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H. Tran, J. Vander Auwera, T. Bertin, W. Fakhardji, O. Pirali, et al.. Absorption of methane broadened by carbon dioxide in the 3.3 $\mu \rm m$ spectral region: From line centers to the far wings. Icarus, 2022, 384, 10.1016/j.icarus.2022.115093 . insu-03726886

HAL Id: insu-03726886 https://insu.hal.science/insu-03726886v1

Submitted on 22 Jul 2024

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- Absorption of methane broadened by carbon dioxide in the 3.3 μm spectral region:
 from line centers to the far wings
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12 Abstract

This work studied the infrared absorption of methane broadened by carbon dioxide, which can 13 contribute to the radiative budget of CO₂-dominated atmospheres. Fourier transform 14 absorption spectra of CH₄ perturbed by CO₂ were recorded in the 3.3 µm spectral region, at 15 room temperature and total pressures ranging from 3 to 25 bars. These experimental data were 16 modeled using a theoretical approach taking collision-induced line mixing into account. 17 18 Comparisons between measured and calculated spectra demonstrate that the proposed model is capable of accurately representing the absorption of methane broadened by collisions with 19 CO₂, from line centers to the far wings. For practical applications, this rigorous spectral 20 21 modeling was used to derive a simple γ -factor model to represent the spectral shape of CO₂-22 broadened CH₄ line wings. Comparisons with experimental values show that, at room 23 temperature, the proposed χ -factor reproduces the measured spectra to within 10% in the band wings where the absorption is mainly due to the far line wings, while the usual Lorentzian 24 profile leads to relative differences several orders of magnitude larger. This line-shape 25 26 correction was successfully validated through comparisons with heavily saturated spectra of 27 the v_3 band of methane recorded at sub-atmospheric pressures.

28

29 **1. Introduction**

Molecular absorption in the far line wings is known to significantly contribute to the radiative 30 budget of many planets (e.g. those of H₂O for the Earth atmosphere, of CO₂ for Venus and of 31 CH₄ for Titan and Jupiter). This has motivated numerous experimental and theoretical 32 laboratory studies of the far wing absorptions of various molecular systems [e.g. (Hartmann et 33 al., 2018; Tran et al., 2018; Tran et al., 2019; Hartmann et al., 2021) and references therein]. 34 To understand the climate of early Mars (Wordsworth 2016; Haberle et al., 2017; Ramirez 35 and Kaltenegger, 2018) at the time when valley networks were carved by liquid water, it has 36 been proposed that significant amounts of hydrogen and methane could have accumulated in 37 the atmosphere 3-4 billion years ago (Ramirez et al., 2014; Chassefièere et al., 2016; Tarnas et 38 al., 2018; Haberle et al., 2019; Kite et al., 2020). With this aim in view, several experimental 39 and theoretical studies have been performed to determine the far-infrared absorptions induced 40

by collisions between CH₄ (and H₂) and CO₂. The first modeling of these collision-induced 41 absorptions (CIAs) was based on those of CH₄-N₂ (and H₂-N₂) interacting pairs (Ramirez et 42 al., 2014). In (Wordsworth et al., 2017), theoretical calculations of these CIAs were 43 performed using a semi-empirical model that assumed that the shape of the CIA spectrum of 44 CH₄-CO₂ (H₂-CO₂) can be calculated as a linear combination of the CO₂-CO₂ and CH₄-CH₄ 45 (H₂-H₂) CIAs (Richard et al., 2012). The CH₄-CO₂ (H₂-CO₂) CIA integrated intensity was 46 then predicted based on ab initio calculations of the zeroth order spectral moment 47 (Wordsworth et al., 2017). The results showed that the obtained CH₄-CO₂ and H₂-CO₂ CIAs 48 are much more intense than those of CH₄-N₂ and H₂-N₂. The first measurements of these far-49 infrared CIAs were then performed in (Turbet et al., 2019) where it was shown that the 50 calculated values of (Wordsworth et al., 2017) are overestimated by a factor of 1.6-1.7. In 51 (Turbet et al., 2020), a new set of measurements of both CH₄-CO₂ and H₂-CO₂ CIAs was 52 53 carried out, confirming the results of (Turbet et al., 2019), with reduced uncertainties. A semiempirical model was also proposed in (Turbet et al., 2019), based on these measurements, 54 which enabled to calculate CH₄-CO₂ and H₂-CO₂ CIAs over broad spectral (0-1500 cm⁻¹) and 55 temperature (100-600 K) ranges. 56

57 The present work complements these studies through the measurement and modeling of absorption in the far wings of the lines of the CH₄ monomer broadened by CO₂ in the 3.3 µm 58 59 spectral region, which can contribute to the greenhouse warming in CO₂-dominated atmospheres. Spectra of CH₄ highly diluted in CO₂ have been recorded, at room temperature 60 and in a wide pressure range, using two different setups. The experimental setups and 61 conditions are presented in Sec. 2. In Sec. 3, the spectral modeling, considering the effect of 62 collisional line mixing, is described, and comparisons between measured and calculated 63 spectra are presented. In Sec. 4, we propose an empirical χ -factor model to represent the 64 spectral shape of CO₂-broadened CH₄ line wings. This model is determined from the most 65 rigorous spectral modeling presented in Sec. 3 and validated using measured spectra. The 66 conclusions of this work are drawn in Sec. 5. 67

68 **2.** Experiments

69 2.1 Measurements at ULB

Unapodized absorption spectra of CH₄+CO₂ mixtures and of pure CO₂ have been recorded 70 using a Bruker IFS 120 upgraded to 125 HR Fourier transform spectrometer (FTS). For all the 71 72 spectra, the instrument was fit with a globar source, a 0.8-mm entrance aperture diameter, a KBr beamsplitter, a band pass filter and an InSb detector cooled down to 77 K. Except when 73 stated otherwise, the spectra were recorded at a resolution of 0.1 cm⁻¹, corresponding to a 74 maximum optical path difference of 9 cm. A band-pass optical filter was used to only record 75 the spectral range from 1800 to 5300 cm⁻¹, covering the entire v_3 band of CH₄. Methane 76 (99.998 % purity) and carbon dioxide (99.995 % purity) were purchased from Air Liquide. 77 During the measurements, the gas mixtures were contained in a high-pressure cell providing 78 79 an absorption path length of 10.16 m, optically coupled under vacuum to the FTS. The methane pressures were measured using MKS Baratron gauges model 690 of 10 Torr full 80 scale range (Spectrum "No. 1" in Table 1) or model 626 of 1000 Torr full scale range (Spectra 81 "No. 5" and "No. 9" in Table 1). The total pressures were measured using a MKS Baratron 82 gauge model 627F of 25000 Torr full-scale range, directly connected to the high-pressure cell. 83 The model 690 and 627F gauges were temperature stabilized at 45° Celsius. All the gauges 84

are characterized by an accuracy of reading conservatively assumed to be 0.5 %. The cell was

wrapped in 5 cm thick stone wool and its temperature measured using 4 PT-111 platinum
sensors (Lakeshore Cryotronics) characterized by an accuracy of reading better than 0.1 %,
fixed at several locations on the outside wall.

Two series of spectra were recorded. In the first one, a mixture of CH₄ and CO₂ (mixture no. 1 89 in Table 1) at an initial total pressure (Ptot) of 24.53 atm and initial partial pressure of CH4 90 (PCH4) of 0.428 mbar was considered. The mixture was prepared in the cell, filling it with 91 methane followed by carbon dioxide. A first spectrum (identified as "No. 1" in Table 1) was 92 recorded under these sample conditions. The total pressure was then reduced to 12.36 atm and 93 a second spectrum recorded at a resolution of 0.05 cm⁻¹ ("No. 2" in Table 1). Using the same 94 procedure, two other spectra were recorded at total pressures of 6.17 atm ("No. 3" in Table 1) 95 and 3.08 atm ("No. 4" in Table 1) and higher resolution. The corresponding transmission 96 spectra being not saturated, these spectra enable to study the effect of pressure on the 97 absorption in the central region and near wings of the band. To achieve good accuracy on the 98 absorption in a wide spectral range, i.e. from the center of the v_3 band to its far wing, the 99 second set of transmission spectra (identified as mixtures nos. 2 and 3 in Table 1) involved 100 different CH₄ partial pressures for a fixed value of the total pressure. For mixture no. 2, the 101 cell was first filled with 404 mbar of CH₄ and the total pressure brought to 24903 mbar with 102 CO₂. A first spectrum ("No. 5" in Table 1) was recorded. The total pressure was then reduced 103 by a factor F of about 10, CO_2 added to bring the total pressure to 25003.6 mbar (effectively 104 dividing by F the partial pressure of methane) and spectrum "No. 6" was recorded. This 105 procedure was repeated two more times, leading to spectra "No. 7" and "No. 8". The cell was 106 evacuated, a new CH₄+CO₂ mixture (no. 3) prepared and spectra "No. 9" to "No. 13" were 107 measured following the procedure applied to mixture no. 2. Pure CO₂ spectra $S_{CO2}(Ptot, \sigma)$ 108 were also recorded at the same total pressures and resolutions as spectra No. 1 to 4 (Table 1). 109

The uncertainties on the methane pressures provided in Table 1 were estimated as follows. 110 The uncertainties listed for spectra "No. 5" and "No. 9" are equal to 0.5% of P_{CH4} . The 111 uncertainty on the low methane pressure measured for spectrum "No. 1" was conservatively 112 set to 2.5% of P_{CH4} (Cacciani et al., 2022). The uncertainties on P_{CH4} for spectra of samples 113 resulting from the dilution (i.e. spectra "No. 6" to "No. 8" and "No. 10" to "No. 13") were 114 estimated by adding in quadrature the estimated uncertainty on P_{CH4} for the previous 115 spectrum and the uncertainty ΔF on the factor F, estimated to be $\Delta F = \sqrt{2(0.5)^2} \approx 0.71\%$ 116 of the reduced methane pressure. 117

The transmissions due to CH4 broadened by CO2 were obtained by dividing the signal 118 obtained with the mixture, $S_{CH4+CO2}(Ptot,\sigma)$, by $S_{CO2}(Ptot,\sigma)$. With this procedure, the 119 contribution of CO₂ absorption (the wings of neighboring CO₂ vibrational bands and some 120 weak collision-induced absorption bands) could be removed. The absorption coefficients were 121 then extracted from the transmissions thus obtained and normalized by the CH₄ partial 122 pressure. Since most of the studied gas samples contain mixing ratios of methane smaller than 123 0.55%, the contribution of CH₄-CH₄ collisions is negligible. The contribution of CH₄-CH₄ 124 collisions being not negligible for spectra No. 5 and 9, it was taken into account assuming that 125 the effects of CH₄-CH₄ and of CH₄-CO₂ collisions are the same. This is justified by the fact 126 that the average ratio of self- to air-broadening coefficients of CH₄ is 1.27 (Pine, 1992), very 127 close to the ratio of CO₂ to air-broadenings (1.3, see Sec.3). As shown in Fig. 1 for mixture 128 no. 2, the whole v_3 band was then reconstructed by combining the spectral windows in which 129

- the transmissions were between 0.2 and 0.8. The analysis of the second series of spectra
- 131 recorded at ULB therefore resulted in two CH₄ partial-pressure-normalized absorption spectra
- of CH₄ broadened by CO₂ measured at average total pressures of 24991 mbar (mixture no. 2)
- and 24978 mbar (mixture no. 3) and an average temperature of 295.7 K.

Mixture	No.	$P_{tot}(\Delta P)$ / mbar	$T(\Delta T) / ^{\circ}C$	$P_{CH4}(\Delta P)$ / mbar	Res. / cm ⁻
					1
1	1	24851 (17)	23.72 (0.12)	0.428 (11)	0.1
	2	12521.2 (2.9)	22.52 (0.01)	0.216 (6)	0.05
	3	6255.1 (2.0)	22.28 (0.08)	0.108 (3)	0.025
	4	3122.9 (1.3)	22.16 (0.02)	0.054 (1)	0.0125
2	5	24903 (12)	23.30 (0.08)	404 (2)	0.1
	6	25003.6 (6.0)	22.68 (0.04)	42.1 (4)	0.1
	7	25029.4 (6.0)	22.68 (0.04)	4.20 (5)	0.1
	8	25029 (13)	22.70 (0.08)	0.428 (6)	0.1
3	9	24764 (14)	22.85 (0.11)	1241 (6)	0.1
	10	24990.2 (8.0)	22.71 (0.06)	130.3 (1.1)	0.1
	11	25025.2 (3.3)	22.62 (0.03)	13.16 (15)	0.1
	12	25052.0 (2.0)	22.62 (0.01)	1.345 (18)	0.1
	13	25074.8 (8.1)	22.63 (0.07)	0.1376 (21)	0.1

134 Table 1: Experimental conditions for the CH₄+CO₂ spectra measured with the ULB setup.

135 Spectra of pure CO₂ were recorded for the same pressure conditions as spectra No. 1 to 4. The

136 numbers between parentheses provided for the pressures and temperatures are the estimated

uncertainties. For the total pressure P_{tot} and the temperature T, they are the variations of these

138 quantities during the recordings, the estimated accuracy of measurement of the pressure (0.5%)

139 of reading) and of the temperature (0.1% of reading) being not included. The estimation of the

140 uncertainties on the methane pressures is detailed in the text.

141 Note that the measured and reconstructed spectra were corrected to get rid of possible errors 142 in the partial pressures of methane. This was done by multiplying the experimental results by 143 a constant factor such that the integrated absorptions of the measured spectrum and of the 144 corresponding calculated one match. The correction factors vary from 0.96 to 1.1 depending 145 on the spectrum. They are larger than the estimated uncertainties on the methane pressures 146 and probably a better assessment of these. No correlation was found between the correction 147 factors and the CH₄ pressures.

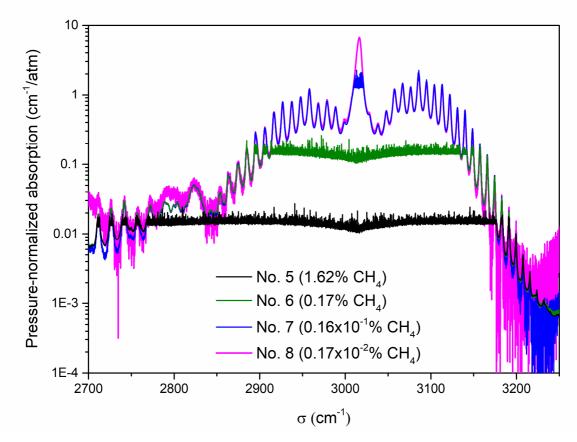


Figure 1: CH₄ partial-pressure-normalized absorption spectra of CH₄ broadened by CO₂
 measured at an average total pressure of 24991 mbar and for various partial pressures of CH₄
 (mixture no. 2 in Table 1).

152 2.2. Measurements at SOLEIL

153 CH₄+CO₂ transmittance spectra were also measured using the facilities at the AILES beam line of the SOLEIL synchrotron. The experimental setup is composed of a Fourier transform 154 spectrometer Bruker IFS 125 HR equipped with a globar source and a KBr beamsplitter from 155 which the incident light beam goes into a 2.5 m long multi-pass cell closed with diamond 156 windows, set for an optical path of 101.3 m. The transmitted light was received by a HgCdTe 157 detector cooled down with liquid nitrogen. Each spectrum has a usable part between 800 and 158 4600 cm⁻¹ and results from the average of 1000 interferograms recorded with an unapodized 159 resolution of 0.5 cm⁻¹. 160

The experiments proceeded as follows. The cell was filled with 54.4 mbar of CH₄ and CO₂ 161 was added until a total pressure of 945.7 mbar was reached (the pressures were measured with 162 a 0 - 1000 mbar Pfeiffer gauge), resulting in a methane mole fraction of about 5 %. The 163 mixture was let to evolve freely for 10 to 15 minutes to reach equilibrium and the absorption 164 spectrum of the mixture was recorded. The total pressure was then reduced to 764.2 mbar, the 165 mixture given 10 to 15 minutes to reach equilibrium and the spectrum of the mixture 166 recorded. The procedure was then repeated for total pressures of 607.3, 480.0, 378.3, 171.6, 167 107.1 and 58.3 mbar. A reference spectrum was also recorded with the cell filled with about 1 168 bar of pure argon to generate transmittance spectra for all the total pressures. 169

170 **3. Spectral modeling: line-mixing effect**

It is well known that collisional interferences (line-mixing) play an important role in the infrared absorption spectrum of methane [see (Pieroni et al., 1999; Tran et al., 2006a) for instance]. Within the impact approximation and including line-mixing effects, the infrared absorption coefficient α (cm⁻¹) at wavenumber σ (cm⁻¹) of CH₄ broadened by CO₂ is given by (Hartmann et al., 2021):

176
$$\alpha^{LM}(\sigma, P_{CH4}, P_{CO2}, T) = \frac{8\pi^2 \sigma}{3hc} \left[1 - \exp\left(-\frac{hc\sigma}{k_B T}\right) \right] \times \frac{P_{CH4}}{k_B T}$$

177
$$\times Im\{\sum_{\ell}\sum_{k}\rho_{\ell}(T)d_{\ell}d_{k}\langle\langle k\|[\mathbf{\Sigma}-\mathbf{L}_{\mathbf{0}}-iP_{CO2}\mathbf{W}(T)]^{-1}\|\ell\rangle\rangle\}.$$
 (1)

In Eq. (1), the sums extend over all the absorption lines ℓ and k of CH₄; P_{CH4} and P_{CO2} are the 178 partial pressures of CH₄ and CO₂, respectively; T is the temperature (in kelvin); h, c and k_B 179 are the Planck constant, the speed of light in vacuum and the Boltzmann constant. ρ_{ℓ} and d_{ℓ} 180 are the relative population of the initial level and the dipole matrix element of line ℓ . Σ , L_0 181 and W are operators in the Liouville space and $\langle \langle k \| \dots \| \ell \rangle \rangle$ designates a matrix element. The 182 183 first two are diagonal, associated with the current wavenumber and the positions of the lines. 184 W is the relaxation matrix which describes the effect of collisions between CH₄ and CO₂ on the CH₄ optical transitions. In this equation, we neglected the self-broadening effect as 185 186 explained in Sec. 2.1, the Doppler term and the influence of velocity changes as well as the 187 speed dependences of the relaxation matrix W. The latter contains the influence of collisions on the spectral shape. It depends on the band, on the temperature and on the collisional 188 189 partner. Its off-diagonal elements lead to collisional interferences between lines while the real 190 and imaginary diagonal terms are the pressure-broadening, γ_{ℓ} , and -shifting, δ_{ℓ} , of the 191 isolated lines, i.e.

192
$$\langle \langle \ell \| \boldsymbol{W}(T) \| \ell \rangle \rangle = \gamma_{\ell}(T) - i \delta_{\ell}(T) .$$
 (2)

193 When line mixing is disregarded, W is diagonal and Eq. (1) becomes the usual sum of 194 Lorentzian contributions.

195 The off-diagonal imaginary elements of W are neglected while the real elements are 196 computed, as done in (Pieroni et al., 1999; Tran et al., 2006), using:

197 $\langle \langle k \| \boldsymbol{W}(T) \| \ell \rangle \rangle$

$$\langle\langle k \| \boldsymbol{W}(T) \| \ell \rangle\rangle = -A_{k,\ell} \times K(i_k \leftarrow i_\ell, T), \tag{3}$$

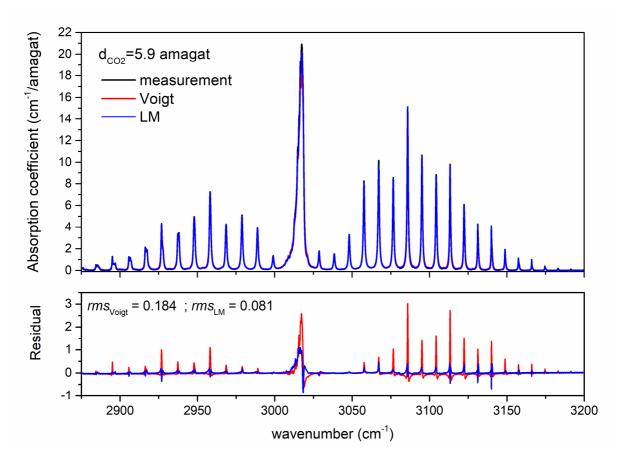
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where $K(i_k \leftarrow i_\ell, T)$ denotes the state-to-state collisional transfer rates from the initial level of line ℓ to the initial level of line k; the parameters $A_{k,\ell}$ enable switching from the state space to the line space, as explained in (Pieroni et al., 1999).

Spectra were calculated for the pressure and temperature conditions of the experiments. Since 201 the considered region is dominated by the v_3 band of ${}^{12}CH_4$, calculations with line-mixing 202 taken into account were performed only for the lines of this band. The other lines (minor 203 isotopologues, hot bands) were calculated with the Voigt profile. The individual line 204 parameters, i.e. the unperturbed line position, integrated intensity, energy of the lower state, 205 were taken from the 2020 edition of the HITRAN database (Gordon et al., 2022). The 206 populations and the dipole matrix elements were deduced from these data [as done in (Tran et 207 al., 2006a)]. The pressure-broadening coefficients of CH₄ lines by CO₂ were obtained by 208 multiplying by 1.3 the air-broadening coefficients provided by the package of (Tran et al., 209 2006a). This factor, which is close to the value of 1.21 obtained by averaging the ratio 210

- between the CO₂-broadening coefficients of 28 CH₄ lines measured by (Fissiaux et al., 2014) 211 and the corresponding values for air-broadening [taken from (Tran et al., 2006a) and 212 references therein], was retained because it gives the best agreement between the spectra 213 calculated and measured in the present work. In the absence of available measured or 214 calculated data, we assumed that the pressure-shifting coefficients of CH₄ by collisions with 215 CO₂ are the same as those of CH₄ by air. The latter were also provided by the package of 216 (Tran et al., 2006a). Note that the air-broadening and -shifting coefficients in (Tran et al., 217 2006a) were obtained from the measured data of (Pine, 1997) in which these parameters were 218 determined with line-mixing effect taken into account. These data are thus consistent with the 219 line-shape model used here. 220
- The off-diagonal elements of the relaxation matrix, including the state-to-state rates $K(i_k \leftarrow i_{\ell}, T)$ and the $A_{k,\ell}$ parameters, were assumed to be proportional to those for CH₄ in N₂, also provided in (Tran et al., 2006a), with the same multiplicative factor as for the line broadening coefficients, i.e. $\langle\langle k \| W^{CH4/CO2}(T) \| \ell \rangle\rangle = 1.3 \times \langle\langle k \| W^{CH4/N2}(T) \| \ell \rangle\rangle$. Detailed information on the state-to-state rates $K(i_k \leftarrow i_{\ell}, T)$ and the $A_{k,\ell}$ parameters can be found in (Tran et al., 2006a) and references therein.
- 227 Figures 2 and 3 show examples of comparison between measured spectra (mixture no. 1 in 228 Table 1) and those calculated as described above for the same pressure and temperature conditions. In addition to the predictions made taking line-mixing into account, the spectra 229 calculated with the same spectroscopic parameters but using the Voigt profile (i.e. neglecting 230 line-mixing effect) are also displayed. The density of CO₂ was determined from the measured 231 pressure and temperature using NIST-recommended equation of state taking into account the 232 fact that the gas is not ideal.ⁱ As can be observed, the line-mixing calculations, although not 233 perfect, lead to much better agreements with the measurements, the root mean square of the 234 obtained residuals being more than twice smaller than that for the Voigt calculations. These 235 results are consistent with those obtained previously for CH₄ broadened by N₂ (Pieroni et al., 236 1999; Tran et al., 2006a) in which a detailed analysis of the influence of line-mixing effects 237 on the absorption spectrum can be found. 238

ⁱ https://webbook.nist.gov/cgi/cbook.cgi?Source=1927LOW%2FERI2729-2734&Units=SI&Mask=7



240 Figure 2: Comparison between the CO₂-broadened CH₄ spectrum measured (black) at room

temperature for a density of CO₂, (d_{CO2}) of 5.9 amagat (spectrum No. 3 in Table 1; 1 amagat corresponding to 2.687 10¹⁹ molec.cm⁻³ at 273.15 K) and values calculated with (blue) and with out (md) line mining token into account

243 without (red) line mixing taken into account.

244

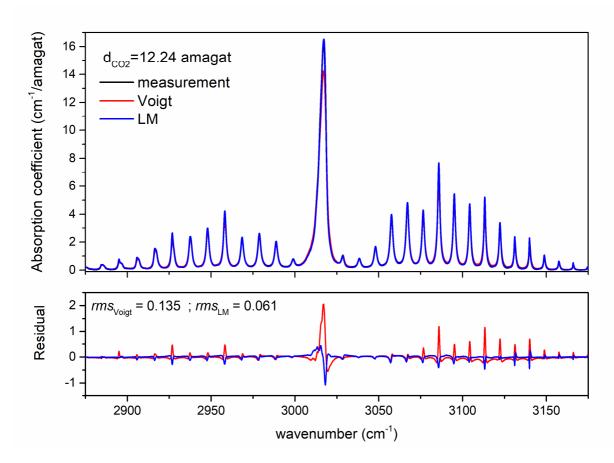


Figure 3: Same as Fig. 2 for d_{CO2} =12.24 amagat (spectrum no. 2 in Table 1).

247 The absorption in the band wing is displayed in Fig. 4 where calculations with and without 248 line-mixing are compared with the spectra reconstructed (see Sec. 2.1) from measurements for the mixtures nos. 2 and 3 of Table 1. Again, the calculation with line-mixing leads to a good 249 250 agreement with the measurements, especially in the blue wing of the band. Absorption in this spectral region is mainly due to the cumulating far wings of the strong v_3 lines close to the 251 center of the band. For CH₄, it is well known that the Voigt profile strongly overestimates 252 absorption in the far line wings (Hartmann et al., 2002; Tran et al., 2006a; Tran et al., 2006b), 253 which is also observed here. Furthermore, in the troughs between the manifolds, particularly 254 255 in the R branch, taking line-mixing into account also significantly improves the agreement with the measurements, when compared with the purely Voigt calculation. This result shows 256 that our model, although quite simple since based on an approach developed for N₂-broadened 257 258 CH₄ lines, is capable to accurately predict the absorption shape from resonances to the far wings of CH₄ perturbed by CO₂. 259

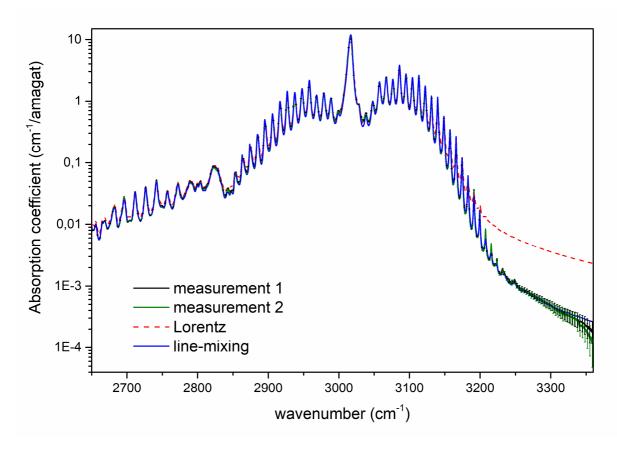


Figure 4: Comparison between CO₂-broadened CH₄ spectra measured at $d_{tot} = 26.5$ amagat (in olive and black, corresponding to mixtures no. 2 and 3 of Table 1) and values calculated with (blue) and without (red) line-mixing taken into account. The error bars on the measured absorption coefficient represent the uncertainties due to those of the baselines and the subtracted CO₂ absorption contributions (estimated to be 1%).

266 4. χ-factor model for the far wings of CO₂-broadened CH₄ lines

For practical applications such as simulations of planetary atmospheres spectra, Lorentzian (or Voigt) line shapes with line wings corrected using a χ -factor are widely used. Within this simple model, the absorption coefficient of a highly diluted CH₄-CO₂ mixture is written as:

270
$$\alpha(\sigma, P_{CH4}, P_{CO2}, T) = \frac{1}{\pi} \sum_{\ell} P_{CH4} S_{\ell}(T)$$

271
$$\times exp\left[\frac{hc(\sigma-\sigma_{\ell})}{2k_{B}T}\right]\frac{\sigma}{\sigma_{\ell}}\frac{1-e^{-\frac{hc\sigma}{k_{B}T}}}{1-e^{-\frac{hc\sigma_{\ell}}{k_{B}T}}}\frac{P_{CO2}\gamma_{\ell}(T)}{(\sigma-\sigma_{\ell})^{2}+(P_{CO2}\gamma_{\ell}(T))^{2}}\chi(|\sigma-\sigma_{\ell}|,T)$$
(4)

The corrective factor χ , which is assumed to be the same for all lines, depends on the frequency detuning with respect to the line center and on the temperature. It is generally (Perrin and Hartmann, 1989; Hartmann et al., 2002; Tran et al., 2011) represented by analytical expressions with parameters deduced from fits of measured spectra. In this work, χ was determined by fitting Eq. (4) to absorption spectra in the region of the v₃ band, calculated using the model described in Sec. 3 (taking line-mixing effect into account) for various pressure and temperature conditions. Note that the model described in Sec. 3 is based on a theoretical approach developed for CH₄ perturbed N₂ (Pieroni et al., 1999; Tran et al., 2006a), which showed that it is capable to accurately represent absorption spectra of CH₄ in N₂ for large range of temperature (i.e. from 200 to 300 K). Here, we assume that the broadening coefficients and the off-diagonal elements of the relaxation matrix of CH₄ in CO₂ have the same temperature dependences as those of CH₄ in N₂. Therefore, the model of Sec. 3 can be used to simulate spectra at various temperatures. The following values for γ were obtained:

285
$$|\sigma - \sigma_{\ell}| = \Delta \sigma < \sigma_1 = 15 \text{ cm}^{-1}, \chi = 1$$

286
$$\sigma_1 < \Delta \sigma < \sigma_2 = 60 \text{ cm}^{-1}, \chi = \exp[-0.0375(\Delta \sigma - \sigma_1)]$$
 (5)

287
$$\sigma_2 < \Delta \sigma$$
, $\chi = \exp[-0.0375(\sigma_2 - \sigma_1) - \beta(T)(\Delta \sigma - \sigma_2)]$
288 with $\beta(T) = -5.3 \times 10^{-4} + 0.0194e^{-0.0041T}$

Figure 5 shows comparisons between the absorption coefficients calculated using Eqs. (4,5)289 and those obtained with the model described in Sec. 3 for three different pressure and 290 temperature conditions. Very good agreements in the blue wing of the band can be observed. 291 Note that the γ -factor as proposed in Eqs. (4,5) allows for the correction of sub-Lorentzian 292 effect in the line wings while it is equivalent to a Lorentzian (or Voigt) line shape in the line 293 centers and near wings. In the band wings, where absorption is mainly due to the far line 294 wings, the χ -factor approach leads to good agreements with measured spectra. For instance, at 295 3400 cm⁻¹ and for $d_{tot} = 26.5$ amagat, the relative difference between the absorption 296 coefficients computed with the γ -factor model and the measured spectra is about 10% while 297 the usual Lorentzian profile leads to a relative difference of several order of magnitude. 298 However, in regions of line centers and near wings, an accurate modeling of the measured 299 absorptions requires taking into account properly line-mixing effect through the use of the full 300 relaxation matrix as described in Sec. 3. The χ -factor model therefore should be used to 301 model absorption in the far line wings only. 302

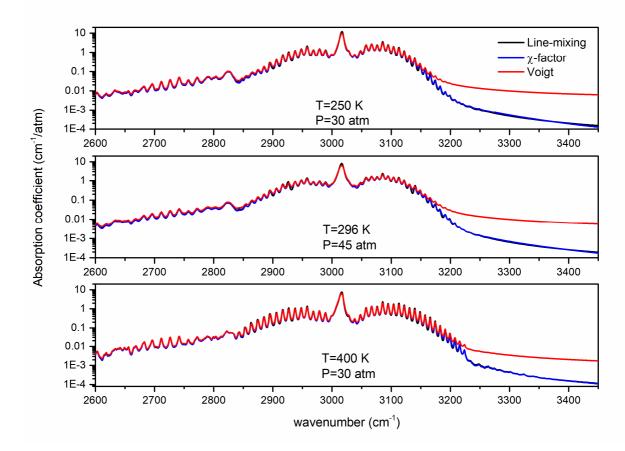
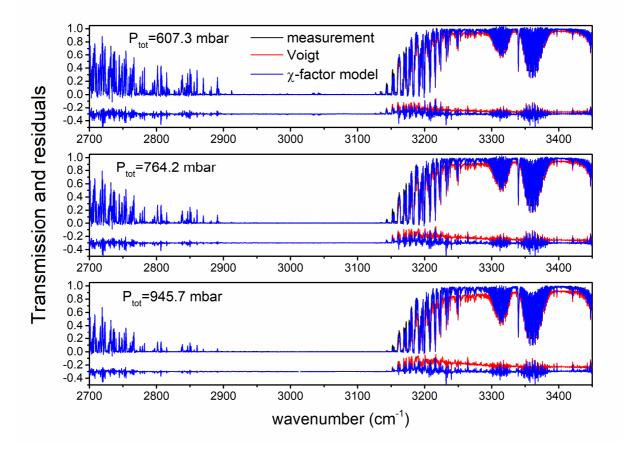


Figure 5: Absorption coefficients of CH₄ highly diluted in CO₂ calculated with (black line) and without (red) line mixing, at 250 K and 30 atm (top panel), 296 K and 45 atm (middle panel) and 400 K and 30 atm (bottom panel). Values calculated with the χ -factor model [see Eqs. (4,5)] are plotted in blue.

In (Gharavi and Buckley, 2005) devoted to the R(3) and R(4) manifolds of the $2v_3$ band, it 308 was shown that the temperature dependences of CO₂- and N₂-broadening coefficients of CH₄ 309 lines can be quite different. The ratio γ_{CO2}/γ_{N2} can indeed vary from 1.36 to 1.24 in the 296-310 670 K range. It is well known that the absorption coefficient in the line wing is proportional to 311 the corresponding line width. Therefore, changing γ_{CO2}/γ_{N2} from 1.3 to 1.2 or to 1.4 leads to 312 a difference of about 10% for the absorption coefficient in the wings. This is negligible 313 compared to the difference between the absorption coefficient calculated with the Voigt line 314 shape and that computed with the full line-mixing modeling (about one order of magnitude at 315 3300 cm⁻¹, for instance, see Figs. 4, 5). In the absence of accurate data for CO_2 -broadening 316 coefficients and their temperature dependences, the χ -factor proposed in this work can be 317 therefore used in the 200 K-400 K temperature range. 318

The obtained χ -factor model was then used to compute spectra for the same conditions as those of the experiments recorded at SOLEIL (see Sec. 2.2). For the comparison with the measured spectra, absorption contributions due to CO₂ and to a small amount of water vapor (a H₂O fraction of $x_{H2O} = 9.5 \times 10^{-5}$ was retrieved from the measured spectra) were included in the calculation. Voigt profiles were used, and the needed spectroscopic data of CO₂ and H₂O were taken from (Gordon et al., 2022). Note that calculations using a χ -factor model for CO₂ line wings (Tran et al., 2011) were also performed to compute CO₂ absorption

contributions, showing that CO₂ line wings have negligible influence on the total absorptions. 326 The CH₄-broadening coefficients of CO₂ lines were assumed to be the same as the air-327 broadenings. This approximation should not affect the results since the mole fraction of CH4 328 is quite small ($x_{CH4} = 0.0575$). Figure 6 shows comparisons between the measured spectra 329 and those computed using the χ -factor model proposed in this work as well as using purely 330 Voigt line shapes, for the v_3 band region. As can be observed, the spectra are largely saturated 331 around the band center. In the red wing of the band, the two calculations are both in good 332 agreement with the measurements. This can be explained by the large contribution of local 333 lines in this region. In the blue wing of the band where the relative contribution of the far 334 wings is more important, the γ -factor model accurately reproduces the measurements, 335 contrary to the Voigt profile that largely overestimates absorptions. These results thus validate 336 the χ -factor model [Eqs.(4) and (5)] at room temperature, which can now be used for 337 applications as those mentioned in Sec. 1. Note that the remaining residuals, namely those 338 observed around 3200 cm⁻¹, are probably due to the uncertainties of the spectroscopic data of 339 the local lines (Gordon et al., 2022) but also to the use of a unique scaling factor to deduce 340 CO₂-broadenings coefficients of methane lines from air-broadening coefficients. 341



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Figure 6: Comparison between the SOLEIL spectra measured (black) in the v_3 band region at

room temperature and various total pressures of a mixture of CH₄ in CO₂ with $x_{CH4} = 0.0575$ and calculations at the same conditions, using the γ -factor model (blue) and the usual Voigt

346 line shape (red). The differences between the measured spectra and calculated values are also

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³⁴⁷ displayed in each corresponding panel (minus 0.3).

349 **5.** Conclusion

350 This article presented the first modeling of the spectral shape of methane absorption lines broadened by CO₂ from the line centers to the far wings. The model, based on a theoretical 351 approach previously developed for N₂-broadened CO₂ absorption lines, accounted for the 352 collision-induced line-mixing effects. Comparison with spectra measured in the CH₄ v₃ band 353 354 region at room temperature and for a broad pressure range showed that the model is capable to accurately reproduce absorption spectra of CH₄ in CO₂, from the line centers to the far 355 wings, in contrast with the Voigt profile that leads to large discrepancies with the 356 measurements. The model was then used to compute spectra of CH₄ in CO₂ for various 357 pressure and temperature conditions (from 200 to 400 K), which enabled the determination of 358 a simple line-shape correction factor γ in order to correct for the sub-Lorentzian behavior in 359 the far line wings. The latter, successfully validated by comparison with experimental spectra, 360 can be straightforwardly used to model absorption in the line wings, for practical applications 361 such as radiative transfer calculations for a planetary atmosphere. Finally, it is worthy to note 362 that the model can be largely improved in the future when accurate CO₂-broadening 363 coefficients of CH₄ lines, their temperature dependences as well as the state-to-state 364 collisional transfer rates are available. 365

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367 Acknowledgements

This work was performed in the frame of the *Agence Nationale de la Recherche* (ANR) project COMPLEAT (ANR-19-CE31-0010-01). T. Bertin thanks the FRIA (*Fonds pour la Formation à la Recherche dans l'Industrie et l'Agriculture*, Belgium) for a PhD fellowship and J. Vander Auwera acknowledges financial support from the *Fonds de la Recherche*

- 372 *Scientifique* FNRS (Belgium, contract no. J.0217.20).
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