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# Habitability on Mars from a Microbial Point of View

Frances Westall,<sup>1</sup> Damien Loizeau,<sup>2,3</sup> Frédéric Foucher,<sup>1</sup> Nicolas Bost,<sup>1,4</sup>  
Marylène Bertrand,<sup>1</sup> Jorge Vago,<sup>2</sup> and Gerhard Kminek<sup>2</sup>

## Abstract

Extraterrestrial habitability is a complex notion. We briefly review what is known about the origin of life on Earth, that is, life based on carbon chemistry and water. We then discuss habitable conditions (past and present) for established life and for the survival of microorganisms. Based on these elements, we propose to use the term *habitable* only for conditions necessary for the origin of life, the proliferation of life, and the survival of life. Not covered by this term would be conditions necessary for prebiotic chemistry and conditions that would allow the recognition of extinct or hibernating life. Finally, we apply this concept to the potential emergence of life on Mars where suitable conditions for life to start, proliferate, and survive have been heterogeneous throughout its history. These considerations have a profound impact on the nature and distribution of eventual traces of martian life, or any precursor, and must therefore inform our search-for-life strategies. Key Words: Mars—Microbial life—Punctuated habitability. *Astrobiology* 13, 887–897.

## 1. Introduction

OUR UNDERSTANDING of the potential of other planets and satellites in our Solar System to host microbial life has increased considerably in the last decade. The icy satellites of the outer planets no longer appear completely inhospitable, especially when viewed as warmer and more active bodies during their early history. Hypotheses concerning the possibility of life on Mars wax and wane as new data and new models related to its aqueous history appear. We tend to automatically equate the availability of water with the guarantee of conditions conducive to life. However, from a microbial point of view, the situation is very different depending on whether we are dealing with the emergence of life, with established or flourishing life, or with life in a survival or dormant mode. The term *habitable* is misleading when applied to conditions in which prebiotic chemistry can lead to the origin of life. The principal requirement of a prebiotic environment would be the simultaneous coexistence of the ingredients of life and the range of chemical and physicochemical reactions necessary to result in a protolife entity—a chemical reactor of sorts. It is entirely possible that an environment conducive to prebiotic chemical processes leading to the origin of life could have been spatially confined and toxic compared to the present terrestrial environment yet able to provide that first spark, with subsequent evolution occurring eventually in

other locations. Similarly, conditions that would allow life to flourish are likely to be different from those in which the molecular building blocks formed. Thus, in this contribution we introduce a concept for the use of the term *habitable* and apply this to life on Mars.

We start with a review of the general understanding concerning the origin of life on Earth. We then consider the ability of life to establish itself opportunistically in any habitable environment, as well as its faculty for survival. Finally, we extend this analysis to Mars from the point of view of the origin and survival of microbial life, recognizing that conditions for both the origin of life and for habitability on the planet would have been heterogeneous.

## 2. Habitability and Life

### 2.1. The building blocks: availability of prebiotic molecules and water

Life as we know it is based on carbon macromolecules and water. Carbon, oxygen, nitrogen, and hydrogen are among the five most common elements in the Solar System and are essential for all organisms on Earth. Chemical reactions are ubiquitous, and one or more of these elements is a constituent in most molecules found in the Universe. However, the universality of chemistry does not necessarily mean the universality of life, because the criteria for the emergence of life (at least the

<sup>1</sup>CNRS-OSUC-Centre de Biophysique Moléculaire, Orléans, France.

<sup>2</sup>ESA-ESTEC, Noordwijk, the Netherlands.

<sup>3</sup>Laboratoire de Géologie de Lyon, Terre, Planètes, Environnement, Villeurbanne, France.

<sup>4</sup>CNRS-OSUC-Institut des Sciences de la Terre d'Orléans-BRGM, Orléans, France.

carbon- and water-based life that we know) are quite particular. In this respect, environmental conditions have a strong influence on the kinds of molecules that can be formed (Brack, 1999).

In the interstellar medium, more than 150 organic molecules have been detected in dense clouds. Among these, 50 are relatively complex, having more than six carbon atoms. These molecules can be formed in the icy mantles of tiny interstellar grains (Herbst and van Dishoeck, 2009; de Marcellus *et al.*, 2011). Important prebiotic molecules (organic molecules formed without the intervention of life, *i.e.*, abiotically) have been detected, such as formic acid (HCOOH), hydrogen cyanide (HCN), and formaldehyde (HCHO) (Oró, 1961; Chyba *et al.*, 1990; Maurette, 1998; Bernstein *et al.*, 1999; Sephton and Botta, 2008), that are incorporated into the carbonaceous matter of interstellar ices and dust particles, comets, meteorites, and micrometeorites. Such materials would have rained down on the early planets in huge quantities, contributing to the inventory of organics and volatiles, including water (Chyba and Sagan, 1992).

On Earth, the prebiotic organic ingredients for life had both endogenic and exogenic sources. Endogenic sources included the primordial, slightly reducing atmosphere and active hydrothermal systems. Organic compounds (such as sugars, amino acids, and nucleobases) may have formed in Earth's early atmosphere via the Strecker reaction of ammonia, formaldehyde, and hydrogen cyanide (Miller, 1953; Johnson *et al.*, 2008). The above reactants may have been produced through photochemistry and/or lightning discharges in a reducing atmosphere. Indeed, Pascal (2013) and Pross and Pascal (2013) noted that only photochemistry and/or the momentary high temperatures produced during electrical discharges in the atmosphere (or by impact shock) can create the necessary energy kinetics to jump-start the process of chemical evolution (or drive to greater "dynamic kinetic stability") leading to abiogenesis and the appearance of simple life. Other compounds, such as hydrocarbons and fatty acids, could have been synthesized in the oceanic crust through hydrothermal Fischer-Tropsch reactions (Baross and Hoffman, 1985; Bougault *et al.*, 1993; Russell *et al.*, 2010). Exogenic sources included comets, meteorites, and micrometeorites. Amino acids in carbonaceous meteorites were probably synthesized by subsequent aqueous alteration of the parent body, also via the Strecker pathway (Bada *et al.*, 1994). Other plausible mechanisms for providing the key molecules necessary for the Strecker reaction are bolide impact shock synthesis in a slightly reducing atmosphere and the pyrolysis of existing bolide organics (Bada *et al.*, 1994). Chyba and Sagan (1992) estimated the flux of carbon during the period of heavy bombardment, about 4 billion years ago, at about  $10^8$  kg/year. With regard to the present, Maurette *et al.* (2001) calculated a flux of 20,000 tons/year of micrometeorites accreting on Earth. Impressive as this may sound, however, when spread over a primordial global ocean, the abundance of delivered organics in the waters must have been the equivalent of "a very dilute primordial soup."

## 2.2. The first spark: origin of life

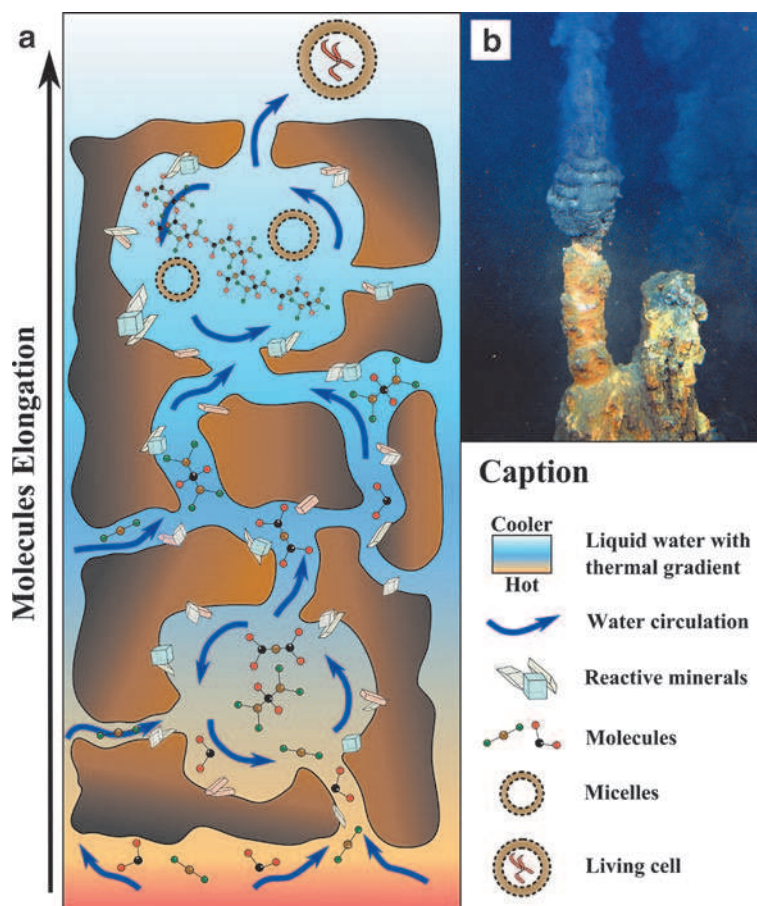
In the previous section, we briefly addressed the reducing chemical environments required for the formation of prebiotic organics. However, the conditions necessary for their subsequent association into more complex molecules, even-

tually leading to functional life-forms (*i.e.*, the process of abiogenesis, Pross and Pascal, 2013), were likely to have been more restrictive (Ourisson and Nakatani, 1999; Pascal *et al.*, 2006; Pollack *et al.*, 2009). Thermodynamically, molecular agitation is necessary to induce macromolecules to adhere to each other, but too much agitation will lead to their disintegration. An effective concentration process is fundamental for this type of chemistry. For example, free-floating organic (macro)molecules could have accumulated by attachment to other molecules passively fixed to the surfaces of reactive minerals, such as pyrite (*e.g.*, Huber and Wächtershäuser, 1998). For these sorts of reactions to take place, organics must be present in large quantities in physicochemical conditions compatible with liquid water and in an environment in which the kinetics of macromolecule formation is higher than that of their degradation—a sort of natural chemical reactor where the required ingredients can flow in, concentrate, react, and where the resulting products can be transported out to undergo further physicochemical evolution. Possible reactors may have included settings combining a regular flow of organic-laden waters possibly including a mild thermal gradient and either mineral grains or a porous medium capable of mediating reactions. Note that macromolecular self-organization can occur at water temperatures below 80°C, but at higher temperatures the molecules risk disaggregation (Larralde *et al.*, 1995). Relatively stable conditions would have been necessary for the assemblage of these molecules into primitive life entities. However, the simple accumulation and organization of macromolecules is not sufficient to form a living cell. Three essential functions must be ensured by the molecules and must coexist for life to arise: confinement, metabolism, and transmission of information.

There are presently two preferred hypotheses regarding the environment in which macromolecules on Earth could have aggregated under sufficiently stable conditions to form a primitive cell: hydrothermal vents and the littoral environment with its tidal swash (Figs. 1 and 2). Yet another environment has been suggested by Benner *et al.* (2010), who proposed streams on dry land containing exposed serpentine crust, the streams being characterized by changing pH conditions, as ideal environments for the formation of one of the key macromolecular components of life, ribose. Given that early Earth was more likely to have resembled an ocean planet than a dry one, this scenario seems less reasonable.

Modern hydrothermal vent systems tend to be episodic, appearing and disappearing relatively quickly. On young Earth, however, hydrothermal activity must have been much more prevalent than today, with hydrothermal systems probably being far larger and more extensive. Although fluid temperatures within modern hydrothermal vents may be far too high for the formation of macromolecules, the temperature gradient immediately adjacent to the vents can be very strong. Today, temperatures range from >350°C at black smoker vent exits to <2°C a little more than 50–100 cm away. On early Earth, silicon and oxygen isotope data suggest that global ocean temperatures were higher than at present, at least in the vicinity of the crustal surface, because of the higher heat flux from the mantle (about three times higher, Sleep, 2010) that drove extensive volcanism and also because of the pervasive hydrothermal circulation in the upper crust (*cf.* Buffett, 2003; Hofmann and Bolhar, 2007; van den Boorn *et al.*, 2007). Even with ocean temperatures

**FIG. 1.** Possible origin of life in porous (beehive), hydrothermal vents. (a) Sketch showing a porous beehive structure where hydrothermal fluids and seawater can circulate, leading to the accumulation of organic molecules. The reduced mineral surfaces within the vent pores could be favorable locations for the structural organization of macromolecules. We hypothesize the formation of lipid micelles in these environments and the incorporation of information-transferring molecules within the micelles, perhaps due to moderate agitation of the hydrothermal effluent. (b) Image of a modern black smoker (image credit: National Oceanographic and Atmospheric Administration). Color images available online at [www.liebertonline.com/ast](http://www.liebertonline.com/ast)



ranging from  $<50^{\circ}\text{C}$  to  $\sim 70^{\circ}\text{C}$ , the temperature gradient near vent structures would still have been steep.

Hydrothermal deposits and rocks in the immediate vicinity of vents contain minerals with very reactive surfaces, such as pyrite and sphalerite, that may have been implicated in the processes leading to the appearance of life (Huber and Wächtershäuser, 1998; Hazen and Sverjensky, 2010). At the very least, redox reactions at their surfaces could have provided the energy needed for primitive life. Hydrothermal edifices formed by the precipitation of vent fluid minerals can be extremely porous (Fig. 1), a characteristic that makes them potential locations in which macromolecules can be concentrated and, thus, give rise to the first cells (Alpermann *et al.*, 2010; Russell *et al.*, 2010). It is noteworthy that recent experiments have shown the ability of primitive cells to retain RNA and DNA oligonucleotides at temperatures up to  $100^{\circ}\text{C}$  (Mansy and Szostak, 2008).

The other scenario for the origin of life on Earth is in the littoral environment (Fig. 2). The possibility of subjecting macromolecules to wetting/drying cycles in the dynamically active environment of a tidal beach could stimulate the association of different macromolecular components. In principle, a sloping beach can act as a natural “chromatographic column” on which simple organics can be deposited and “eluted” through sand grains by the receding waves, over and over again, leading to chemical enrichment (Bada, 2004). For example, Deamer *et al.* (2002) were able to induce the incorporation of RNA into lipid vesicles in an experiment simulating these conditions.

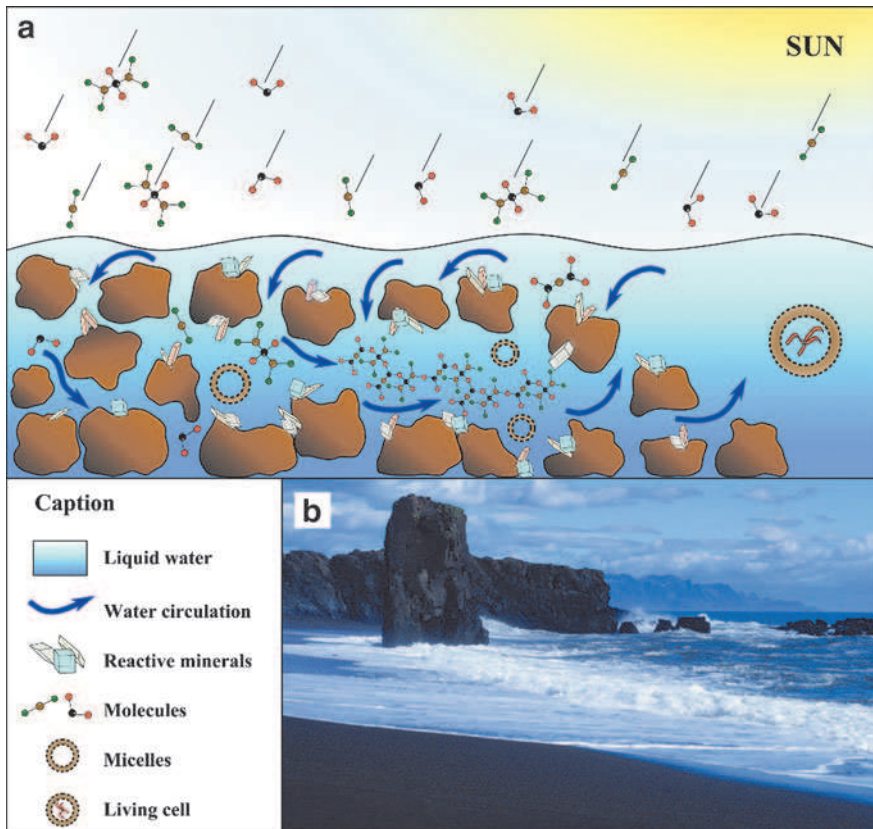
The length of time necessary for macromolecules to self-organize and form primitive cells is not known, but it is believed to be relatively short, of the order of hundreds of thousands to perhaps a few million years (Orgel, 1998). The kinetics of the actual process(es) necessarily had to be rapid, otherwise the macromolecular components of the first cells would have been broken down when exposed to early Earth’s hostile conditions. Once life appeared, colonization of the connected environment would have been rapid, on the scale of years to thousands of years (Sleep, 2010).

Summarizing, for the origin of life, moderately dynamic aqueous conditions, for example in a hydrothermal or a littoral environmental setting influenced by waves, are conditions that facilitate the initial aggregation of available prebiotic molecules into macromolecules. Thereafter, calmer environments would have been necessary for the final assemblage of the macromolecules into very simple, self-reproducing and evolving cells. This is assumed to have taken place on a timescale of  $10^5$  to some  $10^6$  years.

### 2.3. Sustaining conditions: established life

Once established, simple cells would have been able to rapidly take advantage of any habitable conditions, even ephemeral ones. Microbial cells are the archetypal opportunists. Microbes are adaptable because they have the ability to switch from one energy transfer pathway to another, depending upon their availability. Provided the necessary ingredients and the physicochemical conditions for liquid





**FIG. 2.** Possible origin of life in the swash zone. (a) Sketch showing the possible role of the reactive surfaces of volcanic detrital deposits for the structural organization of macromolecules within the swash zone. Wetting and drying cycles have been shown to be effective for the incorporation of information-transferring molecules into lipid micelles (e.g., Deamer *et al.*, 2002). (b) Photo of the modern swash zone on a volcanic beach, Iceland (image credit: Philippe Lefebvre). Color images available online at [www.liebertonline.com/ast](http://www.liebertonline.com/ast)

water are present, simple living cells can rapidly “invade” any newly available habitat. The spatial scales associated with this kind of colonization are the spatial scales of the cells themselves. Anaerobic microbes form colonies that can vary in size depending on the availability of carbon, nutrients, and energy. They may form small colonies of the order of tens of microns on the surfaces of particles (e.g., Westall and Southam, 2006; Westall *et al.*, 2011), or they can form vast mats, for example in the vicinity of hydrothermal vents (Sievert and Vetriani, 2012) (N.B. these mats form at the interface with oxygenated and sulfate-rich seawater; seawater on early Earth was sulfate poor because of the lack of oxygen; Grotzinger and Kasting, 1993). Microbes will live in an environment for as long as favorable conditions persist.

Given the volcanic and hydrothermal environments available on early Earth, the most likely sources of energy for the earliest life-forms would have been  $H_2$  from the hydrolysis of water or serpentinization and oxidants ( $O_2$ ,  $NO_3^-$ , or  $SO_4^{2-}$ ) from subsurface brines. These electron donors would have fueled chemoautotrophic life-forms (Sleep and Bird, 2007; Gaucher *et al.*, 2010) that used metabolisms such as methanogenesis or sulfite reduction. The first life-forms could have rapidly colonized the terrestrial oceanic environment, including the seafloor, and subsurface habitats, such as fractures, pores, deep sediments, and groundwater aquifers.

Earth’s subsurface is today, and was probably in the past, an important habitat for life. Indeed, Sleep (2010) suggested that subsurface refuges for life during/after sterilizing asteroid impacts were  $\sim 1$  km deep. Abyssal ocean sediments host 5–15% of the total terrestrial biomass (Kallmeyer *et al.*, 2009). Living cells have been collected from a depth of nearly

2000 m in submarine sediments (Ciobanu, 2012) and from nearly 4 km depth in South African gold mines (Onstott *et al.*, 2006). It is interesting to note that, compared to most cells in surface environments, the average doubling time of subsurface microorganisms is very slow, of the order of 1 cell/1000 years (Jørgensen and D’Hondt, 2006)!

#### 2.4. Challenging conditions: life in survival mode

Modern microbial life is remarkably resilient and capable of flourishing under extreme (for us) conditions, provided the necessary requirements are available (temperature, pressure, salinity, water activity, pH, radiation, etc.; Rothschild and Mancinelli, 2001; Pikuta *et al.*, 2007). For the purposes of this discussion, we distinguish thriving life from life in survival mode. When the necessary conditions for life are no longer available, many species are capable of slipping into a survival or dormant mode. Under such conditions microbial cells either metabolize very slowly as noted above (Jørgensen and D’Hondt, 2006) or they do not metabolize at all. All the cell functions are reduced, and the cell simply waits for more clement conditions to return. Estimations regarding the length of times cells can survive on Earth vary. There are some reports of viable cells in 5 million-year-old permafrost (Gilichinsky *et al.*, 2007) (although there is a strong likelihood of rejuvenation during the different interglacial cycles over the last 5 million years, for instance) and in 40 million-year-old amber (Cano and Borucki, 1995). Permian–Triassic (300–200 Ma) microbes have also been detected in salt deposits (Vreeland *et al.*, 2000), although they are not necessarily as old as the formation in which they are found (McGenity *et al.*, 2000). Whatever the respective merits

of these observations, survival of microorganisms on geologically long timescales must necessarily be influenced and limited by the background radioactivity in the crust (Kminek *et al.*, 2003).

### 2.5. Habitability: a working definition

The habitability of any rocky planet or satellite can be simply summarized as its capacity to host the necessary ingredients and conditions for life to appear (Southam and Westall, 2007; Hoehler and Westall, 2010). These are liquid water, endogenic and/or exogenic prebiotic molecules for constructing the macromolecules necessary for cellular life, a viable energy transfer mechanism (chemical or photonic), and bioessential elements and nutrients.

## 3. Example of Habitability on Mars

### 3.1. The possible appearance of life on Mars

There are an increasing number of reports of liquid water at the surface of Mars in the past from the instruments in orbit and on the ground, although it is still uncertain how much water there was on the surface or what was the importance of underground water reservoirs. Valley networks have been observed for tens of years and have been mapped over large parts of Noachian and some parts of Hesperian Mars (*e.g.*, Hynek *et al.*, 2010). Numerous alluvial fans occur in craters (di Achille and Hynek, 2010). Furthermore, the presence of significant water is suggested by the detection of phyllosilicate minerals, which likely represent surface weathering profiles produced by aqueous alteration of the basaltic crust (Poulet *et al.*, 2005; Bibring *et al.*, 2006; Fairén *et al.*, 2010; Gaudin *et al.*, 2011; Carter *et al.*, 2012; Le Deit *et al.*, 2012; Loizeau *et al.*, 2012a). The mineralogical associations being analyzed *in situ* at the John Klein site by Curiosity provide more detailed information about the characteristics of the water producing the alteration assemblages and indicate deposition of a smectite phase under neutral pH (Grotzinger *et al.*, 2013). Detected evaporite minerals indicate at least ephemeral liquid water at the surface (Gendrin *et al.*, 2005; Tosca and McLennan, 2006). Possible weathering sequences in various regions of Mars could indicate developed pedogenesis during the Noachian (Bishop *et al.*, 2008; Gaudin *et al.*, 2011; Le Deit *et al.*, 2012; Loizeau *et al.*, 2012a; Carter *et al.*, 2013). But deep alteration in extended regions of Mars (Ehlmann *et al.*, 2011a; Loizeau *et al.*, 2012b; Carter *et al.*, 2013) also shows that groundwater circulation has been an important process in the martian crust. The presence of liquid water on Mars' surface requires a denser atmosphere than exists at present (6 mmb CO<sub>2</sub>), but the exact quantity of CO<sub>2</sub> in the Pre-Noachian/Noachian atmosphere is not known. Recent models have shown that, even with a dense CO<sub>2</sub> atmosphere, early Mars could not have been "warm and wet" and was, in fact, cold (Forget *et al.*, 2013). However, Wordsworth *et al.* (2013) suggested that transiently "warm" conditions related to impacts or volcanism could have produced conditions favoring the formation of the late Noachian valley networks, even though the mid-late Noachian global conditions were probably icy.

Ehlmann *et al.* (2011a, 2011b) proposed a hydrological regime in which early Mars was characterized by ephemeral surficial water, driven largely by subsurface hydrothermal

circulation. However, simulations show that it would have been possible to maintain a water-table level close to the surface over large areas for hundreds of millions of years (Andrews-Hanna and Lewis, 2011). This surface water regime, fueled by impact and volcanic heating (Wordsworth *et al.*, 2013), would have culminated at the end of the Late Noachian; and from the Hesperian–Amazonian onward, groundwater processes only would have dominated. The level of the groundwater table was probably modulated by Mars' orbital parameters, which would have influenced the volume of water ice in the polar caps and in the midlatitudes. Local warming of the environment by volcanism and impacts would have been sporadic, but there should have been many events of water flowing at the surface of Mars, as evidenced by the numerous dendritic valley networks incising Noachian and early Hesperian terrains. Lakes would have also been episodically filled when fed by the rising water table. For example, this is what is assumed to have happened in Meridiani Planum (Grotzinger *et al.*, 2005; Squyres and Knoll, 2005; Andrews-Hanna *et al.*, 2010) and could explain detections of clays and sulfates in Gale Crater from the ground on the crater floor and from orbit at the base of Mount Sharp (Milliken *et al.*, 2010).

Both the mineralogical and the morphological evidence indicates a degradation in the surface conditions of Mars conducive to the proliferation of life from the mid Hesperian onward (Bibring *et al.*, 2006; Carr and Head, 2009). The exact causes are not well understood but may be related to a combination of processes including loss of early atmosphere, whether from solar wind erosion (Jakosky and Phillips, 2001) or impact erosion (Brain and Jakosky, 1998) or both. The loss of the atmosphere was accompanied by decreasing temperatures. A part of the liquid water inventory sublimed and was transported to the poles, while the rest became trapped in a subsurface cryosphere (Clifford, 1993; Clifford and Parker, 2001). Thereafter, any low-lying areas that may have been episodically inundated, for example, by catastrophic outflows, would have lost their water very rapidly through sublimation, evaporation in the tenuous atmosphere, and infiltration in the crust. Crater lakes, however, if fed by an active hydrothermal system, may have lived longer, but this activity still requires the presence of underground water (liquid or ice). Observations on Earth (Osinski *et al.*, 2005) and numerical modeling (*e.g.*, Abramov and Kring, 2005; Schwenzer and Kring, 2009) have shown that a crater-forming impact can result in the creation of a long-lasting hydrothermal system when ice is present in the crust. For example, a 130 km large crater could sustain an active hydrothermal system for up to 2 million years (Schwenzer and Kring, 2009), that is, long enough for life to emerge. Finally, even in the relatively recent past, there is some geomorphological evidence for extremely short-lived aqueous activity (Fenton *et al.*, 2007; Mangold *et al.*, 2007). It is interesting to note that whatever the fate of conditions on the surface of Mars through time, Jakosky and Shock (1998) estimated that there would not be sufficient geochemical energy to sustain abundant, flourishing life for more than 100 million years.

For the largest portion of martian history, the presence of surface water was patchy and infrequent. As a consequence, conditions favorable for the origin of life, compared to Earth, would have been heterogeneous in space and time. Cockell

*et al.* (2012) also pointed out the lack of connectivity between potential habitats on Mars. Even if the right ingredients and conditions would have been available long enough for the origin of life and the development of primitive organisms, early life on Mars would have been more heterogeneous than that of Earth, where the global ocean provides for rapid transport and mixing. A certain amount of mixing between habitable environments on Mars could be envisaged if supported by an efficient subsurface hydrological regime, but generally the surface habitats were probably simply isolated environments. There are some differences between Earth and Mars in the environments in which life could have appeared. On Earth, hydrothermal environments and intertidal areas are the favored hypotheses. Hydrothermal environments associated with volcanism and impact craters would have been abundant on early Mars. The swash zone, related to tidal pull, would be inexistent. On the other hand, the fact that the planet was basically dry means that the scenario envisaged by Berner *et al.* (2010) for life arising in a stream on exposed serpentinized crust could be more likely than on Earth.

### 3.2. Conditions for established life on Mars

As on early Earth, the volcanic and hydrothermal environment on early Mars would have supported chemoautotrophic life-forms (*cf.* Sleep and Bird, 2007). In this respect, the recent detection of serpentine in some outcrops associated with impact ejection of rocks from hundreds of meters' or kilometers' depth (the serpentine could also be due locally to postcrater hydrothermal activity) and with an olivine deposit near Nili Fossae (Ehlmann *et al.*, 2010; Bultel *et al.*, 2013) is of interest. Serpentinization produces  $H_2$  that serves as a source of energy (electron donor) for chemosynthetic life-forms (Schulte *et al.*, 2006). Other sources of energy could have included  $H_2$ ,  $O_2$ ,  $NO_3^-$ ,  $SO_4^{2-}$ , or organic sources, such as the degraded remains of dead organisms, acetate, or abiotic/prebiotic organics, for heterotrophs (Southam and Westall, 2007; Des Marais *et al.*, 2008; Hoehler and Westall, 2010; Summons *et al.*, 2011). Sunlight was and is still a potential source of energy on Mars, but martian microbes, if present, may not have had the necessary conditions for developing phototrophy. On Earth, photosynthesis appeared on a continuously habitable planet that had more or less uninterrupted access to energy, carbon, and nutrients, although the possibility of sterilization of the surface of Earth (but not the subsurface) by massive impacts during the Late Heavy Bombardment ( $\sim 3.9$  Ga) has been proposed by Zahnle and Sleep (2006) (and refuted by Abramov and Mojzsis, 2009). The continued habitability of at least some parts of the upper crust meant that terrestrial life-forms would not have undergone the stop-and-start scenarios that may well have characterized the fate of any early martian life. Photosynthesis developed in terrestrial microorganisms that had direct access to sunlight, that is, in shallow coastal areas (littoral). It was present on Earth at least 3.5 billion years ago (Tice and Lowe, 2004; Allwood *et al.*, 2006; Westall *et al.*, 2006a) and possibly by 3.8 billion years ago (Mojzsis *et al.*, 1996; Rosing, 1999; Rosing and Frei, 2004). These kinds of shallow water environments were probably common at various times on early Mars, but were they suitable for the evolution of photosynthetic microorgan-

isms? Apart from the question of the time necessary for continuous evolution to result in photosynthesis, another factor that may have been important is the distance between Mars and the Sun. Mars is farther away from the Sun, whose luminosity was fainter in the early phases of its history (Sagan and Chyba, 1997); that is, Mars was likely outside the Sun's habitable zone. What impact did this have on standing bodies of water on Mars? Despite the geomorphological and mineralogical evidence for water at the surface of the planet, particularly during its early history, attempts at modeling the evolution of the atmosphere suggest that Mars was always very cold and that liquid water would not have been present on its surface for long (Forget *et al.*, 2013; Wordsworth *et al.*, 2013). The implication is that, where present, the bodies of water may have been ice-covered, thus hindering the development of photosynthesis. Martian life, if it existed, may have remained at a very primitive state of evolution.

The rock record does not record the appearance of life on Earth, but it seems that the ingredients were available already by  $\sim 4.4$  Ga given that there is evidence of low-temperature hydrothermal alteration of the crust (Cavosie *et al.*, 2005). The first living cells would have had only a minimal number of metabolic and reproductive components. They were probably very small, similar to micelles (*cf.* Luisi *et al.*, 2006), and possibly comparable in size to viruses. Even by 3.45 Ga, a time from which we have the first morphologically preserved microorganisms (chemotrophic), the cells are still very small,  $< 1 \mu m$  in size (Westall *et al.*, 2006b, 2011) (similar to the sizes of modern subsurface chemotrophic microorganisms,  $\sim 0.2 \mu m$ ; Ciobanu, 2012). Primitive martian microorganisms, if in existence at the time, might have had (and may still have) similar characteristics, remaining very small. The amount of biomass that can be produced by chemosynthesis is much lower than that produced by photosynthesis. Sleep and Bird (2007) calculated that less than  $0.056 \text{ Tmol y}^{-1}$  of organic carbon (biomass) would have been buried in sediments on Earth in the period prior to the appearance of photosynthesis, compared with  $10 \text{ Tmol y}^{-1}$  today! They thus concluded that the record of chemosynthetic life must be "sparse and dispersed."

If life was established on Mars, it could have colonized even ephemeral habitats, especially if the habitats were interconnected (*cf.* Cockell *et al.*, 2012). Microbial life could have been hosted in short-lived habitats, such as small impact craters, playa lakes, and so on, or transiently protected habitats, such as under the surfaces of ice sheets, rocks, or in cracks where water could be trapped (*cf.* terrestrial endoliths; Friedmann and Koriem, 1989). These habitats may have involved much smaller spatial and temporal scales than those necessary for supporting the emergence of life. As soon as the habitable conditions disappeared—for example due to loss of water through freezing/sublimation or exhaustion of carbon source, nutrients, or energy—any putative martian organisms would have first gone into survival mode, eventually dying, unless the conditions changed and the missing ingredients were replenished. Opportunistic colonization of a habitat could have occurred at any time in martian history, if life were able to remain in a viable state and be transported to the newly available habitat. But, as noted by Cockell *et al.* (2012), a habitable environment may remain uninhabited, if no viable forms of life can be transported to it.



TABLE 1. SPATIAL AND TEMPORAL SCALES OF THE DIFFERENT CATEGORIES FOR MICROBIAL HABITABILITY

Habitability scales	Time	Space
Origin of life	$10^5$ to $10^6$ y?	$10^1$ to $10^3$ km
Established life	Day to weeks	$10^2$ to $10^3$ $\mu$ m
Survival	$10^{-1}$ to $n10^6$ y?	$10^2$ to $10^3$ $\mu$ m

Images: Top left, black smoker; top right, swash zone. Middle, outflow stream from a hydrothermal spring in Yellowstone. The barren sinter is colonized by photosynthetic microbial mats growing in the habitable outflow channel. Bottom right, halophiles in a fluid inclusion in a salt crystal. (Image credits: Top left, Ifremer; Top right, F. Westall; Middle, F. Westall; Bottom left and right, H. Stan-Lotter.)

The most likely refuge for life on Mars after the degradation of habitable conditions at the surface is the subsurface, where it may have spread via groundwater to even greater crustal depths than on Earth (microbes have been retrieved from depths in the order of 3 km; Onstott *et al.*, 2006), owing to the thicker martian crust and lower heat flux. Whether present in

a viable form in subsurface liquid oases or in frozen form in the cryosphere, martian life could have been brought back to the surface at various times during the history of the planet, whenever subsurface aquifers or the liquefied cryosphere erupted to the surface, for example through hydrothermal systems linked to volcanism or impacts.

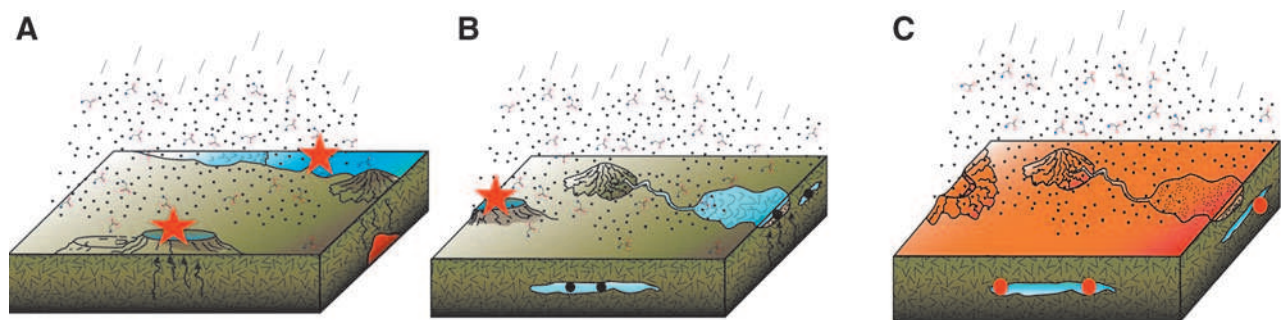


FIG. 3. Variable conditions favoring life on Mars. (A) Sketch representing early Mars with two types of unconnected habitable environments that could have supported the emergence of life (indicated by the red stars)—a lake in an impact crater and a larger water body. (B) Changing habitability on Mars. In one location a crater lake could host conditions conducive to the emergence of life, while simultaneously in another unconnected habitat extant life may be flourishing in the sediments at the bottom of an ice-covered body of water (black dots). Extant life is also present in the subsurface. (C) No habitable conditions at the surface of Mars. The frozen cryosphere may still host dormant cells (red dots). In all scenarios there is a continuous rain of extraterrestrial organics.



### 3.3. Surviving on Mars

If martian life did reach the subsurface, it is in this environment where we may have the best opportunity to discover it, especially considering the restrictive surface conditions after about 3.5 Ga (Jakosky *et al.*, 2007; Sleep, 2010). However, microbes in a subsurface environment would need to be able to repair and renew their genetic machinery to counteract the background radiation effects over time. Without such ongoing or regular repairs, the survival of dormant life-forms is limited to several 100,000 years close to the surface (but protected from direct UV radiation) and to several hundred million years if protected from the ionizing radiation from space and only exposed to natural radioactivity (Kminek *et al.*, 2003). It is worth pointing out that there are natural cycles on Mars; that is, obliquity changes, which may result in windows affording conditions for such repairs. To take advantage of such brief, clement episodes, martian microbes would need to be “ready and waiting” in the subsurface, that is, in a viable state. Upon being “revived,” they would have to divide and metabolize before either dying out or seeping back into the upper crust with the water flow, once again becoming cryofrozen.

Table 1 summarizes the ranges of spatial and temporal scales for the origin of life and for established or surviving life scenarios, while Fig. 3 summarizes the different categories of habitability on Mars, which could have coexisted. The existence of isolated habitable environments means that, at any one time during the early history of the planet, life could have emerged in one location while it was well established in another or dormant in a third. The sketches also have a sequential implication related to the changing (degrading) conditions of surface habitability with time. The prime environment of recharge was probably the subsurface, with an alternating water table/cryosphere activated by heat of geothermal, hydrothermal, volcanic, or impact origin. Precipitation and water ice deposition would have been additional sources of water.

### 4. Conclusions

In principle, life should be able to emerge on any rocky planet having water in contact with rocks and an abundant supply of prebiotic molecules, hence constituting a universal phenomenon. However, a closer look at the conditions supporting prebiotic chemistry, the origin and evolution of life, and its survival shows just how delicate this conclusion can be. On Mars, there is plenty of evidence indicating water in contact with rocks. Additionally, early Mars had the same inventory of exogenous prebiotic molecules as young Earth—although today a number of surface physical and chemical agents are likely responsible for degrading at least the more volatile organic components. Nevertheless, there is a major difference in the habitable conditions of the two planets in terms of spatial and temporal scales. For life to emerge, favorable conditions and ingredients need to be present for up to a couple of million years. For established life, the temporal and spatial scales are less restrictive. Conditions allowing active metabolism and replication come about every few hundred or thousand years on scales as reduced as some tens of microns to perhaps decimeters. Following a relatively short period during which life may

have gained a foothold on the Red Planet, when conditions may have been similar to those on young Earth, the number of potentially habitable environments for established life throughout Mars’ history was probably very large. Suitable conditions may still exist in the subsurface, but they may not necessarily have been inhabited.

This heterogeneity in the distribution of environments supporting the origin, evolution, and survival of life over Mars’ history will have a significant impact on the planning of missions seeking to find traces of life, in particular because locations supporting one habitable element, for example the initial conditions giving rise to life, might not support another element, such as ephemeral habitats that might support life for very brief periods of time (Westall *et al.*, in preparation).

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Address correspondence to:

Frances Westall  
CNRS-OSUC-Centre de Biophysique Moléculaire  
Rue Charles Sadron  
45071 Orléans cedex 2  
France

E-mail: frances.westall@cnrs-orleans.fr

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