## **Supplementary Materials**

## Rate coefficients for the reaction of ozone with 2- and 3-carene

H. Chen<sup>1,2</sup>, Y. Ren<sup>2</sup>, M. Cazaunau<sup>2</sup>, V. Daële<sup>2</sup>, Y. Hu<sup>1,2</sup>, J. Chen<sup>1,3</sup>, A. Mellouki<sup>2,3</sup>\*

<sup>1</sup> Shanghai Key Laboratory of Atmospheric Particle Pollution and Prevention (LAP<sup>3</sup>), Department

of Environmental Science & Engineering, Fudan University, Shanghai 200433, China

<sup>2</sup> Institut de Combustion, Aérothermique, Réactivité et Environnement/OSUC, CNRS,

45071 Orléans Cedex 02, France

<sup>3</sup>School of Environmental Science & Engineering, Shandong University,

Shandong 250100, China

\* Corresponding author:

A. Mellouki, ICARE-CNRS, 1C Avenue de la Recherche Scientifique, 45071 Orléans Cedex 02,France (Fax: (33) 238 696004, e-mail: mellouki@cnrs-orleans.fr)

## 1 1. Materials

2 The purities of the chemicals were as follows: synthetic air,  $O_2$  and  $N_2$ , UHP certified

- 3 to >99.9995% (Alphagaz); 2-carene (Sigma-Aldrich,  $\geq$ 97); 3-carene (Sigma-Aldrich,  $\geq$ 98.5%);
- 4 cyclohexene (Fluka,  $\geq$ 99.5); 2-butanol (Fluka,  $\geq$ 99.5).
- 5 The purity of 3-carene was 95% determined by GC-MS, as shown in Fig. S1. The 3-carene
- 6 sample also consists of impurity of camphene (2%), limonene (1%) and terpinolene (2%).





8 Supplementary Figure 1: Chromatogram spectrum for 3-carene sample.



**Supplementary Figure 2:** Example for (a) concentration profile and (b) decay rate of ozone

13 before and after correction in the presence of excess 3-carene in SC.



**Supplementary Figure 3:** Example of PTR-TOF mass spectra of Exp. 2CAR-3T using TMB as







**Supplementary Figure 4:** Example of PTR-TOF mass spectra for isobaric peaks: m/z 83.05 21  $(C_5H_7O^+)$  and m/z 83.09  $(C_6H_{11}^+)$  during measurements of relative rates.

Experiment <sup>(a)</sup>	Temperature	carene	ozone <sup>(b)</sup>	c-hexane <sup>(c)</sup>	TMB	2-butanol	Injection sequences
	Κ	ppbv	ppbv	ppbv	ppbv	ppbv	
2-carene							
2CAR-1C	$298 \pm 1$	11.3	104	1700			2car; c-hexane; $O_3^{(c)}$
2CAR-2C	$300 \pm 1$	22.7	254	3990			2car; c-hexane; $O_3^{(c)}$
2CAR-3T	$305 \pm 1$	19.1	190		288		TMB; O <sub>3</sub> ; 2car
2CAR-4B	$293\pm1$	23.9	255			2640	2-butanol; 2car; O <sub>3</sub>
2CAR-5BR <sup>(d)</sup>	$293\pm1$	6.7	410			4290	2-butanol; 2car; O <sub>3</sub>
<u>3-carene</u>							
3CAR-1C	$299 \pm 1$	29.2	298	6820			c-hexane; O <sub>3</sub> ; 3car <sup>(c)</sup>
3CAR-2C	$295\pm1$	19.1	179	4490			c-hexane; 3car; O <sub>3</sub> <sup>(c)</sup>
3CAR-3C	$295\pm1$	11.3	96	2240			c-hexane; 3car; O <sub>3</sub> <sup>(c)</sup>
3CAR-4B	$294\pm1$	9.6	347			1320	2-butanol; 3car; O <sub>3</sub>
3CAR-5B	$295 \pm 1$	7.7	213			1150	2-butanol; O <sub>3</sub> ; 3car
3CAR-6BR <sup>(d)</sup>	$296\pm1$	9.7	347			3960	2-butanol; 3car; O <sub>3</sub>
3CAR-7BR <sup>(d)</sup>	$296 \pm 1$	29.1	<sup>(e)</sup>			6630	2-butanol; 3car; O <sub>3</sub>

22 Supplementary Table 1: Initial experimental conditions performed in HELIOS.

<sup>(a)</sup> C = cyclohexane as a scavenger; B = 2-butanol as a scavenger; BR= cyclohexene as a reference in the presence of excess 2-butanolas a scavenger; T = 1,3,5-trimethylbezene as a scavenger.

<sup>(b)</sup> [O<sub>3</sub>] was monitored by APOA-360 UV-absorption monitor, which is in agreement with FTIR within 5% uncertainty.

26  $^{(c)}$  c-hexane = cyclohexane.

27 <sup>(d)</sup> In the presence of 29.6 ppbv cyclohexene as a reference.

28 <sup>(e)</sup> O<sub>3</sub> was injected into HELIOS in several additions.