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Organic Matter Oxidation of the Tégulines Clay formation, (Paris Basin, France): Spatial Heterogeneities

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Abstract

Samples of Albian Tégulines Clay outcropping near Brienne-le-Château in the eastern part of the Paris Basin were subjected to a geochemical characterization of organic matter (OM), combining OM bulk Rock-Eval analysis (RE), organic petrography and molecular analysis. In this study, the objective was to investigate the oxidation of natural organic matter in critical zone developed in early reduced marine clay formation. For that purpose, core samples from three boreholes crosscutting the Tégulines Clay formation down to the Greensands were selected. Several meters of Brienne marls preserved Tégulines Clay from atmosphere and therefore from high weathering and high oxidation in AUB240 borehole, whereas Tégulines Clay were weathered in their upper part in AUB1010 and in AUB230 boreholes as a result to their vicinity to the surface. The results showed that the Total Organic Carbon (TOC) contents in all clays are very low, ranging between 0.3 and 0.7 wt.%. The RE bulk data and Tmax parameter of OM in preserved Tégulines Clay indicate that the OM is immature. The organic petrography shows a predominance of gel-like amorphous OM preserved by natural sulphurisation in anoxic environments. Organic matter in Tégulines Clay from the AUB230 and AUB1010 boreholes show similar characteristics. The OM at the top of Tégulines Clay contains large amounts of oxidized lignocellulosic debris and has high RE Oxygen-Index values. In the first upper 10 m of the clays, the RE bulk, organic petrography and molecular data revealed oxidative events (high OI and high oxidized debris contents). Below 10 m, the OM is almost preserved. The OM in the reduced marine Tégulines Clay formation is very little functionalized and seems to be slightly reactive but not negligibly with the inorganic fractions in presence of oxygen.

Keywords: Tégulines Clay formation, weathering profile, oxidation, Bulk organic characterization, Palynofacies, Molecular organic characterization

1 Introduction

Clay-rich geological formations are considered by many countries as natural barriers for radioactive waste isolation due to their low permeability and their sorption properties preventing the release of radioactive substances towards the biosphere. Most of previous investigations concerning clay-rich formations focus on deep systems (> 100 m) without direct interface with biosphere-atmosphere (Hendry and Wassenaar 2000; Hautevelle et al., 2007; Altmann et al. 2012). In the context of sub-surface waste repository, the vicinity of the surface can affect the solid-fluid equilibrium system of the geological formation due to weathering and oxidation phenomena involving in turn potential effects on solute transport properties with depth. Such phenomena in clay-rich formations are scarce and have been recently evidenced in the Lower Cretaceous Tégulines marine-clay formation (Tégulines Clay) near Brienne-Le-Chateau (Aube department, France) investigated for a decade in the context of sub-surface repository for low-level radioactive waste (Debure et al., 2020; Debure et al., 2018; Duffy et al., 2014; Lerouge et al., 2018, 2020). Like most of the marine clay-rich formations, Tégulines Clay evolved under reduced geochemical conditions during early diagenesis and its burial (Lerouge et al. 2018). In the studied area, weathering and oxidation phenomena including changes of mineralogy, chemistry, and petrophysical properties occur in Tégulines Clay at vicinity of the surface, due to interactions with oxygen and meteoric waters. A gradual oxidation profile developed in the clay formation and the resulting effects on pore water chemistry were evidenced based on several indicators as sulfur and iron oxidation states (Lerouge et al., 2018) and the changes of CO_2 partial pressure (Lerouge et al., 2020). Reactivity of Organic Matter (OM) with oxygen is suspected to play a role on CO_2 and redox regulation of the Tégulines Clay at vicinity of the surface.

In this study, we propose to investigate the reactivity of natural OM in Tégulines Clay as a marker of weathering and oxidation, using bulk and molecular techniques in the objective to better evaluate how atmosphere interacts with the rock and regulates its solid-fluid systems with depth. Organic Matter (OM) bulk analysis is firstly achieved by Rock-Eval analyses associated with palynofacies. This analytical approach allows determining the OM composition and its quality in term of maturity, refractory and redox state. Additional molecular organic geochemistry secondly completes the OM characterization, in term of origins (marine/terrestrial), maturity (saturated hydrocarbons) and oxidation (deposition mode

and weathering). Such type of approach was previously applied in deep clay-rich geological formation (Deniau et al., 2008; Elie et al. 2000; Elie and Mazurek, 2008; Hautevelle et al. 2007; Landais et al. 1991; Landais and Elie, 1999) and is here challenged in a sub-surface context.

2 Materials and methods

2.1 Geological setting and mineralogy

The studied area is located near Brienne-le-Chateau in the eastern part of the Paris Basin (Aube department, Figure 1 (a,b)). The Gault clay formation (Gault Clay) consists of siliciclastic shales deposited in an open marine environment from Middle to Upper Albian (Lower Cretaceous) on the Greensands formation (Amedro et al., 2014). The stratotype of the Gault Clay defined in the Aube department consists of Tégulines Clay (82 m) overlaid by the Brienne marls (43 m). At the maximum burial of the Paris Basin, temperatures reached 30-36°C in the Gault Clay (Lerouge et al., 2018). In Early Miocene, the Paris Basin chalk formations were eroded. Nowadays the Gault Clay outcrops as an 8-10 km wide and 80-km long band of terranes-oriented NE-SW through the Aube department. The Brienne marls only subsist in the western part of the studied area, elsewhere the Tégulines Clay outcrops. Its thickness increases from east to west between 60 and 120 m. Locally few meters of Quaternary surficial formations overly Gault Clay.

The unweathered Tégulines Clay essentially consists of clay minerals (47-72 %) associated with quartz-feldspar (28-43 %), carbonates (0-22%), and minor pyrite and phosphate nodules. Mineral variability in unweathered Tégulines Clay is mainly vertical (Fig. 1 (c)). The first 9-12 meters at the bottom of the clay formation is characterized as a clay-quartz rich unit (named UAQ), with the highest quartz – feldspar silty content (46 %) and the lowest carbonate content (3.5 %). The next 35-40 m constitutes a clay-rich unit (named UA) with a clay average content of 51 %, and intermediate carbonate and quartz-feldspar silty contents. The last 20-25 meters at the top of the clay formation constitutes a carbonate-clay rich unit (named UAC), characterized by the highest carbonate content (9-25 %) and the lowest quartz – feldspar silty content (30 %). The UAC unit constitutes a transitory term with the Brienne marls (Lerouge et al. 2018).

2.2 Materials, sampling campaigns

Fifty-three core samples of Tégulines Clay were selected among three boreholes for the study of organic matter: 7 from the AUB240 borehole (2018 drilling campaign), 31 from the AUB230 borehole (2017 drilling campaign) and 15 from the AUB1010 borehole (2015 drilling campaign). Additional samples of soil and Quaternary surficial formations were collected in a trench (TPH 1-1) corresponding to the first meters of the AUB230 borehole, due to the difficulty to recover the first meters during the drilling. In the AUB240 borehole, surficial formations and ~12 meters of Brienne marls overlaid the Tégulines Clay, allowing a protection for the Tégulines Clay. Consequently, core samples collected in the Tégulines horizon from this borehole can be considered as the most representative of compacted pristine Tégulines Clay. The AUB230 borehole, located on a ridge top in cropland, crosscut ~ 30 cm of brown soil, ~ 5 m of surficial formations (sandy loam) and ~ 63 m of Tégulines Clay down to Greensands. The AUB1010, located down the valley, crosscut less than 2 m of surficial formations and ~ 33 m of Tégulines Clay down to Greensands (Fig. 3).

Major mineralogy of the Tégulines Clay core samples from the AUB230 and AUB1010 deduced from XRD data treatment are consistent (Lerouge et al., 2018, 2020) (fig. 1 (c)).

At the top of Tégulines Clay from the AUB230 and AUB1010 boreholes, claystone is sticky and grey with iron oxi-hydroxides. Interactions between Tégulines Clay and surficial oxidizing waters induce mineralogical changes such as pyrite oxidation, calcite dissolution and precipitation of iron hydroxides and gypsum between the top of the formation down to ~11 m. Detailed mineralogical, geochemical and petrophysical work on Tégulines Clay of the AUB1010 and AUB230 boreholes rather indicated an extension of the oxidation front down to ~20 m depth (Lerouge et al., 2018, 2020). Current investigations provide evidence of ancient root network development down to ~12 m depth at least.

2.3 Methodology

2.3.1 Rock-Eval analysis (RE)

The type, quality and amount of OM preserved in TCF samples before and after thermal maturation were determined using new Rock-Eval 6® pyrolysis (Vinci Technologies, Rueil Malmaison). About 50 to 60 mg of powdered and dried samples were used for the analysis depending on the estimated OM

contents. The pyrolysis program involved an isothermal stage at 200 °C, held 2 min under inert gas (helium); the temperature was then raised to 650 °C (30 °C.min⁻¹) and held for 3 min. The oxidation phase started with an isothermal stage at 400 °C under purified air. Then, the oven temperature was raised to 750 °C (at 30 °C.min⁻¹) and held for 5 min. New Rock-Eval 6® classical parameters and their significance were explained by Espitalie et al. (1985a,b) and Lafargue et al. (1998). 4 parameters were used in the present study: (i) the total organic carbon content (TOC, wt %), which accounts for the quantity of OM calculated from the integration of the amount of thermo-evaporated free hydrocarbons (S1 peak), hydrocarbons produced by kerogen pyrolysis (S2), S3CO and S3CO₂ produced by the breakdown of kerogen under inert gas and S4CO and S4CO₂ produced by the oxidation and the pyrolysis of residual carbon under purified air; (ii) the oxygen index (OI, mg CO₂.g⁻¹ TOC), which is the oxygenated quality of OM calculated from the S3 peak; (iii) the hydrogen index (HI, mg HC.g⁻¹ TOC), which represents the amount of HC quality produced during pyrolysis (calculated from the S2 peak); (iv) the T_{max}, which is deduced from the T_{pic} of Rock-Eval VI. T_{pic} is the temperature of the peak oil production during pyrolysis. T_{max} is often used as an OM maturity indicator (Espitalie et al., 1985a,b). The Total Organic Carbon (TOC), the Hydrogen Index (HI) and the Oxygen Index (OI) are the 3 main parameters used in environmental science (Di-Giovanni et al. 1998, Boussafir et al. 2012). T_{peak} deduced from S2 spectra is also used in environmental science. It is the temperature of the oven for the maximum hydrocarbon released.

Thanks to the deconvolution of the S2 pyrolysis spectra using Peakfit software (SPSS®), we were able to show the presence of 5 elemental Gaussian distributions describing the release of the different pyrolytic molecular families at T_{peak} temperatures (+/- 20°C) of 260 °C, 310 °C, 370 °C, 460 °C and 560 °C. The relative contribution of these different pyrolytic molecular compounds to the signal was also semi-quantified by the calculation of peak surfaces of the corresponding Gaussians plotted to those of the entire S2 signal surface.

2.3.2 Microscopic investigations

Initially developed by Combaz (1964), the quantitative organic petrography is a method that consists to study thin slides under a microscope that permit the identification and the quantification of the organic

compounds present in a constant amount of sediments (1 cm³ or 1 gram) after the elimination of carbonate and silicate by hydrochloric and hydrofluoric treatments (Graz et al. 2010). Optical investigations were performed with a DMR XP Leica microscope by using the transmitted light mode. The approach is based on the distinction of different categories of petrographic components characterized by their optical, form properties and, origins (Combaz 1980; Tyson 1995). This quantification was carried out on a minimum of 500 particles with a 50x magnification objective (Di-Giovanni et al. 1998, Sebag et al. 2006 a,b, Graz et al. 2010). To achieve absolute quantification of palynofacies, a pollen standard of precise concentration has been introduced in the samples. (Battarbee and Kneen 1982; Vernal et al. 1987) In this study, the standard chosen is *Cupressus sp.* Because of its strong reaction under UV excitation thus eliminating any confusion. Three main classes were distinguished in the different samples: the amorphous organic matter, without any biological apparent structures, lignocellulosic debris inherited from higher plant and other particles distinguished by their easily recognizable morphology. Different categories were observed in these three main classes and are detailed in Table 1 and Figure 2.

2.3.3 Lipid molecular analysis

Lipids were extracted from ca. 5 g of dried sediment using a Dionex ASE 200 with a dichloromethane (DCM): methanol (MeOH) mixture (9:1 v/v; at 100 °C and 1000 psi for 5 min in 3 cycles). After solvent evaporation, the total extract was then separated into neutral, acid and polar fractions using liquid chromatography with aminopropyl bonded silica as solid phase (Jacob et al., 2005). 5 α (H)-cholestane was added as an internal standard to the neutral and acid fractions and dried under N₂ flux. The acid fraction samples were then methylated by adding a mixture of anhydrous MeOH and acetyl chloride kept at 60 °C for 1 h. The obtained methyl ester and neutral fractions were then derivatized by reacting 40 μ l *N,O*-bis(trimethylsilyl)trifluoroacetamide (BSTFA) in 60 μ l pyridine at 60 °C for 1 h. and analyzed by GC-MS.

GC-MS analyses were performed on a Trace GC coupled to a TSQ Quantum XLS mass spectrometer (both from Thermo Scientific). Gas chromatograph (GC) was fitted with a Trace Gold TG-5MS capillary column (60 m x 0.25 mm id, 0.25 μ m film thickness). The sample was dissolved in toluene and injected

with a splitless mode in a 2 μ l volume. The injector temperature set at 280 °C. Helium was the carrier gas at a flow rate of 1 ml.min⁻¹. The GC operating conditions were as follows: temperature increase from 40 to 120 °C at 30 °C/ min and 120 to 310 °C at 3 °C/min with a final isothermal hold at 310 °C for 68 min. The mass spectrometer was operated in the electron ionization (EI) mode at 70 eV and scanned from 50 to 650 Daltons. The identifications of the different peaks were based on a comparison of mass spectra with published data. Lipids concentration was estimated by the comparison of TIC area of each lipid with that of the internal standard and normalized to the weight of the extracted sample.

3. Results and discussion

3.1 Rock Eval data

3.1.1 Quantitative and qualitative characterization of the OM

For all the analyzed samples, the results obtained by Rock-Eval analysis are presented in Figure 3 and Supplementary data 1. While, the TOC values indicates an organic content below 1 wt.% for the Tégulines Clay, the content is varying with the depth with an increase from 0.36 to 0.59 wt.% in the first 10 meters and with a small variation around a 0.53 mean value in the deeper part. The Hydrogen Index (HI) and the Oxygen Index (OI) were reported both in the Figure 3 as a function of depth and in a Pseudo van Krevelen diagram to characterize the OM type (Fig. 4). Again, as detailed hereafter the first 10 m show a different behavior with a decrease of HI and OI while below 10 m OI and HI are homogeneous and showed almost no variation with depth

Soil and Quaternary formation

Soil and Quaternary formation are available only for AUB230. The soil at the top of the AUB230 borehole (TPH1-1 0-30 cm) has a TOC of 1.45 wt.%, and is characterized by high HI and OI values, revealing oxygenated immature OM. These data are in good agreement with soil data of the literature (Copard et al. 2006).

Quaternary surficial formation at the top of the Tégulines Clay in the AUB230 borehole does not contain any analyzable OM, except the samples at 4 and 5 m, that corresponds to the transition with Tégulines

Clay. The latter show low but valid TOC, and a hydrogenated quality (HI) and oxygenated (OI) average ranging respectively between 100 and 286 mg HC. g⁻¹ TOC and 268 and 307 mg CO₂. g⁻¹ TOC.

Tégulines Clay

Tégulines Clay in the AUB240 borehole (Fig. 3) shows homogenous RE parameters (TOC, HI, OI) suggesting stable OM and can be considered as a reference in term of types and preservation of OM in the clay formation for this study. TOC values are extremely low, ranging between 0.4 and 0.7 wt.%. The HI through the clay formation varies between 17 mg HC. g⁻¹ TOC and 30 mg HC. g⁻¹ TOC. The OI is ranging between 122 and 176 mg CO₂.g⁻¹ TOC.

TOC values of Tégulines Clay in the AUB230 borehole (AUB230 6.10 - 68.45 m) range between 0.4 – and 0.7 % and are almost similar to that of Tégulines Clay in borehole AUB240. At the top of the clay formation at 6.00-6.10 m, TOC value of 0.4 is low compared to the soil (TPH1-1 0-30) and remains almost homogeneous until 45 m. In the deepest part (>45 m), the TOC increases slightly to reach 0.8 wt. %. At 70 m toward the transition with Greensands, the TOC decreases to 0.5 wt.% (Supplementary data 1). The HI decreases between 6 and 9 m (98 – 25 mg HC. g⁻¹ TOC) and remains stable, until 55 m where the HI increases again to reach 118 mg HC g⁻¹ TOC at 68 m. The OI is relatively high (mean around 180 mg O₂ g⁻¹ TOC) between 6 and 9 m and remains stable until the depth of 45.84 m (around a mean of 76 mg O₂ g⁻¹ TOC). However, one pulse of OI can be observed at 21.3 m (117 mg CO₂ g⁻¹ TOC). Below 45.84 m the OI values decrease (37 to 48 mg O₂ g⁻¹ TOC). The RE parameters measured in Tégulines Clay of the AUB230 borehole allow discriminating three main parts corresponding to the top (6 – 9 m), the center (10 – 45 m) and the bottom of the borehole (55 – 68 m). At the top of Tégulines Clay, the HI/OI relationship is quite classical with an obvious anti-correlation: the HI decreases and OI increases in this section demonstrating a classical oxidation profile (Marchand et al. 2008).

Tégulines Clay of the AUB1010 borehole (Fig. 3) are characterized by low TOC content at the surface and then remains homogenous with values comprised between 0.3 % and 0.6 %. The HI is high at the top (192 mg HC. g⁻¹ TOC) and decreases rapidly at 3.31 m (81 mg HC. g⁻¹ TOC). Deeper, the HI remains homogenous (17 – 29 mg HC g⁻¹ TOC). The HI value at 1.88 m is high (192 mg HC. g⁻¹ TOC) as for the soil sample (TPH1-1 0-30) revealing more immature well-preserved OM. The OI values are

extremely high at the surface (883 mg CO₂.g⁻¹ TOC and 912 mg CO₂.g⁻¹ TOC at 1.88 m and 3.31 m respectively) due to low TOC content and probably the input of surficial pedogenetic OM. Deeper, the OI decreases rapidly and remains relatively stable along with the profile (122 – 241 mg CO₂.g⁻¹ TOC). The trends of the HI/OI ratios in Tégulines Clay of the AUB1010 and the AUB230 boreholes are similar. Tégulines Clay OM of the three boreholes plot in type III OM field, revealing the terrestrial origin and/or well-oxidized type II OM. Tégulines Clay OM of the borehole AUB240 shows intermediate OI and HI values indicating the most preserved conditions of OM. Tégulines Clay OM of the AUB230 borehole is a mixture of type II- and type III- OM, revealing 1) either the ratio of inherited marine OM (type II) and terrestrial input (type III) or 2) a degradation process of marine OM allowing a shift from type II- OM to an intermediate type with a decrease in the HI and a slight increase in the OI compared to the original quality of the OM at the time of its deposition. Tégulines Clay of borehole AUB1010 shows oxygenated OM revealing more oxic conditions resulting in a shift from type II or more terrestrial oxidized (Fig. 4).

3.1.2 T_{max} and shape of the pyrolysis S2 peak

The S2 signal delivered by the Rock-Eval pyrolysis is a spectrum that allows defining the rate of pyrolysate hydrocarbons expressed as a function of temperature (Copard et al. 2006). Deconvoluting the S2 signal permits us to better understand the evolution of the pyrolysis Temperature (T_{peak}) to unravel the composition of the natural organic matter, which is generally composed of a complex mixture of organic compounds. The S2 peak is non-Gaussian and is accompanied by shoulders for the immature samples. The different peaks and shoulders correspond to the cracking of the different organic compounds (Disnar 1982; Disnard & Trichet 1984; Disnar et al., 2003). The literature concerning the shape of the pyrolysis S2 peak is mainly focused on soils samples. In this study, the different peaks of OM in the soil sample (TPH1-1 0-30 cm) can be attributed to specific compounds. The S2 peak shows a broad shape and two T_{peak} at 370 °C and 460 °C (C3 and C4) (Fig. 5a), typical of a soil sample and in agreement with the bulk RE results. The cluster C3, at 370 °C can be attributed to the thermal breakdown of biological constituents such as polysaccharides and lignin (Disnar et al. 2003). The C4 cluster shows more preserved OM. For this sample, cluster C1 is absent. In their study, Copard et al. (2006) showed

that the C1 cluster (260 °C) is more labile and easily mineralized and could be easily absent in the most diagenetically or pedogenetically evolved samples.

For OM in Tégulines Clay, the shape of the pyrolysis S2 peak does not permit to attribute a specific compound but will permit to interpret the results in term of preservation. The mathematical deconvolution of S2 peak show five molecular RE pyrolytic families named C1 to C5 with 260, 310, 370, 460 and 560 °C T_{peak} respectively (Fig. 5, 6 and supplementary data 2).

Tégulines Clay OM in AUB240 borehole shows very homogeneous cluster contribution with depth (Fig. 5b; Fig. 6a). C4 is the major cluster encountered (80 %), followed by C3 and C5. The distribution of these contributions reveals preserved conditions with diagenetically well stabilized OM and are in accordance with the bulk RE values (Copard et al. 2006).

Tégulines Clay OM in AUB230 and AUB1010 boreholes show variations of the cluster distribution with depth. In AUB230 borehole, OM between 9 and 55 m show a high contribution of the C4 cluster (Fig. 5d, Fig. 6b supplementary data 2) that is almost similar to Tégulines Clay OM in AUB240 borehole,, revealing stable OM input in probably good preserved conditions. At 21 m, the C2 cluster is more represented indicating the presence of labile OM. At the top of the clay formation (6.10 m), OM is particular with a high contribution of the C1 cluster (Fig. 5c). This could be due to the presence of light recycled OM composed of a small molecular structure easily thermopyrolysable. At the bottom of the clay formation (67 and 68 m) (Fig. 5e), the higher contribution changes to C3 cluster also revealing stable and preserved OM (Copard et al. 2006).

Tégulines Clay OM in AUB1010 borehole (Fig. 5; Fig. 6c) shows an S2 deconvolution similar to that in AUB230 borehole. The most upper sample (AUB1010 1.88, Fig. 5f) is characterized by the contribution of C1, C3 and C4 (35, 45 and 20% respectively) clusters. The C1 cluster reveals the presence of thermally labile biological compounds and shows more immature OM that could be, as for the AUB230 6.10 m, a similar light recycled OM. Between 3 and 16 m, the major cluster is C4 (Fig. 5g, h): stable OM with a variable contribution of the other clusters. Between 17 and 34 m (Fig. 5i), the major cluster is C4, followed by C3 and C5. These conditions are similar to the AUB240 borehole and suggest more stabilized and preserved OM.

The R400 parameter (proportion of the S2 peak integrated before 400°C, Disnar et al. 2003) and depending on the proportion of labile and probably more diagenetically reactive biomolecules was also calculated (supplementary data 1). This parameter varied between 0.1 and 0.6 and was higher for surface samples (0.6 at 0-30 cm) indicating the presence of 60 % of labile OM in the surface. This parameter was low in the AUB240 borehole (0.10 – 0.14) which is related to the 90 % of more stable OM (10 % of labile OM) with highly refractory character.

T_{\max} is also largely used in the literature in order to characterize the maturity of the OM. A general overview of the samples reveal that the T_{\max} is lower than 430 °C, suggesting that the OM is considered immature.

3.2 Quantitative palynofacies investigations

The particles observed in the different samples were identified following the different morphological and textural criteria presented in Fig. 2 and table 1. All the acronyms cited in the next section are detailed in the table 1.

The quantitative palynofacies results for TPH1-1 are presented in Fig.7a. As expected from the Rock-Eval results, the sample collected in the first 30 cm (0-30 cm) reveals a predominance of preserved lignocellulosic fragments (plc), slightly degraded lignocellulosic fragments (slc) plc+slc (4.41 mg. g⁻¹) and divers particles (div) (1.70 mg.g⁻¹). The pyrite is absent at this depth. These lignocellulosic compounds are, in their majority under gel-like forms, suggesting the sequestration of these particles in an anoxic hydromorphic environment permitting protection artificially the OM (Boussafir et al 2012). These parameters are in accordance with the RE data (high TOC, HI) and S2 deconvolution are typical of a soil sample. The samples from the trench collected below 30 cm (30 cm to 5 m) did not permit to perform any quantitative analysis due to the too-small number of residual particles in the palynofacies. Tégulines Clay OM in AUB240 borehole (Fig. 7b) shows a homogeneous palynofacies content with depth. The major particles encountered are the grAOM (1.11 – 2.31 mg.g⁻¹) followed by the OgAOM (0.06 – 0.69 mg.g⁻¹) and do (0.26 – 0.51 mg.g⁻¹). OgAOM content increases between 29 and 80 m (0.06 mg.g⁻¹ – 0.69 mg.g⁻¹) and pyrite is observed in inclusion. All these features suggest preserved anoxic conditions of the initial deposit and agree with the RE and deconvolution results.

For the borehole AUB230 (Fig. 7c), at 6.10 m, the palynofacies is dominated by grAOM (1.03 mg.g⁻¹). There is also the presence of pyrite (0.59 mg.g⁻¹) and od (0.27 mg.g⁻¹) with small particles (between 10 and 20 µm). In addition, the div category decreases in comparison with the surface. At 9.36 m, the palynofacies is characterized by small particles (around 1 µm) and pyrite is absent. The presence of small black particles as opaque-debris and/or oxidized Lignocellulosic particles suggests more oxidative conditions. Between 10 and 18 m the grAOM is present in a higher proportion (between 1.47 and 4.51 mg.g⁻¹). The proportion of OgAOM is higher than the 9.36 m sample (between 0.090 and 0.24 mg.g⁻¹). The od proportion is homogeneous in this depth range. At 25 m, the proportion of OgAOM increases (0.35 mg.g⁻¹) and the pyrite content also increases (0.49 mg.g⁻¹). Between 30 and 55 m, the proportion of grAOM is higher than the other samples (between 1.99 and 4.83 mg.g⁻¹) there is also the presence of pyrite and the od category detains bigger size (~100 µm). At the end of the borehole, (55-68 m), the proportion of grAOM is the highest (4.83-6.83 mg.g⁻¹), the OgAOM proportion is also high (0.57-0.84 mg.g⁻¹) with bigger size and pyrite in inclusion. The palynofacies show a higher contribution of grAOM and OgAOM with pyrite inclusion. This feature also shows more preserved conditions. The increase of the AOM particles is correlated with the increase of HI and the principal contribution of C3 cluster suggesting that this AOM is the principal component of kerogen and suggest more preserved conditions of the formation.

The palynofacies for the AUB1010 borehole was difficult to interpret because the particles were present in very low proportion and could not permit any quantitative work. The particles were mainly composed of very small (around 1 µm) OgAOM and od particles. The samples of the AUB1010 borehole are similar to the AUB230 (6.10 m) sample and suggest more oxidative conditions.

The palynofacies is representative of the organic matter composition after the sedimentation and the organic diagenesis (Tyson 1993, Zhang et al. 2015). The different assemblages of refractory particulate organic matter permit to record environmental evolution and depositional characteristics. In particular, the different distributions can reflect the conditions of the oxygenation (Tyson 1993, Disnar et al. 2003, Di-Giovanni et al. 1999a). As mentioned in the geological setting, the Gault Clay was deposited in an open marine environment (Lerouge et al. 2018). Indeed, the presence of Glauconie, Siderite and Pyrite are characteristic of an anoxic marine deposition at shallow depth. In our study, AUB230 (at 55-68 m)

and the AUB240 samples present the well-preserved palynofacies assemblage. These samples show the presence of AOM with pyrite in inclusion (Fig. 7b, c). (Tyson 1993, Boussafir et al 1995a, b) showed that the presence of gel-like AOM is generally associated with high preservation rates and low energy environments and result to anoxic deposits. This OM is similar to that observed in Kimmeridgian clays Formation (Boussafir et al, 1995, a, b, Boussafir et al. 1997) and has an ultrafine monoscopically amorphous structure derived from the preservation by natural sulfuration in an anoxic environment. The H_2S produced during the sulphate-reduction process allows the formation of pyrite in association with the iron present in the sediments and preserve a part of the initially metabolizable OM and therefore becomes resistant to degradation and oxidation. This process is at the origin of the highly stabilized nature of the OM of these clays. Consequently, the OM of Tégulines Clay is insensitive to physicochemical conditions and would probably not have a direct action on the redox system of this clay formation.

3.3 Molecular investigation in Tégulines Clay OM

3.3.1 Total lipids extract

Tégulines Clay samples from the AUB230 borehole were selected for the molecular investigation of organic matter, based on $S1$ (> 0) and TOC values (~ 0.2) of the RE results.

The extract concentrations in Tégulines Clay are characterized by low content in labile compounds (0.3 to 1.0 mg.g^{-1} of dry sediment), and are consistent with the Rock-Eval quantitative (TOC) and qualitative (HI and OI) parameters and the palynofaciès composition (supplementary data 1), indicating the immaturity of OM (low free HC) and preserved insoluble kerogen. The extract concentrations in Tégulines Clay samples from the AUB240 and AUB1010 boreholes are very low and close to 0 mg/g (Rock-Eval, $S1$, supplementary data 1), giving evidence of refractory and residual OM essentially.

3.3.2 Chromatograms patterns: n-alkanes

The molecular composition of the Tégulines Clay samples for the AUB230 borehole was investigated. Three chromatograms patterns are presented in Figure 8 (sample at 6.10 m representing the oxidized area, sample at 25 m and, sample at 66 m representing preserved area). Very few compounds are present

in the different samples. Phthalates compounds were found in the samples (55 – 60 min retention time). the origin of these phthalates is unknown and could be the result of the extremely low concentration of natural organic compounds in the samples that could exacerbate the phthalates peaks and are not related to laboratory contamination. Several steroids were also encountered. In the biosphere, sterols contain essentially 27, 28 and 29 C atoms. C₂₇-sterols produced by animals and C₂₈-sterols produced by microalgae and fungi, C₂₉-sterols are synthesized by land plants (Huang and Meinschein 1979; Volkman 1986). The samples from the top and intermediate depths (6.10 m to 45 m) are composed of sitosterol, sitostanol, cholesterol, cholestanol coprostanol and epi-coprostanol. These sterols have a terrigenous origin. The end of the borehole is exempt from sterols.

This study mainly focuses on the alkanes composition in the different samples as they are more abundant and can lead to several calculations such as the Carbon Preference Index (CPI). The n-Alkanes are ubiquitous in the environment and are directly synthesized by living organisms or derived from straight-chain biolipids (Eglinton and Hamilton, 1967). They can be used to discriminate the sources and the preservation of the sedimentary immature OM (Brocks and Summons, 2003).

The total abundance of n-Alkanes varies between 0.29 and 9.14 µg/g sed, and increases from the bottom to the top of the clay formation (Table 4). In the AUB230 borehole, n-alkanes were dominated by n-C₂₉ to n-C₃₃ (Fig. 8) rather typical of terrestrial OM. The CPI defined by Bray and Evans, (1961) was determined with the following equation: $CPI = 2(\text{odd } n\text{-C}_{23} \text{ to } n\text{-C}_{31}) / (\text{even } n\text{-C}_{22} \text{ to } n\text{-C}_{30} + \text{even } n\text{-C}_{24} \text{ to } n\text{-C}_{32})$. The n-alkanes CPI index is generally interpreted in terms of the origin. A CPI ~ 1 indicates marine origin, whereas a CPI > 1 reflects more terrestrial contribution. Between 9.36 and 45 m, the CPI value is ~1.0 suggesting more marine contribution, however, this value could also reflect the oxidation of the OM. At the end of the borehole, the CPI is higher than 1, reflecting a mixture of marine and terrestrial organic compounds, more representative of the sedimentation conditions. CPI can also be used in order to characterize the degree of preservation of the n-alkanes in the sediments (Xie et al. 2004, Zheng et al. 2007, Jansen and Nierop, 2009 Zhou et al. 2010; Garel et al., 2013). CPI values between 3 and 10 refer to preserved OM. Values between 1 to 3 are the first signs of early degradation. Values close to 1 indicate more bacterial activity or recycled OM. At the start of the diagenesis, the molecular compounds are mainly composed of oxygen functional groups. During the diagenesis, the

defunctionalization of the compounds leads at the end to the formation of hydrocarbons. The lipids that are stocked in the sediments lead to the loss of many compounds and short chains molecules are more degraded than the long chains ones.

The CPI values, calculated from the n-alkanes concentrations showed variations with depth. The CPI was high for the trench sample and decreased after to values around 1 (9.36 – 21.30 meter deep). At the end of the borehole, the CPI values increased again. These features suggest more preserved OM at the top, in relation to the immature state of the OM, more degraded conditions between 9.36 and 21.30 m and again, more preserved conditions (or slightly degraded) at the end of the borehole. These results are in accordance with the RE and palynofacies results. The top of the AUB230 borehole is oxidized and after 21 m, more preserved OM is observed. However, the CPI ratio shows oxidized conditions until 21 m whereas the palynofacies and at a lesser extent, the RE data shows an oxidation profile until 10 m. The CPI ratio is in accordance with the mineralogical data (Lerouge et al. 2018) where the oxidation profile is observed until 21 m.

3.4 Comparison with other marine clay formations – Impact of the oxidation

From a general overview, the TOC values show that the Tégulines Clay are very poor in organic content. Guo et al. (2021) also studied the Tégulines clay in order to investigate the mobility of organic compounds in this formation. Despite a low TOC content (as observed in this study), they observed the retention of certain compounds such as lipophilic compounds. Other marine clay-rich formations that have experimented low temperature such as Tégulines Clay have been extensively investigated in the literature concerning their organic matter composition and variations. The Opalinus Clay from northern Switzerland did not display a higher temperature of burial than 85 °C (Elie and Mazurek, 2008). They also display quite low C contents from 0.1 to 1 wt% (Todorof et al. 1993; Nagra, 2002). Callovo-Oxfordian claystones is another formation, located in the Eastern part of the Paris basin, that experienced low temperature (50 °C) (Landais and Elie, 1999). Their total organic C content is also generally low (<1 wt.% of the bulk rock). The T_{\max} recorded for the Tégulines Clays is essentially < 430 °C. This temperature is revealing immature OM. The RE T_{\max} of the Opalinus Clay also displays temperatures

424 ≤ 435 °C in accordance with low thermal maturity. The thermal immaturity of the Callovo-Oxfordian
425 formations was demonstrated by the predominance of odd/even long-chain n-alkanes with CPI higher
426 than 2 (Hautevelle et al. 2007). In the case of the Tégulines Clay, the CPI cannot estimate the immaturity
427 as most of the formation is under oxidation. However, CPI can be used for the estimation of the
428 preservation. Our study emphasizes that the RE technique is suited to be used for oxidized formations.
429 The HI versus OI plot (pseudo Van Krevelen plot, Fig. 4) is usually employed to separate the marine
430 and terrestrial contributions in preserved conditions. In Elie and Mazurek (2008), the Opalinus Clay
431 falls into the Type III kerogen area, revealing more terrestrial contribution. The Callovo-Oxfordian
432 formations van Krevelen diagram indicate type II and III OM mixture, showing a marine origin, with a
433 contribution of terrestrial-derived materials (Disnar et al. 1996; Hautevelle et al. 2006). For the
434 Tégulines Clay, no conclusion permits to settle between marine or terrestrial contribution because the
435 HI and OI values can also suggest the presence of oxidized marine contribution, leading to the transfer
436 to the type III kerogen area.

437 The molecular investigation of the Tégulines Clay seems also to be perturbed by the oxidation profile.
438 The yields of OM extract in Tégulines Clay are low 0.3 to 1.0 mg. g⁻¹ of dry sediment. By comparison,
439 the extract yield for Opalinus Clay is comprised of 0.9 and 48 mg. g⁻¹ (Elie and Mazurek 2008). The
440 yield fractions of soluble OM ranges between 0.01 and 0.65 mg of organic extract/g of rock for the
441 Callovo-Oxfordian formations (Hautevelle et al. 2007). Despite the low yield of soluble OM, Callovo-
442 Oxfordian formations displayed a large variety of compounds deciphered by GCMS (n-alkanes, steroids,
443 hopanoids among others), permitting to get information on the OM maturity and origin. In contrast, the
444 chromatograms patterns of the different samples from the AUB230 borehole show the presence of a low
445 number of peaks, revealing a low number of molecular compounds (Fig. 8). Only n-alkanes could be
446 easily deciphered. However, the results concerning the n-alkanes distribution and the CPI are
447 questionable. Indeed, the predominant n-alkanes are ranging between C₂₉ and C₃₃, revealing terrestrial
448 origin such as for the Opalinus Clay (Odd/Even predominance of C₂₅-C₃₅ range (Elie and Mazurek,
449 2008)). By contrast, the Tegulines CPI is ~ 1 revealing marine origin. These opposite results could stem
450 from an artifact caused by the oxidation of the OM, shifting the short-chain of n-alkanes. These results
451 can question the use of certain techniques when dealing with oxidized OM.

3.5. Comparison of the TCF OM with mineralogical and dissolved inorganic carbon (DIC) changes through the TCF weathering profile

In this study, the AUB230 borehole enabled to evidence the three main facies (from the bottom to the top: UAQ, UA, and UAC units, see §2.1.) that can be encountered in the TCF, whereas the AUB1010 borehole only crosscut the UAQ and UA units (Lerouge et al., 2018). In the both boreholes, the oxidation profile was observed between the sub-surface and ~21 m depth (Lerouge et al., 2018, 2020) (Fig. 9).

The brown and granulous textures, the bulk and molecular characteristics of the OM at the surface (0-30 cm borehole AUB230) revealed high amounts of organic matter showing typical oxidative degradation in soil, according to the following reaction: $\text{CH}_2\text{O} + \text{O}_2 = \text{H}_2\text{O} + \text{CO}_2$. These observations are in good agreement with the DIC of 1.4 mEq/L measured in soil pore waters, the high CO_2 soil degassing and the low $\delta^{13}\text{C}$ value of degassed CO_2 (-25.5 ‰ PDB) (Lerouge et al., 2020).

Between 30 cm and 5 m deep in the AUB230 borehole, in ocher carbonate-free sandy clay loams overlying TCF, the bulk and OI index of the OM gave evidence of extremely low TOC contents and high oxidative conditions consistent with the altered pedogenetic OM surficial bring (Fig. 4). These observations are in good agreement with the low DIC (<0.9 mEq/L) measured in pore waters and the low $\delta^{13}\text{C}$ values of DIC (~ -21 ‰ PDB) (Lerouge et al., 2020).

Below ~5 m in the AUB 230 borehole and ~2-3 m in AUB1010 borehole, the top of TCF is ocher, plastic and marked by an intense and visible oxidation of pyrite and other iron-bearing minerals (iron-rich clays, ankerite) that decreases down to 11 m depth. Pyrite oxidation leads to iron oxyhydroxide and gypsum precipitation, but also to a small calcite dissolution due to slight acid pH. Ankerite oxidation leads to iron oxyhydroxide precipitation. The clay fraction in TCF is quite homogeneous and composed by ~16±5 % of detrital kaolinite, ~16±4 % of illite-smectite, ~12±5 % of illite-mica, and minor chlorite (<2 %), and ~8±6 % of diagenetic glauconite. The study of clay minerals in the 5-10 m zone, based on microscopic observations, XRD and electron microprobe gave evidence of a change of color from greenish to ocher due to oxidation of iron, a loss of iron to form iron oxides and an increase of the smectite component, in particular concerning chlorite and glauconite (Lerouge et al. 2018, 2020). In the same way, the OI of the OM is high and confirms oxidative conditions. Combined mineralogical and OM data are consistent with the high DIC values (up to 8.5-8.6 mEq/L) measured in Tegulines Clay

pore waters in the 5-10 m zone of the both AUB230 and AUB1010 borehole. The $\delta^{13}\text{C}$ values of degassed CO_2 (-19.5 to -12.9 ‰ PDB) corresponds to a mixing between OM matter degradation and CO_2 due to calcite dissolution (Lerouge et al., 2020). The OM study gave a supplementary information with the OI index, which is consistent with a mixing of the altered pedogenetic OM surficial supply and the oxidation of original organo-clays fraction of the initial marine sediments (Fig. 4). The identification of OM surficial supply down to ~9-10 m is in good agreement with the rare presence of small roots down to 10-12 m in boreholes (Lerouge et al., 2020). The oxidation of pristine marine OM down to ~9-10 m in the weathering profile is also an important point, which is a subject of debate for chemical modelling of weathering profile in marine clay formations (Bao et al., 2017).

Between 9-10 m and 55 m, the OM oxidation state was homogenous and low, whereas changes of consistency from plastic to hard and of mineralogy (slight oxidation of glauconite marked by yellowish color) were still observed at ~ 20-25 m (Debure et al. 2018; Lerouge et al. 2018). The variation between plastic/non plastic behavior of the TCF seems to not play a role for OM preservation.

The bottom of the TCF evolved under reducing conditions, and are characterized by an enrichment in clay minerals, including glauconite and smectite (Lerouge et al., 2018). The OM data also showed more preserved OM with higher HI, the microscopic data also show more preserved conditions with higher grAOM and OgAOM contents with the inclusion of pyrite suggesting the role of bacterial sulphate reduction in an anoxic sedimentary environment. Indeed, the association of pyrite with the amorphous orange gel-like organic matter is interpreted as a non-biodegradable and stable organic matter preserved by incorporation of sulfur by natural sulfurization (Boussafir et al. 1994, 1995, 1997).

At interface with the Greensand, TCF shows a drop of the TOC, S1, S2 and an increase of OI (Supplementary data 1). The green sand detains coarser particles than the Tégulines Clay and could store and/or preserve less OM.

Overall, the OM characterization (RE, palynofacies) of TCF in the different boreholes supports a well-developed oxidation profile in the ~ 0-10 m and preservation of reduced conditions in the deeper parts, in agreement with the previous mineralogical and geochemical studies. As for sulfur and iron oxidation states, the oxidized or preserved state of the OM will finally help to redox regulation with depth in the Tégulines Clay. Indeed, despite a similar TOC content all along the investigated boreholes, the

508 preservation of the organic matter and especially its reducing capacity will be one of the parameters that
509 might influence the mobility of sensitive-redox chemical elements (Debure et al. 2018, 2020). The small
510 amount of preserved, very little functionalized and unreactive OM and the refractoriness and stability
511 of OM of almost all the borehole samples studied plead for a slight influence of OM in the Tégulines in
512 a redox system.

513 Clay minerals, and particularly smectite, can physically protect the OM in these sediments (Hedges and
514 Oades 1997; Nelson et al. 1999, Drouin et al. 2010; Mahamat et al. 2016). The comparison with the
515 organic data (TOC, HI, OI) shows that we have, at the bottom of the borehole an increase of clay
516 minerals, including smectite / smectite-illite and glauconite, and an IH increase. Bruun et al. (2010)
517 stated that smectite have higher capacity to stabilize organic matter than illite due to higher specific
518 surface area and cation exchange capacity. All these information permits to state that the protection
519 of the organic matter at the bottom of the AUB230 borehole (66 – 68 m) can be the presence of
520 smectite in higher proportion.

521 The OM is preserved in the AUB240 borehole while it was more oxidized in the AUB1010 at least
522 between 3 and 16 m and also observed from mineral characterizations. The OM preservation is higher
523 in the AUB240 borehole, as a 12 m layer of Brienne Marls overlies it themselves covered by 8 m of
524 surficial layers (loess, clay sediments and carbonated alluvium). By contrast, the surficial layer in the
525 AUB230 borehole represents only 5 m and even less (3 m) in the AUB1010 borehole. Therefore,
526 meteoric water reaches faster the Tégulines in the AUB1010 and AUB230 boreholes. Finally, the
527 difference between the AUB230 and the AUB1010 borehole relies on their position in the valley. The
528 AUB230 borehole is located at the hilltop while the AUB1010 is located at the bottom where the water
529 can stream from the hills. The position of the AUB1010 borehole is therefore favorable to a higher
530 alteration compared to the other boreholes and all these characteristics explain the higher oxidation depth
531 observed for organic matter in that borehole.

532

Conclusion

This study investigated the OM composition of the Tégulines Clay formation using bulk, microscopic and molecular techniques. All the profiles studied show that Tégulines Clay are poor in organic content in general. The bulk results using RE analyses revealed that the OM of all clays studied samples was immature. The RE parameters, deconvolution of S2 peak, R400, palynofacies and molecular investigations were all positively correlated. Indeed, at the surface (TPH1-1 0-30), the OM showed relatively high HI and TOC with a large amount of preserved particles revealing typical soil litter OM. The deconvolution of the S2 peak curve showed a dominant contribution of the C3 cluster that is typical of the breakdown of polysaccharides and lignin in the surface part of the studied profile. The other trench samples collected below 30 cm showed very oxidized conditions (low TOC and high OI). The AUB240 borehole, from the bulk and microscopic investigation, revealed more preserved conditions, mainly due to the overlaying of the Brienne marls and is presented as a reference borehole in term of OM preservation. The investigation of the AUB230 borehole clearly showed an oxidation profile at the top and more pristine, preserved and reductive conditions deeper. The AUB1010 borehole showed similar conditions to the AUB230 borehole but a higher oxidation depth. These different oxidation depths could be explained by the different spatial topological location of the boreholes. This study showed that the Tégulines Clay OM is: very low in amount, stable in its preserved section, essentially residual and little-sensitive to the biotic and abiotic degradation and thus would be very weakly reactive. The OM influence would also be limited in a redox system supporting previous results obtained on the behavior of sensitive-redox chemical elements. Finally, this is supported on the one hand by the small amount of preserved, very little functionalized and unreactive OM and on the other hand, by the refractoriness and stability of OM of almost all the borehole samples studied. This study also shows that a good set of techniques permit to deal with the complex atmosphere / Geological formations interactions in the critical zone.

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References

- Altmann, S., Tournassat, C., Goutelard, F., Parneix, J. C., Gimmi, T., & Maes, N. (2012). Diffusion-driven transport in clayrock formations. *Applied Geochemistry*, 27(2), 463-478.
- Bao, Z., Haberer, C. M., Maier, U., Amos, R. T., Blowes, D. W., & Grathwohl, P. (2017). Modeling controls on the chemical weathering of marine mudrocks from the Middle Jurassic in Southern Germany. *Chemical Geology*, 459, 1-12.
- Battarbee, R.W., Kneen, M.J. (1982). The use of electronically counted microspheres in absolute diatom analysis. *Limnology and Oceanography* 27, 184 – 188.
- Boussafir M., Lallier-Vergès E., Bertrand P. & Badaut-Trauth D. (1994). Structure ultrafine de la matière organique des roches mères du Kimméridgien du Yorkshire. *Bulletin de la Société Géologique de France*, 165, (4) 355-363.
- Boussafir, M., Gelin, F., Lallier-Vergès, E., Derenne, S., Bertrand, P., & Largeau, C. (1995a). Electron microscopy and pyrolysis of kerogen from the Kimmeridgian Clay Formation, UK: Source organisms, preservation processes, and origin of the microcycles. *Geochimica et Cosmochimica Acta* 59, 3731-3747.
- Boussafir M., Lallier-Vergès E., Bertrand, P., & Badaut-Trauth, D. (1995b). SEM and TEM studies on isolated organic matter and rock microfacies from a short-term organic cycle of the Kimmeridge Clay Formation (Yorkshire, GB), *Organic matter accumulation*, Springer, Berlin, Heidelberg , 15-30.
- Boussafir, M., Lallier-Vergès, E. (1997). Accumulation of Organic Matter in the Kimmeridge Clay Formation (KCF): An Update Fossilisation Model for Marine Petroleum Source-Rocks. *Marine and Petroleum Geology*, 14 (1), 75–83.
- Boussafir, M., Sifeddine, A., Jacob, J., Foudi, M., Cordeiro, R. C., Albuquerque, A. L. S., ... & Turcq, B. (2012). Petrographical and geochemical study of modern lacustrine sedimentary organic matter

- 586 (Lagoa do Caçò, Maranhã, Brazil): Relationship between early diagenesis, organic sedimentation and
 587 lacustrine filling. *Organic Geochemistry*, 47, 88-98.
- 588 Bray, E.E., Evans, E.D. (1961). Distribution of n-paraffins as a clue to recognition of source beds.
 589 *Geochimica et Cosmochimica Acta* 22, 2–15.
- 590 Brocks, J.J., Summons, R.E. (2003). Sedimentary hydrocarbons, biomarkers for early life. In: Holland,
 591 H.D., Turekian, K. (Eds.), *Treatise in Geochemistry*, vol.8. Elsevier, Amsterdam, pp. 63 – 115.
- 592 Bruun, T. B., Elberling, B., & Christensen, B. T. (2010). Lability of soil organic carbon in tropical
 593 soils with different clay minerals. *Soil Biology and Biochemistry*, 42(6), 888-895.
- 594 Combaz, A. (1964). Les palynofaciès, *Revue de micropaléontologie* 7, 205 – 218.
- 595 Combaz, A. (1980). Les kérogènes vus au microscope. In : Durant, B. (Ed.) *Kerogen*. Technips, Paris,
 596 pp. 55 – 111.
- 597 Copard, Y., Di-Giovanni, C., Martaud, T., Albéric, P., & Olivier, J.E. (2006). Using Rock-Eval 6
 598 pyrolysis for tracking fossil organic carbon in modern environments: implications for the roles of erosion
 599 and weathering. *Earth Surface Processes and Landforms* 31, 135 – 153.
- 600 Debure, M., Tournassat, C., Lerouge, C., Madé, B., Robinet, J. C., Fernández, A. M., & Grangeon, S.
 601 (2018). Retention of arsenic, chromium and boron on an outcropping clay-rich rock formation (the
 602 Tégulines Clay, eastern France). *Science of The Total Environment*, 642, 216-229.
- 603 Debure, M., Grangeon, S., Robinet, J. C., Madé, B., Fernández, A. M., & Lerouge, C. (2020).
 604 Influence of soil redox state on mercury sorption and reduction capacity. *Science of The Total*
 605 *Environment*, 707, 136069.
- 606 Deniau, I., Devol-Brown, I., Derenne, S., Behar, F., & Largeau, C. (2008). Comparison of the bulk
 607 geochemical features and thermal reactivity of kerogens from Mol (Boom Clay), Bure (Callovo–
 608 Oxfordian argillite) and Tournemire (Toarcian shales) underground research laboratories. *Science of*
 609 *the Total environment*, 389(2-3), 475-485.
- 610 Di-Giovanni, C., disnar, J.R., Bichet, V., & Campy, M. (1998). Sur la présence de matière organique
 611 méso-cénozoïque dans les humus actuels (bassin de Chaillexon, Doubs, France). *Comptes Rendus de*
 612 *l'Académie des Sciences Paris*, 326, 553 – 559.

- 613 Di-giovanni, C., Disnar, J.R., Campy, M., & Macaire, J.J. (1999a). Variability of the ancient organic
614 supply in modern humus. Analysis 27, 398 – 402.
- 615 Disnar, J.R. (1982). Etude expérimentale de la fixation de divers métaux sur une matière organique
616 sédimentaire d'origine algaire. Maturation thermique des composés formés. Thèse d'Etat, Université
617 d'Orléans.
- 618 Disnar, J.R., Trichet, J. (1984). The influence of various divalent cations (UO_2^{2+} , Cu^{2+} , Pb^{2+} , Co^{2+} ,
619 Ni^{2+} , Zn^{2+} , Mn^{2+}) on thermally induced evolution of organic matter isolated from an algal mat.
620 Organic Geochemistry 6, 865 – 874.
- 621 Disnar, J. R., Le Strat, P., Farjanel, G., & Fikri, A. (1996). Sédimentation de la matière
622 organique dans le nord-est du Bassin de Paris: conséquences sur le dépôt des argilites carbonées
623 du Toarcien inférieur (Organic matter sedimentation in the northeast of the Paris Basin:
624 consequences on the deposition of the lower toarcian black shales). Chemical geology, 131(1-
625 4), 15-35.
- 626 Disnar, J.R., Guillet, B., Keravis, D., Di-Giovanni, C., & Sebag, D. (2003). Soil organic matter (SOM)
627 characterization by Rock-Eval pyrolysis : scope and limitations. Organic Geochemistry 34, 327 – 343.
- 628 Drouin, S., Boussafir, M. Robert, J.L., Albéric P. & Durand, A. (2010). Carboxylic acid sorption on
629 synthetic clays in sea water: In vitro experiments and implications for organo-clay behaviour under
630 marine conditions. Organic Geochemistry 10.1016/j.orggeochem.2009.10.006.
- 631 Duffy, C., Shi, Y., Davis, K., Slingerland, R., Li, L., Sullivan, P. L. & Brantley, S. L. (2014).
632 Designing a suite of models to explore critical zone function. Procedia Earth and Planetary Science,
633 10, 7-15.
- 634 Eglinton, G., Hamilton, R. J. (1967). Leaf epicuticular waxes. Science, 156 (3780), 1322-1335.
- 635 Elie, M., Faure, P., Michels, R., Landais, P., & Griffault, L. (2000). Natural and laboratory oxidation
636 of low-organic-carbon-content sediments: comparison of chemical changes in hydrocarbons. Energy &
637 Fuels, 14 (4), 854-861.

- 638 Elie, M., Mazurek, M. (2008). Biomarker transformations as constraints for the depositional
639 environment and for maximum temperatures during burial of Opalinus Clay and Posidonia Shale in
640 northern Switzerland. *Applied Geochemistry*, 23 (12), 3337-3354.
- 641 Espitalié, J., Deroo, G., & Marquis, F. (1985a). La pyrolyse Rock-Eval et ses applications. *Oil & Gas*
642 *Science and Technology*, 40, 563 – 579.
- 643 Espitalié, J., Deroo, G., & Marquis, F. (1985b). La pyrolyse Rock-Eval et ses applications. *Oil & Gas*
644 *Science and Technology* 40, 755 – 783.
- 645 Garel, S., Schnyder, J., Jacob, J., Dupuis, C., Boussafir, M., Le Milbeau, C., Storme, J.Y., Iakovleva,
646 A.I., Yans, J., Baudin, F., Fléoch, C., & Quesnel, F. (2013). Paleohydrological and paleoenvironmental
647 changes recorded in terrestrial sediments of the Paleocene–Eocene boundary (Normandy, France).
648 *Palaeogeography, Palaeoclimatology, Palaeoecology* 376, 184–199.
- 649 Graz, Y., Di-Giovanni, C., Copard, Y., Laggoun-Défarge, F., Boussafir, M., Lallier-Vergès, E., Baillif,
650 P., Perdereau, L., & Simmoneau, A. (2010). Quantitative palynofacies analysis as a new tool to study
651 transfers of fossil organic matter in recent terrestrial environments. *International Journal of Coal*
652 *Geology* 84, 49 – 62.
- 653 Guo, N., Disdier, Z., Thory, É., Robinet, J. C., & Dagnelie, R. V. (2021). Mobility of organic compounds
654 in a soft clay-rich rock (Tégulines clay, France). *Chemosphere*, 275, 130048.
- 655 Hautevelle, Y., Michels, R., Malartre, F., & Trouiller, A. (2006). Vascular plant biomarkers as proxies
656 for palaeoflora and palaeoclimatic changes at the Dogger/Malm transition of the Paris Basin (France).
657 *Organic Geochemistry*, 37(5), 610-625.
- 658 Hautevelle, Y., Michels, R., Malartre, F., Elie, M., & Trouiller, A. (2007). Tracing of variabilities within
659 a geological barrier by molecular organic geochemistry. Case of the Callovo-Oxfordian sedimentary
660 series in the East of the Paris Basin (France). *Applied Geochemistry* 22, 736 – 759.
- 661 Hedges, J.I, Oades, J.M. (1997). Comparative organic geochemistries of soils and marine sediments.
662 *Organic Geochemistry* 27, 319-361.
- 663 Hendry, M. J., Wassenaar, L. I., & Kotzer, T. (2000). Chloride and chlorine isotopes (^{36}Cl and $\delta^{37}\text{Cl}$)
664 as tracers of solute migration in a thick, clay-rich aquitard system. *Water Resources Research*, 36(1),
665 285-296.

- 666 Huang, W. Y., & Meinschein, W. G. (1979). Sterols as ecological indicators. *Geochimica et*
 667 *cosmochimica acta*, 43(5), 739-745.
- 668 Mahamat, A. M., Boussafir, M., Le Milbeau, C., Guegan, R., Valdes, J., Guinez, M., Sifeddine, A., &
 669 Le Forestier, L. (2016). Organic matter-clay interaction along a seawater column of the Eastern Pacific
 670 upwelling system (Antofagasta bay, Chile): Implications for source rock organic matter preservation.
 671 *Marine Chemistry* doi:10.1016/j.marchem.2016.01.003
- 672 Lafargue, E., Marquis, F., & Pillot, D. (1998). Rock-Eval 6 applications in hydrocarbon exploration,
 673 production and soil contamination studies. *Revue de l'Institut Français du Pétrole* 53, 421 – 437.
- 674 Jacob, J., Disnar, J.-R., Boussafir, M., Sifeddine, A., Albuquerque, A.L.S., & Turcq, B. (2005).
 675 Pentacyclitriterpenemethylethers in recent lacustrine sediments (Lake Caço, Brazil). *Organic*
 676 *Geochemistry* 36, 449–461.
- 677 Jansen, B., & Nierop, K.G.J. (2009). Methyl ketones in high altitude Ecuadorian Andosols confirm
 678 excellent conservation of plant-specific n-alkane patterns. *Organic Geochemistry* 40, 61–69.
- 679 Landais, P., & Elie, M. (1999). Utilisation de la géochimie organique pour la détermination du
 680 paléoenvironnement et de la paléothermicité dans le Callovo-Oxfordien du site de l'Est de la France.
 681 *EDP Sciences*, 1999, 35-58.
- 682 Landais, P., Michels, R., Kister, J., Dereppe, J. M., & Benkhedda, Z. (1991). Behavior of oxidized
 683 type II kerogen during artificial maturation. *Energy & fuels*, 5(6), 860-866.
- 684 Lerouge, C., Debure, M., Henry, B., Fernandez, A. M., Blessing, M., Proust, E., & Robinet, J. C.
 685 (2020). Origin of dissolved gas (CO₂, O₂, N₂, alkanes) in pore waters of a clay formation in the
 686 critical zone (Tégulines Clay, France). *Applied Geochemistry*, 116, 104573.
- 687 Lerouge, C., Robinet, J.C., Debure, M., Tournassat, C., Bouchet, A., Maria Fernandez, A., Flehoc, C.,
 688 Guerrot, C., Kars, M., Lagroix, F., Landrein, P., Madé, B., Negrel, P., Wille, G., & Claret, F. (2018). A
 689 deep alteration and oxidation profile in a shallow clay aquitard : example of the Tégulines Clay, East
 690 Paris basin, France. *Geofluids* <https://doi.org/10.1155/2018/1606753>.

- 692 Marchand, C., Lallier-Vergès, E., Disnar, J. R., & Kéravis, D. (2008). Organic carbon sources and
693 transformations in mangrove sediments: a Rock-Eval pyrolysis approach. *Organic Geochemistry*,
694 39(4), 408-421.
- 695 Nagra. 2002. Project Opalinus Clay. Safety Report. Nagra Technical Report NTB 02-05. Nagra,
696 Wettin-gen, Switzerland.
- 697 Nelson, P. N., Baldock, A., Oades, J. M., Churchman, G. J., & Clarke, P. (1999). Dispersed clay and
698 organic matter in soil: their nature and associations. *Soil Research*, 37(2), 289-316.
- 699 Sebag, D., Di-Giovanni, C., Ogier, S., Mesnage, V., Laggoun-Défarge, F., & Durand, A (2006a).
700 Inventory of sedimentary organic matter in modern wetland (Marais Vernier, Normandy, France) as
701 source –indicative tools to study Holocene alluvial deposits (Lower Seine Valley, France). *International*
702 *Journal of Coal Geology* 67, 1 – 16.
- 703 Sebag, D., Copard, Y., Di-Giovanni, C., Durant, A., Laignel, B., Ogier, S., & Lallier-Verges, E. (2006b).
704 Palynofacies as useful tool to study origins and transfers of particulate organic matter in recent terrestrial
705 environments : synopsis and prospect. *Earth Science Reviews* 79, 241 – 259.
- 706 Todorov, I., Schegg, R. A., & Wildi, W. (1993). Thermal maturity and modelling of Mesozoic
707 and Cenozoic sediments in the south of the Rhine Graben and the Eastern Jura (Switzerland).
708 *Eclogae Geologicae Helvetiae*, 86(3), 667-692.
- 709 Tyson, R. V. (1993). Palynofacies analysis. In *Applied micropalaeontology* (pp. 153-191). Springer,
710 Dordrecht.
- 711 Tyson, R.V. (1995). *Sedimentary organic matter: organic facies and palynofacies*. Chapman and Hall,
712 London. 615 pp.
- 713 Vernal, A., Larouche, A., & Richard, J.H. (1987). Evaluation of palynomorph concentrations: do the
714 aliquot and the marker-grain methods yield comparable results? *Pollen and spores* XXIX, 291 – 304.
- 715 Volkman, J. K. (1986). A review of sterol markers for marine and terrigenous organic matter. *Organic*
716 *geochemistry*, 9(2), 83-99.
- 717

Xie, S., Nott, C., Avsejs, L., Maddy, D., Chambers, F., & Evershed, R. (2004). Molecular and isotopic stratigraphy in an ombrotrophic mire for paleoclimate reconstruction. *Geochimica et Cosmochimica Acta* 68, 2849–2862.

Zhang, M., Ji, L., Wu, Y., & He, C. (2015). Palynofacies and geochemical analysis of the Triassic Yanchang Formation, Ordos Basin: Implications for hydrocarbon generation potential and the paleoenvironment of continental source rocks. *International Journal of Coal Geology*, 152, 159-176.

Zheng, Y., Zhou, W., Meyers, P.A., & Xie, S. (2007). Lipid biomarkers in the Zoigê-Hongyuan peat deposit: indicators of Holocene climate changes in West China. *Organic Geochemistry* 38, 1927– 1940.

Zhou, W., Zheng, Y., Meyers, P.A., Jull, A.J.T., & Xie, S. (2010). Postglacial climate-change record in biomarker lipid compositions of the Hani peat sequence, Northeastern China. *Earth and Planetary Science Letters* 294, 37–46.

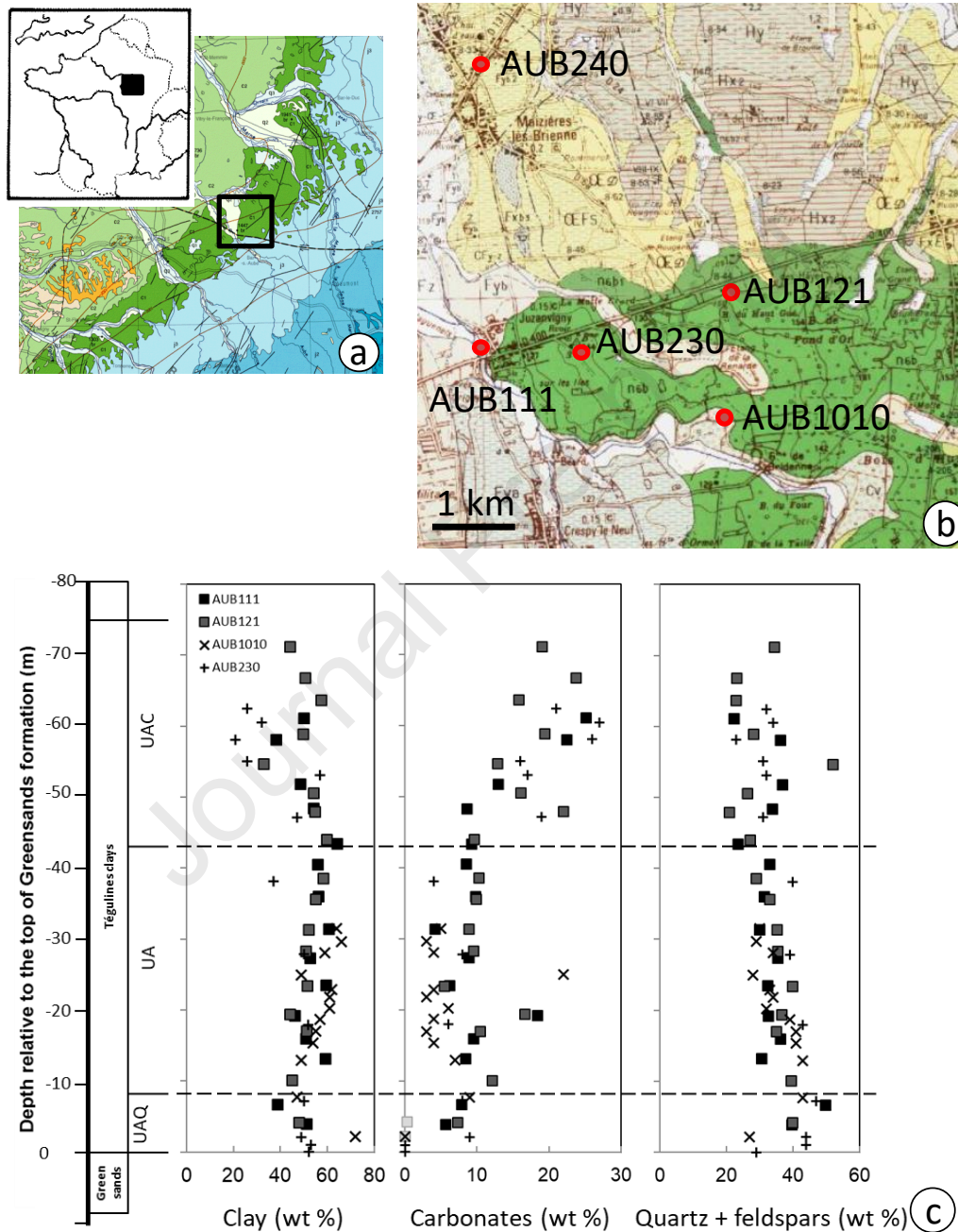
1 Table 1 *Characteristics of the different particles encountered in the different samples*

Classes	Categories	Characteristics
Amorphous particles	GrAOM: granular Amorphous Organic Matter	Diffuse limits, grey to yellow in transmitted light, opaque or slight orange in reflected light, size: 10-100 μm
	gAOM: Gelified Amorphous Organic Matter	Particle entirely gelified, no internal structure, orange to red in transmitted light, dark orange in reflected light, size: can reach very large size over 100 μm
	Gd: Gelified debris	Sharp contours, high reflectance, size: 10 μm
	Od: Opaque debris	Sharp contours, high reflectance, dark in transmitted light, size: 10 μm
Ligno cellulosic debris	Plc: Preserved lignocellulosic	Elongated particles with visible internal structures, yellow in transmitted light (10-100 μm)
	Salc: Slightly amorphised lignocellulosic	Elongated particles with visible internal structures, yellow to orange in transmitted light (10-100 μm), trace of gelification
	Glc: Gelified lignocellulosic	Elongated particles with visible internal structures, yellow to orange in transmitted light (10-100 μm), higher state of gelification
	Olc: Opaque lignocellulosic	No internal structure visible, elongated, show high reflectance (10-100 μm)
Other organic particles	Div: divers	Mycelium fragments: elongated particles, brown color; Cuticular fragments: particles with homogeneous limits, brown, with typical botanic structures, size variable and can exceed several hundred μm ; Spores and pollens: particles from the terrestrial and aquatic flora, translucent or yellow for fresh particles and brown-orange for preserved particles; acritarchs, foraminifera
	pyrite	framboidal, opaque, with geometrical forms under transmitted light

Table 2 Distributions, abundances and ratios of *n*-alkanes in the Trench TPH1-1 (0 – 30 cm sample) and AUB230 borehole.
^aMajors *n*-alkanes in core samples, ^bTotal = total abundance of *n*-alkanes (µg/g sediment), ^c odd/even ratio (odd/even = $\sum \text{conc. odd alkanes} / \sum \text{conc. even } n\text{-alkanes}$) ^d*n*-alkanes carbon preference index CPI = $2(\text{odd } n\text{-C23 to } n\text{-C31}) / (\text{even } n\text{-C22 to } n\text{-C30} + \text{even } n\text{-C24 to } n\text{-C32})$.

	Major alkanes ^a	Total of alkanes (µg/g sed) ^b	odd/even ^c	CPI ^d
TPH1-1 0-30cm	n-C29, n-C31, n-C33	5.08	2.97	4.95
AUB230 9.36m	n-C29, n-C31, n-C33	9.14	1.26	1.08
AUB230 21.30m	n-C29, n-C31, n-C33	0.95	1.06	0.97
AUB230 25.32m	n-C29, n-C29, n-C31	0.61	1.37	1.19
AUB230 45.84m	n-C29, n-C31, n-C33	0.56	1.34	1.15
AUB230 66.35m	n-C29, n-C31, n-C33	0.47	0.97	1.57
AUB230 67.45m	n-C25, n-C31, n-C31	0.29	1.27	1.74

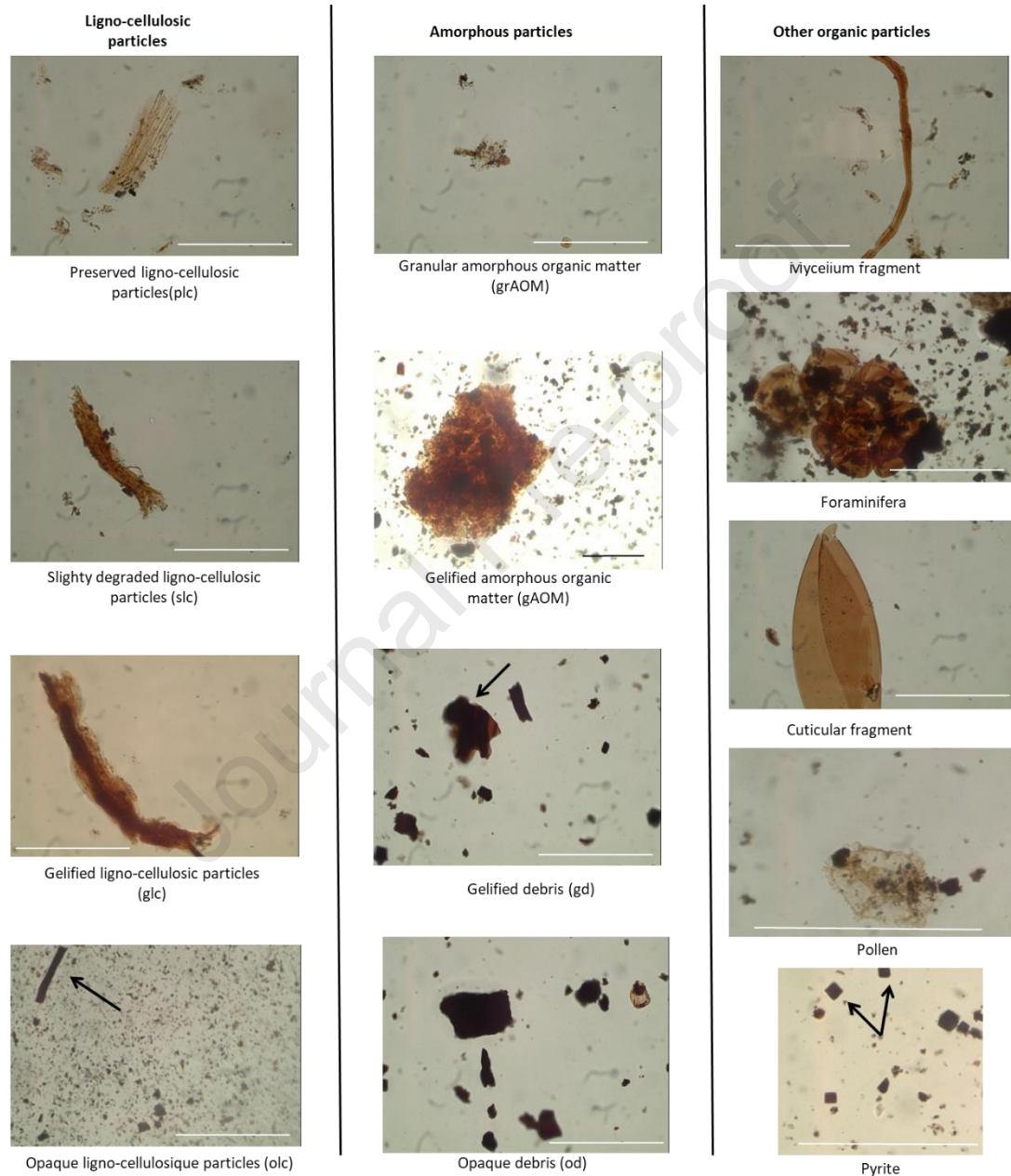
Fig. 1 (a) Geological map of the east part of the Paris Basin showing the studied area; (b) Geological map of the studied area showing the location of the AUB240, AUB230 and AUB1010 boreholes selected for the organic matter study; (c) Major mineralogy of the Tégulines clay from the AUB230 and AUB1010 boreholes deduced from XRD data treatment, compared with previous data of AUB111 and AUB121 boreholes (Lerouge et al., 2018); UA: clay rich unit; UAC: carbonate-clay rich unit; UAQ: clay-quartz rich unit



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13 Fig. 2 The different categories of particulate organic matter distinguished from their morphology, the scale of the white bars
 14 corresponds to 100 μm ; **Ligno-cellulosic fragments:** - preserved ligno-cellulosic fragments (plc),- slightly degraded ligno-
 15 cellulosic fragments (slc), - gelified ligno-cellulosic fragments (glc),- opaque ligno-cellulosic fragments (olc); **amorphous**
 16 **particles:** - granular amorphous organic matter (grAOM),- gelified amorphous organic matter (gAOM),- gelified debris
 17 (gd),- opaque debris (od); **Others organic particles:** - mycelium fragments (Myc); - foraminifera, - cuticular fragments
 18 (Cut), - Spores and pollens (sp), - (c) Pyrite.

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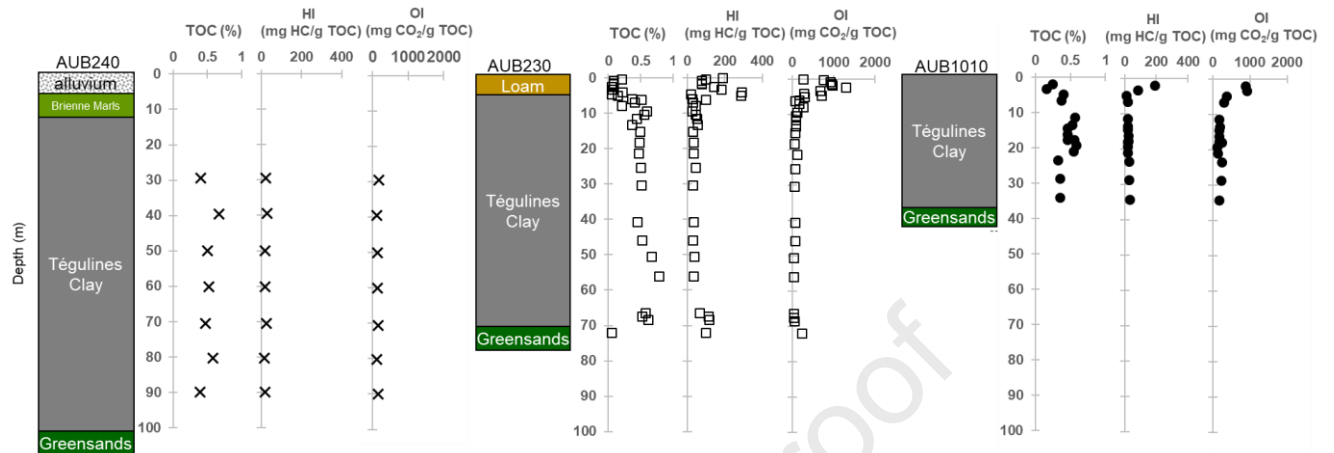
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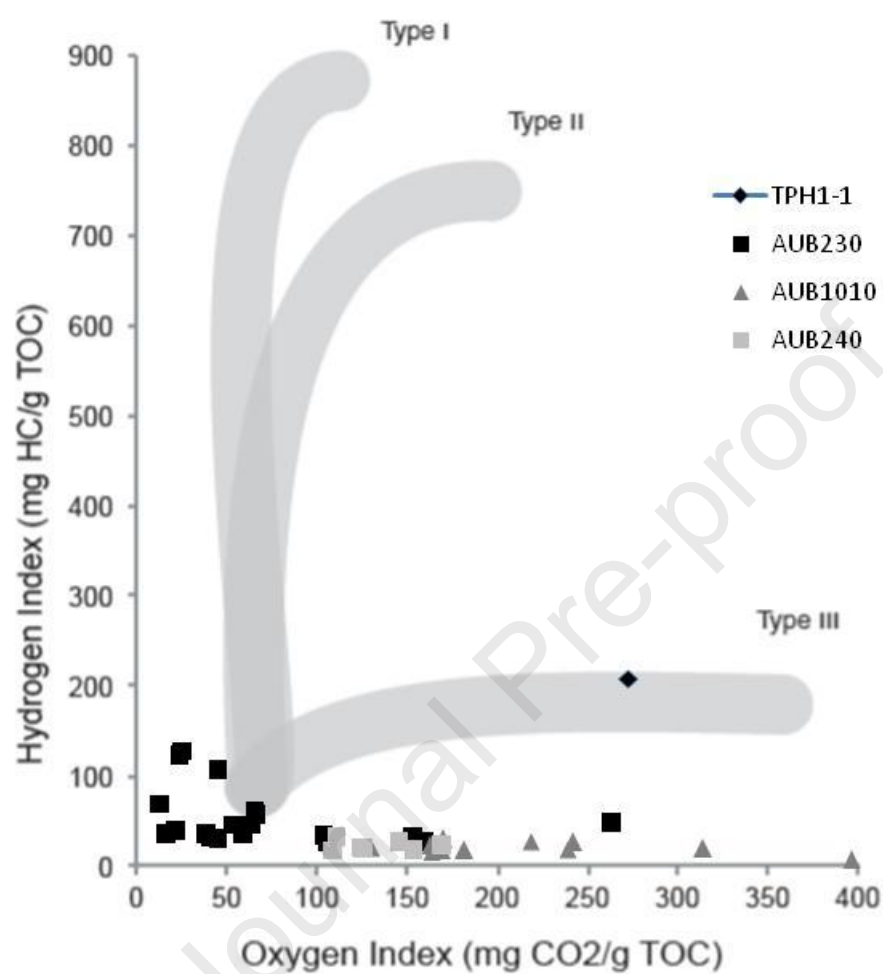
Fig. 3 Total Organic Carbon (TOC, %), Hydrogen Index (HI, $mg\ HC.g^{-1}TOC$), and Oxygen Index (OI, $mg\ CO_2.g^{-1}TOC$) in relation with depth for the AUB240, AUB230 and AUB1010 boreholes.



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50 Fig. 4 Pseudo van-Krevelen diagram type for the different samples.

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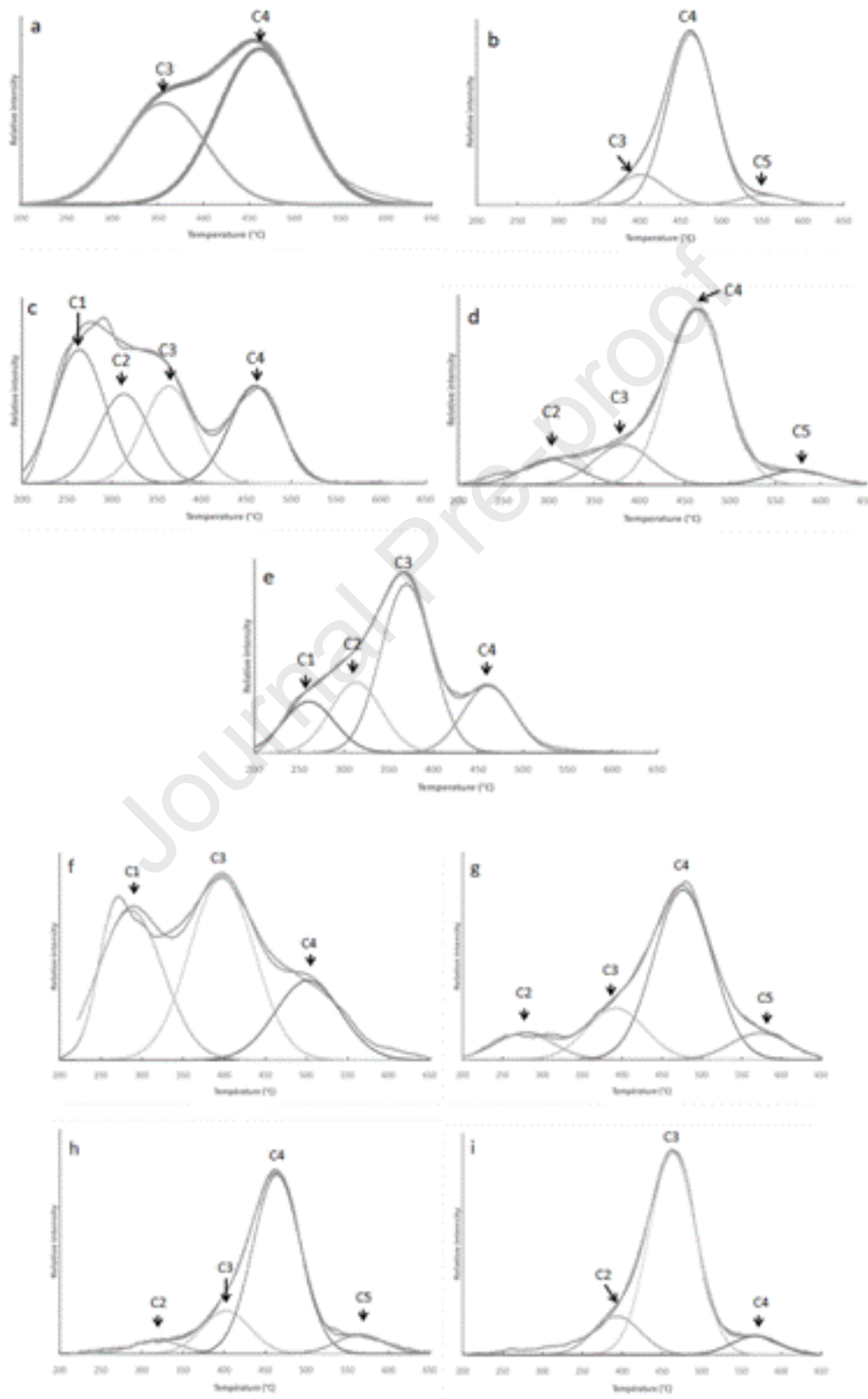
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64 Fig. 5 Deconvolution of multilobed S2 signals from RE pyrolysis for *a* TPH1-1 030 ($r^2 = 0.996$), *b* AUB240 39 ($r^2 = 0.999$), *c*
 65 AUB230 6.10 ($r^2 = 0.994$), *d* AUB230 25.32 ($r^2 = 0.997$), *e* AUB230 67.45 ($r^2 = 0.998$), *f* AUB1010 1.88 ($r^2 = 0.974$), *g* AUB1010
 66 3.31 ($r^2 = 0.996$), *h* AUB1010 14.47 ($r^2 = 0.998$), *i* AUB1010 20.95 ($r^2 = 0.995$). Black-grey curves are the rebuilt S2 signal from
 67 deconvolution.
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Fig. 6 Contribution of the different clusters (C1-C5, %) AUB240, AUB230 and AUB1010 samples

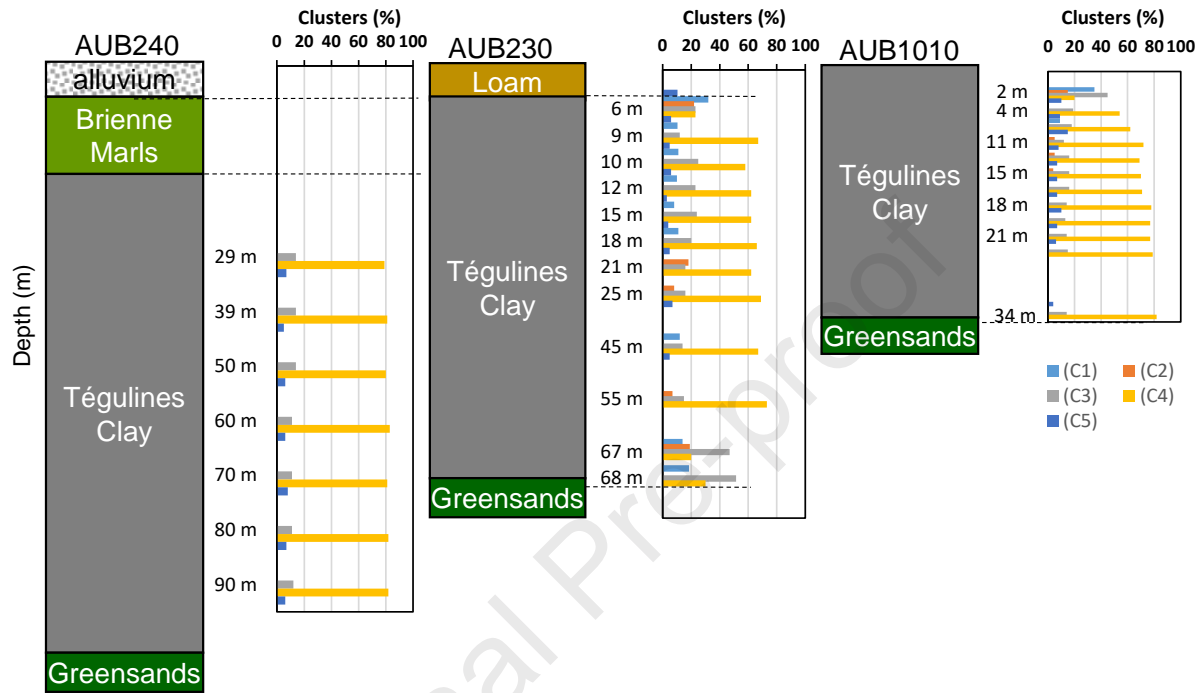


Fig. 7 Distribution of the different particles in: (a) TPH1-1 0-30 sample; (b) AUB240 borehole (29,39,70 and 80 m deep); (c) AUB230 borehole (6.10, 18.22, 25.32, 45.84, 67.45 m and comparison with the TPH1-1 0-30)

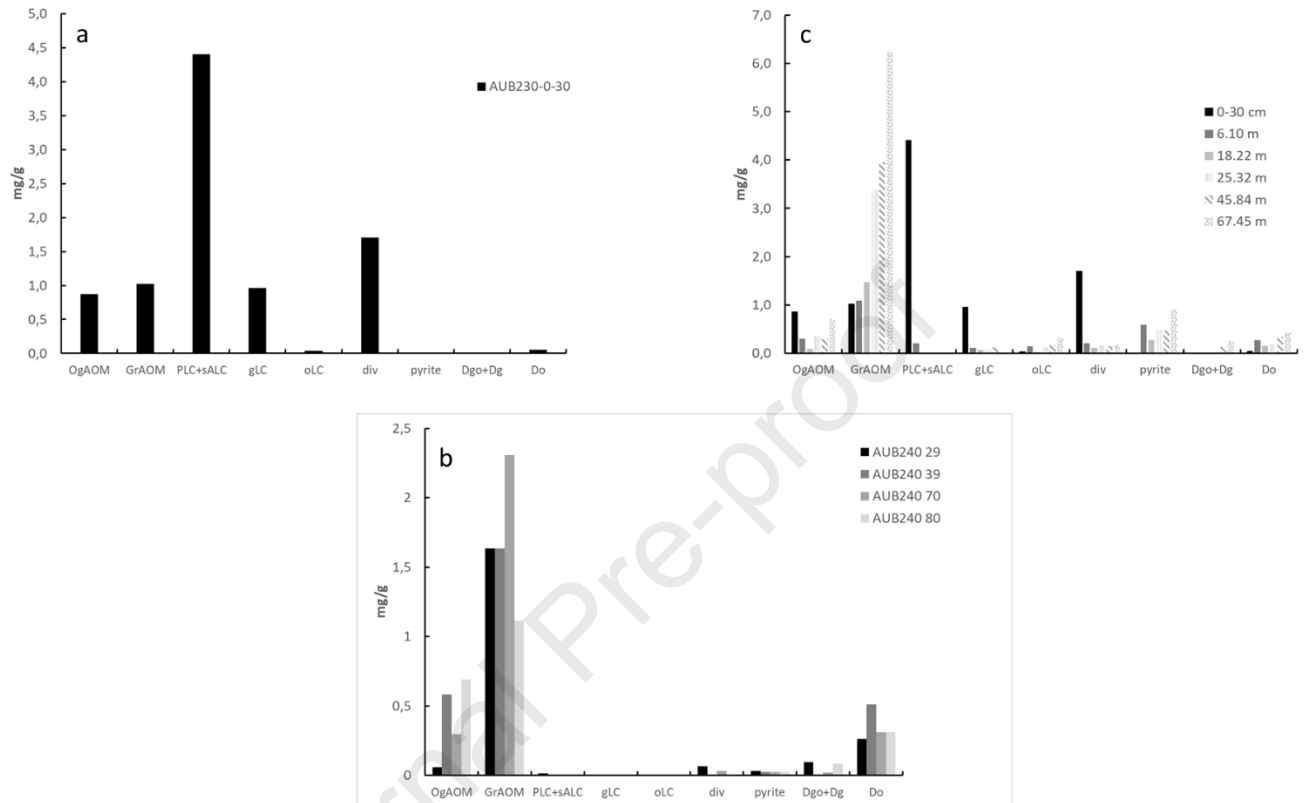


Fig. 8 Chromatograms of the Tégulines Clay formation at 67 m; 25 m and 6.10 m

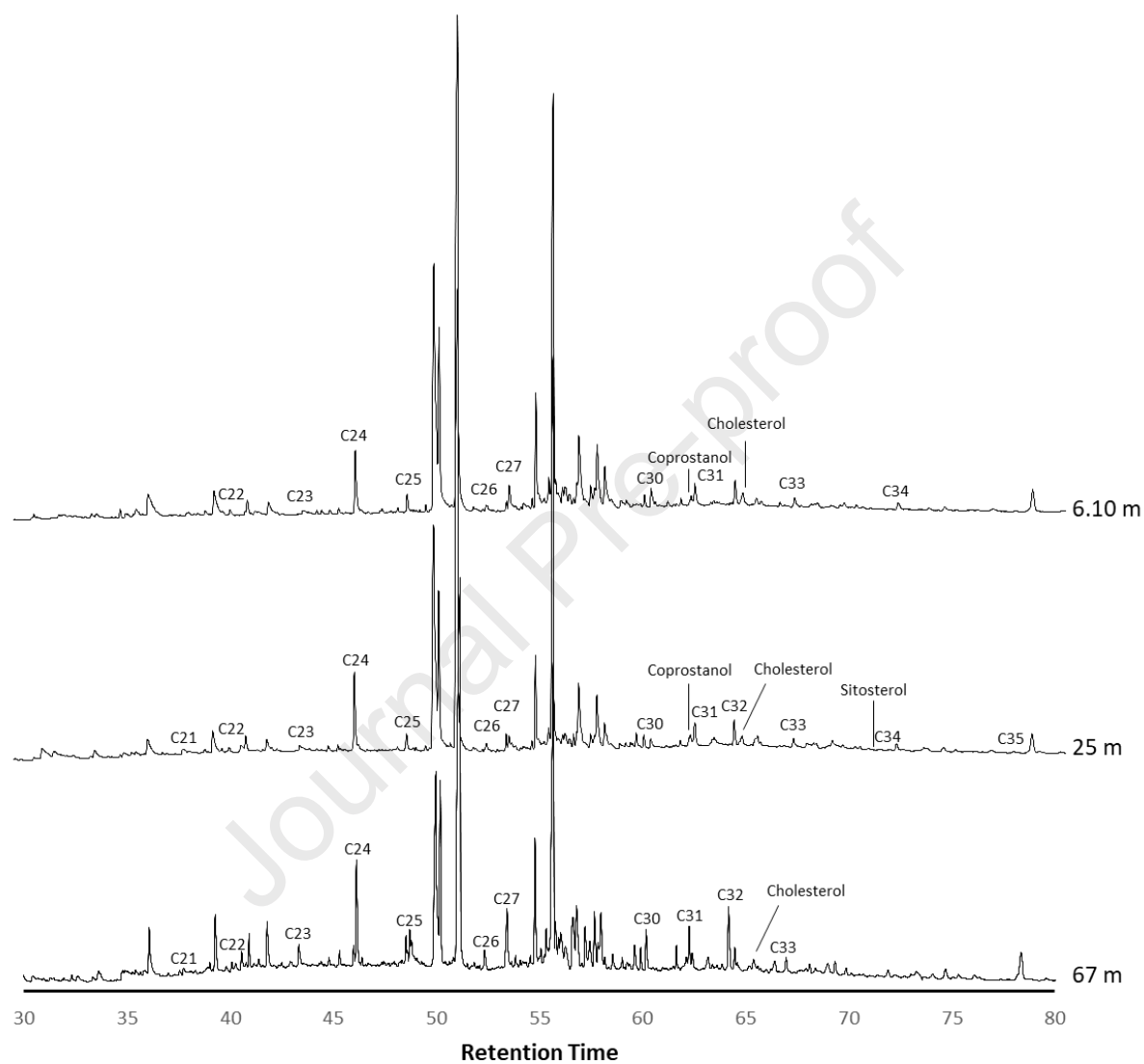
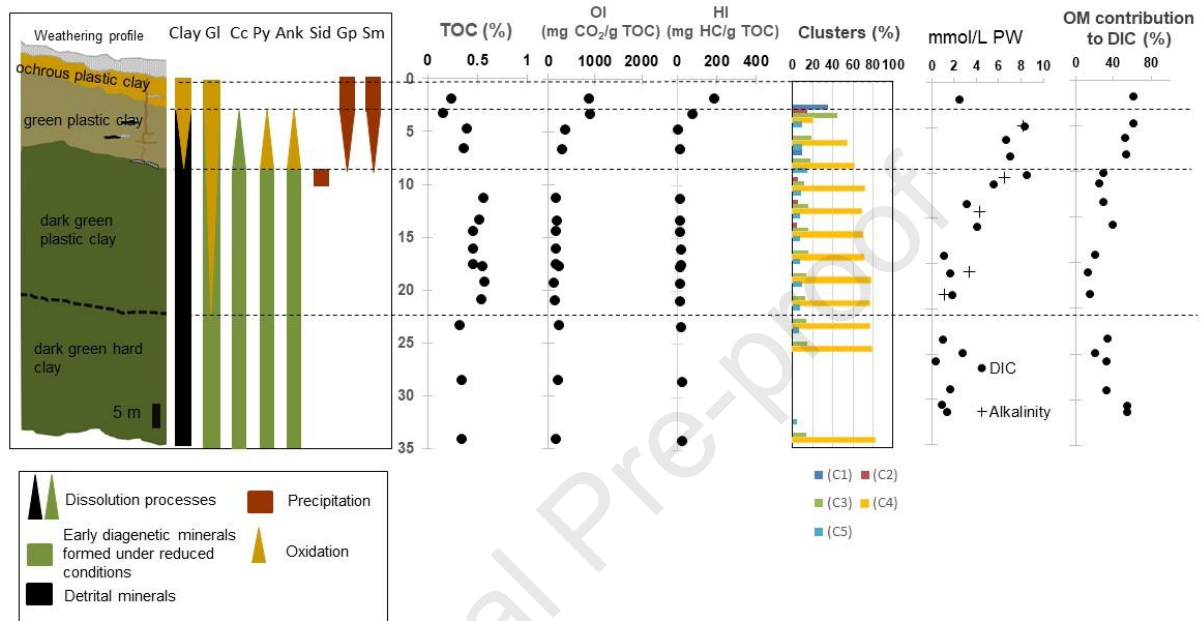


Fig. 9 Main characteristics of the organic matter through the weathering profile developed in Tégulines clay, compared with mineral changes (Lerouge et al., 2018), dissolved inorganic carbon concentration (DIC) in pore waters (PW) and contribution of degradation of organic matter (OM) to the DIC (Lerouge et al., 2020).



Journal Pre-proof

- OM characterization of the Tégulines Clay formation using bulk and molecular data
- Vertical variation of the oxidation profile as a result to the vicinity to the surface
- First upper 10 m of the clays subjected to oxidative events, below 10 m OM preserved
- Spatial variations of the oxidation state due to different topography of the boreholes
- OM bulk and molecular techniques are a good association for critical zone studies

Conflict of Interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.