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# Distribution of In and other rare metals in cassiterite and associated minerals in $Sn \pm W$ ore deposits of the western Variscan Belt

CATHERINE LEROUGE<sup>1,\*</sup>, ERIC GLOAGUEN<sup>1,2</sup>, GUILLAUME WILLE<sup>1,2</sup> and LAURENT BAILLY<sup>1</sup>

<sup>1</sup> BRGM, 3 Avenue Claude Guillemin, 45060 Orléans, France

\*Corresponding author, e-mail: c.lerouge@brgm.fr

<sup>2</sup> ISTO, UMR 7327 - CNRS/Université d'Orléans, 1A rue de la Férollerie, 45071 Orléans Cedex 2, France

**Abstract:** We present data available on rare metal and indium distributions in cassiterite and associated minerals from thirteen  $\text{Sn} \pm W$  granite-related ore deposits in the western Variscan Belt (Massif Central and Armorican Massif, France; Galicia, Spain; and SW England). Cassiterite and associated minerals including sulfides and titanium oxides were analysed using an electron probe microanalyzer (EPMA). Significant indium contents were only measured in cassiterite from hydrothermal vein-type mineralizations associated with peraluminous granites of Montbelleux, Abbaretz (French Armorican Massif) and Marcofán (Galicia); they correlate with the highest Nb, Ta and Fe substitutions. Two coupled substitutions are proposed: (1)  $2(\text{Sn}^{4+}, \text{Ti}^{4+}) \leftrightarrow (\text{Fe}^{3+}, \text{In}^{3+}) + (\text{Nb}^{5+}, \text{Ta}^{5+})$  and (2)  $\text{Fe}^{2+} + (\text{Nb}, \text{Ta})^{5+} \leftrightarrow \text{In}^{3+} + (\text{Ti}, \text{Sn})^{4+}$  depending on Fe valence state. Sulfides (stannite, chalcopyrite, pyrite, arsenopyrite and sphalerite) and rutile associated with cassiterite contain significant amounts of indium, even when cassiterite is indium-free, suggesting preferential partitioning into the sulfides.

Key-words: cassiterite; tin; indium; tungsten; ore deposit; trace element; sulfides; mineral chemistry.

#### 1. Introduction

The element indium (In) was discovered by Ferdinand Reich and Theodor Richter in 1863 at the Bergakademie Freiberg in polymetallic Ag-base metal ores of the local Freiberg district (Erzgebirge, Germany) with a spectrograph (Seifert & Sandmann, 2006). Indium is a chalcophile element used in industry for its semiconductor and optoelectronic performance. Due to its softness, In alloys can be plated onto metal or evaporated onto glass in a variety of high-tech applications, including screens and monitors, infrared detectors, high-speed transistors, diodes, and photovoltaics. Indium is present as traces in the Earth's crust, but does not form primary ore deposits, and is currently extracted as a by-product of Zn, Sn and Cu deposits (Werner et al., 2017). The growing use of In in high-tech applications and a restricted supply dependent on only a few present-day producers (e.g., China, Korea, Japan, Canada, Zhang et al., 2015) created challenges to In sourcing worldwide (EU Commission, 2014; Werner et al., 2017).

During the two last decades, mineralogical studies of ore deposits in numerous countries have shown that In is a dispersed element present in various deposit types (Schwartz-Schampera & Herzig, 2002; Schwarz-Schampera, 2014; Werner *et al.*, 2017) and usually shared between several mineral phases (*e.g.*, Pavlova *et al.*, 2015; Andersen *et al.*, 2016; Frenzel *et al.*, 2016; George *et al.*, 2016; Werner

et al., 2017). Eighteen In minerals are currently identified, among them, roquesite, petrukite, sakuraiite, indite, laforêtite, dzhalindite (Andersen et al., 2016). In many In-rich deposits, the In minerals are not abundant or even absent. The Kawazu Au-Ag-Cu-Mn-Te epithermal ore deposit in Japan is an exception. Roquesite and dzhalindite are its main ores (Shimizu et al., 2008). In other In-rich deposits, the element is instead substituted in a wide variety of minerals, including base-metal sulfides (Fe-rich sphalerite), copper sulfides (chalcopyrite, bornite), copper-tin sulfides (chalcostannate group, stannite), tin oxides (cassiterite) (e.g., Seifert & Sandmann, 2006; Cook et al., 2009, 2011a, b, c; Dill et al., 2013; Pavlova et al., 2015; Andersen et al., 2016; George et al., 2016; Frenzel et al., 2016; Valkama et al., 2016); at this time sphalerite represents about 95% of global In production. The major In-rich deposits are, by decreasing order of importance, sediment-hosted massive sulfides, volcanogenic massive sulfides (VMS), skarn, epithermal, and porphyry; sediment-hosted Pb-Zn and VMS represent more than 60% of In resources (Werner et al., 2017). Magmatic-hydrothermal mineralizations mostly associated with post-collisional magmatic pulses, including skarn-, greisen-, and vein-type mineralization, also represent promising exploration targets for In (SE Finland; Erzgebirge/Krušné Hory; Far East Russia; SW England; South China Tin Belt) (Seifert, 2008; Cook et al., 2011c; Pavlova et al., 2015; Seifert et al., 2015; Andersen et al., 2016; Valkama et al., 2016).

The Variscan Belt contains several ore deposits identified for their significant In resources (Werner *et al.*, 2017), including VMS in Portugal such as Neves Corvo (Pinto *et al.*, 2014) or Lagoa Salgada (Figueiredo *et al.*, 2012), different ore-deposit types in the Erzgebirge, eastern Germany (skarn-type ores in the Pöhla district: Schuppan & Hiller, 2012; Bauer *et al.*, 2017; Jeske & Seifert, 2017), polymetallic Sn(-Ag)-base metal vein- and greisen-type deposits in the old Freiberg, Marienberg, Annaberg, and Ehrenfriederdorf-Geyer mining districts (Seifert *et al.*, 1992; Jung & Seifert, 1996; Seifert & Sandmann, 2006; Seifert, 2015) and Sn deposits in SW England (Andersen *et al.*, 2016). According to these recent investigations, late-Variscan granite-related Sn mineralizations might also represent potential interesting In targets.

In this work, we present data available on rare-metal and In distribution in cassiterite and associated minerals of  $Sn \pm W$  ore deposits from Massif Central and Armorican Massif in France, Galicia in Spain and SW England (Table 1; Fig. 1). The role of cassiterite concentrates for the actual world production of In is minor, and the question of whether In can also be hosted in cassiterite remains poorly defined (Pavlova *et al.*, 2015; Andersen *et al.*, 2016). We analysed cassiterite and associated minerals including sulfides and titanium oxides using an electron probe micro-analyzer (EPMA). We paid particular attention to the EPMA analysis method for trace elements in cassiterite related to interferences between Sn and In X-ray emission lines and to a high detection limit described in previous work (Benzaazoua *et al.*, 2003; Pavlova *et al.*, 2015).

Three topics of interest are (1) substitutions of In and other minor and trace elements, such as, Fe, Ti, Nb, Ta and W in the cassiterite lattice, (2) the In distribution among cassiterite and associated minerals, and (3) the trace element content in cassiterite, and more specifically the In content, in terms of ore deposit type.

## 2. Metallogeny in the Variscan Belt, samples and deposits

The European Variscides, extending from the Iberian Peninsula to Sudetic Mountains, are the result of continental collision between Gondwana and Laurussia. This collision caused the development of a polyphased orogeny, lasting more than 100 Ma from the Early Devonian to the Early Permian, and comprising three major successive stages (e.g., Faure et al., 2009; Maierová et al., 2016) of metallogenesis (Marignac & Cuney, 1999). The first stage of the Variscan orogeny (Early Devonian-Early Carboniferous) corresponds to the closure of Rheic Ocean followed by continental subduction. It is during this stage that sediment-hosted Pb-Zn and VMS were deposited (e.g., Lescuyer et al., 1998; Marignac & Cuney, 1999). The second stage (Mississippian) corresponds to the Variscan continental collision. The third stage of the Variscan Belt (Pennsylvanian to Lower Permian) is a synto post-orogenic collapse, marked by normal faults, development of granulite metamorphism of the lower crust and emplacement of granites (*cf.* Williamson *et al.*, 1996; Seifert, 2008; Simons *et al.*, 2016). Most of the magmatic-hydrothermal rare-metal mineralizations and hydrothermal W, Sn, Sn–F(Li) sulfide mineralizations are associated with the emplacement of peraluminous granites and rare-metal granites of this stage (*e.g.*, Marignac & Cuney, 1999; Bouchot *et al.*, 2005).

Samples of cassiterite-bearing ores were selected from twelve Late Variscan and the only Ordovician granitehosted Sn mineralizations/deposits in the Armorican Massif and Massif Central (France), in Galicia (NW-Spain), and in SW England (Table 1; Fig. 1). In the French Armorican Massif,  $Sn \pm W$  ore deposits were exploited for Sn (Abbaretz, Saint Renan, La Villeder) and for Sn+W (Montbelleux, Chauris & Marcoux, 1994). The Sn deposits La Villeder and Abbaretz are related to the Variscan leucogranites of the South Armorican zone belonging to the ilmenite serie (Chauris & Marcoux, 1994). Mineralization occurs as quartz – cassiterite  $\pm$ sulfide veins associated with tourmaline + muscovite +  $bervl + apatite + albite \pm fluorine hydrothermal alteration.$ At the Sn deposit La Villeder, metre-thick ore veins are hosted by the Lizio leucogranite at its periphery in contact with an Ediacarian metapelitic schist. The density of the ore veins increases toward the contact of the granite with schist. At the Abbaretz Sn deposit, low-angle quartz-veins are associated with the cupola of the Nozay leucogranite in contact with the schist of low metamorphic grade. The Saint Renan Sn deposit is located in the Leon block (Faure et al., 2008), a specific geodynamic domain, where late Hercynian intrusives are polyphased (two-mica potassic granite, locally injected by a leucogranite of probable Permian age). Mineralization occurs as wolframite-quartz veins with minor cassiterite and well-developed metagranite-greisen ore bodies in the northern part of the Saint Renan Massif (Chauris & Marcoux, 1994). Hydrothermal alteration is essentially represented by tourmalinization and muscovitization. Scheelite (but no cassiterite or wolframite) has been found in the porphyritic granite of the central part and in the fine-grained granite of the southern part. The Montbelleux district is different from the other Sn-W mineralizations by its Ordovician age (Chauris & Marcoux, 1994). Mineralization occurs as quartz-cassiterite-wolframite stockwork in a sodic granite and quartz-wolframite veins in schist near the granite contact (Chauris et al., 1989). The stockwork ore type also contains minor stannite, molybdenite, chalcopyrite, arsenopyrite and sphalerite.

The studied Sn deposits of the French Massif Central are a Cu–Sn–Fe ore deposit (Charrier), W(-Sn) ore deposits (Vaulry, Chataigneraie district) and rare-metal ore deposits disseminated within the rare-metal granites (as defined by Černý *et al.*, 2005) of Beauvoir and Montebras. Charrier is a complex ore deposit (Picot & Pierrot, 1963), which is composed of: (1) Cu orebodies hosted by Devonian metasediments and metavolcanics in contact with microgranite; the massive sulfide ores consist of chalcopyrite, and bornite with inclusions of roquesite, sphalerite, wittichenite, tennantite–tetrahedrite; and (2) magnetite–cassiterite–lep-

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Deposit	Resource	Sn total (t) prod. + reserves	Ore type and hostrock	Main mineralogy	Age (Ma)	Keterences
French Armorican Massif, F. Abbaretz	<i>rance</i> Sn	2800 + 9100	Qtz-Cst vein-type in schists and	Apy, Py, Ccp, Mo	~315 Ma	(1)
La Villeder	Sn	160 + 700	some in the Nozay granite Qtz-Cst vein-type and greisen	Sch, Apy, Sp, Ccp,	$316\pm 6$ Ma	(1), (2)
Montbelleux	W-Sn	25 + 4800	the the Lizio granue Qtz-Wolf vein-type in schist; Qtz-Wolf-Cst stockwork in granite sills and	Stn. Mo, Ccp, Apy, Sp	$470 \pm 10$ to $426 \pm 15$ Ma	(3)
Saint Renan	W-Sn	5000	associated greisens Qtz-Wolf-Cst vein-type and greisen ore bodies in granite		$\sim$ 321 $\pm$ 5 Ma	(1), (4)
French Massif Central, Fran Beauvoir	<i>ce</i> Rare metals (Sn, Ta, Nb, Do, T : )	2000 + 20000	Magmatic-hydrothermal dissemination in Beauvoir rare metal granite		317±6 Ma	(5), (6)
Montebras	Be, LI) Rare metals (Sn, Ta, Nb, Be Ti)	300 + 1000	Magmatic-hydrothermal dissemination in Montebras rare metal granite		314±4Ma	(5), (6)
Charrier	Cu-Fe-Sn	760	Mag-Cst-Lepidomelane orebodies in	Rqt, Bn, Ccp, Cst		(8)
Vaulry Le Prunet, Chataigneraie Entraygues, Chataigneraie	Sn-W Sn-W Sn-W	50	metaoasau Qtz-Cst-Wolf vein-type in Blond granite Dissemination in metapelitic schist Dissemination in metapelitic schist	Py, Apy Ccp, Sp, Apy, Py	298±1.2 Ma	
<i>Galicia, NW Spain</i> Marcofan, Beariz district			Vein-type and associated greisens dominantly in granite	Apy, Mo, Py, Ccp, Stn, Sch, Gn,	~320 Ma	(11), (12), (13)
Magros, Beariz district			Vein-type and associated greisens emplaced both in leucogranite and schist	Apy, Mo, Py, Ccp, Stn, Sch, Gn, Bis, Bi	~320 Ma	(11), (12), (13), (13)
Cornwall (England) St Agnes	Sn	1000-2000	Vein-type and greisen in topaz peraluminous granite	Apy, Sp, Ccp, Bn		(14)
References: (1) Chauris & Ma (8) Picot & Pierrot (1963); (9) arsenopyrite; Py, pyrite; Qtz, bismuth; Sth, stannite; Wolf,	rcoux (1994); (2) Ta Harlaux <i>et al.</i> (201 quartz; Mag, magn wolframite.	rtèse <i>et al.</i> (2011); (3) Chauris <i>et</i> 7); (10) Lerouge <i>et al.</i> (2007); ( etite; Mo, molybdenite; Bn, bor	<i>al.</i> (1989); (4) Faure <i>et al.</i> (2008); (5) Melleton <i>e</i> 11) Gloaguen <i>et al.</i> (2003); (12) Gloaguen (2006 nite; Sp, sphalerite; Gn, galena; Cst, cassiterite;	<i>et al.</i> (2015); (6) Cuney <i>et al.</i> (); (13) Sizaret <i>et al.</i> (2009); ( Sch, scheelite; Rqt, roquesi	(2002); (7) Vallance (14) Andersen <i>et al.</i> te; Bmt, bismuthini	<i>et al.</i> (2001); (2016). Apy, te; Bi, native

Table 1. Description of the Sn deposits of the Variscan Belt and the Ordovician Sn deposit of Montbelleux, selected for this study.

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Rare metals in cassiterite

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Fig. 1. (a) Schematic map of the Variscan terranes in Europe (after Seifert & Sandmann, 2006); squares indicate the four studied metallogenic provinces. Detailed maps of each province show the distribution of late Hercynian granites and the positions of the studied Sn mineralizations in: (b) French Massif Central (after Melleton *et al.*, 2010), (c) Corwall district in SW England (after Simons *et al.*, 2016), (d) The Beariz district in Galicia, Spain (after Gloaguen *et al.*, 2014) and (e) French Armorican Massif.

idomelane orebodies hosted by metabasalts near the contact with the La Burnolle granite; molybdenite, scheelite, bismuthinite, native bismuth and pyrite are also present.

At Vaulry (northwestern Massif Central), the W–(Sn–Cu) mineralization is hosted by the 310 Ma Li–F-rich Blond rare-metal leucogranite close to its contact with Paleozoic schists (Boulandon, 1989). The Vaulry deposit is characterized by subvertical ore veins, which show three major stages (Vallance *et al.*, 2001): (I) a stage of barren quartz; (II) infill of small fractures by younger quartz, cassiterite and wolframite, and (III) late microcracks infilled with loellingite, chalcopyrite, and pyrite. Stannite, stannoidite, arsenopyrite, bornite with inclusions of roquesite, mawsonite, scheelite, luzonite and molybdenite were also described (Cantinolle *et al.*, 1985; Boulandon, 1989).

The  $W \pm Sn$  ore deposits of the Chataigneraie district have limited economic potential for W (30 000 t of known WO<sub>3</sub> production distributed across eight deposits: Béziat & Bornuat, 1995), but are not economic for their Sn content. The Sn mineralization is hosted by metamorphic schists in the vicinity of large monzogranites and leucogranitic stocks related to W mineralization at 305 Ma (Lerouge & Bouchot, 2005; Lerouge et al., 2000, 2007). It occurs as late magmatic cassiterite in leucogranite stocks (Entraygues leucogranite) and as hydrothermal cassiterite associated with tourmaline, sulfides and rutile, disseminated in biotite-muscovitechlorite quartz-rich schist (Le Prunet, Entravgues schist). Typically, sulfides essentially consist of chalcopyrite, sphalerite, arsenopyrite and minor pyrite. The magmatic rare-metal (Sn, Ta, Nb, Be, Li) deposits of Beauvoir and Montebras in northwestern Massif Central are economic (Cuney et al., 1992; Raimbault et al., 1995). The Sn mineralization consists of dominantly disseminated postmagmatic cassiterite associated with a magmatic-hydrothermal paragenesis: lepidolite, topaz, columbo-tantalite, and amblygonite formed at around  $570 \pm 50$  °C (Fouillac & Rossi, 1991). It is hosted by high-phosphorus, peraluminous small albite-lepidolite granitic stocks emplaced between  $317 \pm 6$  Ma for the Beauvoir granite and  $314 \pm 4$  Ma for the Montebras granite (U/Pb columbite-tantalite, Melleton et al., 2015).

The Sn-bearing deposits in central Galicia, NW Spain, are of several types, disseminated within LCT pegmatites or in quartz veins associated with granite intrusions in a district where intrusion-related gold deposits also occur (e. g., Gloaguen et al., 2014). Tin-W ore deposits (Mina Vella mine – Marcofán, Mina Soriana mine – Magros) are located on the eastern side of the late-Variscan Beariz granite (Gloaguen, 2006; Sizaret et al., 2009). Mineralization occurs in N060°E-trending large normal-faulting quartz veins emplaced both in granite and surrounding micaschist. Large hydrothermal alteration zones occur in micaschist as massive tourmalinite haloes closely related to intrusive bodies and quartz veins. Both cassiterite and wolframite occur in the first mineral assemblage that comprises the infill of quartz veins. Marcofán is essentially hosted by granite, whereas Magros consists of veins hosted by schist.

The Saint Agnes Sn–W ore deposit is hosted by a small peraluminous granitic stock associated with the post-Variscan Cornubian Batholith outcropping in SW England (Andersen *et al.*, 2016; Simons *et al.*, 2016). The mineralization consists of quartz–tourmaline–cassiterite– wolframite–chalcopyrite–stannite–sphalerite–löllingite– arsenopyrite veins associated with greisen developed in the apical part of the granitic stock. The greisen alteration minerals include topaz, beryl, apatite and fluorite (Andersen *et al.*, 2016). Textural relationships provide evidence of a high-temperature cassiterite–wolframite± arsenopyrite–quartz–tourmaline stage followed by arsenopyrite–chalcopyrite with minor stannite stage (Andersen *et al.*, 2016). Roquesite was described in different ore deposits of the Cornwall district (Andersen *et al.*, 2016).

#### 3. Analytical techniques

Images in cathodoluminescence have been acquired on a MIRA 3 XMU (TESCAN, Brno, Czech Republic) equipped with a panchromatic cathodoluminescence detector (350-650 nm) (TESCAN BSE/CL detector) under a low vacuum mode (P=20 Pa nitrogen).

Analyses of cassiterite and sulfides were performed at the BRGM using a Cameca SX50 EPMA with an accelerating voltage of 20 kV. The spot size was  $\sim 1 \,\mu$ m. The system was calibrated with a variety of synthetic oxides and pure elements. Matrix corrections were made with the phi-rho-Z computing program PAP (Pouchou & Pichoir, 1984).

Tin and trace elements in cassiterite were analysed with a beam current of 150 nA. Ti– $K\alpha$ , Sn– $L\alpha$ , In– $L\alpha/\beta$ , Nb–  $L\alpha$  were measured on PET, Fe– $K\alpha$  on LiF and Ta– $M\alpha/\beta$ , W-L $\alpha$  on TAP. Counting times on peak and background were 10s for Ti and Fe, and 40s for other elements. Standards of calibration were natural minerals (cassiterite for Sn, roquesite for In), synthetic oxides (MnTiO<sub>3</sub> for Ti,  $Fe_2O_3$  for Fe), and pure elements (Nb, Ta and W). Detection of In and Ta in cassiterite is complicated because of interference between the Sn X-ray emission lines and those of In and Ta during EPMA-WDS analyses. The major  $L\alpha$  line of In (Fig. S1a and b in Supplementary Material, linked to this article and freely available online at the GSW website of the journal: http://eujmin.geoscience world.org) and the major  $M\alpha$  line of Ta (Fig. S1d) interfere with the positions of the  $L\eta$  and the  $L\alpha^{2nd \text{ order}}$  lines of Sn, respectively. The contributions of the Sn X-rays to In- $L\alpha$ and Ta- $M\alpha$  were approximated by analysis of In-free and Ta-free cassiterite and of Sn metal at around 1400 ppm for In and 3000 ppm for Ta. The contribution of the Sn X-ray to the In-L $\alpha$  is a value similar to that estimated by Benzaazoua et al. (2003). To avoid interference with Sn Xrays, In and Ta were measured on the second major In- $L\beta$ and Ta- $M\beta$  even though their intensities are lower than the intensities of the In-L $\alpha$  and Ta-M $\alpha$  (Fig. S1c and d). The In-L $\beta$ /In-L $\alpha$  intensity ratio is 0.45. The detection limits of the elements are (in ppm): 218 (Ti), 369 (Fe), 238 (Sn), 77 (Sc), 184 (Ta), 144 (Mn), 253 (In), 208 (W) and 313 (Nb).

Elemental mapping of Sn (Sn- $L\alpha$ ), Ta (Ta- $L\alpha$ ), Nb (Nb- $L\alpha$ ), Fe (Fe- $K\alpha$ ), Ti (Ti- $K\alpha$ ), and In (In- $L\beta$ ) in cassiterite were performed on a CAMECA SXFive-FE electron microprobe equipped with a Schottky Field-Emission Gun (FEG) (CAMECA, Gennevilliers – France), using an acceleration voltage of 20 kV and beam current of 200 nA.

Sulfides were analyzed for major and trace elements (S, Cu, Fe, Sn, As, Cd, Zn, In), using a beam current of 20 nA and counting time of 40 s for each element. Standards of calibration were natural minerals (pyrite for Fe, galena for S, sphalerite for Zn, cassiterite for Sn, roquesite for In), synthetic oxides (MnTiO<sub>3</sub> for Ti, Fe<sub>2</sub>O<sub>3</sub> for Fe), pure elements (Cu) and AsGa (As). The Cd- $L\alpha$ , In- $L\alpha$ , S- $K\alpha$  and Sn- $L\alpha$  lines were measured on PET, Cu- $K\alpha$ , Fe- $K\alpha$ , Zn- $K\alpha$  on LiF, and W- $L\alpha$ , As- $L\alpha$  on TAP. The detection limits of the elements were (in ppm): 281 (Cu), 178 (Fe), 494 (Sn), 438 (As), 294 (Cd), 293 (Zn), and 306 (In).

#### 4. Ore-mineral composition

#### 4.1. Cassiterite

Cassiterite grains in seventeen samples of the thirteen Sn ore deposits/occurrences were analysed for In and other trace elements. At least 50 analyses were performed in all the cassiterite samples, except for two samples where cassiterite content was very low (Le Prunet, Marcofán6). More analyses were done when In was detected in only small amounts and the amounts of cassiterite allowed it. All data are available in a supplementary electronic file (Table S1) and are presented in summary form with averages and standard deviations in Table 2. Sorting and visual inspection of the data give a good idea of trace element distribution. Averages are low and standard deviations are high compared to averages in most of the cassiterite ores studied because in many analyses, contents were too low to be detected. To provide more representative data, the lowest and highest values, the number of analyses, and the percentage of analyses above the limits of detection are also given. The presence of an element was arbitrarily considered as representative when 20% at least of the analyses were above the limit of the detection. The distribution of Fe, Nb, Ta, Ti and In in cassiterite is highly variable depending on the ore deposit (Fig. 2).

Cassiterite from Saint Agnes is quite pure and only contains traces of Fe and W; it will not be discussed further. Iron is absent or present in small amounts in cassiterite from Beauvoir and Montebras, Magros and Marcofán. Iron is present in significant concentrations in cassiterite from other mineralizations. Considerable concentrations of Nb are present in cassiterite from Abbaretz (1082 ppm), Montbelleux (3901 ppm), Beauvoir (1764 ppm) and Vaulry (590 ppm), and in two samples from Marcofán (#1: 620 ppm; #13: 880 ppm). Tantalum and Ti are both significant trace elements in cassiterite from all the mineralizations, with Ti contents ranging between 326 ppm (Beauvoir) and 4494 ppm (Entraygues) and Ta contents ranging between 393 (Le Prunet) and 5809 ppm (Montbelleux).

Indium is present in cassiterite from Abbaretz, Montbelleux, and two samples of Marcofán (#1, #13): the highest In content is measured in cassiterite of the Ordovician ore deposit of Montbelleux (average:  $519 \pm 204$  ppm, 91% of analyses above the detection limit, max. In content: 1092 ppm). About 20% of the cassiterite analyses of Marcofán (#13, #11) and Abbaretz are above the In detection limit. Indium is absent or systematically lower than the detection limit in cassiterite from Saint Renan, Beauvoir, Entravgues, Le Prunet and Marcofán (#1, #6). Samples from the Sn deposits La Villeder, Charrier, Montebras, Vaulry and Magros had less than 16% of analyses above the detection limit. Cathodoluminescence (CL) and elemental mapping of a cassiterite grain from the Montbelleux Sn-W deposit provide evidence of chemical zoning in Ta, Nb, Ti, Fe and a fine zoning in W, which, however, are not related to the In distribution: the darker zones in CL are richer in Fe (Fig. 3).

#### 4.2. Chemical composition of sulfides

A limited number of EPMA analyses were performed on sulfides that are associated with cassiterite in the thin sections from five Sn ore deposits (Table 3). Chalcopyrite, pyrite and sphalerite in the cassiterite-sulfide dissemination of Entravgues show In contents up to 940 ppm. 930 ppm and 520 ppm respectively. Sphalerite is also characterized by a homogeneous Cd content ~5200 ppm. Chalcopyrite and bornite in the Sn-polymetallic deposit of Charrier show average In contents of ~800 ppm and ~430 ppm, respectively. Chalcopyrite, stannite (Cu<sub>2</sub>FeSnS<sub>4</sub>) and pyrite from the Marcofán Sn deposit have significant In contents (up to 1220 ppm, 620 ppm and 1540 ppm, respectively), whereas sphalerite shows very low In and high Cd (1.5 wt%) contents. Chalcopyrite (up to 450 ppm), arsenopyrite (up to 640 ppm) and pyrite (up to 740 ppm) from the Magros Sn deposit have lower In contents than sulfides from Marcofán. Chalcopyrite and stannite from the Saint Agnes Sn deposit are characterized by high average In contents of 1560 ppm and 2590 ppm, respectively.

#### 5. Discussion

### 5.1. Incorporation of trace elements in the cassiterite lattice

For further investigations on major substitutions of trace elements in cassiterite, structural formulae were calculated on the basis of six oxygen atoms. The 1:1 correlation between  $3-Sn^{4+}$  and  $Fe^{3+} + In^{3+} + Nb^{5+} + Ta^{5+} + Ti^{4+}$  (correlation coefficient  $r^2$  of 0.99) shows that mineral stoichiometry is respected. Tin substitutions by Ti, Fe, Nb and Ta are limited in these cassiterites, the number of Sn atoms varying between 2.88 and 2.99. Binary diagrams of Ti, Fe, Nb and Ta atomic contents reported as a function of (3–Sn) show that all the analyses exhibit the same order of incorporation of trace elements in cassiterite for most of

Table 2. EPMA analyses of cassiterite from the different ore deposits. For each trace element average value, standard deviation and the number of analyses are given. The number of analyses is given to estimate detection limit for a set of data. The detection limits are 369 ppm for Fe, 253 ppm for In, 313 ppm for Nb, 184 ppm for Ta and 218 ppm for Ti.

Ore deposit	Number of analyses		$SnO_2$		Trace e	element conten	nts (ppm)	
			wt%	Fe	In	Nb	Та	Ti
Abbaretz	98	Min Max Average Std. dev. % >dl	94.20 99.71 98.97	350 3288 1292 462 99	<dl 419 154 122 22</dl 	<dl 2517 1082 401 90</dl 	<dl 1998 444 302 79</dl 	1421 5419 3006 770 100
Montbelleux	45	Min Max Average Std. dev. % >dl	95.43 99.31 97.83	<dl 5666 2568 1402 98</dl 	<dl 1020 519 204 91</dl 	<dl 12 282 3901 3487 84</dl 	597 20 490 5809 4325 100	647 5827 2635 1377 100
La Villeder	50	Min Max Average Std. dev. % >dl	98.69 100.25 99.70	<dl 1337 451 326 26</dl 	<dl 379 77 112 5</dl 	<dl <dl< td=""><td>753 1957 1222 200 50</td><td>403 8369 3728 1471 50</td></dl<></dl 	753 1957 1222 200 50	403 8369 3728 1471 50
Beauvoir	46	Min Max Average Std. dev. % >dl	96.61 101.53 99.96	<dl 979 227 672 16</dl 	<dl <dl< td=""><td><dl 5991 1764 1794 67</dl </td><td><dl 3407 1255 591 90</dl </td><td><dl 1900 326 537 41</dl </td></dl<></dl 	<dl 5991 1764 1794 67</dl 	<dl 3407 1255 591 90</dl 	<dl 1900 326 537 41</dl 
Charrier	57	Min Max Average Std. dev. % >dl	92.48 100.10 99.06	630 18 647 3818 4040 100	<dl 597 109 139 16</dl 	<dl <dl< td=""><td><dl 1204 662 378 79</dl </td><td><dl 24 897 2018 4045 58</dl </td></dl<></dl 	<dl 1204 662 378 79</dl 	<dl 24 897 2018 4045 58</dl 
Entraygues	124	Min Max Average Std. dev. % >dl	91.35 100.41 98.95	<dl 22 191 1935 2972 94</dl 	<dl <dl< td=""><td><dl 860 120 191 5</dl </td><td><dl 2383 569 384 97</dl </td><td>1271 9820 4494 2408 100</td></dl<></dl 	<dl 860 120 191 5</dl 	<dl 2383 569 384 97</dl 	1271 9820 4494 2408 100
Montebras	49	Min Max Average Std. dev.	97.77 100.70 100.26	<dl 1718 186 284 14</dl 	<dl 1023 63 201 6</dl 	<dl 2398 72 356 4</dl 	<dl 1409 1059 386 96</dl 	468 2938 1483 466 100
Le Prunet	21	Min Max Average Std. dev. % >dl	98.62 101.21 100.31	<dl 7788 1284 1769 52</dl 	<dl <dl< td=""><td><dl <dl< td=""><td>188 557 393 102 100</td><td><dl 10 527 2355 2617 86</dl </td></dl<></dl </td></dl<></dl 	<dl <dl< td=""><td>188 557 393 102 100</td><td><dl 10 527 2355 2617 86</dl </td></dl<></dl 	188 557 393 102 100	<dl 10 527 2355 2617 86</dl 
Saint Renan	197	Min Max Average Std. dev. % >dl	98.13 101.28 99.83	<dl 7718 658 1365 30</dl 	<dl <dl< td=""><td><dl 2503 271 520 16</dl </td><td><dl 1671 491 271 94</dl </td><td><dl 11 109 2065 2230 82</dl </td></dl<></dl 	<dl 2503 271 520 16</dl 	<dl 1671 491 271 94</dl 	<dl 11 109 2065 2230 82</dl 
Vaulry	50	Min Max Average Std. dev. % >dl	94.51 100.63 99.57	<dl 7431 1215 1202 76</dl 	<dl 617 94 119 8</dl 	<dl 2181 590 703 34</dl 	<dl 1499 1185 233 98</dl 	<dl 5198 2787 1800 74</dl 

Ore deposit	Number of analyses		$SnO_2$		Trace e	element conten	nts (ppm)	
			wt%	Fe	In	Nb	Та	Ti
Magros	50	Min	98.68	< dl	< dl	< dl	401	1007
iviug105	50	Max	99.93	847	425	1601	3456	3741
		Average	99.30	182	78	285	1569	1716
		Std. dev.	<i>уу</i> .50	196	98	335	908	360
				16	4	16	100	100
Marcofán 1	93	Min	99.78	<dl< td=""><td><dl< td=""><td><dl< td=""><td><dl< td=""><td>1127</td></dl<></td></dl<></td></dl<></td></dl<>	<dl< td=""><td><dl< td=""><td><dl< td=""><td>1127</td></dl<></td></dl<></td></dl<>	<dl< td=""><td><dl< td=""><td>1127</td></dl<></td></dl<>	<dl< td=""><td>1127</td></dl<>	1127
		Max	102.09	707	<dl< td=""><td>1202</td><td>2293</td><td>3327</td></dl<>	1202	2293	3327
		Average	100.35	119		620	799	2031
		Std. dev.		134		300	383	571
		% >dl		5		56	99	100
Marcofán 6	11	Min	99.57	<dl< td=""><td><dl< td=""><td><dl< td=""><td>7136</td><td>1229</td></dl<></td></dl<></td></dl<>	<dl< td=""><td><dl< td=""><td>7136</td><td>1229</td></dl<></td></dl<>	<dl< td=""><td>7136</td><td>1229</td></dl<>	7136	1229
		Max	100.65	389	< dl	692	2997	3831
		Average	100.13	228		201	1345	2151
		Std. dev.		114		221	708	1030
		% >dl		9		9	100	100
Marcofán 11	97	Min	99.38	<dl< td=""><td><dl< td=""><td><dl< td=""><td><dl< td=""><td>839</td></dl<></td></dl<></td></dl<></td></dl<>	<dl< td=""><td><dl< td=""><td><dl< td=""><td>839</td></dl<></td></dl<></td></dl<>	<dl< td=""><td><dl< td=""><td>839</td></dl<></td></dl<>	<dl< td=""><td>839</td></dl<>	839
		Max	100.90	583	400	< dl	2310	2578
		Average	100.38	147	148		489	1561
		Std. dev.		144	113		352	435
		% >dl		9	20		94	46
Marcofán 13	52	Min	98.51	<dl< td=""><td><dl< td=""><td><dl< td=""><td>532</td><td>1709</td></dl<></td></dl<></td></dl<>	<dl< td=""><td><dl< td=""><td>532</td><td>1709</td></dl<></td></dl<>	<dl< td=""><td>532</td><td>1709</td></dl<>	532	1709
		Max	99.52	1399	578	2579	7404	3963
		Average	98.98	571	271	880	2913	3100
		Std. dev.		239	134	730	1849	500
		% >dl		83	48	48	100	100

Table2. (continued).

the Sn mineralizations (Fig. S2). The  $Sn^{4+} \leftrightarrow Ti^{4+}$ substitution is a major one, which can be considered as the easiest substitution due to the similar properties of Sn and Ti cations and similarity between the rutile [TiO<sub>2</sub>] and cassiterite structures. Other  $Sn^{4+}$  substitutions by Fe (Fe<sup>2+</sup> and Fe<sup>3+</sup>) and rare metals (dominant Ta<sup>5+</sup> and minor  $Nb^{5+}$ ) are more complex due to the need to maintain electroneutrality in the crystal. Chemical compositions of cassiterite from different ore occurrences and mining districts are plotted in (Sn, Ti) - (Nb, Ta) - Fe composition diagrams (Fig. 4a-d). Cassiterite from the Sn deposits Beauvoir, Magros and Marcofán, and some cassiterite analyses from the Montbelleux deposit plot along the  $(Sn, Ti) - (Fe, Mn)(Nb, Ta)_2$  join, which defines the  $Fe^{2+} + 2$  (Nb, Ta)<sup>5+</sup>  $\leftrightarrow$  3 (Sn, Ti)<sup>4+</sup> ideal substitution (substitution 1) (Černý et al., 1985; Lerouge et al., 2007). This substitution (1) is also defined in minerals of the tantalite supergroup (Breiter et al., 2005), and microinclusions of minerals of the tantalite group have already been observed in cassiterite in Nb-Ta rich systems (Neiva, 2008). That probably implies a limit to Nb, Ta incorporation in cassiterite, which is not well defined to our knowledge. Cassiterite from the Abbaretz Sn deposit and some cassiterite analyses from the Montbelleux Sn deposit plot along the (Sn, Ti) - Fe (Nb, Ta) join,

suggesting that incorporation of Fe, Nb, Ta and Ti is rather defined by the  $\text{Fe}^{3+} + (\text{Nb}, \text{Ta})^{5+} \leftrightarrow 2$  (Sn, Ti)<sup>4+</sup> substitution (substitution 2). Analyses of the Fe-rich cassiterites of the Sn deposits Charrier, Saint Renan, and Vaulry that plot along the (Sn, Ti) – (Fe, Mn) join might correspond to a  $\text{Fe}^{3+} + \text{OH}^- \leftrightarrow \text{Sn}^{4+} + \text{O}^{2-}$  substitution (Möller *et al.*, 1988), but could be due to micrometre-sized inclusions of Fe oxides or ilmenite. Backscattered electron images of cassiterite from these ore deposits did not provide evidence of any inclusions of Fe-bearing phases, favoring the assumption of the substitution.

To conclude, two ideal coupled substitutions (1)  $Fe^{2+} + 2$  (Nb, Ta)<sup>5+</sup>  $\leftrightarrow 3$  (Sn, Ti)<sup>4+</sup>, and (2)  $Fe^{3+} +$  (Nb, Ta)<sup>5+</sup>  $\leftrightarrow 2$  (Sn, Ti)<sup>4+</sup> are at least possible for incorporation of Nb and Ta in the cassiterite studied. That has several implications: (1) incorporation will remain low in Fe-poor cassiterite, (2) incorporation will be limited by the formation of tantalite minerals, and (2) incorporation will highly depend on the Fe valence, and consequently on the redox condition in the system.

In regard to In incorporation in cassiterite, homogeneous spot values and mapping of Montbelleux cassiterite indicate that In is in the cassiterite lattice. Assuming that In is present as  $In^{3+}$  in the crust (Smith *et al.*, 1978), and that the electroneutrality of the crystal needs to be maintained,



Fig. 2. (a) Histogram showing the average values of Fe, Ti, Nb, Ta and In contents in cassiterite from the thirteen Sn ore deposits; (b) histograms of indium grade distribution (ppm) for some cassiterite from Abbaretz, Marcofán and Montbelleux deposits. Bin size = 10 ppm, EPMA detection limit of 253 ppm is represented by the vertical dotted line on each histogram. The disrupted zero bins represent 78, 78 and 27 analyses for Abbaretz, Marcofán11 and Marcofán13, respectively.

In incorporation may follow the coupled substitution 1: (Fe, In)<sup>3+</sup> + (Nb, Ta)<sup>5+</sup>  $\leftrightarrow$  2 (Sn, Ti)<sup>4+</sup>. Incorporation of In<sup>3+</sup> in cassiterite via coupled substitution (2) could be possible by modifying the exchange vector as follows: Fe<sup>2+</sup> + (Nb, Ta)<sup>5+</sup>  $\leftrightarrow$  In<sup>3+</sup> + (Ti, Sn)<sup>4+</sup>.

#### 5.2. Distribution of In in mineral parageneses

Among the studied deposits, the In mineral roquesite is present at the Sn deposits Charrier, Vaulry and Saint Agnes. In these three deposits, cassiterite shows very low In contents. At the Charrier deposit, the order of In distribution in Zn-poor Cu–Sn ore is discrete roquesite, followed by sphalerite (0.8 wt%), chalcopyrite (800 ppm) and bornite (430 ppm). At Saint Agnes, the In distribution in the studied sample is discrete roquesite followed by stannite (2190 ppm) and chalcopyrite (1565 ppm), in good agreement with Andersen *et al.* (2016).

Cassiterite contains In in only three of the thirteen ore deposits studied: Montbelleux ( $519 \pm 204$  ppm), Abbaretz (up to 420 ppm) and Marcofán (up to 580 ppm). Indium contents measured in those cassiterites are consistent with literature data (Briskey, 2005; Pavlova *et al.*, 2015). No sulfides were observed in samples from Montbelleux and Abbaretz. In the Marcofán ore deposit, cassiterite contains In in sulfide-bearing samples (#1: up to 400 ppm, #13: up to 578 ppm); this has already been described by Pavlova *et al.* (2015). The In distribution in



Fig. 3. Mapping of trace elements in cassiterite from the Montbelleux ore deposit. (a) CL image of a single cassiterite grain; (b) to (g) Element mapping of Fe, Ti, Ta, Nb, W and In, using electron microprobe equipped with a field-emission gun (FEG).

Zn-poor Sn–Cu ore of the Marcofán Sn deposit is chalcopyrite (150–730 ppm), followed by cassiterite (150–270 ppm) and stannite (140 ppm).

In other ore samples containing sulfides (Magros and cassiterite–sulfide dissemination of Entraygues), no In mineral is observed and few analyses of cassiterite have detectable In contents, whereas associated sulfides and rutile (when it is analysed) contain In.

The comparative In contents of the different phases from each ore deposit confirm that chalcopyrite, bornite, stannite, pyrite, arsenopyrite, rutile can also contain In. In the studied Sn ore deposits, ores are Zn-poor and Cu-rich, and they are of interest because of the high average In contents in Cu-sulfides: chalcopyrite (up to 2780 ppm); stannite (up to 2670 ppm); bornite (up to 670 ppm).

## 5.3. Trace element contents in cassiterite and ore deposit type

Mineralizations of the Sn deposits Montebras and Beauvoir consist of magmatic cassiterite disseminated in rare-metal (Sn, Ta, Nb, Be, Li) granites (Černý et al., 2005). Mineralizations of the other Sn deposits of this study are dominantly hydrothermal vein-type (except the mineralizations of Le Prunet and Entraygues schists, which are disseminated in schists) and spatially associated with late Variscan peraluminous granites. These granites result from partial melting of the crust and fractional crystallisation (Cornwall: Simons et al., 2016; French Armorican Massif: Bernard-Griffiths et al., 1985; Chauris & Marcoux, 1994; Tartèse & Boulvais, 2010; Galicia: Gloaguen, 2006; French Massif Central: Williamson et al., 1996). Partial melting was initiated by increased crustal temperature and by F-Li-P fluids derived from granulite metamorphism of the lower crust in relation with processes of underplating of mantle magmas (cf. Williamson et al., 1996; Seifert, 2008; Simons et al., 2016).

Even though Sn hydrothermal mineralizations are spatially associated with granites, they are not systematically genetically linked to them (Marignac & Cuney, 1999; Vallance *et al.*, 2001). Case studies of mineralizations provided evidence of a genetic link between mineralizations and peraluminous granites in La Chataigneraie district French Massif Central (Lerouge *et al.*, 2007), in the Beariz district, Galicia (Gloaguen *et al.*, 2014), in Saint Agnes, SW England (Andersen *et al.*, 2016), and in the South Armorican Massif (Chauris & Marcoux, 1994).

In the Sn deposits studied here, two types of cassiterite may be distinguished according to their trace element (Fe, Ti, Nb, Ta) contents. The first population of cassiterite, which is Fe–Ti-rich and Nb–Ta-poor, mostly corresponds to hydrothermal mineralizations hosted by pelitic schists (Entraygues schists, Le Prunet), metabasalts (Charrier) or granite not genetically linked to mineralizations (Vaulry, Vallance *et al.*, 2001). The second population, which is Nb–Ta-rich, corresponds to magmatic mineralizations disseminated in rare-metal granites (Beauvoir and Montebras) and to hydrothermal vein-type mineralisations genetically associated with peraluminous granites and hosted by both granite and pelitic schists (Marcofán and Magros in the Beariz district, Galicia; Abbaretz and LaVilleder, South Armorican Massif; Ordovician Mon-

Table 3. Chemical compositions of sulfides associated with cassiterite in the different studied Variscan Sn ore deposits. The detection limits are 281 ppm for Cu, 494 ppm for Sn, 293 ppm for As, and 306 ppm for In.	1 E	
Table 3. Chemical compositions of sulfides associated with cassiterite in the different studied Variscan Sn ore deposits. The detection limits are 281 ppm for Cu, 494 ppm for Sn, 293 p. <sup>7</sup> n, 438 ppm for As, and 306 ppm for In.	bin	
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Deposit	Mineral	Number of	Cu	Fe	Sn	Zn	As	s	Total		Cd (in	(mqq			In (in p	(mc	
		analyses				in wt%	,0			Mean	Std dev	Min	Max	Mean	Std dev	Min	Max
Charrier	Bornite Chalcopyrite	5 9	63.30 35.44	10.70 28.69	-dl 0.06	<dl 0.05</dl 	<dl-0.10 <dl-0.09< td=""><td>25.84 34.55</td><td>99.84 98.77</td><td></td><td></td><td></td><td></td><td>434 797</td><td>154 263</td><td><math display="block">\begin{array}{c} 310\\ 0\end{array}</math></td><td>670 1100</td></dl-0.09<></dl-0.10 	25.84 34.55	99.84 98.77					434 797	154 263	$\begin{array}{c} 310\\ 0\end{array}$	670 1100
Entraygues	Chalcopyrite Pyrite Sphalerite	4 m Γ	34.68 0.06 0.49	28.93 45.90 10.80	0.13 <dl <dl< td=""><td><dl <dl 53.54</dl </dl </td><td><dl-0.07 <dl-0.09 <dl-0.09< td=""><td>34.82 53.69 33.41</td><td>98.55 99.65 98.24</td><td>5266</td><td>368</td><td>4580</td><td>5720</td><td>495 540 163</td><td>363 342 188</td><td>000</td><td>940 930 520</td></dl-0.09<></dl-0.09 </dl-0.07 </td></dl<></dl 	<dl <dl 53.54</dl </dl 	<dl-0.07 <dl-0.09 <dl-0.09< td=""><td>34.82 53.69 33.41</td><td>98.55 99.65 98.24</td><td>5266</td><td>368</td><td>4580</td><td>5720</td><td>495 540 163</td><td>363 342 188</td><td>000</td><td>940 930 520</td></dl-0.09<></dl-0.09 </dl-0.07 	34.82 53.69 33.41	98.55 99.65 98.24	5266	368	4580	5720	495 540 163	363 342 188	000	940 930 520
Magros	Arsenopyrite Chalcopyrite Pyrite	18 4 2	<dl <="" td=""><dl< td="">34.01<dl< td=""></dl<></dl<></dl>	33.74 29.22 46.05	<dl>0.07<td>0.04 <dl <dl< td=""><td>47.21 <dl-0.15 <dl-0.84< td=""><td>19.19 34.41 53.55</td><td>100.19 97.71 99.60</td><td>3420</td><td>10 685</td><td>0</td><td>33 830</td><td>395 154 400</td><td>288 172 349</td><td>000</td><td>740 450 640</td></dl-0.84<></dl-0.15 </td></dl<></dl </td></dl>	0.04 <dl <dl< td=""><td>47.21 <dl-0.15 <dl-0.84< td=""><td>19.19 34.41 53.55</td><td>100.19 97.71 99.60</td><td>3420</td><td>10 685</td><td>0</td><td>33 830</td><td>395 154 400</td><td>288 172 349</td><td>000</td><td>740 450 640</td></dl-0.84<></dl-0.15 </td></dl<></dl 	47.21 <dl-0.15 <dl-0.84< td=""><td>19.19 34.41 53.55</td><td>100.19 97.71 99.60</td><td>3420</td><td>10 685</td><td>0</td><td>33 830</td><td>395 154 400</td><td>288 172 349</td><td>000</td><td>740 450 640</td></dl-0.84<></dl-0.15 	19.19 34.41 53.55	100.19 97.71 99.60	3420	10 685	0	33 830	395 154 400	288 172 349	000	740 450 640
Marcofàn	Chalcopyrite Pyrite Sphalerite Stannite	16 6 5	33.07 <dl <dl 29.30</dl </dl 	29.27 58.18 9.40 12.01	0.06 <dl <dl 27.18</dl </dl 	1.69 1.17 55.27 0.74	<dl-0.018 0.08-0.16 <dl-0.06 <dl-0.07< td=""><td>35.05 39.55 33.19 29.70</td><td>99.15 98.90 97.86 98.93</td><td>14 282 326</td><td>563 260</td><td>0 13 700 80</td><td>2130 14 740 720</td><td>727 597 142</td><td>265 611 270</td><td>360 0 0</td><td>1220 1540 620</td></dl-0.07<></dl-0.06 </dl-0.018 	35.05 39.55 33.19 29.70	99.15 98.90 97.86 98.93	14 282 326	563 260	0 13 700 80	2130 14 740 720	727 597 142	265 611 270	360 0 0	1220 1540 620
Saint Agnes	Chalcopyrite stannite	8 7	34.14 30.07	26.22 11.19	3.71 26.39	$0.30 \\ 1.97$	<dl-0.05 <dl-0.04< td=""><td>33.96 29.35</td><td>98.33 98.96</td><td>59 211</td><td>120 198</td><td>0 0</td><td>330 550</td><td>1563 2187</td><td>594 372</td><td>700 1790</td><td>2780 2670</td></dl-0.04<></dl-0.05 	33.96 29.35	98.33 98.96	59 211	120 198	0 0	330 550	1563 2187	594 372	700 1790	2780 2670

tbelleux). Reported in a binary Fe vs. Nb+Ta diagram, analyses of hydrothermal Fe–Ti-rich and Nb–Ta-poor cassiterites plot in the field of hydrothermal cassiterite of Tindle & Breaks (1998) (Fig. 4f). Analyses of magmatic-hydrothermal Nb–Ta-rich cassiterites associated with rare-metal granites (Beauvoir, Montebras) plot in the field of rare-element granites and pegmatites of Tindle & Breaks (1998). On the contrary, analyses of hydrothermal Nb–Ta-rich cassiterites from Montbelleux, Marcofán, Magros, Abbaretz and La Villeder plot in the field of rare-element granites and pegmatites, rather than in the hydrothermal field. Thus the field of rare-element granites and pegmatites and pegmatites defined by Tindle & Breaks (1998) is not so restrictive and would rather correspond to cassiterites that show the coupled substitutions (1) Fe<sup>2+</sup> + 2 (Nb, Ta)<sup>5+</sup>  $\leftrightarrow$  3 (Sn, Ti)<sup>4+</sup> and (2) Fe<sup>3+</sup> + (Nb, Ta)<sup>5+</sup>  $\leftrightarrow$  2 (Sn, Ti)<sup>4+</sup>.

It is also noteworthy that the Nb-Ta contents of hydrothermal cassiterites are lower in the Sn deposit La Villeder (granite/schist-hosted) than in the Sn deposit Abbaretz (schist-hosted), although the Sn mineralizations are considered to be associated with the same types of granite. In the La Chataigneraie district, hydrothermal Fe-Ti-rich and Nb-Ta-poor cassiterites disseminated in schists (Le Prunet, Entraygues) and Nb-Ta-rich magmatic cassiterite in the Entraygues leucogranite derived both from magmatic fluids (Lerouge et al., 2007). These chemical variations of cassiterite from the different granite-related hydrothermal Sn mineralisations probably reflect the complex processes of Sn-ore deposition, including chemical heterogeneities of the peraluminous magmas (due to, e.g., different degrees of partial melting, fractional crystallisation, different sediment sources), but also interactions of magma-derived fluids with various host-rocks (various compositions, P-T conditions) and mixing of magma-derived fluids with external fluids (meteoric, metamorphic) (Lehmann, 1990; Linnen & Cuney, 2005).

On this basis, the Fe vs. Nb+Ta diagram of Tindle & Breaks (1998) could be interesting to efficiently target hydrothermal Nb-Ta-rich cassiterites, which have a favourable chemistry for In incorporation in the cassiterite lattice.

#### 6. Conclusion

The purpose of this work was to evaluate the distribution of In within Sn-polymetallic mineralizations in the western Variscan Belt, to discuss criteria favouring the presence of In within cassiterite, and to determine how In may be substituted. The EPMA analyses of cassiterite from the twelve Variscan Sn-polymetallic ore deposits and from the Ordovician Sn–W ore deposit of Montbelleux provide evidence of low substitution of Sn by Ti, Fe, Nb, Ta and In, highly dependent on ore type, magmatic processes, host rock, fluid/rock interaction processes, and fluid mixing. The highest Ti and Fe substitutions in cassiterite are observed in ores hosted by metapelitic schists. The highest Nb and Ta substitutions in cassiterite are observed in magmatic ores



Fig. 4. (a)  $(Sn^{4+}, Ti^{4+})-(Nb^{5+}, Ta^{5+})$ -Fe ternary diagram showing chemical composition of all analysed cassiterite; (b) detail of this ternary diagram for cassiterite of Galicia (Spain); (c) detail of this ternary diagram for cassiterite of French Massif Central; (d) detail of this ternary diagram for cassiterite of French Armorican Massif; (e) detail of this ternary diagram for indium-bearing cassiterite of Abbaretz, Marcofán and Montbelleux; (f) binary diagram of (Nb + Ta) vs. (Fe + Mn) for cassiterite from all the studied deposits. The fields of hydrothermal cassiterite and cassiterite in rare element granites and pegmatites were defined by Tindle & Breaks (1998).

associated with rare-metal granites and in hydrothermal vein-type mineralizations genetically linked with peraluminous granites (Montbelleux, Marcofán and Abbaretz). The significant In content in hydrothermal cassiterites from Montbelleux, Marcofán and Abbaretz correlates with the highest levels of Nb, Ta and Fe substitution, allowing two coupled substitutions to be proposed: (1) 2 (Sn<sup>4+</sup>, Ti<sup>4+</sup>)  $\leftrightarrow$  (Fe<sup>3+</sup>, In<sup>3+</sup>) + (Nb<sup>5+</sup>, Ta<sup>5+</sup>) and (2) Fe<sup>2+</sup> + (Nb, Ta)<sup>5+</sup>  $\leftrightarrow$  In<sup>3+</sup> + (Ti, Sn)<sup>4+</sup>, depending on the Fe valence.

These substitutions may be limited by the co-formation of tantalite or other Ta-bearing minerals. Further micro-X-ray fluorescence spectrometry coupled with micro-X-ray diffraction and near-edge X-ray absorption fine structure (XANES) measurements at the Fe *K*-edge could be of major interest to further constrain substitution mechanisms and Fe valency, and the consequence of these substitutions on the cassiterite lattice. In all deposit types, sulfides are the dominant In carriers, notably stannite (up to 3000 ppm In),

chalcopyrite (up to 2800 ppm In), pyrite (up to 1490 ppm In), sphalerite (up to 700 ppm In) and arsenopyrite (up to 740 ppm In). Interestingly, other oxides host In, notably rutile (up to 1100 ppm In).

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