















# Using the 3D MOCAGE CTM to simulate the chemistry of halogens in the volcanic plume of Etna's eruption in December 2018 at the regional scale

Herizo Narivelo<sup>1</sup>, Virginie Marécal<sup>1</sup>, Paul David Hamer<sup>2</sup>, Luke Surl<sup>3,4</sup>, Tjarda Roberts<sup>3,4</sup>, Claire Lamotte<sup>1</sup>, Mickaël Bacles<sup>1</sup>, Simon Warnach<sup>5</sup>, and Thomas Wagner<sup>5</sup>



<sup>&</sup>lt;sup>1</sup>Centre National de Recherches Météorologiques (CNRM), Toulouse, France

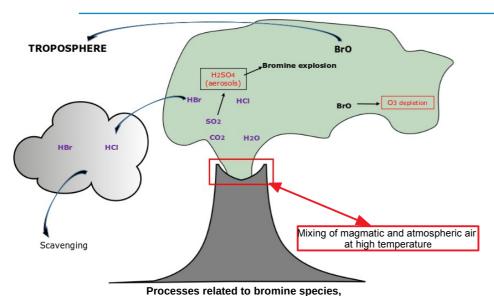
<sup>&</sup>lt;sup>2</sup> Norsk institutt for luftforskning (NILU), Kjeller, Norway

<sup>&</sup>lt;sup>3</sup>Laboratoire de Physique et Chimie de l'Environnement et de l'Espace (LPC2E), Orléans, France

<sup>&</sup>lt;sup>4</sup>Laboratoire Atmosphères, Milieux, Observations Spatiales (LATMOS), Paris, France

<sup>&</sup>lt;sup>5</sup> Max-Planck-Institut für Chemie (MPI-C), Mainz, Germany

## Halogen chemistry in volcanic plumes



occurring in the volcanic plume

- (R1)  $HBr + OH \longrightarrow Br + H_2O$
- (R2) Br + O<sub>3</sub>  $\longrightarrow$  BrO + O<sub>2</sub>
- $(R3) \qquad BrO + HO_2 \longrightarrow HOBr + O_2$
- (R4) BrO + NO<sub>2</sub>  $\longrightarrow$  BrONO<sub>2</sub>
- (R5)  $HOBr + HBr (sulfate aerosols) \longrightarrow Br_2 + H_2O$
- (R6)  $HOBr + HCl (sulfate aerosols) \longrightarrow BrCl + H_2O$
- (R7) Brono<sub>2</sub> + HBr (sulfate aerosols)  $\longrightarrow$  Br<sub>2</sub> + HNO<sub>3</sub>
- (R8)  $BrONO_2 + HCl (sulfate aerosols) \longrightarrow BrCl + HNO_3$
- (R9)  $Br_2 + h\nu \longrightarrow 2Br$
- (R10)  $BrCl + h\nu \longrightarrow Br + Cl$
- (R11)  $Br + O_3 \longrightarrow BrO$
- (R12) Br + HCHO  $\longrightarrow$  HBr + HCO
- $(R13) \quad Br + HO_2 \longrightarrow HBr + O_2$
- (R14) BrO + HO<sub>2</sub> + HBr (sulfate aerosols) + 2 O<sub>3</sub>  $\longrightarrow$  2 BrO + 3 O<sub>2</sub> + H<sub>2</sub>O
- $(R15) \quad BrO + NO_2 + HBr \left( sulfate aerosols \right) + 2\,O_3 \, \longrightarrow 2\,BrO + 3\,O_2 + HNO_3$

**Bromine-explosion cycle** 

- Volcanoes emit large amounts of gases in particular SO<sub>2</sub> and halogens (HCl and HBr)
- Rapid mixing of magmatic and atmospheric gases at temperatures ~1000°C:
  - ⇒ Formation of primary sulfate aerosol and new bromine species from HBr
- Plume measurements and modeling work show :
  - ⇒ HBr rapidly converts into BrO in plumes resulting in depletion of ozone (O3)

This is the process of bromine-explosion:

⇒ Significant and rapid production of BrO leading to a destruction of O<sub>3</sub>





# State of the art on the modeling of halogens chemistry in volcanic plumes

- Halogen chemistry in the volcanic plume was mainly modelled/studied at the local scale close to the volcanoes and over short time scales (one- two hours)
- Only one modelling study at the regional scale : Jourdain et al. (ACP 2016) on the Ambrym volcano emissions (Vanuatu) :
  - Ambrym volcano (southwest Pacific) located in the tropics where the troposphere is characterized by low ozone levels
  - Main results of this modeling study ⇒ the effects of the halogen emissions on the tropospheric chemistry is found even far from the source
    - need for more regional studies on other volcanoes in different environments





#### **General objectives and strategy**

#### **Objectives:**

- To test the capability of the regional Chemistry-transport model MOCAGE to simulate the « bromine-explosion » and associated ozone depletion with a ~ 20km x ~20km horizontal resolution on a case study: Mt Etna emissions around Christmas 2018
- To analyse the impact of halogen species on air composition from this event at the regional scale

#### Strategy:

- Part a : Evaluate the results of MOCAGE against the WRF-Chem simulations at finer resolution using idealised emissions
- Part b : Simulation of the Christmas 2018 event in the most realistic way ; i.e. with varying emission fluxes and altitude, and evaluation against TROPOMI satellite observations





## Case study: Mt Etna 'Christmas' 2018 eruptions

2 small eruptions on the 24th of December

Strong passive degassing over a few days after

- BrO signal captured by TROPOMI satellite measurements even far from the volcano
  - Signature of bromine explosion



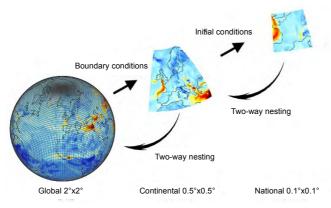
Eruption of Mt Etna (Sicily, Italy) on 24th of December 2018 Source: Corradini et al, (2020), Remote Sens. 2020, 12(8), 1336; https://doi.org/10.3390/rs12081336

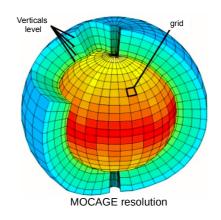




#### **Description of MOCAGE CTM model and general configuration**

- MOCAGE : Chemistry-Transport Model developed at CNRM (Météo-France/CNRS)
- Describes the evolution of the composition of the troposphere and stratosphere for gas and aerosol species
- Possibility to make nested domains :





- Resolution: 2° (Global), 0.2° (Mediterranean region)
- Vertical grid uses a hybrid coordinates : 47 verticals level with a top at 5 hPa
- Using MOCAGE CTM model in 'chemistry mode' (no explicit representation of aerosols)
- Still, it considers the sulfate aerosol surface for heterogeneous reactions by:
  - Calculating the oxidation of SO<sub>2</sub> in aqueous phase  $\Rightarrow$  SO<sub>4</sub><sup>2</sup>·
  - Then, SO<sub>4</sub><sup>2-</sup> is assumed to be entirely converted in the form of sulfate aerosols.
  - Sulfate aerosols are assumed to have a uniform distribution of radius 1μm
     (consistent with aerosol measurements in volcanic plumes)

#### **Emissions for the model simulations:**

- Numerical experiments for Part a and Part b :
  - Duration of simulations: 2018/12/24/ 00:00 UTC to 2018/12/12/26 00:00 UTC
  - Simulation without volcanic emission to serve as a reference: NOVOLC
  - Simulation with volcanic emission: VOLC
- Ratios with respect to SO<sub>2</sub> volcanic emissions used in the VOLC simulations :

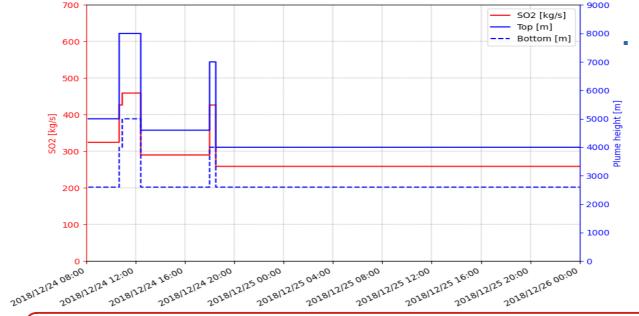
Flux ratio	Value	Note
Molar ratios HCI/SO <sub>2</sub>	0.44	Average ratios at Bnand NEC craters from Wittmer et al. (2014)
Brtot/SO <sub>2</sub> HBr/Brtot Br/Brtot	6.6E-04 0.75 0.25	To take into account the effects of high temperature processes at vent
Mass ratio Aerosol/SO <sub>2</sub>	0.03	Roberts et al. (2018) – assumed all sulfate





#### Specific model setup for part a and part b:

- Part a: Idealized volcanic source of SO<sub>2</sub> and halogens with constant emissions from 24<sup>th</sup> of Dec.at 00UTC to 26<sup>th</sup> of Dec.at 00UTC
  - 320 kg SO<sub>2</sub>/s flux rate beginning at 11h15 on December 24
  - 4km injection height into single model layer (~300m thick)
  - Other species emitted at fixed ratio to SO<sub>2</sub> (as given in the previous slide)
- Part b: Simulation with varying emission fluxes and altitude as used in Lamotte et al. EGU presentation based on Corradini et al. (2020)
  - Other species emitted at fixed ratio to SO<sub>2</sub> (as given in the previous slide)



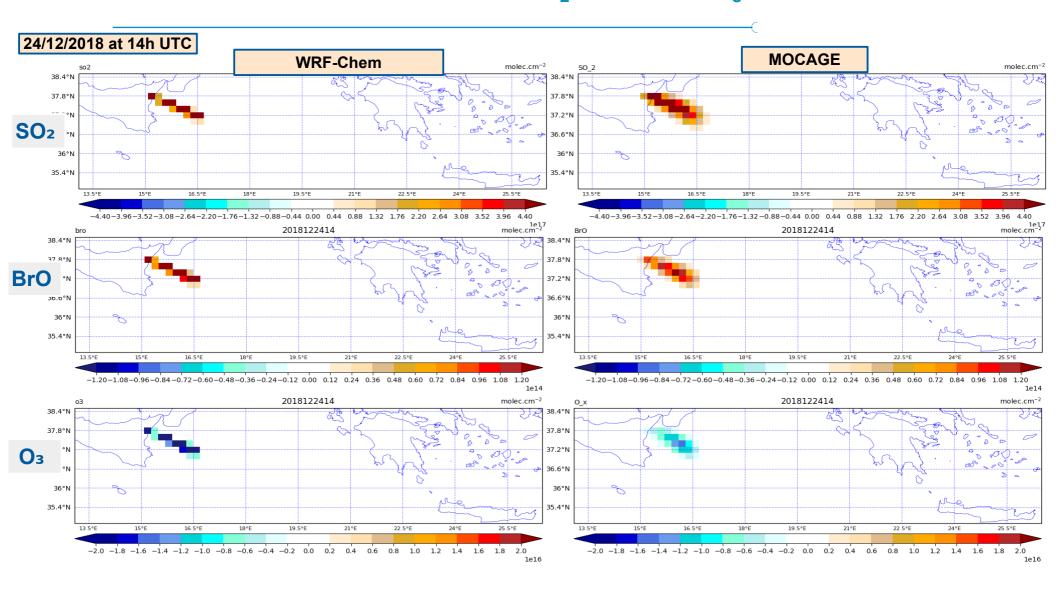
Etna Christmas 2018 emissions with two small eruptions on the 24th of December and strong passive degasing over a few days after



METEO FRANCE

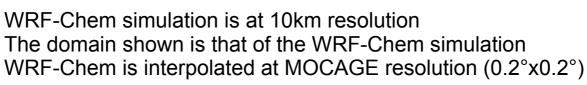
All figures shown in the next slides are the difference of the tropospheric column between VOLC and NOVOLC [molec/cm²]

# Part a: Results of the WRF-Chem-MOCAGE comparison Tropospheric columns of SO<sub>2</sub>, BrO and O<sub>3</sub> [molec/cm2]



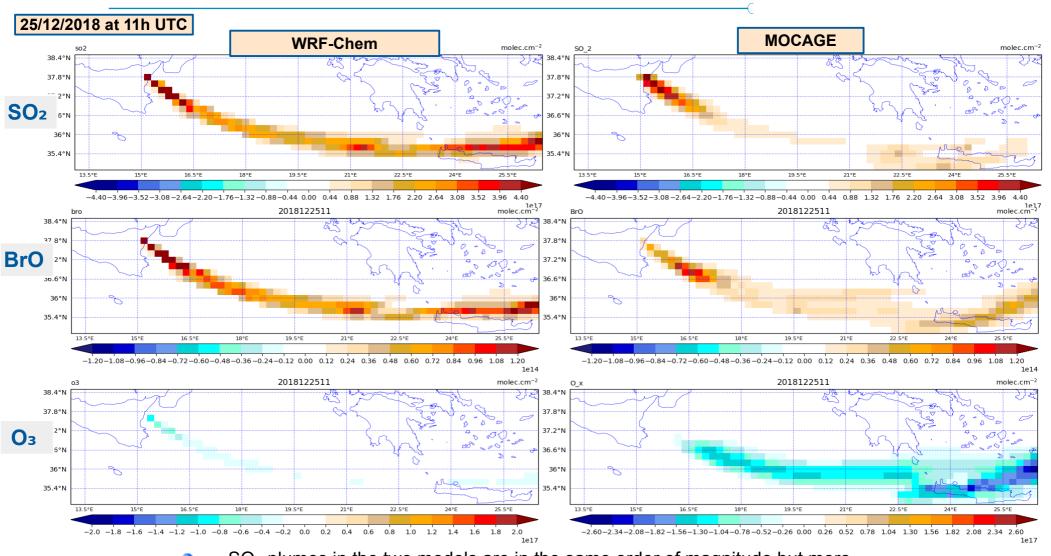








# Part a: Results of the WRF-Chem-MOCAGE comparison Tropospheric columns of SO<sub>2</sub>, BrO and O<sub>3</sub> [molec/cm2]



SO<sub>2</sub> plumes in the two models are in the same order of magnitude but more spread in MOCAGE

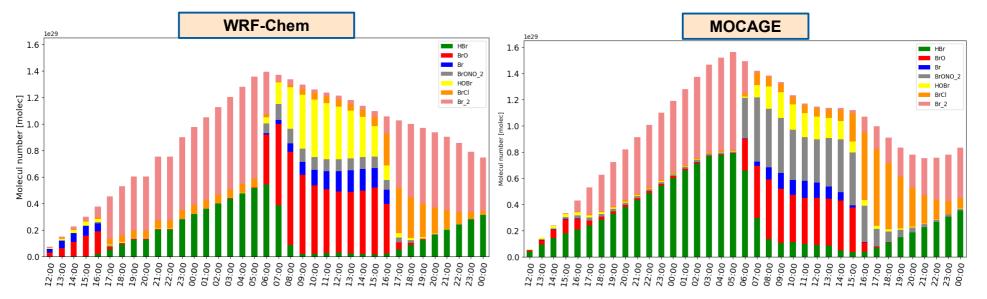


- Bromine explosion cycle appears in the two models.
  - BrO is more quickly formed in WRF-Chem (closer to the volcano)
    - Ozone depletion along the plume found in the two models but later and much stronger in MOCAGE



### Partition of bromine compounds from WRF-Chem and MOCAGE

Comparison of the total molecule number of the bromine compounds between WRF-Chem simulation with 10km horizontal resolution and MOCAGE (~20km resolution)



- During the daytime on the 24<sup>th</sup>
  - BrO starts to form and rapidly increases in WRF-Chem
  - Bromine-explosion cycle is less efficient in MOCAGE
- During nightime (24th to 25th)
  - BrCl and Br2 become reservoirs of bromine species while HBr is still emitted
- The day after (25th), good consistency between MOCAGE and WRF-Chem if taking into account the fact that MOCAGE has higher NO<sub>2</sub> in background air leading to more BrONO<sub>2</sub> and less HOBr
- During nighttime (25th-26th), the ratio between Br<sub>2</sub> and BrCl is a bit different in the two models





### What you learn from part a?

#### Overall, the behaviour of the 2 models is fairly similar

The comparison of the tropospheric columns from MOCAGE and WRF-Chem have the same order of magnitude and similar location

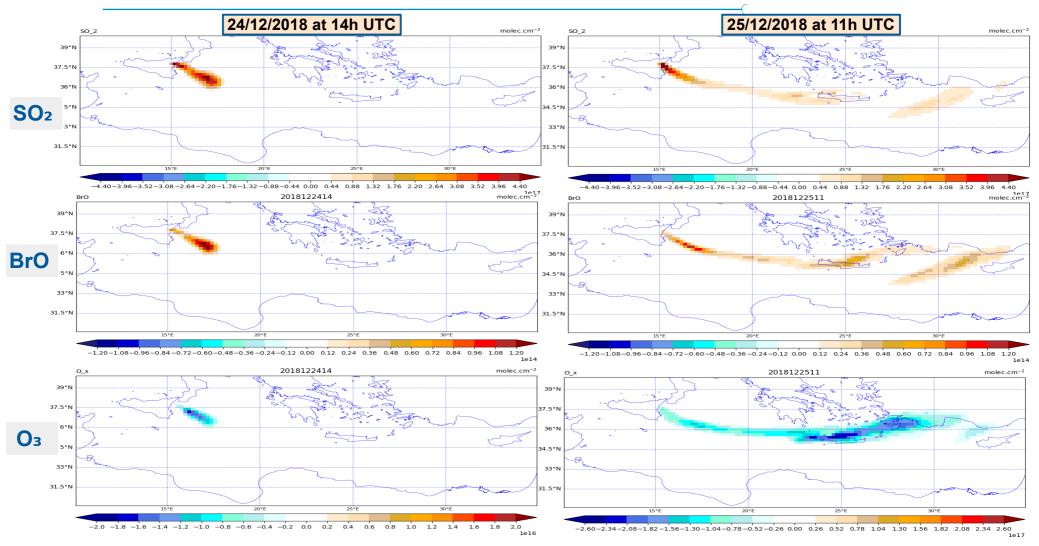
#### But with some differences

- BrO starts to form and increases rapidly in WRF-Chem while MOCAGE forms BrO later
- Ozone depletion is stronger in MOGAGE than in WRF-Chem
- Part of the differences comes from the (initial) background air composition (NOx and ozone) which is different in WRF-Chem and MOCAGE and possibly to the fact that WRF-Chem used OH emissions which are not yet included in MOCAGE simulations
- Differences on bromine and ozone can also be related to differences between WFR-Chem and MOCAGE chemical schemes, in particular some chemical reactions not yet present in MOCAGE (for example, the hydrolysis of BrONO<sub>2</sub>)





# Part b: Results from MOCAGE simulation Tropospheric columns of SO<sub>2</sub>,BrO and O<sub>3</sub> during daytime [molec/cm<sup>2</sup>]

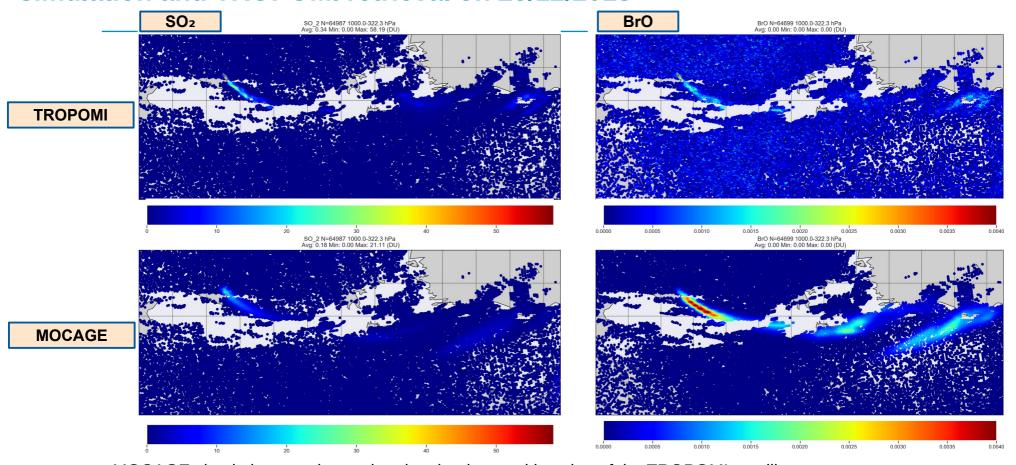


- SO<sub>2</sub> emission and transport of the plume
- Bromine-explosion cycle during daytime
  - Depletion of ozone along the plume → bromine explosion
    - Results similar to those obtained in Part a, except a slightly different location of the plume





# Part b: Comparison of SO<sub>2</sub> and BrO tropospheric columns between MOCAGE simulation and TROPOMI retrieval on 25/12/2018



- MOCAGE simulations are interpolated at the time and location of the TROPOMI satellite measurements
- The comparison shows
  - Overall good agreement on the plume location between MOCAGE and TROPOMI but with a more spread plume in MOCAGE
  - Similar order of magnitude for SO<sub>2</sub> and BrO between MOCAGE and TROPOMI
  - More rapid formation of BrO "Bromine-explosion cycle" in TROPOMI observations than in MOCAGE, consistent with results of the comparison with WRF-Chem (Part a). In MOCAGE, BrO formation is slower but high BrO persists further downwind



### What you learn from Part b?

- The concentrations of the different halogens species are in the same order of magnitude as found in Part a
- The volcanic plumes of SO<sub>2</sub> and BrO is better located using the variation with time of the quantity of species emitted and height of the plume
- TROPOMI vs MOCAGE 3D model :
  - ⇒ because MOCAGE resolution is coarser than TROPOMI pixels and because of model diffusion, the volcanic plume is more spread out in MOCAGE (as expected)
  - ⇒ Tropospheric column of BrO (TROPOMI) is lower than the BrO from MOCAGE 3D model
    - differences may be related to uncertainties on SO<sub>2</sub>, HBr emissions and on assumptions on changes of emissions at high temperature
    - uncertainty/missing reactions in MOCAGE halogen chemistry





### **Conclusion & perspectives**

#### Conclusion

- These preliminary results show that MOCAGE CTM (~20km x ~20km of horizontal resolution) is able to represent the bromine explosion
- BrO formation is associated with tropospheric ozone depletion along the volcanic plume but ozone depletion is too strong in MOCAGE
- Farily good consistency of MOCAGE results with WRF-Chem results
- Fairly good consistency of MOCAGE results with TROPOMI SO2 and BrO observations
- However, MOCAGE simulations still need to be improved to better match with WRF-Chem and TROPOMI

#### Perspectives

- Complete the chemical halogen scheme to be more consistent with WRF-Chem model
- Use of the MOCAGE version including the secondary aerosol scheme with explicit formation of sulfate aerosols
- Sensitivity studies to improve even more the location of the plume by testing slight modifications of the emission height (Part b)
- Emissions used as input in MOCAGE can be improved (Part b), in particular :
  - Sensitivity studies to adjust SO<sub>2</sub> emissions in mass to improve plume SO<sub>2</sub> column with respect to TROPOMI
  - Sensitivity studies to adjust HBr emissions to improve plume BrO column with respect to TROPOMI
  - Emissions from high temperature processes at vent:
    - Use of additional OH emissions as in WRF-Chem
    - Test the sensitivity to HBr/Br ratio



