

Development of the Analysis of Fecal Stanols in the Oyster Crassostrea gigas and Identification of Fecal Contamination in Shellfish Harvesting Areas

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17 Abstract

The objective of this work was to study the effects of washing and purification steps 18 on qualitative and quantitative analysis of fecal stanols in the oyster Crassostrea gigas using 19 20 either single or combination of lipid purification steps on silica gel or aminopropyl bonded silica gel (NH₂) or a washing step. Among the three analytical pathways compared, the two 21 22 including water extraction or NH₂ purification did not lead to higher recoveries and decreased 23 repeatabilities of extractions compared to the single purification on silica gel. This latter led to similar recoveries (ca. 80%) and repeatabilities (ca. 10%) for both spiked standards 24 (coprostanol and sitostanol). This analytical pathway has been applied to oysters collected in a 25 26 harvesting area in Brittany (France) where fecal contaminations are important and allowed to quantify eight stanols in oysters. The relative proportions of fecal stanols of these oysters 27 were combined with principal component analysis in order to investigate the usefulness of 28 their stanol fingerprints to record a fecal contamination and to distinguish its source between 29 human, porcine and bovine contaminations. Oysters non-fecally contaminated by Escherichia 30 31 coli did not present specific stanol fingerprints while oysters fecally contaminated had a bovine fingerprint, suggesting a contamination of these samples by bovine sources. As a 32 consequence, the method developed here allows the use of stanol fingerprints of oysters as a 33 34 microbial source tracking tool that can be applied to shellfish harvesting areas subjected to fecal contaminations in order to identify the different sources of contamination and improve 35 watershed management. 36

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39 Abbreviations

40	BSTFA	N,O-bis- (trimethylsilyl)trifluoroacetamide
41	DCM	Dichloromethane
42	DW	Dry weight
43	GC-MS	Gas chromatography- mass spectrometry
44	MeOH	Methanol
45	NH ₂	Aminopropyl-bonded silica gel
46	PCA	Principal component analysis
47	TMCS	Trimethylchlorosilane
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50 1. Introduction

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52 Coastal and shellfish harvesting areas are subjected to fecal contaminations from human 53 and animal waste leading to sanitary risks due to the presence of source-specific microbial 54 pathogens in contaminated waters and shellfish [1,2,3]. Among shellfish, several species of 55 bivalves such as mussels and oysters have been used as biological models for research in 56 ecotoxicology and biomonitoring since they are suspension filter-feeders which may 57 bioaccumulate and record environmental contaminants into their tissues [4, 5].

Fecal contamination of shellfish is particularly acute in France which is the first European producer of oysters (mainly *Crassostrea gigas*) [6]. In Brittany, one of the main areas of production of *Crassostrea gigas*, shellfish can be subjected to fecal contamination leading to the closure of shellfish harvesting areas [7]. Therefore, to limit i) sanitary risks linked to the consumption of contaminated shellfish and especially *Crassostrea gigas* and ii) economic loss due to the closure of shellfish harvesting areas, it is crucial to improve watershed management by controlling and limiting the sources of fecal contamination within these environments.

For this purpose, the actual European Shellfish Directive on shellfish harvesting (854/2004/EC) imposes the classification of shellfish and requires the assessment of potential pollution sources upstream of shellfish harvesting. The actual classification is based on the fecal indicator bacteria *Escherichia coli* and enterococci which are not species-specific.

In order to distinguish human and animal sources of fecal contamination within environment, microbial source tracking methods have been developed during the last decade. Based on specific microbial or chemical markers from human or animals, they have been successfully applied to field studies to identify the sources of fecal contamination in water, soil and sediment [7-16]. Among chemical markers, fecal stanols have proven their usefulness as direct fecal markers [17]. Indeed, the distribution of fecal stanols in animal faeces relies on three main factors: i) the animal's diet, ii) the ability of animals to biosynthesize endogenous sterols and iii) the composition of the intestinal flora responsible for sterol biohydrogenation into stanols. Consequently, the fecal stanol fingerprint allows to distinguish between different fecal sources in environmental matrix by the use of stanol ratios [18-20] or multivariate analyses [14, 15, 21, 22].

In Brittany, the main sources of fecal contamination in water are human wastewater 80 treatment effluent, porcine and bovine manure or slurry [12]. In this region, the transfer time 81 82 of water in coastal watersheds from streams to sea can last for one day [23] and it has been shown that a stanol fingerprint associated with a specific source of contamination can last for 83 six days in fresh and seawaters microcosms [24, 25]. Therefore, the specificity of a stanol 84 fingerprint can be transferred from inland waters to receiving seawater, which could allow the 85 identification of the sources of fecal contamination in water in such areas [14, 20]. In shellfish 86 87 harvesting areas, shellfish can bio-accumulate microbial pathogens by filtration of contaminated surrounding waters, which enables the identification of contamination sources 88 using microbial markers [7]. However, it is still unknown whether chemical markers such as 89 90 fecal stanols and corresponding stanol fingerprints allow the identification of fecal contamination sources in oysters. 91

Indeed, studies dealing with the occurrence of fecal stanols in shellfish mainly have focused on coprostanol as a marker of human fecal contamination in fresh or seawater mussels [26-31]. However, the identification of fecal sources with stanol fingerprints requires the accurate analysis of several compounds [14, 17, 22]. Therefore, it is necessary to develop an analytical pathway that allows the quantification of different fecal stanols in shellfish matrix, which constituted the first goal of the present study.

Among the main studies focusing on coprostanol quantification in mussels, only Cathum and Sabik [27] have tested the extraction efficiency of their method and found recoveries of about 48 % for wet mussel samples. As a consequence, the efficiencies of such methods remain largely unknown. In this present study, the efficiencies and repeatabilities of three analytical pathways have been compared for the stanol extraction of the oyster *Crassostrea gigas* using two recovery standards:

-The first method consisted of three steps: i) extraction of lipids from oyster matrix, ii)
purification of lipids on silica gel and iii) analysis of stanol fraction by gas chromatographymass spectrometry (GC-MS).

-The second method comprised of a washing step of oyster tissue with water prior to lipid
extraction. Indeed, shellfish such as oysters can be constituted of more than 90 % of nonlipid compounds such as glycogen and proteins that can potentially interact with lipids and
decrease their recoveries [32, 33]. Therefore, a washing step of matrix with water prior to
the lipid extraction step allows the removal of the non-lipid compounds and could improve
extraction efficiencies of stanols [34, 35].

-The third method comprised of a second purification step of lipids on aminopropyl-113 bonded silica gel after that on silica gel. The separation of lipid classes from the total 114 extract is mandatory in complex environmental matrix to improve the analysis accuracy of 115 target compounds. The lipid fraction of oysters is a complex mixture containing several 116 117 lipid classes such as phospholipids, triacylglycerols, free fatty acids, sterols and stanols [36]. Therefore, the addition of a purification step could be particularly interesting in order 118 to remove as much of lipids of non-interest as possible. Aminopropyl-bonded silica gel 119 (NH₂) was chosen as the second sorbent because of its affinity and subsequent selective 120 retention of acidic phospholipids [37]. 121

To the best of our knowledge, no method has been developed for the analysis of several fecal stanols in shellfish. Among the three methods tested here, the one leading to the highest and similar recoveries and highest repeatabilities for both recovery standards was chosen to analyse the concentration of several fecal stanols in oysters from northern Brittany, France. The stanol fingerprint of these oysters were analyzed by principal component analysis (PCA) in order to identify the fecal contamination sources in this shellfish harvesting area.

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129 **2. Experimental**

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131 *2.1. Reagent and chemicals*

Organic solvents were of high performance liquid chromatography 132 grade. Dichloromethane (DCM) was purchased from Carlo-Erba SDS (Val de Reuil, France), 133 methanol (MeOH) and cyclohexane were purchased from VWR (West Chester, PA). N,O-bis-134 (trimethylsilyl)trifluoroacetamide and trimethylchlorosilane (99:1, by vol) (BSTFA + TMCS) 135 136 and SPE disks (Supelco ENVI-18DSK, 47mm diameter) were purchased from Supelco (St. 137 Quentin Fallavier, France). Coprostanol (5β-cholestan-3β-ol), cholestanol (5α-cholestan-3βol), 5a-cholestane and aminopropyl-bonded silica gel were purchased from Sigma (St. 138 139 Quentin Fallavier, France). Sitostanol (24-ethyl-5a-cholestan-3β-ol) was purchased from Steraloids (Newport, RI). 24-Ethylcoprostanol (24-ethyl-5β-cholestan-3β-ol) and 24-140 ethylepicoprostanol (24-ethyl-5\beta-cholestan-3a-ol) were purchased from BCP Instruments 141 (Irigny, France). Silica gel (40-63 µm) was purchased from Merck (Darmstadt, Germany). 142 Cholesterol d₆ ([2,2,3,4,4,6⁻²H₆]-cholest-5-en-3 β -ol) was purchased from CDN Isotopes 143 (Pointe-Claire, Quebec, Canada). 144

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146 *2.2. Sample preparation*

To compare the three analytical pathways, 90 oysters (*Crassostrea gigas*) were purchased at Cancale (Brittany, France) in November 2012. After purchasing, oysters were opened, the intervalvular liquid discarded, the flesh of 10 individuals were pooled as one sample (ca. 4 g of dry weight flesh), frozen, freeze-dried and finally ground with an agate mortar for homogenization.

As the main goal of this study is to analyse several fecal stanols in shellfish tissues, two 152 recovery standards were used to determine the reliability of the tested methods. The common 153 154 human marker coprostanol was the first one and sitostanol was chosen as the second recovery standard because it is a fecal stanol rather representative of a bovine contamination [17]. 155 Coprostanol and sitostanol were spiked on the freeze-dried flesh pool just before organic 156 extraction (methods 1 and 3, see below) or aqueous extraction (method 2, see below) at a 157 concentration of 10 μ g g⁻¹ dry weight (DW). This concentration is in the range of coprostanol 158 159 concentration recorded in bivalves after human fecal contamination [26-31]. For both blanks and spiked samples, each extraction method was performed in triplicates. 160

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2.3. Application to a study case: the Fresnaye bay

Among the three methods tested, the one leading to the better recoveries of spiked 163 coprostanol and sitostanol and to better repeatabilities was used to determine the 164 concentration and the distribution of fecal stanols of oysters from the Fresnaye bay (Brittany, 165 France, Figure 1). This bay is an intensive shellfish harvesting area with an annual production 166 of ca. 550 tons of Crassostrea gigas intended for human consumption. The Fresnaye 167 watershed covers 121 km² and its number of human inhabitants is estimated at 14 000. The 168 169 potential sources of fecal contaminations originate from the seven wastewater treatment plants of the watershed and multiple sources of untreated wastewater, and its agricultural area covers 170 ca. 70 % of watershed area with intensive livestock farming of pigs (ca. 235.000 head in 171

172 2010), and cows (ca. 5300 head in 2010) [38]. In the last decade, this shellfish harvesting area 173 is subjected to increasing fecal contaminations by these different sources leading to the 174 degradation of the quality of oysters. In February, March and August 2013, oysters were 175 sampled at two locations on the bay and analysed for each sampling date.

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2.4. Analytical pathways for stanol analysis

Figure 3 summarizes the different steps involved in the four analytical pathways investigated for the analysis of fecal stanols in oyster tissues. Each step is described in detail below. Briefly, the method 1 consisted of an extraction of lipids with DCM followed by a purification step on silica gel and analysis of fecal stanols by GC-MS. The method 2 consisted of a first purification step of samples with water prior to lipid extraction followed by a purification on silica gel and GC-MS analysis. The method 3 involved a second purification step on aminopropyl-bonded silica gel prior to GC-MS analysis.

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2.4.1. Lipid extraction

For the three analytical pathways, lipids were extracted using an Accelerated Solvent 187 Extractor (ASE 200, Dionex, Courtaboeuf, France) with DCM. For each sample, about 4 g of 188 freeze-dried tissue were extracted 3 times in pre-washed (with DCM) 33 mL extraction cells. 189 Each extraction consisted in 2 cycles of 5 min at 100°C and 100 bar followed by a 40 second 190 191 flush step and a 30 second purge step. Each extract was then concentrated under reduced pressure and the 3 extracts were pooled. Then, total lipid extracts were dried, weighed and 192 dissolved in 20 mL of DCM to obtain a concentration of ca. 20 mg mL⁻¹ of lipids compounds 193 194 and stored at -20° C until fractionation.

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196 *2.4.2. Silica gel purification*

About 10 g of silica gel was preconditioned with ca. 50 mL of a mixture of 197 cyclohexane/DCM (2:1, by vol) and loaded into a 35 mL chromatography column equipped 198 with a glass frit and a pre-washed (with DCM) cotton wool at the bottom. Aliquots of 5 mL of 199 200 total lipid extract, corresponding to ca. 100 mg, were made up to a final volume of 15 mL in cyclohexane to obtain of final ratio of cyclohexane/DCM of 2:1 by volume before loading on 201 the silica column. Nonpolar compounds were eluted with 30 mL of a mixture of 202 cyclohexane/DCM (2:1, by vol) and the stanol-containing polar fraction was eluted with 40 203 204 mL of a mixture of DCM/Methanol (MeOH, 1:1, by vol). On average, this fraction accounted for 70 % of total lipids. For all samples, the elution was completed with pressurized air. For 205 methods 1 and 3, the polar fraction of interest containing stanols was then dried under reduced 206 pressure and weighed for quantification. 207

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2.4.3. Aqueous extraction

For the method 2, freeze-dried tissues were extracted with 50 mL of ultra-pure water to 210 211 obtain a ratio of 10 mL per gram of sample, and extractions were performed with stirrers at 212 ambient temperature overnight [39]. Then the separation of the solid residue from the aqueous extract was performed by centrifugation (2 x 15 min at 3500 rpm and 10°C, Rotenta 460 R 213 centrifuge, Hettich, Tuttlingen, Germany). The solid residue was freeze-dried prior to lipid 214 215 extraction and the aqueous extract filtered through a 0.7µm glass-fiber filter. In order to analyse the amount of fecal stanols removed from the sample by the water purification step, 216 solid phase extractions were performed on aqueous extracts as described by Jeanneau et al. 217 [39]. 218

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2.4.4. Aminopropyl-bonded silica gel purification

For method 3, the polar fraction eluted on the silica gel column was loaded on a chromatography column containing about 10 g of aminopropyl-bonded silica gel (NH₂) preconditioned with ca. 50 mL of a mixture of DCM/MeOH (1:1, by vol). The fraction containing stanols was eluted with 30 mL of a mixture of DCM/MeOH (1:1, by vol), dried under reduced pressure and weighted for quantification.

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2.2.5. Stanols analysis by gas chromatography-mass spectrometry (GC-MS)

Stanols were derivatizated using a mixture of BSTFA + TMCS (99:1, by vol) at 60°C
 during 20 minutes to convert hydroxyl groups into trimethylsilyl ether groups.

Stanols as trimethylsilyl ethers were analysed by GC-MS with a Shimadzu QP2010 + 230 MS gas chromatograph/mass spectrometer (Shimadzu, Tokyo, Japan). 1 µL of samples was 231 232 injected in splitless mode at 310°C. The temperature of the ionization source was set at 200°C. The temperature of the transfer line was set at 250°C, and molecules were ionized by 233 234 electron impact using an energy of 70 eV. Separation was achieved using a fused silica 235 column coated with SLB-5 MS (Supelco, 60 m, i.d. 0.25 mm, film thickness 0.25 µm) with helium as carrier gas at a flow of 1 mL min⁻¹. The GC oven temperature was maintained at 236 70°C for 1 min, then increased to 130°C at 15 °C min⁻¹, then to 300°C at 3°C min⁻¹ and held 237 at this temperature for 15 min. 238

Identification of stanols was based on the comparison with mass spectra and retention
times of standards. Analyses were performed in selective ion monitoring mode, the identified
and quantified stanols were coprostanol, cholestanol, campestanol, stigmastanol, 5βstigmastanol, sitostanol, 24-ethylcoprostanol and 24-ethylepicoprostanol and (Table 1 and 2).
Figure 2 presents the structures of coprostanol, sitostanol and other stanols involved in this
study. As 24-ethylcoprostanol eluted with other compounds, the mass fragmentogram of this

signal (main fragments m/z: 253, 296, 343, 386, 470) was a combination of the mass fragmentogram of those coeluted compounds. Therefore, the 215 fragment used to quantify stanols could originate from another compound and was not used here. As the 398 fragment is used as an identification fragment for 24-ethylcoprostanol and as its intensity was similar to that of the 215 fragment for all calibration solutions, it has been used here as the quantification fragment.

251 Quantification was based on the internal standard 5α -cholestane, which was added to samples after extraction and fractionation steps and prior to derivatization [12, 14, 15, 22, 39] 252 In opposition to recovery standards spiked in oysters (coprostanol and sitostanol) that were 253 254 used to quantify the efficiency of extraction procedures and to evaluate matrix effects, the internal standard was used to evaluate losses of sensivity of the detection with GC-MS. The 255 quantification method used a five-point calibration curve (standards: coprostanol, cholestanol 256 and sitostanol) at concentrations of 1, 3, 5, 8, and 10 μ g mL⁻¹ with a constant internal standard 257 concentration of 5 µg mL⁻¹. Considering the mass of samples and dilutions performed during 258 the analytical procedures, the limits of quantification for stanols analysed in oysters ranged 259 from 5 to 50 μ g g⁻¹ DW. Linearity of calibration curves, detection limits of GC-MS and 260 fragment used for the quantification of stanols are described in the Table 1. 261

262 The recoveries of spiked coprostanol and sitostanol (recovery standards) were calculated263 as follows:

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265 *2.5. Escherichia coli* analysis

In order to investigate the level of fecal contamination of oysters from the Fresnaye bay, the concentration of the fecal indicator bacteria *Escherichia coli* was determined by IFREMER (Laboratoire National de Référence , Nantes, France) using the impedance method [41].

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271 *2.6. Statistical analyses*

The analyses were conducted on three replicates for each sample for the comparison of the efficiencies of the three analytical pathways on the recoveries of both recovery standards and on two replicates for oysters from the Fresnaye bay. As non-parametric tests can lead to the conclusion that observed differences are not significant whereas qualitative differences are evident for low replication, comparison of stanol concentrations between samples were only qualitative.

278 Stanol fingerprints of oysters from the Fresnaye bay were investigated using the principal component analysis (PCA) model set up by Derrien et al. [22] with XLSTAT 2013 279 (Addinsoft, Paris, France). Briefly, this model is based on the distribution of six main fecal 280 stanols (i.e., coprostanol, epicoprostanol, 24-ethylcoprostanol, 24 ethylepicoprostanol, 281 282 campestanol, and sitostanol) of 88 various samples from bovine, porcine (faeces, manures, slurries...) and human origin (raw and diluted waste water treatment plant effluent, sewage 283 284 sludges). The PCA plot is a two-dimensional graphic representation of the correlations between the 6 stanols (variables). This plan is built on two axis (principal components) F1 and 285 F2, which explain 78.3 % of the total variance of the model. Each of the 6 stanol distribution 286 contributes to F1 and F2 axis. This model distinguish the stanol fingerprints from the three 287 previous origins into three distinctive clusters. Based, on their abundance of the 6 previous 288 stanols, the coordinates of samples on the PCA plots are calculated as follows: 289

• F1 coordonate = 0.497(%coprostanol) - 0.347(%epicoprostanol) +
0.295(%ethylepicoprostanol) - 0.460(%ethylepicoprostanol) - 0.422(%sitostanol) 0.395(%campestanol)

• F2 coordonate = -0.074(% coprostanol) + 0.565(% epicoprostanol) + 0.531(% ethylcoprostanol) - 0.303(% ethylepicoprostanol) - 0.288(% sitostanol) + 0.468(% campestanol).

These equations allow the identification of the origin of fecal contamination in environmental matrix between, bovine, porcine and human contaminations using this PCA model.

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- **300 3. Results and Discussion**
- 301 *3.1. Method comparison*

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303 3.1.1. Water extraction

Non-lipid compounds can interact with lipids and decrease the efficiency of their extraction from the sample matrix or their separation during solid-phase chromatography. Thus, the addition of an extraction step with water is expected to increase the recovery of target lipids. To investigate the effects of water extractions on the recovery of coprostanol and sitostanol in oysters, we compared the method 1 (organic extraction and silica gel purification) to the method 2 (water extraction, organic extraction and silica gel purification). Figure 4 presents recoveries of both coprostanol and sitostanol for the two methods tested.

The recovery of coprostanol extracted with the method 2 (59 ± 10 %, mean \pm standard deviation, SD) was lower than that extracted with the method 1 (79 ± 8 %) and the two methods led to similar repeatabilities. Similarly, the recovery of sitostanol extracted with the

method 2 (47 \pm 13 %) was lower than that extracted with the method 1 (84 \pm 8 %) and the 314 repeatability of the method 2 was lower than that of the method 1. Contrary to our hypothesis, 315 the recoveries of both coprostanol and sitostanol extracted with the method 2 tended to be 316 317 lower than those of the method 1. This result suggests that the extraction step with water removed more coprostanol and sitostanol than it decreased the potential interactions of these 318 two molecules with non-lipid compounds. Interestingly, the addition of coprostanol and 319 sitostanol quantities analysed in aqueous extracts by solid phase extraction to the quantities of 320 coprostanol and sitostanol in oyster tissues extracted with the method 2 led to quite similar 321 recoveries (61 versus 59 % for coprostanol and 54 versus 47 % for sitostanol) and remained 322 323 lower than those without the water extraction (method 1). This imbalance could be attributed to the low efficiency of solid phase extraction on aqueous extracts that are very rich in 324 hydrophilic organic compounds, which greatly decrease the efficiency of this method [40]. 325 326 Unfortunately, the efficiency of solid phase extractions could not be checked because of the coelution of the recovery standard cholesterol d_6 with cholesterol. 327

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3.1.2. Aminopropyl-bonded silica gel (*NH*₂) *purification*

Oysters contain high amounts of lipids from different classes that can potentially interact each other's and decrease the efficiency of their analysis [32, 33, 37]. The effects of a purification step with NH₂ was tested in order to remove as much as compounds of noninterest as possible. The comparison of the efficiency of the methods 1 (organic extraction and silica gel purification) and 3 (organic extraction, silica gel and NH₂ purifications) allows the investigation of the impact of this second purification step.

The recoveries of coprostanol were 79 ± 8 % and 89 ± 15 % for the methods 1 and 3, respectively and the recoveries of sitostanol were 84 ± 8 %, 103 ± 70 % for the methods 1 and 3, respectively (Figure 4). The addition of this second chromatographic step involving

aminopropyl-bonded silica seems to induce an increase of the recoveries of both coprostanol and sitostanol. However, the standard deviation between the triplicates highlights that the values of the recoveries belong to the same group and that the methods appear to be not different. The repeatability of the methods 1 and 3 can be inferred from the value of the relative standard deviation. For the method 1, the relative standard deviation represented 10 and 9% of the mean value for coprostanol and sitostanol, respectively, while for method 3 it represented 17 and 68% of the mean value for coprostanol and sitostanol, respectively.

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3.1.3. Comparison of the three methods

348 The first goal of the present study was to determine an efficient method for the analysis 349 of fecal stanols in the oyster *Crassostrea gigas*.

Water extraction led to opposite trends on coprostanol and sitostanol recoveries and increased their respective standard deviation. Thus, the addition of this step on the extraction pathway i) did not increase the recovery of both recovery standards and ii) decreased their repeatabilities. As a consequence, water extraction prior to organic extraction is not reliable for the analysis of fecal stanols in oysters.

The addition of a purification step on NH_2 increased the recoveries of coprostanol and sitostanol. Nevertheless, this step strongly decreased the repeatabilities of the methods tested, especially for sitostanol. Therefore, this step does not appear to be reliable for the analysis of fecal stanols in oysters.

Finally, among the three methods tested, the method 1, which included a lipid extraction step with organic solvent, a purification step on silica gel and analysis by GC-MS, led to i) statistically similar recoveries than the others two methods, ii) the higher repeatability and iii) similar recoveries for both recovery standards. Moreover, the recovery of coprostanol with the method 1 (79%) is higher than that found by Cathum and Sabik [27] (48%) probably because
these authors analysed coprostanol by GC-MS as underivatized compound.

365 In order to further improvement of this method and solvent and sorbent savings, stanol 366 extraction efficiency could be investigate using pre-packed silica cartridges available for 367 solid-phase extraction.

Finally, the method 1 has been chosen to analyse the concentration of fecal stanols in natural oysters sampled at the Fresnaye bay.

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371 *3.2. Stanol occurrence and concentrations in natural oysters from the Fresnaye bay*

Table 2 presents the concentrations of the stanols quantified in oysters sampled inFebruary, March and August 2013.

Eight stanols were detected and quantified in the samples analysed. Cholestanol was the 374 major compound and ranged from 58.4 μ g g⁻¹ DW (August) to 221.8 μ g g⁻¹ DW (February). 375 As these concentrations were above the upper limit of quantification of our method (i.e. 50 µg 376 g^{-1} DW), they were just qualitatively discussed in comparison to the concentrations of the 377 other stanol found in ovsters. The other stanols detected and quantified were coprostanol, 5β-378 stigmastanol, 24-ethylcoprostanol, 24-ethylepicoprostanol, campestanol, stigmastanol and 379 sitostanol and their concentrations ranged from 7.5 to 21.7 μ g g⁻¹ DW. The predominance in 380 Crassostrea gigas of cholestanol compared to other stanols is not surprising since its 381 precursor, cholesterol, is the main sterol in oysters [42-44]. Dunstan et al. [43] found 382 concentration of cholestanol of ca. 112 μ g g⁻¹ DW for *Crassostrea gigas* that is our range of 383 concentrations for cholestanol. Since cholestanol have been rarely found or in very low 384 concentrations in the diet of bivalves largely dominated by phytoplankton [45, 46], its high 385 relative abundance in bivalves might be due to the bioconversion of cholesterol during the 386 digestive process by the presence of gut bacteria [47]. However, cholesterol might not be the 387

only dietary sterol supplied by food leading to the formation of cholestanol within bivalves. 388 Indeed, it has been shown that marine bivalves are able to bioconvert several dietary sterols 389 into cholesterol for physiological needs [43, 48]. Furthermore, the high variability of 390 cholestanol concentrations between oyster samples from date to date are accompanied by high 391 variations of total stanol concentrations, which ranged from 145.0 μ g g⁻¹ DW (August) to 392 297.6 µg g⁻¹ DW (February). These variations might be due to the differences in physico-393 chemical conditions of surrounding seawater between sampling dates that could have led to 394 395 different metabolic responses of oysters resulting in different stanol concentrations.

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397 3.3. Fecal contamination and stanol fingerprint of oysters

The concentration of *Escherichia coli* in oysters sampled in February and March was respectively 67 and 220 Most Probable Number 100 g⁻¹ (Table 2). According to the European Shellfish Directive on shellfish harvesting (854/2004/EC), these amounts of *Escherichia coli* classified the two previous samples in the A class and oysters collected in February and March were considered as non-fecally contaminated. With a concentration of *Escherichia coli* of ca. 9150 Most Probable Number 100 g⁻¹, oysters sampled in August were classified in the B class and considered here as fecally contaminated.

In order to investigate the ability of oysters to record a species-specific fecal 405 contamination by bioaccumulation using their stanol fingerprint, the relative proportions of 406 fecal stanols of the three samples were injected in the PCA developed by Derrien et al. [14]. 407 Stanol fingerprints of oysters sampled in February (F1 and F2) and March (M1 and M2) were 408 located between the bovine and the human clusters (Figure 5). This absence of a specific 409 fingerprint is consistent with the absence of a fecal contamination of these samples measured 410 411 with Escherichia coli. By contrast, the fecally contaminated oysters sampled in August (A1 and A2) showed specific stanol fingerprints located in the bovine cluster (Figure 5). The 412

absence of a specific stanol fingerprint of oysters when they are not fecally contaminated in 413 addition to the specific stanol fingerprint of oysters fecally contaminated suggests that these 414 organisms could be able to record a species-specific stanol fingerprint when they are exposed 415 416 to a fecal contamination high enough. The bovine fingerprint of oysters sampled in August suggests that the fecal contaminations transferred from the watershed to seawater and 417 bioaccumulated by oysters during this period would mainly originate from bovine sources. 418 This hypothesis is consistent with the agricultural activity of the watershed where livestock 419 farming of cows is not negligible with ca. 5300 heads of livestock in 2010 [38]. The 420 contamination of oysters by bovine sources suggested by their specific stanol fingerprint in 421 422 August could be explained by agricultural practices and manure spreading calendar. Indeed, during summer, cows are grazing on grassland and thought lixiviation and erosion of soils 423 during raining events, even low, their faeces can be directly transported to streams that flow 424 425 into the bay. In August, pig slurry spreading is forbidden, so the large quantities of pig slurry produced by pig farming (ca. 235.000 pigs in 2010 in the watershed) remain stored, limiting 426 427 the fecal contamination of soils, streams and finally shellfish by this source.

In conclusion, the method developed here enables to analyse the concentration of eight fecal stanols in oysters and to record and identify the main source of fecal contamination of oysters using their stanol fingerprint with PCA. Stanol fingerprint could then be used as a microbial source tracking tool in oysters to track the origin of the fecal contamination in shellfish in order to enhance watershed management and reduce health risks linked to the consumption of contaminated shellfish.

434

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589 Figure legends:

590

Figure 1: Location of oyster sampling sites, streams, towns and waste water treatment plants(WWTP) in the Fresnaye watershed, Brittany, France.

593

594 Figure 2: Generalized structure of a stanol. The distinction between main fecal stanols

involves four points: i) the orientation (α or β) of the hydrogen atom at the position C-5

596 (mediated by the anaerobic reduction of the double bond located at the same position in the

597 corresponding unsaturated sterol precursor), ii) the orientation (α or β) of the hydroxyl group

at the position C-3, iii) the occurrence of methyl or ethyl groups at position C-24 (denoted by

599 R), and iv) the occurrence of a double bond at position C-22.

600

Figure 3: Schematic representation of the three analytical pathways used for the extraction
and the purification of fecal stanols from oyster samples (blanks and spiked ones) and their
subsequent analysis by GC–MS.

604

Figure 4: Comparison of coprostanol and sitostanol recoveries in oysters for the three extraction methods. Error bars are standard deviations (n = 3). Values under the error bars are relative standard deviations. The dotted line represents the 100% recovery threshold.

608

Figure 5: Plot of the principal component analysis comparing the 88 source samples and the 6oyster samples using the 6 most discriminant stanol compounds proposed by Derrien et al.

- 611 (2012). Each source samples was used as individual and oyster samples were used as
- 612 supplementary individuals. F1 axis: principal component 1; F2 axis: principal component 2.





Figure 2 :



Figure 3 :











Observations (axis F1 and F2 : 78.27 %)

Table 1: Trivial and IUPAC names, retention times relative to cholestanol (RRT), m/z values used for the identification and quantification of stanols, and information on quantification compounds (standard used, linearity of the corresponding calibration curve, and limit of detection).

Trivial name	IUPAC name	RRT -	Fragment m/z		Quantification details		
			Identification	Quantification	Standard	Linearity	LD ^a (ppb)
Coprostanol	5β-Cholestan-3β-ol	0.956	257, 355, 370	215	Coprostanol	0.998	5
Cholestanol	5α-Cholestan-3β-ol	1.000	257, 355, 384	215	Cholestanol	0.997	10
Campestanol	24-Methyl-5α-cholestan-3β-ol	1.050	369, 398, 484	215	Coprostanol	0.998	5
Stigmastanol	24-Ethyl-5α-cholesta-22-en-3β-ol	1.063	215, 383	215	Coprostanol	0.998	5
5β-Stigmastanol	24-Ethyl-5β-cholesta-22-en-3β-ol	1.019	257, 353, 486	215	Coprostanol	0.998	5
Sitostanol	24-Ethyl-5α-cholestan-3β-ol	1.097	383, 398, 473	215	Sitostanol	0.995	10
24-Ethylcoprostanol	24-Ethyl-5β-cholestan-3β-ol	1.041	257, 383, 398	398	Coprostanol	0.994	10
24-Ethylepicoprostanol	24-Ethyl-5β-cholestan-3α-ol	1.047	257, 283, 398	215	Coprostanol	0.998	5

^a Limit of Detection

	Sampling date					
Concentration	February	March	August			
Coprostanol	7.9 ± 1.4	7.4 ± 0.1	10.4 ± 1.2			
Cholestanol	221.8 ± 83.0	110.3 ± 16.3	58.4 ± 12.0			
Campestanol	16.1 ± 0.7	10.8 ± 0.2	12.2 ± 0.3			
Stigmastanol	8.0 ± 2.0	8.9 ± 1.0	10.7 ± 0.8			
5β-Stigmastanol	9.6 ± 0.8	7.8 ± 0.1	10.6 ± 1.0			
Sitostanol	13.1 ± 1.8	13.9 ± 1.2	21.7 ± 0.3			
24-Ethylcoprostanol	7.6 ± 1.7	7.5 ± 0.5	9.6 ± 1.3			
24-Ethylepicoprostanol	13.4 ± 0.2	9.4 ± 0.5	11.4 ± 0.6			
Total	297.6 ± 91.7	175.9 ± 19.7	145.0 ± 17.5			
Escherichia coli	67 ± 0	220 ± 0	9150 ± 9687			

Table 2 : Stanol (μ g g⁻¹ DW) and *Escherichia coli* (Most Probable Number 100 g⁻¹ of meat and shell liquor) concentrations of oysters from the Fresnaye bay. Errors are standard deviations (n = 2).