Tibetan-Himalayan range are essentially based on those from experiments with peak temperatures of 850°C (PP213, PP214 and PP216; Table 2). During these experiments, we linearly increased the temperature to 500°C and then increased by 50°C steps up to 850°C , enabling us to perform resistivity measurements at different constant temperatures. After annealing the samples at 850°C , the temperature was decreased to 800°C , which corresponds to established temperatures of melt extraction as suggested by geochemical, experimental and field observations in the Himalayas (Patiño Douce and Harris, 1998; Searle et al., 1997). Long static resistivity measurements or viscosity experiments were thus conducted separately at 800°C after the annealing stage. Electrical resistivities were then measured during cooling cycles (Fig. 4 and Supplementary Fig. 8).

4. Results

4.1. Chemical composition of the produced melt

EMP analyses show that the melt produced during our partial melting experiments is granitic in composition and analogous to the High Himalayan Leucogranites (Scaillet et al., 1995; Table 1). It has typical high SiO_2 contents (>70 wt. %) and high water contents (>6 wt. % based on the difference to 100%, see Table 1). This composition was analysed in run PP214 after equilibration at a peak temperature of 850°C . Liquid compositions analysed in other experiments were relatively similar, but the low melt fraction did not allow us to obtain precise chemical analyses.

4.2. Electrical resistivity

Replicated experiments showed a good reproducibility during heating stages (Fig. 3) in spite of the variable sample geometries (length, layering configuration). As temperature increases from 550 to 825°C, the electrical resistivity of the metapelites decreases dramatically from ~ 2000 to $20 \text{ } \Omega\cdot\text{m}$. The onset of partial melting, optically identified from the PM-experiments presented in section 3, is also indicated by a slope shift (i.e. a change in electrical properties) between 650 and 700°C on Fig. 3.

Figure 4 presents the resistivity measurements from experiment PP216 during both heating and cooling cycles (after a peak T at 850°C). As illustrated on this figure, the resistivity/temperature path is shifted toward less resistive values during cooling cycles. The difference between heating and cooling cycles is significant but remains reasonably low except for temperatures $< 650^\circ\text{C}$, where this discrepancy reaches at least one order of magnitude. Furthermore, between 850 and 550°C during the cooling cycle, melt percentage remains constant at 23 vol. %.

4.3. Viscosity

Image analyses (Supplementary Figs. 3 and 4) revealed no migration and accumulation of melt throughout the deformed samples. The samples deformed homogeneously since no strain localization was observed.

We have determined the stress exponent, n , of our material, corresponding to the slope of a logarithmic plot of shear strain rate versus maximum measured torque (Supplementary Fig. 9). Despite our small strain rate/torque data set, the calculated n ($n = 5.44$) is in good agreement with a previous study on the rheology of synthetic metapelitic samples undergoing partial melting ($n = 5.15$; Misra et al., 2009).

The mechanical behaviour of our metapelitic samples is summarized in a shear stress versus shear strain plot (Fig. 5). These rheological measurements show a rapid linear increase, corresponding to the elastic response of the stressed sample, followed by either steady state

flow or moderate weakening during plastic deformation (Fig. 5). Peak stress values vary with peak temperature: as temperature increases, melt percentage increases, thus weakening the metapelitic sample. The significant peak stress value difference between runs PP187–PP202 and runs PP213–PP216 is therefore directly related to their corresponding peak temperatures (825°C and 850°C, respectively). Figure 5 and Table 2 also indicate good reproducibility for equivalent strain rates and peak temperatures (experiments PP213 and PP216 for $T_{\text{peak}} = 850^{\circ}\text{C}$ and $\dot{\gamma} = 2.5 \cdot 10^{-4} \text{ s}^{-1}$).

Experiments PP187 and PP202 ($T_{\text{peak}} = 825^{\circ}\text{C}$ and $\dot{\gamma} = 1 \cdot 10^{-5} - 1 \cdot 10^{-4} \text{ s}^{-1}$) present calculated apparent viscosities, η_A , ranging from $10^{12.32}$ to $10^{11.38} \text{ Pa.s}$, whereas experiments PP213 and PP216 ($T_{\text{peak}} = 850^{\circ}\text{C}$ and $\dot{\gamma} = 1 \cdot 10^{-5} - 2.5 \cdot 10^{-4} \text{ s}^{-1}$) present η_A ranging from $10^{11.69}$ to $10^{10.56} \text{ Pa.s}$ (Table 2).

5. Discussion

5.1. Equilibrated melt percentage/temperature/resistivity paths

At 300 MPa during heating cycles, melt percentage increases continuously from 0 vol. % at 550°C to 23 vol. % at 850°C (Fig. 4; Supplementary Table 1), indicating slow melting kinetics because melt fraction is expected to abruptly increase once muscovite breakdown temperature is exceeded. Likewise, during cooling cycles, the low resistivity/temperature paths were interpreted as slow kinetics of crystallization, as melt percentage remains at 23 vol. % even at the lowest temperature (500°C, Fig. 4). We have thus calculated the variations of equilibrium melt percentage as a function of P and T by creating a P-T pseudosection (Supplementary Fig. 10), using the collection of thermodynamic modelling programs Perple_X 6.6.6. (Connolly, 1990, 2005; Connolly and Kerrick, 1987; Connolly and Pettrini, 2002). The bulk composition of the system used for calculations is that of our metapelitic sample together with 1.75 wt. % water. Melt percentage is modelled by using the program PyWerami (<http://petrol.natur.cuni.cz/~ondro/pywerami:home>, Fig. 6). At 300 MPa, the

calculated equilibrium melt percentage increases from 0 to more than 15 vol. % within a few degrees as a consequence of the dehydration-melting of muscovite in fluid absent conditions. This strong and abrupt increase in melt mode is not observed during heating cycles of the present study due to a sluggish melting rate. The P-T pseudosection (Supplementary Fig. 10) indicates that above the solidus temperature ($T_S = 650^\circ\text{C}$ at 300 MPa), the partially molten samples in our experiments are in disequilibrium, thus overestimating the electrical resistivity values. Measurements collected during the cooling cycles therefore represent a very close assessment of the resistivity of equilibrated partially molten metapelites at temperatures just above the temperature of muscovite dehydration-melting ($650^\circ\text{C} < T < 700^\circ\text{C}$ at 300 MPa).

Therefore, in our experiments ($P = 300$ MPa), we considered that the equilibrated resistivity/temperature path should follow the solid-state path during the heating cycle until the solidus temperature is reached ($650^\circ\text{C} < T_S < 700^\circ\text{C}$). Above T_S , resistivity should be shifted within a few degrees Celsius to values corresponding to 23 vol. % partial melting during the cooling cycle (bold dashed line at 300 MPa for $T > 650^\circ\text{C}$ in Fig. 4). The equilibrated resistivity/temperature path varies with pressure, as T_S increases with increasing pressure (from 675°C at 500 MPa to 745°C at 900 MPa; Fig. 6). This T_S /pressure-dependence therefore shifts the abrupt decrease in electrical resistivity to higher temperatures. As an example, the equilibrated resistivity/temperature path at 700 MPa is shown to abruptly decrease by 1-log unit at $T_S = 720^\circ\text{C}$ (Fig. 4).

5.2. Electrical resistivity of solid and liquid end-members

Since HHL pods are presently outcropping in the GHS (Fig. 1), we have calculated the electrical resistivity of crystal-free leucogranite magma bodies as extrapolated from our partial melting experiments. Electrical resistivity is commonly estimated by using mixing models where the bulk resistivity of the sample depends on the electrical resistivities of the

present phases (in this study, melt and solid), as well as their relative volumes and geometrical distributions in the sample.

We were able to extrapolate the electrical resistivity of the produced melt by using our experimental and previously published data and mixing models (Table 3; Gaillard and Marziano, 2005; Yoshino et al., 2012). For brevity, only a few mixing models were used in this section: Archie's law, which is an empirically derived relation (Archie, 1942); the modified Archie's law, which is derived from the Archie's law using geometrical constraints (Glover et al., 2000); Hashin-Shtrikman upper bound (Hashin and Shtrikman, 1962); the cubes model (Waff, 1974) and the tubes model (Grant and West, 1965; Schmeling, 1986).

The electrical resistivity of the solid was adjusted from the low temperature data (<650°C) collected during our experiments as follows:

$$\ln \rho_s = -\ln \rho_0 + (-E_a / (RT)) \quad (3)$$

with

$$\rho_0 = 2.90 \pm 0.20 \text{ and } E_a / R = -8581.80 \pm 600.73$$

where ρ_s is the resistivity of the solid rock in $\Omega.m$; ρ_0 , the pre-exponential term in $\Omega.m$; E_a , the activation energy in J; R , the gas constant; T , the temperature in K.

The graphite in our sample shows no noticeable effect on the electrical resistivity of our solid-state samples during heating stages (Fig.4). Indeed, measured conductivities are typical of solid-state crustal rocks (Ferri et al., 2013; Schilling and Partzsch, 2001), whereas interconnected graphite in our sample should enhance conductivities by several orders of magnitude (Shankland et al., 1993; Yoshino and Noritake, 2011). This experimental observation on natural metapelite is in agreement with previous experimental studies performed on graphite-rich xenoliths (Ferri et al., 2013) and C-rich synthetic quartz aggregates (Yoshino and Noritake, 2011).

We used the bulk resistivity value collected at 800°C during the cooling cycle, once the samples presented an electrical steady state ($\rho_B = 11.97 \text{ } \Omega\cdot\text{m}$) to calculate the resistivity of the melt. Image analyses showed that this sample contained 23 vol. % partial melting.

Calculations show that melt resistivities are essentially similar for Archie's law, the Hashin-Shtrikman upper bound and the cubes model (between 1.87 and 2.13 $\Omega\cdot\text{m}$, Table 3). Melt resistivity values calculated using the tube model and the modified Archie's law are however slightly lower and higher, respectively (0.97 and 2.71 $\Omega\cdot\text{m}$, respectively, Table 3). Electrical resistivity of hydrated granitic melts was also calculated following equations by Gaillard (2004):

$$\rho = (\sigma_0 * \exp (-E_a / (RT)))^{-1} \quad (4)$$

with

$$\sigma_0 = -78.9 * \ln (H_2O) + 754 \text{ and } E_a = -2925 * \ln (H_2O) + 70132$$

where T is the temperature in K and H_2O the water content in wt. % (natural logarithm). For granitic liquids containing 6 wt.% of dissolved H_2O at 800°C, equation (4) yields 2.35 $\Omega\cdot\text{m}$, which is in remarkable agreement with the liquid resistivity independently estimated from partial melting experiments and mixing models.

5.3. Cross-checking petrological, geophysical and laboratory constraints

Laboratory work and thermobarometric studies (Patiño Douce and Harris, 1998) indicate that melting of the Himalayan metapelites occurred at a pressure equivalent to 18-24 km depth and a temperature of 750-800°C (Fig. 7), which according to our experimental data correspond to a resistivity of 10 $\Omega\cdot\text{m}$ (Fig. 8). As shown in Fig. 7, such resistivity-depth-temperature conditions are fully compatible with the northwestern Himalayan electrical anomaly. Furthermore, the high melt percentage deduced from our study yields the

appropriate shear wave velocities observed in this region (Caldwell et al., 2009; Caricchi et al., 2008).

Electrical resistivity was calculated during cooling and crystallization of leucogranites (Fig. 8; Scaillet et al., 1995). Due to the near-eutectic composition of such magmas, the resistivity moderately increases to $\sim 15 \Omega\cdot\text{m}$ as temperature decreases to 650°C (i.e. 30 vol.% crystals), whereas in the temperature range $650\text{--}645^\circ\text{C}$, where the bulk crystallization occurs, the resistivity increases by several orders of magnitude (Fig. 8). Experimental studies show that HHL were emplaced as near-liquidus magma with a depth of $\sim 12 \text{ km}$ and a temperature of about 800°C (Fig. 7; Scaillet et al., 1995), which yields a resistivity of $3 \Omega\cdot\text{m}$ (Fig. 8). Such resistivity-depth-temperature conditions remarkably match the electrical anomaly detected beneath Southern Tibet (Fig. 7). Since shear waves do not propagate through liquid states, we suggest that spatial resolution of seismic analyses may not allow identification of fully molten bodies of limited vertical extents, such as sill-shaped or laccolith bodies.

Magnetotelluric profiles enable us to constrain the depth of the top of the electrical anomaly but can hardly constrain their thicknesses. We have based our experimental modelling exclusively on peak resistivity values from MT data of northwestern Himalaya and southern Tibet (10 and $3 \Omega\cdot\text{m}$, respectively; Aurora et al., 2007; Unsworth et al., 2005). The resistivity values implied in this study yield anomaly thicknesses of about 23 km for northwestern Himalaya and 12 km for southern Tibet (Supplementary Fig. 11). There is however a tradeoff between the thickness of a low resistivity layer and the value of resistivity of that layer. We decided to adopt the values given in Aurora et al. (2007) and Unsworth et al. (2005) therefore knowingly ignoring this tradeoff. We suggest that a more realistic MT-modelling should be carried out by using our experimental electrical results, complemented by seismic data, but such considerations go beyond the scope of the present work.

5.4. Viscosity implications

Presence of partial melting and fully molten bodies has serious impact on the rheological behaviour of the middle crust. Indeed, our results show that crustal rocks with 20-25 vol. % partial melting (corresponding to the electrical anomaly of northwestern Himalaya) imply a viscosity, η , of 10^{11} Pa.s (Table 2). Our data seems in good agreement with the experimental study of Misra et al. (2009), where the apparent viscosity of their synthetic sample yielded a value of $10^{11.62}$ Pa.s at a strain rate of $3 \cdot 10^{-4} \text{ s}^{-1}$, at a temperature of 750°C and for comparable melt percentages.

Pure melts, corresponding to the electrical anomaly beneath southern Tibet, have been shown to present lower viscosities, i.e. 10^4 Pa.s (Scaillet et al., 1996; Whittington et al., 2004). These two values (i.e. 10^{11} Pa.s and 10^4 Pa.s) are several orders of magnitude lower than previous estimations used in thermo-mechanical models of channel flow and crustal extrusion ($\eta = 10^{18}\text{-}10^{19}$ Pa.s) (Beaumont et al., 2001, 2004, 2006; Jamieson et al., 2004; Medvedev and Beaumont, 2006). Such low-viscosity regions provide a natural mechanism for localizing strain into the weakest parts of the deforming layers, as shown by previous field observations and demonstrated by experimental deformation studies (Brown, 2007; Kohlstedt and Holtzman, 2009; Marchildon and Brown, 2003; Rosenberg and Handy, 2005; Vanderhaeghe, 2009). Crustal melting in the Himalayan middle crust must therefore be an essential process in forming low-viscosity regions and thus crustal-scale thrusts and/or high strain shear zones, such as the STDs and the MCT (Fig. 1).

5.5. The aqueous fluid hypothesis in conflict with geochemical and geophysical observations

We have found that the aqueous fluids hypothesis cannot completely satisfy combined constraints from metamorphic fluid salinity, electrical resistivities and seismic velocities. Firstly, the question of the geochemical origin of brines is raised since the presence of Cl-rich sources in the Himalayan-Tibetan crust remains unproved. Secondly, fluid inclusions within

rocks from the GHS show high salinity content (25 wt. % NaCl; Spencer et al., 2011), which implies a resistivity value of $\sim 75 \cdot 10^{-4} \Omega \cdot \text{m}$ (Nesbitt, 1993). By using electrical mixing models (as in section 5.2), we have found that the electrical anomalies beneath the Himalayan-Tibetan orogenic system ($10 \Omega \cdot \text{m}$ - $3 \Omega \cdot \text{m}$ in Northwestern Himalaya and Southern Tibet, respectively) are best explained by ~ 0.8 vol.% of connected brines for the aqueous fluid hypothesis (Supplementary Table 2). Such a low brine percentage cannot explain the corresponding low seismic velocity zone, which requires at least 10 vol.% fluids (Makovsky and Klemperer, 1999).

Furthermore, by using the electrical mixing models we have determined that 10 vol.% interconnected aqueous fluids would yield a resistivity value of $\sim 0.11 \Omega \cdot \text{m}$. This calculated value is two orders of magnitude lower than the values assessed by MT-field campaigns ($10 \Omega \cdot \text{m}$ – $3 \Omega \cdot \text{m}$ in Northwestern Himalaya and Southern Tibet, respectively; Arora et al., 2007; Li et al., 2003; Unsworth et al., 2005; Wei et al., 2001). The aqueous fluids hypothesis therefore yields a substantial discrepancy between seismic and electrical constraints.

6. Conclusion

New electrical resistivity measurements on partially molten metapelites have shown that the electrical anomalies detected by geophysical surveys beneath southern Tibet ($3 \Omega \cdot \text{m}$) and northwestern Himalaya ($10 \Omega \cdot \text{m}$) remarkably match fully molten leucogranite bodies and 25 vol. % partially molten metapelites, respectively. Our study therefore corroborates that the processes (i.e. partial melting of metapelites and subsequent melt segregation) that generated the 25-13 Ma HHL must still be operating at depth beneath the Himalayan-Tibetan belt. Whether these processes are local or regional phenomena remains to be ascertained (Harrison, 2006; Hetényi et al., 2011).

The high melt percentages and corresponding viscosity values acquired during our experiments represent a benchmark for thermo-mechanical models, which must be calibrated

against this inescapable constraint. The implementation of the electrical-seismic-petrological-thermal-rheology relationships made possible with our study constitutes a requirement for understanding which factors in such thermo-mechanical models are most likely to influence orogenic processes and therefore in improving our knowledge of the development and sustainment of large-scale orogens.

We finally suggest that forward modelling based on our laboratory electrical measurements and petrological-thermobarometric models of the Himalayan-Tibetan range should be confronted to field data from MT studies. Such an integrated methodology might lead to a revision of the minimum resistivity, translating into revised melt fractions, viscosity and strain localisations in the mid-crustal range.

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Figure captions

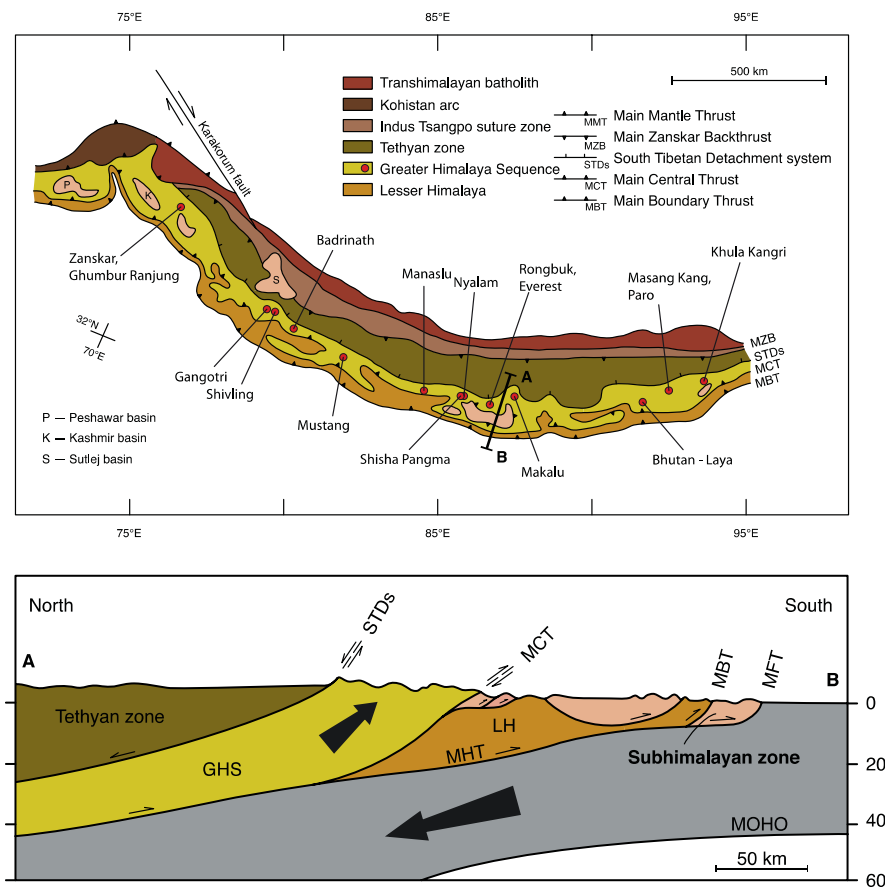


Figure 1. Simplified geological map and cross-section of the Himalaya showing the main High Himalayan Leucogranites (red dots) along the Greater Himalaya Sequence (adapted after Law et al. (2004) and Searle et al. (2006)). Abbreviations - GHS: Greater Himalaya Sequence; LH: Lesser Himalaya; STDs: South Tibetan Detachment system; MCT: Main Central Thrust; MBT: Main Boundary Thrust; MFT: Main Frontal Thrust; MHT: Main Himalayan Thrust; MOHO: Mohorovicic discontinuity.

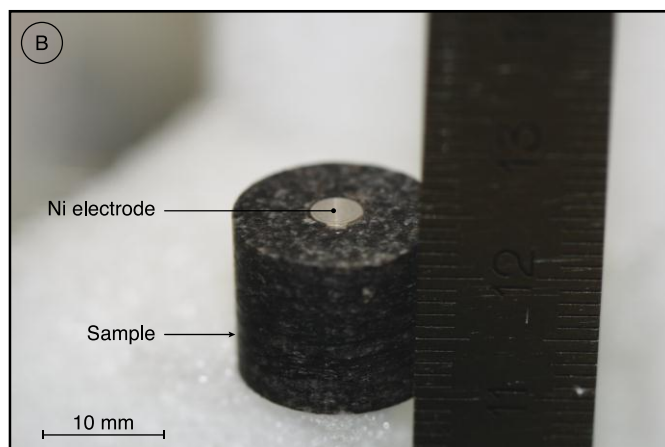
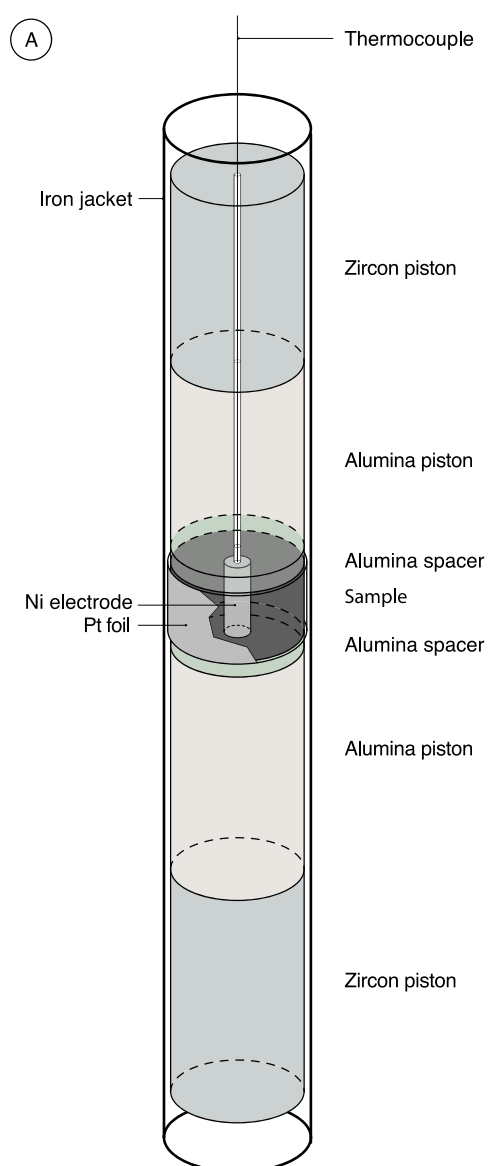


Figure 2. (A) Modified Paterson assembly used for electrical resistivity measurements. (B) Zoom of sample showing inner Ni electrode.

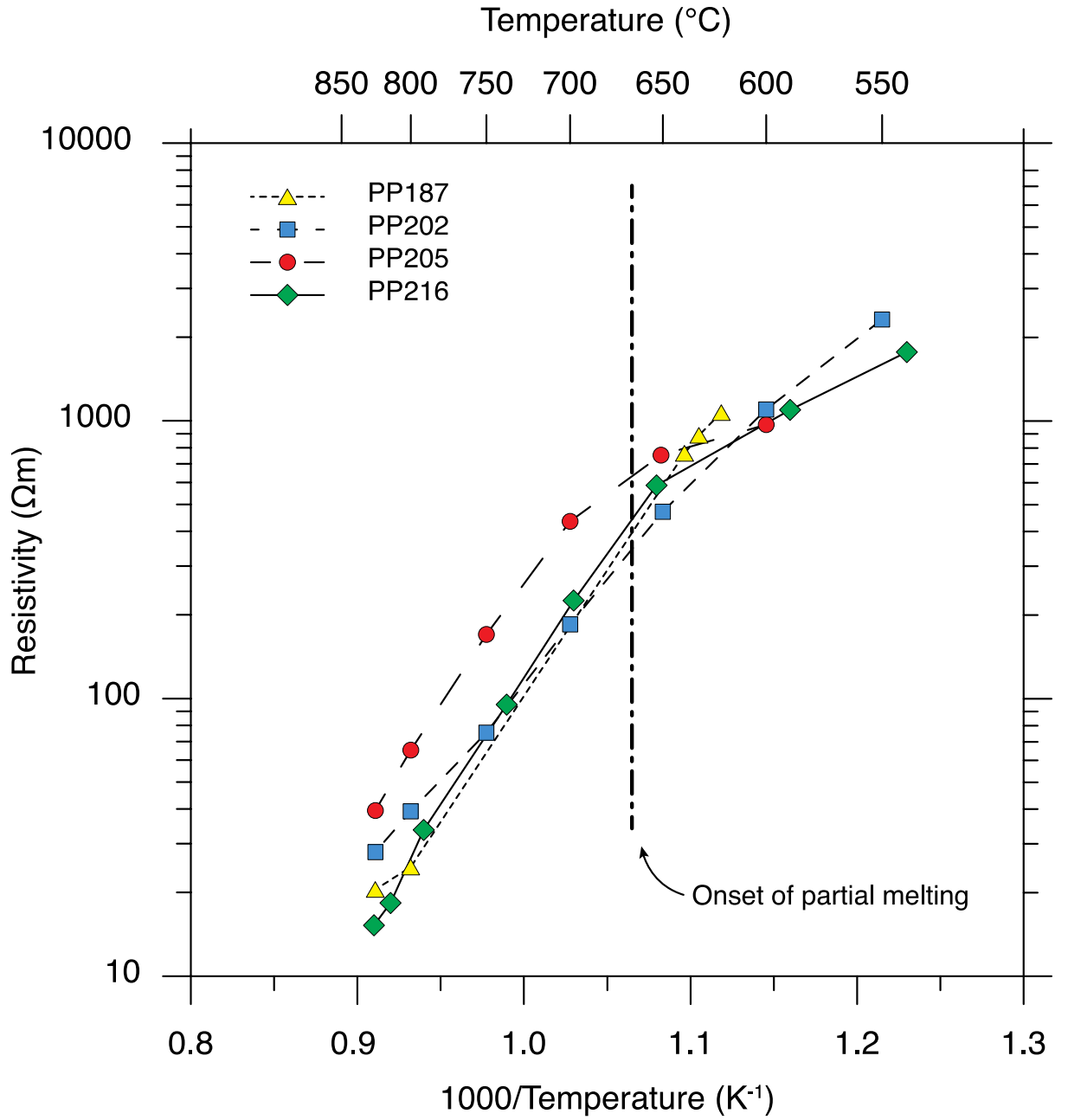


Figure 3. Measured electrical resistivities of runs PP187, PP202, PP205 and PP216 during heating cycles at 300 MPa. The plotted values represent electrical measurements after $t > 60$ min. Error on the plotted electrical resistivity values is within the size of the data points.

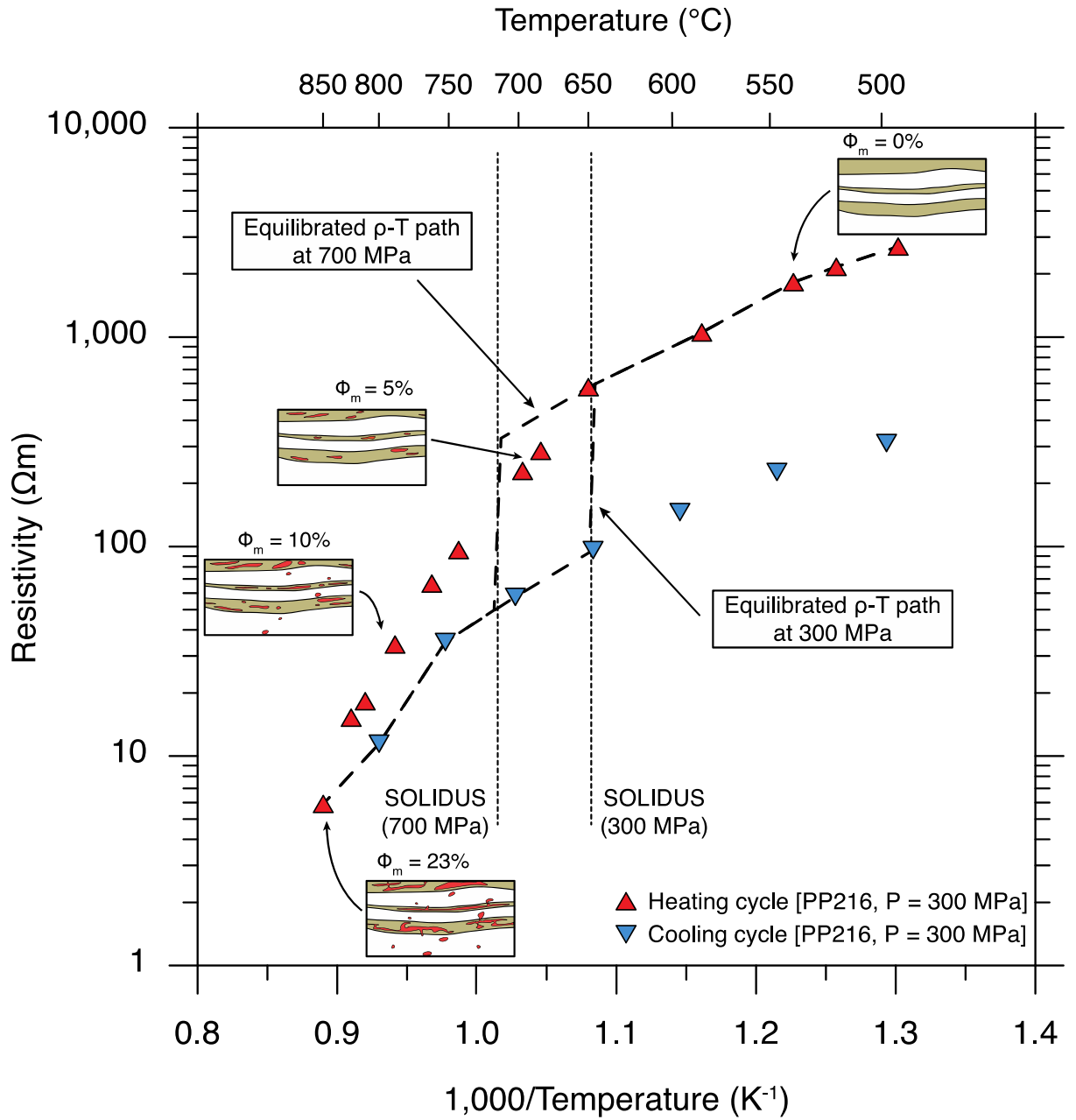


Figure 4. Electrical resistivity vs. temperature of experiment PP216 at 300 MPa confining pressure. During the heating cycle (red triangles), melt percentage increases moderately from 0 to 23 vol. %. During the cooling cycle (blue triangles), the sample remains partially molten (23 vol. % melt), even at the lowest temperature (500°C). This hysteresis cycle shows that the sample is in electrical disequilibrium. The equilibrated resistivity/temperature path (bold dashed lines) should follow the solid-state sample until the solidus temperature (e.g. solidi at 300 and 700 MPa), where the resistivity is shifted within a

few degrees Celsius to lower values corresponding to 23 vol. % partial melting. See text for more details. Error on the plotted electrical resistivity values is within the size of the data points. Abbreviation – Φ_m : melt percentage; P: pressure.

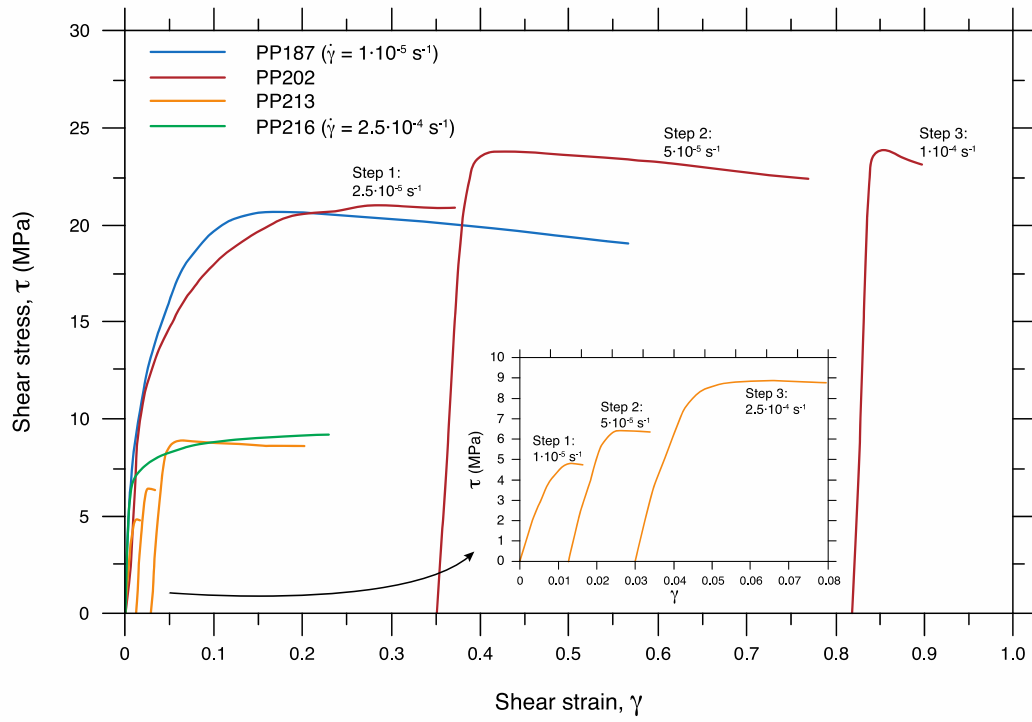


Figure 5. Mechanical behaviour of the metapelite samples during runs PP187, PP202, PP213 and PP216. Experiments PP187 and PP216 were deformed at constant strain rate. Stepping strain rate experiments were undertaken during runs PP202 and PP213.

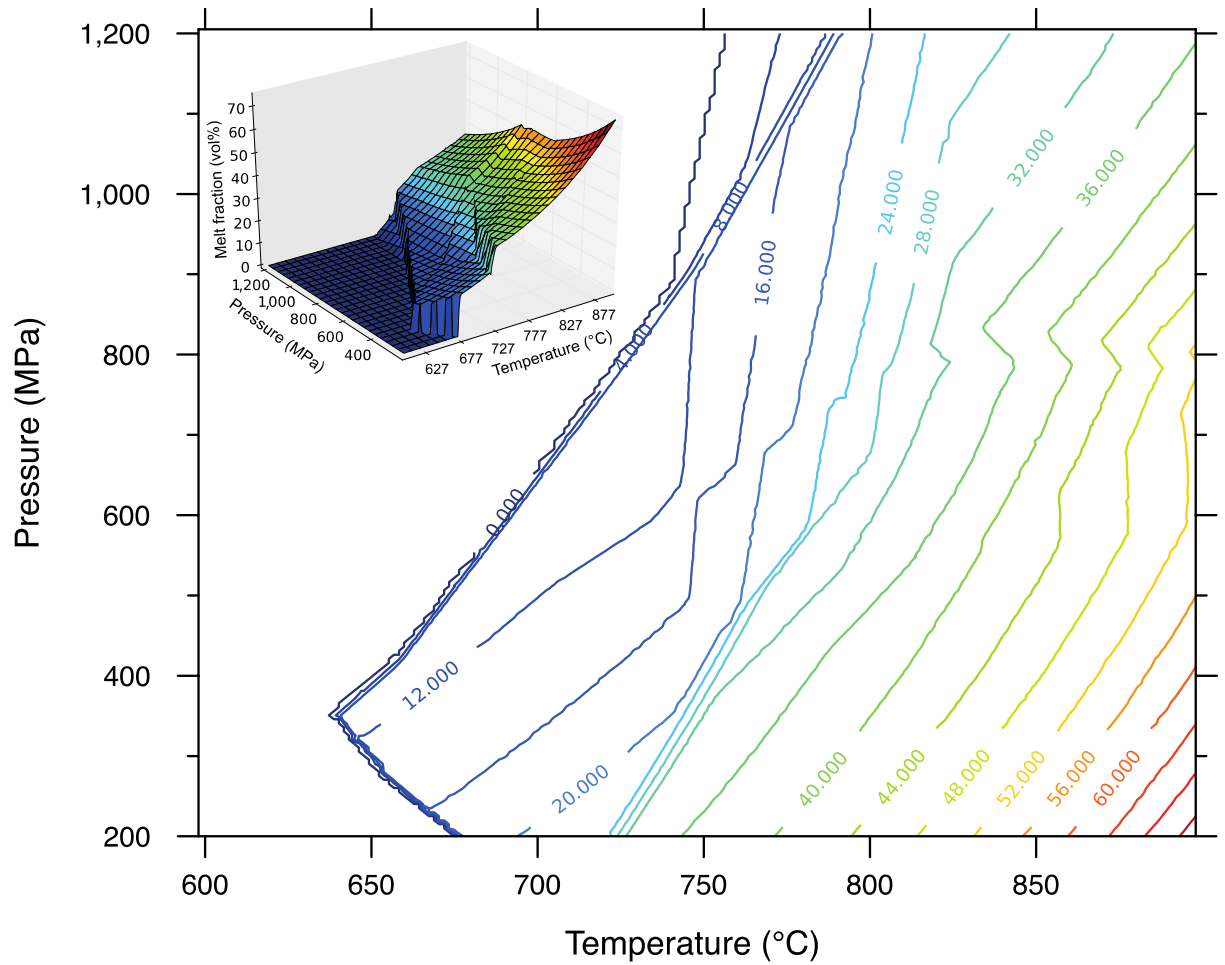


Figure 6. Melt percentages (vol. %) calculated with Perple_X 6.6.6. for the chemical composition of the Sioule sample used as starting material in the present study. The water content of the system was fixed at 1.75 wt. %. The results are plotted in P-T diagrams (2D contours and 3D surface) with the program PyWerami.

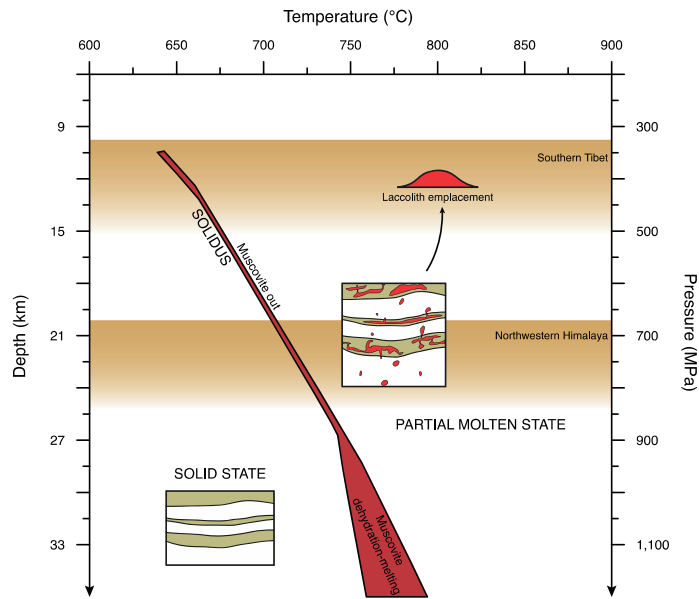


Figure 7. Pressure/depth versus temperature plot for partial melting of Himalayan metapelites. The red zone represents the dehydration-melting of muscovite in fluid absent conditions and isolates solid-state rocks (on the left-hand side of the red zone) from partially molten rocks (on the right-hand side of the red zone). In the Himalayan geological context, metapelites produce granitic melts in the temperature range 750-800°C and pressures between 600-800 MPa (Patiño Douce and Harris, 1998). High Himalayan Leucogranites emplace as near-liquidus magma at 800°C and 400 MPa (Scaillet et al., 1995). Resistivity-depth-temperature conditions for Himalayan partial melting and leucogranite emplacement remarkably match the MT anomalies beneath this orogen.

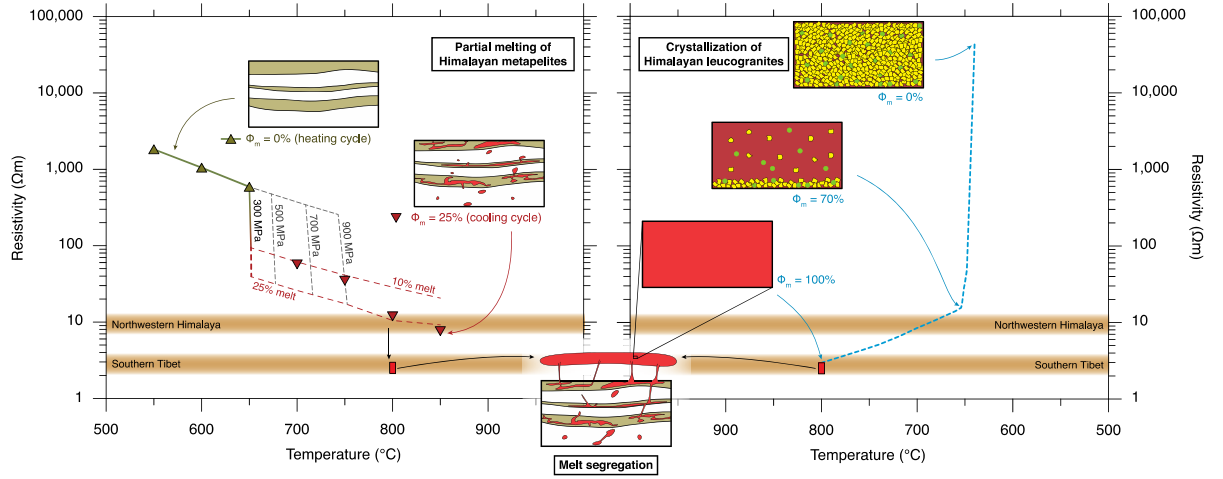


Figure 8. Electrical resistivity versus temperature for partial melting of Himalayan metapelites and crystallization of Himalayan leucogranites. Our experimental data is represented by the triangles in the left panel. At 650°C and 300 MPa, melt percentage increases from 0 to 10 vol.% within 5°C, sharply decreasing the electrical resistivity. With increasing pressure, melting temperature increases, therefore shifting the sharp decrease in resistivity to higher temperatures. Resistivity mixing models are shown in red dashed lines, considering fully connected melt (10 and 25 vol.% melt) in a solid matrix (HS^+). Extrapolation of fully molten bodies at 800°C from our partial melting experiments and mixing models is represented by red rectangles in both panels (lower and upper boundaries calculated using HS^+ and equation 5 by Gaillard (2004), respectively). The blue dotted line in the right panel corresponds to cooling and crystallization of leucogranitic bodies (experimental data from Scaillet et al. (1995) using HS^+). MT anomalies can be explained by either 25 vol.% partial melting of the middle crust at 800°C or leucogranites containing 20 vol.% crystals at 680°C for northwestern Himalaya and by emplacement of leucogranitic bodies at 800°C for southern Tibet. Abbreviation – Φ_m : melt percentage.

Table 1

	Starting Sample						Produced melt	
	Bulk rock	Quartz	K-Feldspar	Plagioclase	Muscovite	Biotite	DK	Melt
Modal abundances	100	22	2	30	23	23	—	—
SiO ₂	62.36 (± 0.26)	100.00	63.63 (± 4.54)	67.25 (± 1.04)	48.27 (± 0.69)	36.41 (± 2.02)	73.04	74.82 (± 3.59)
Al ₂ O ₃	19.27 (± 0.18)	0.00	19.87 (± 3.73)	20.21 (± 0.57)	37.29 (± 0.82)	19.73 (± 1.43)	15.32	14.92 (± 2.39)
Fe ₂ O ₃	—	—	—	—	—	—	0.17	nd
FeO	4.56 (± 0.22)*	0.00*	0.18 (± 0.22)*	0.14 (± 0.08)*	1.73 (± 1.62)*	22.00 (± 1.47)*	0.74	0.66 (± 0.30)*
MnO	0.08 (± 0.07)	0.00	0.01 (± 0.01)	0.03 (± 0.03)	0.08 (± 0.02)	0.32 (± 0.03)	0.01	nd
MgO	3.28 (± 0.03)	0.00	0.04 (± 0.15)	0.01 (± 0.03)	0.74 (± 0.08)	9.14 (± 0.47)	0.20	0.13 (± 0.05)
CaO	1.06 (± 0.07)	0.00	0.03 (± 0.07)	0.88 (± 0.47)	0.00 (± 0.00)	0.08 (± 0.16)	0.85	0.70 (± 0.63)
Na ₂ O	2.63 (± 0.02)	0.00	1.21 (± 1.66)	9.62 (± 2.19)	0.49 (± 0.05)	0.12 (± 0.05)	3.85	2.65 (± 1.15)
K ₂ O	5.33 (± 0.09)	0.00	15.00 (± 2.38)	1.84 (± 3.66)	10.77 (± 0.18)	9.92 (± 0.93)	4.96	5.92 (± 1.68)
TiO ₂	1.23 (± 0.05)	0.00	0.03 (± 0.13)	0.02 (± 0.02)	0.63 (± 0.03)	2.28 (± 0.43)	0.13	0.06 (± 0.04)
P ₂ O ₅	nd	nd	nd	nd	nd	nd	0.14	nd
CO ₂	0.001 [‡]	nd	nd	nd	nd	nd	0.05	nd
H ₂ O	1.75 [‡]	nd	nd	nd	nd	nd	0.65 [¥]	nd
F	0.19 (± 0.02)	nd	nd	nd	nd	nd	0.09	0.14 (± 0.02)
Cl	0.01 (± 0.01)	nd	nd	nd	nd	nd	nd	nd
Total	96.09 (± 0.30)	100.00	97.97 (± 1.75)	98.37 (± 0.64)	93.52 (± 1.80)	91.99 (± 6.12)	100.20	92.49 (± 1.39)

Table 2

Run	Length (mm)	Diameter (mm)	Type	T _{peak} (°C)	$\dot{\gamma}$ (s ⁻¹)	γ	τ_{peak} (MPa)	η_A (Log Pa.s)	Φ_M (vol. %)	Phases	Total run duration (h)	Maximum error on resistivity measurements (%)
PP187	12.40	14.93	Torsion	825	1·10 ⁻⁵	0.57	22.0	12.32	23	Qz + Pl + Kfs + Bt + Ms + Sil + Melt	24.2	7.0
PP202	16.93	14.91	Torsion	825	2.5·10 ⁻⁵ 5·10 ⁻⁵ 1·10 ⁻⁴	0.35 0.77 0.91	22.0 24.6 24.6	11.93 11.67 11.38	24	Qz + Pl + Kfs + Bt + Ms + Sil + Melt	17.0	0.1
PP205	8.33	14.86	Static	825	–	–	–	–	10	Qz + Pl + Kfs + Bt + Ms + Sil + Melt	30.0	2.2
PP213	14.71	14.91	Torsion	850	1·10 ⁻⁵ 5·10 ⁻⁵ 2.5·10 ⁻⁴	0.02 0.03 0.20	4.9 6.5 8.9	11.69 11.11 10.55	30	Qz + Pl + Kfs + Bt + Ms + Sil + Melt	29.6	1.3
PP214	13.23	14.99	Static	850	–	–	–	–	23	Qz + Pl + Kfs + Bt + Ms + Sil + Melt	23.6	0.1
PP216	19.65	14.98	Torsion	850	2.5·10 ⁻⁴	0.23	9.2	10.56	30	Qz + Pl + Kfs + Bt + Ms + Sil + Melt	7.6	28.2

Table 3

Electrical mixing law	Melt resistivity ($\Omega\cdot\text{m}$)
Archie's law	1.87
Modified Archie's law	2.71
Hashin-Shtrikman upper bound	2.13
Cubes model	1.91
Tubes model	0.97