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Editorial

# Critical Raw Materials Recovery through Bio/Hydrometallurgy from Secondary Resources

Manivannan Sethurajan <sup>1</sup> and Eric D. van Hullebusch <sup>1,2,\*</sup> 

<sup>1</sup> Department of Environmental Engineering and Water Technology, IHE Delft Institute for Water Education, Westvest 7, 2611 AX Delft, The Netherlands; biotek\_mani@yahoo.co.in

<sup>2</sup> Université de Paris, Institut de physique du globe de Paris, CNRS, F-75005 Paris, France

\* Correspondence: vanhullebusch@ipgp.fr; Tel.: +33-1-83-95-78-36

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## 1. Introduction and Scope

Demand for critical raw materials (CRMs) to be used in consumer products is growing rapidly. However, in the past couple of decades, the world's high-grade metal reserves have been depleted considerably. As a result, alternative resources are currently being explored for metal extraction. In this regard, secondary resources have received considerable attention as they contain a considerable amount of valuable metals. Conventional pyrometallurgical processes are not really of use for resource recovery from secondary resources because of their high energy and cost requirements.

On the other hand, bio/hydrometallurgy is a fast developing, eco-friendly, and cost-effective technology for the extraction of base and precious metals and rare earth elements (REE). Hydrometallurgy consists of leaching and recovery unit operations. Leaching is the solubilization of metals from a solid phase using chemicals or biological agents, whereas recovery is the extraction of metals from polymetallic leachate using physicochemical processes, electrowinning, or biological processes [1]. Bio/hydrometallurgy can be successfully applied not only to a variety of mineral ores such as high grade, low grade, and lean grade ores, but also to secondary resources (such as slags, sludges, red mud, dusts, fly and bottom ashes, and electronic wastes) [2].

## 2. Contribution to the Special Issue

Researchers around the globe investigating the bio and hydrometallurgy of critical, precious, and rare earth elements have been invited to submit research papers so that readers can recognize the common points between them. Among the submitted manuscripts, six papers have been published in the issue.

### 2.1. CRMs from Waste Electrical and Electronic Equipment (WEEE)

Nowadays, recovery of CRMs from waste electrical and electronic equipment (WEEE) is a top priority even in developed countries. These WEEE such as printed circuit boards (PCB), spent liquid crystal displays (LCDs) and light emitting diodes (LEDs), are fast generating waste groups and they contain several CRMs such as Ga, In, W, Nd, Pd, Ta, and other Platinum Group Element (PGE), REE, along with base and heavy metals. Articles that deal with metal recovery from WEEE have been considered for this special issue.

Chen et al. [3] reported the recycling of gallium nitride (GaN) from spent LEDs using pressurized acidic leaching. More than 98% of Ga can be leached from spent LEDs under optimum conditions (i.e., 0.25 mol·L<sup>-1</sup> of HCl with a liquid–solid ratio of 30 mL·g<sup>-1</sup> at 200 °C (15 atm)) within 3 h [3]. Sethurajan et al. [4] investigated the leaching and selective recovery of Cu (as Cu sulfide) from waste PCBs. It was reported that more than 98% of Cu can be leached under optimum conditions (ferric

sulfate concentration (100 mM), agitation speed (300 rpm), temperature (20 °C), and solid-to-liquid ratio (10 g·L<sup>-1</sup>), and more than 95% of Cu can be selectively recovered as Cu sulfide from the poly metallic leachate by selective metal sulfide precipitation technique.

### 2.2. CRMs from Industrial Solid Wastes and Acid Mine Drainage

Metallurgical wastes from metallurgical industries such as slags, dross, spent catalysts, and sludges contain economically significant levels of precious metals and REE. Acid mine drainage (AMD) also contains economically significant levels of toxic elements and high value metals. Articles focusing on the recovery of these metals from metal bearing wastes materials have also been included in this special issue.

Drzazga et al. [5] investigated and reported the leaching of CRMs such as In and Ge from Polish Zn dross using sulfuric and oxalic acids. It was observed that more than 80% of In and Ge can be leached using sulfuric acid at high temperature (80 °C), while the leaching by oxalic acid was comparatively less than sulfuric acid leaching. Ding et al. [6] used response surface methodology (Box-Behnken experimental design) to optimize the parameters that could influence the leaching of Pt from the spent catalysts, and reported that more than 98% of Pt can be leached under optimum conditions (HCl 1.45 mol·L<sup>-1</sup>, NaCl 4.55 mol·L<sup>-1</sup>, 10% H<sub>2</sub>O<sub>2</sub>/spent catalysts of 0.66 mL<sup>-1</sup>, and S/L ratio 1:4.85). Silva et al. [7] studied and reported the recovery of Cu (as covellite) from acid mine drainage using sulfide precipitation technique. Biogenic sulfide (produced by sulfate reducing bacteria) was used to precipitate soluble Cu in AMD to insoluble CuS. It should be noted that the precipitates formed after 96 h exposure had better ordination than the ones precipitated after 48 h exposure.

### 2.3. CRMs from Low Grade Ores

A publication dealing with biotechnological innovations in extractive metallurgy such as the use of microorganisms for the leaching and recovery of metals (i.e., biohydrometallurgy) from primary ores and secondary resources has also been presented in this special issue. Nascimento et al. [8] proposed the use of *Acidithiobacillus ferrooxidans* in column and stirred tank reactors for bioleaching of copper from a mineral sulfide (mostly chalcopyrite) of the Brazilian Amazon region. After 47 days, the column's copper bioleaching efficiency was 1% and 0.95% for 2.00/4.75 mm sulfide ore, respectively, whereas the stirred reactors bioleaching displayed better results with 4% Cu leaching for the sulfide ore.

## 3. Conclusions

The Special Issue, "Critical Raw Materials Recovery through Bio/Hydrometallurgy from Secondary Resources" deals with research articles covering advanced approaches to bio and hydrometallurgy of critical raw materials. The guest editors believe that this collection of papers may be useful to people who are actively involved (directly or indirectly) in this field. This special issue was successful already, but if the articles in this issue can inspire and invite more research studies, debates, and discussion in the field, that will make it even more special.

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