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1 Pervasive aqueous alteration in the early Solar System revealed by 2 potassium isotopic variations in Ryugu samples and carbonaceous chondrites

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111 **Abstract**

112 C-type asteroids are the presumed home to carbonaceous chondrites, some of which contain
113 abundant life-forming volatiles and organics. For the first time, samples from a C-type asteroid
114 (162173 Ryugu) were successfully returned to Earth by JAXA's Hayabusa2 mission. These
115 pristine samples, uncontaminated by the terrestrial environment, allow a direct comparison with
116 carbonaceous chondrites. This study reports the stable K isotopic compositions (expressed as
117 $\delta^{41}\text{K}$) of Ryugu samples and seven carbonaceous chondrites to constrain the origin of K isotopic
118 variations in the early Solar System. Three aliquots of Ryugu particles collected at two
119 touchdown sites have identical $\delta^{41}\text{K}$ values, averaged at $-0.194 \pm 0.038\text{‰}$ (2SD). The K isotopic
120 composition of Ryugu falls within the range of $\delta^{41}\text{K}$ values measured on representative CI
121 chondrites, and together, they define an average $\delta^{41}\text{K}$ value of $-0.185 \pm 0.078\text{‰}$ (2SE), which
122 provides the current best estimate of the K isotopic composition of the bulk Solar System.
123 Samples of CI chondrites with $\delta^{41}\text{K}$ values that deviate from this range likely reflect terrestrial
124 contaminations or compositional heterogeneities at sampled sizes. In addition to CI chondrites,
125 substantial K isotopic variability is observed in other carbonaceous chondrites and within
126 individual chondritic groups, with $\delta^{41}\text{K}$ values inversely correlated with K abundances in many
127 cases. These observations indicate widespread fluid activity occurred in chondrite parent bodies,
128 which significantly altered the original K abundances and isotopic compositions of chondrules
129 and matrices established at their accretion.

130

131 **Keywords:** Asteroid Ryugu; Hayabusa2 mission; Carbonaceous chondrites; Stable potassium
132 isotopes; Aqueous alteration

133 **1 Introduction**

134 Primitive asteroids accreted varying amounts of ice that melted to water-rich fluids due to
135 heat released from the radioactive decay of short-lived nuclides (e.g., ²⁶Al, McSween et al.,
136 2002) and possibly from impacts (Rubin, 2012). Consequently, fluid activity is believed to have
137 been pervasive in the early history of planetesimal and planet development (Brearley, 2006;
138 Brearley, 2014; Clayton, 1993; Zolensky and McSween, 1988). Potassium (K) is a fluid-mobile,
139 large-ion lithophile element that resides primarily in **chondrule** mesostasis and fine-grained
140 matrix in carbonaceous asteroids. These components react readily with aqueous fluids, leading to
141 redistribution of K⁺ and its isotopes. In addition, K is a moderately volatile metal that is variably
142 depleted in carbonaceous chondrites and terrestrial planetary bodies. Because both volatility-
143 driven depletion and fluid-rock interaction can lead to mass-dependent fractionations between
144 the two stable K isotopes (i.e., ³⁹K and ⁴¹K), it is essential to understand how these processes
145 affect δ⁴¹K variations in meteorites to facilitate the interpretation of K isotopic signatures of
146 planetary materials.

147 Carbonaceous Ivuna-type (CI) chondrites are the chemically most primitive meteorites and
148 have been widely used to infer the isotopic composition of the bulk Solar System (Palme et al.,
149 2014). Despite their primitive bulk composition, CI chondrites underwent near-complete aqueous
150 alteration of anhydrous precursors to secondary mineral assemblages dominated by
151 phyllosilicates (e.g., Tomeoka and Buseck, 1988) and display significant small-scale chemical
152 heterogeneities (e.g., Barrat et al., 2012; Morlok et al., 2006; Palme and Zipfel, 2022). Because
153 CI chondrites are limited in numbers and masses, **previous K isotopic analyses focused on the**
154 **largest witnessed fall (Orgueil, 14 kg) and the CI-type specimen (Ivuna, 0.7 kg).** Earlier analyses
155 of Orgueil returned a δ⁴¹K value (−0.534 ± 0.097‰, 2SE) that is similar to the bulk Silicate
156 Earth (Wang and Jacobsen, 2016); **however, this sample shows elemental enrichments (e.g., K,**
157 **Rb, Ba, Th, and U) that indicate significant contamination on Earth (Ku and Jacobsen, 2020).**
158 **Reported δ⁴¹K values of other Orgueil samples vary between −0.290‰ and −0.039‰ (Hu et al.,**
159 **2022, 2023; Koefoed et al., 2022, 2023; Ku and Jacobsen, 2020; Nie et al., 2021), with the six**
160 **distinct stones reported by Koefoed et al. (2022; 2023) displaying more uniform δ⁴¹K values**
161 **ranging from −0.290‰ to −0.170‰. However, while Koefoed et al. (2023) reported a similar**
162 **δ⁴¹K value of −0.180 ± 0.060‰ for Ivuna, a markedly lower value of −0.460 ± 0.046‰ was**
163 **reported for another Ivuna sample with a similar K abundance (Nie et al., 2021). The cause of**

164 this intra-group difference remains unclear. It may reflect an unrepresentative sampling of the
165 bulk composition due to sample heterogeneity caused by aqueous alteration in the parent body(s)
166 of the CI chondrites or remobilization of K in terrestrial environments. A better-defined K
167 isotopic composition of CI chondrites is critical for constraining the bulk composition of the
168 Solar System.

169 In December 2020, JAXA's Hayabusa2 mission returned the first samples from the
170 carbonaceous asteroid 162173 Ryugu, which had not been exposed to terrestrial environments
171 prior to laboratory analysis. These samples contain abundant organic material plausibly inherited
172 from the protosolar nebula and/or the interstellar medium, which attests to their primitive nature
173 (Nakamura et al., 2022; Pilorget et al., 2022). Furthermore, chemical and isotopic analyses
174 suggest that Ryugu has a similar bulk composition to CI chondrites (Hopp et al., 2022; Moynier
175 et al., 2022; Nakamura et al., 2022; Paquet et al., 2022; Yokoyama et al., 2022). In particular, the
176 Ryugu samples do not contain chondrules or refractory inclusions (Pilorget et al., 2022;
177 Yokoyama et al., 2022) and display no depletion of volatile metals relative to the solar
178 photosphere (Yokoyama et al., 2022), which are essential features of CI chondrites. Therefore,
179 the Ryugu samples are a direct asteroidal analogue to CI chondrites. Here, we report the K
180 isotopic compositions of three aliquots of Ryugu particles collected from two touchdown sites
181 and seven carbonaceous chondrites measured with the Nu Sapphire collision-cell multi-collector
182 inductively-coupled plasma mass spectrometer (CC-MC-ICP-MS) **to further constrain the**
183 **pristine composition of CI chondrites and the origin of K isotopic variation in the Solar System.**

184

185 **2 Samples and methods**

186 **2.1 Sample description and pre-chemistry preparation**

187 Ryugu formed as a rubble-pile asteroid (approximately 1 km in diameter) reaccumulated
188 from the shattered debris of a larger parent body (Sugita et al., 2019); therefore, Ryugu likely
189 sampled rock fragments from all depths of its parent body. The Hayabusa2 mission retrieved 5.4
190 g of pebbles and sands at two locations on Ryugu (Tachibana et al., 2022; Yada et al., 2022).
191 Around 3 g of surface material was collected from the touchdown site 1 and stored in Chamber
192 A (Morota et al., 2020). The second landing targeted at collecting impact ejecta from a ~ 15-m
193 diameter crater artificially made by the onboard Small Carry-on Impactor (SCI) (Arakawa et al.,

194 2020). Around 2 g of samples containing subsurface material was collected from the touchdown
195 site 2 and stored in Chamber C. Samples collected from these two sites are considered
196 representative of the compositionally homogeneous Ryugu surface based on their consistent
197 spectral features and physical properties (color, shape, surface morphology, and structure) with
198 onboard measurements made by the spacecraft (Kitazato et al., 2019; Nakamura et al., 2022;
199 Pilorget et al., 2022; Tachibana et al., 2022).

200 Three aliquots of Ryugu particles were analyzed for K isotopes. Sample A0106 is from
201 Chamber A, while samples C0107 and C0108 are from Chamber C. Preliminary sample
202 treatment has been performed at the Tokyo Institute of Technology (Tokyo Tech). Samples
203 A0106 and C0107 were first treated to extract soluble organic matter (Naraoka et al., 2023).
204 Then, together with C0108, they were dissolved in pre-cleaned perfluoroalkoxy (PFA) vials in a
205 mixture of concentrated HF and HNO₃, as described in Yokoyama et al. (2022). All three Ryugu
206 samples were first processed to separate the Zn fractions from other constituent elements in the
207 samples (Paquet et al., 2022). This was followed by a three-step sequential chemical separation
208 procedure for the isotopic measurements of Fe (Hopp et al., 2022), Ti, and Cr (Yokoyama et al.,
209 2022), after which we obtained a fraction containing **mainly** K, Mg, and Ni. Aliquots of the K-
210 Mg-Ni fraction for the Ryugu samples were transferred to the Institut de Physique du Globe de
211 Paris (IPGP) for K isotopic analyses. The amounts of K finally obtained from the three Ryugu
212 samples ranged from 1365 to 2867 ng. To verify whether the multi-stage chemical separation
213 fractionated the K isotopes, the K-Mg-Ni fractions of three carbonaceous chondrites, including
214 Tarda (C2-ungrouped), Tagish Lake (C2-ungrouped), and Allende (CV3), which were subjected
215 to the same processing as the Ryugu samples at Tokyo Tech, were chemically purified for K
216 isotopic analyses at IPGP.

217 We also analyzed four geostandards (BIR, BCR-2, AGV-2, and seawater) and seven in-house
218 carbonaceous chondrites at IPGP to examine possible sample heterogeneity and to compare with
219 the K isotopic compositions of the Ryugu samples. The additional chondrite samples were:
220 Orgueil (CI1), Tagish Lake (C2-ungrouped), Tarda (C2-ungrouped), Cold Bokkeveld (CM2),
221 Murchison (CM2), Lancé (CO3.5), and Allende (CV3). At IPGP, approximately 8-15 mg of the
222 sample powders were dissolved in Savillex screw-top PFA beakers using sequential addition of
223 concentrated HF-HNO₃ (3:1), HCl-HNO₃ (3:1), and HNO₃. The dissolved sample solutions were
224 refluxed with and redissolved in 0.5 mol/L HNO₃.

225

226 2.2 Potassium purification and isotope measurement

227 Potassium was isolated from other elements by cation-exchange chromatography described
228 in Xu et al. (2019). Two mL of Bio-Rad AG 50W-X8 cation exchange resin (200-400 mesh) was
229 filled into pre-cleaned Bio-Rad Poly-Prep columns. The resin was cleaned by three passes with 6
230 mol/L HCl, one pass with 6 mol/L HNO₃, and one pass with 1 mol/L HNO₃. Each acid pass was
231 rinsed with 10 mL of Milli-Q deionized water. The resin was then conditioned with 10 mL of 0.5
232 mol/L HNO₃. Dissolved sample solutions were loaded onto the resin in 1 mL of 0.5 mol/L
233 HNO₃. A volume of 13 mL of 0.5 mol/L HNO₃ was passed through the column, and the K
234 fraction was collected in the subsequent 22 mL of the same acid. The sample solutions were
235 passed through the column four times. This chromatography protocol provides a K yield greater
236 than 99.3% and effectively separates matrix elements from the K fractions, as have been
237 evaluated in previous studies (Xu et al., 2019; Hu et al., 2022). The K blank introduced by each
238 pass of column chemistry is ~ 0.2 ng, and the whole-procedure K blank varies between 3.7 ng
239 and 6.2 ng (Hu et al., 2022), which is insignificant (< 1%) compared to the amounts of K in
240 processed samples.

241 Potassium isotopic analyses followed the protocol described by Moynier et al. (2021).
242 Purified K fractions were diluted in 0.5 mol/L HNO₃ to have K concentrations between 60 ppb
243 and 70 ppb. They were introduced into the mass spectrometer using an Apex Omega desolvation
244 nebulizer system fitted with a 100 µL/min nebulizer. The ion beam was first decelerated to enter
245 a hexapole collision cell to remove interfering species (e.g., ⁴⁰ArH⁺ on ⁴¹K⁺), and the ⁴¹K⁺ and
246 ³⁹K⁺ ion beams were accelerated and directed through the magnet to the Faraday collectors. Each
247 analysis consisted of a 90-second transfer time, a 60-second zero measurement in a blank
248 solution, and 50 cycles of sample measurement with 5-second integration. A 150-second wash
249 was performed between analyses. Each sample solution was paired with a standard solution
250 diluted to the same K concentration (to within 2%). Standard and sample solutions were
251 analyzed alternatively during a sequence to correct for instrumental mass bias. Potassium
252 isotopic data are reported as the average of repeated analyses (4 to 8 times) relative to NIST
253 SRM 3141a in delta notation:

254
$$\delta^{41}\text{K} (\text{‰}) = \left\{ \frac{({}^{41}\text{K}/{}^{39}\text{K})_{\text{sample}}}{({}^{41}\text{K}/{}^{39}\text{K})_{\text{NIST SRM 3141a}}} - 1 \right\} \times 1000$$

255 Analytical uncertainties are reported as 2SD (standard deviation) and 95% c.i. (confidence
256 interval) in Table 1 and Table S1. In the following sections, analytical uncertainty used for
257 individual sample is 95% c.i. and that used for group average is 2SD unless stated otherwise.

258

259 **3 Results and discussions**

260 **3.1 Analytical accuracy**

261 We first evaluate the accuracy of our analyses by comparing the $\delta^{41}\text{K}$ values of
262 representative terrestrial standards and chondrites from analyses in this study to previous results.
263 The $\delta^{41}\text{K}$ values of basalt BIR ($-0.418 \pm 0.031\text{‰}$) and BCR-2 ($-0.426 \pm 0.030\text{‰}$), andesite
264 AGV-2 ($-0.455 \pm 0.030\text{‰}$), and Pacific seawater ($0.126 \pm 0.028\text{‰}$) analyzed during the course
265 of this study agree well with published results (Chen et al., 2019; Hille et al., 2019; Hu et al.,
266 2018; Moynier et al., 2021; Xu et al., 2019). In addition, Orgueil ($-0.081 \pm 0.066\text{‰}$), Murchison
267 ($-0.250 \pm 0.043\text{‰}$), Allende ($-0.169 \pm 0.030\text{‰}$), Cold Bokkeveld ($-0.267 \pm 0.035\text{‰}$), and
268 Lancé ($-0.382 \pm 0.040\text{‰}$) processed and analyzed in this study yielded consistent values with
269 our previous analyses (Hu et al., 2022; Hu et al., 2023). This consistency confirms the accuracy
270 of our analyses.

271 We then evaluate whether the multi-stage chemical separation procedures performed at
272 Tokyo Tech affected the isotopic compositions of K in the samples. For Tagish Lake, the K
273 extracted from the K-Mg-Ni fraction has a $\delta^{41}\text{K}$ value of $-0.180 \pm 0.049\text{‰}$, which is
274 indistinguishable from the values obtained from two new dissolutions of separate pieces of this
275 meteorite from Tokyo Tech ($-0.215 \pm 0.043\text{‰}$) and IPGP ($-0.215 \pm 0.032\text{‰}$). Likewise, the K
276 purified from the K-Mg-Ni fraction of Tarda ($-0.223 \pm 0.037\text{‰}$) is isotopically identical to that
277 purified from a separate piece of this meteorite at IPGP ($-0.205 \pm 0.024\text{‰}$). This similarity in
278 $\delta^{41}\text{K}$ values suggests that the preliminary chemical processing performed at Tokyo Tech affected
279 K negligibly. In support of this inference, the anion resin used to separate Zn from other
280 constituent elements in the samples should not retain any K^+ , which would be eluted from the
281 resin. Therefore, it is reasonable to assume that the $\delta^{41}\text{K}$ values of K extracted from the K-Mg-Ni
282 fraction of the three Ryugu samples are representative of these samples. **While the K-Mg-Ni**

283 fraction of Allende ($-0.292 \pm 0.044\text{‰}$) is isotopically different from our in-house Allende
284 sample ($-0.169 \pm 0.030\text{‰}$), the $\delta^{41}\text{K}$ values of both samples fall within the reported range from
285 $-0.620 \pm 0.050\text{‰}$ to $-0.082 \pm 0.030\text{‰}$ (Bloom et al., 2020; Ku and Jacobsen, 2020; Nie et al.,
286 2021; Jiang et al., 2021; Hu et al., 2023; Koefoed et al., 2023). The discrepancy between the two
287 Allende samples thus likely reflects sample heterogeneity.

288

289 **3.2 Stable K isotopic compositions of CI chondrites and the bulk Solar System**

290 The three Ryugu samples have K abundances that are comparable to those of the CI
291 chondrites (Fig. 1A). In addition, they display consistent $\delta^{41}\text{K}$ values (Fig. 1B), with no
292 significant differences between the samples collected from the touchdown site 1 (Chamber A
293 sample A0106 = $-0.172 \pm 0.039\text{‰}$) and the touchdown site 2 (Chamber C samples C0107 =
294 $-0.204 \pm 0.048\text{‰}$ and C0108 = $-0.207 \pm 0.055\text{‰}$). Furthermore, the two samples from Chamber
295 C have essentially identical $\delta^{41}\text{K}$ values, indicating that the extraction of soluble organic matter
296 from samples C0107 and A0106 prior to K separation did not affect their K isotopic
297 compositions. The three Ryugu samples yield an average $\delta^{41}\text{K}$ value of $-0.194 \pm 0.038\text{‰}$ (2SD),
298 which falls within the range of CI chondrites (Fig. 1B). This result is consistent with previous
299 observations of CI-like isotopic compositions of Ryugu for the moderately volatile elements Zn
300 and Cu (Paquet et al., 2022) and the more refractory elements Fe, Cr, Ti, and Ca (Hopp et al.,
301 2022; Moynier et al., 2022; Yokoyama et al., 2022). The stable K isotope ratios thus further
302 strengthen the compositional links between Ryugu and CI chondrites.

303 The new data from Ryugu samples provide critical constraints on the K isotopic composition
304 of the bulk Solar System, which can otherwise only be inferred from CI chondrites. Previous
305 analyses of CI chondrites show highly variable $\delta^{41}\text{K}$ values from -0.534‰ to -0.039‰ ,
306 covering almost the entire variation measured in carbonaceous chondrites (Hu et al., 2022, 2023;
307 Koefoed et al., 2022, 2023; Ku and Jacobsen, 2020; Nie et al., 2021; Wang and Jacobsen, 2016).
308 Ku and Jacobsen (2020) found that the lowest $\delta^{41}\text{K}$ values were measured for a piece of Orgueil
309 contaminated with 8% continental crust material, resulting in its substantially higher K
310 abundance and lower $\delta^{41}\text{K}$ value than other Orgueil samples (Fig. 1). In contrast, the 11 other
311 Orgueil samples show a limited K isotopic range from -0.290‰ to -0.039‰ , with six stones
312 reported by Koefoed et al. (2023) having similar values between -0.290‰ and -0.170‰ (Fig.

313 1B). All these analyses together give an average $\delta^{41}\text{K}$ value of $-0.182 \pm 0.148\text{‰}$ (2SD),
314 suggesting a largely homogeneous K isotopic composition in Orgueil. The small variability
315 between different Orgueil samples could be due to aqueous redistribution, as suggested by a
316 slightly negative correlation between the $\delta^{41}\text{K}$ value and K abundance (Fig. 1B).

317 Unlike Orgueil, however, there is a large discrepancy in the $\delta^{41}\text{K}$ values reported for two
318 separate analyses of the CI-type specimen Ivuna. The lower value of $-0.460 \pm 0.046\text{‰}$ does not
319 appear to reflect terrestrial contamination because this Ivuna sample has a normal K abundance
320 (458 ppm, Nie et al., 2021) as the uncontaminated Orgueil samples and the other Ivuna sample,
321 which has a $\delta^{41}\text{K}$ value ($-0.180 \pm 0.060\text{‰}$, Koefoed et al., 2023) similar to the Orgueil average
322 calculated above (Fig. 1B). Given that this lower Ivuna value is significantly lower than those of
323 the Ryugu samples, it likely reflects an unrepresentative sampling of the bulk Ivuna composition.
324 This inference is supported by the heterogenous distribution of soluble elements in CI chondrites,
325 in particular for Ivuna (Barrat et al., 2012; Brearley, 2006; Morlok et al., 2006; Palme and Zipfel,
326 2022) (Fig. 1A). Excluding the anomalous Ivuna value, representative averages of Ryugu
327 ($-0.194 \pm 0.022\text{‰}$, 2SE) and Orgueil ($-0.182 \pm 0.045\text{‰}$, 2SE) and a single Ivuna value (-0.180
328 $\pm 0.060\text{‰}$, 2SE, Koefoed et al., 2023) define an arithmetic mean of $-0.185 \pm 0.078\text{‰}$ (2SE).
329 This $\delta^{41}\text{K}$ mean agrees with the recently proposed average CI value of $-0.21 \pm 0.05\text{‰}$ (Koefoed
330 et al., 2023) calculated by averaging 15 published CI data (including the anomalous Ivuna
331 value). The agreement between the two estimates further supports a relatively homogeneous K
332 isotopic composition of the bulk Solar System.

333

334 3.3 Origin of K isotopic variations in carbonaceous chondrites

335 A well-established average $\delta^{41}\text{K}$ value of CI chondrites is an important reference for
336 understanding the origin of K isotopic variations among different carbonaceous chondritic
337 groups. This is because CI chondrites are generally used to approximate the isotopic composition
338 of the matrix component in carbonaceous chondrites (e.g., Alexander, 2005; Braukmüller et al.,
339 2018), and a robust $\delta^{41}\text{K}$ value for CI chondrites is critical for explaining the K isotopic
340 variations in carbonaceous chondrites (Bloom et al., 2020; Ku and Jacobsen, 2020; Nie et al.,
341 2021; Hu et al., 2022, 2023; Koefoed et al., 2022, 2023). The average $\delta^{41}\text{K}$ values established for
342 CI chondrites and Ryugu samples in this study agree with each other within uncertainty (Fig. 2).

343 Their $\delta^{41}\text{K}$ values also overlap within uncertainty with those of the two ungrouped C2 chondrites
344 (Tagish Lake and Tarda), which contain the second-highest matrix volume among carbonaceous
345 chondrites (Fig. 2). These observations indicate that the CI-like matrix in carbonaceous
346 chondrites is largely homogeneous in $\delta^{41}\text{K}$.

347 Compared with CI chondrites and Ryugu samples, carbonaceous chondrites from the CM
348 (Murchison), CO (Lancé), and CV (Allende) groups are variably depleted in the heavier K
349 isotope, and their $\delta^{41}\text{K}$ values correlate positively with the $\delta^{66}\text{Zn}$ and $\delta^{65}\text{Cu}$ values (Fig. 2).
350 Positive correlations are also evident between $\delta^{41}\text{K}$ with isotopes of other MVEs, including Rb,
351 Sn, and Te, in a larger dataset of carbonaceous chondrites (Hu et al., 2023; Koefoed et al., 2023;
352 Nie et al., 2021). **In addition**, a broad negative correlation between $\delta^{41}\text{K}$ and $1/[\text{K}]$ was reported
353 by Nie et al. (2021) for seven carbonaceous chondrites, from which they extrapolated a $\delta^{41}\text{K}$
354 value of $-0.33 \pm 0.12\text{‰}$ for the chondrule component and of $0.04 \pm 0.08\text{‰}$ for the matrix
355 component. These systematic correlations reflect the early formation of two nebular reservoirs,
356 with a volatile-depleted refractory component (e.g., chondrules) carrying the light isotopic
357 signatures, and a CI-like reservoir with the heaviest isotopic compositions, as initially proposed
358 by Luck et al. (2005; 2003). When all available data are considered, there is a coinciding
359 increase in $\delta^{41}\text{K}$ with water content (as a proxy for matrix content) and a decrease in $\delta^{41}\text{K}$ with
360 chondrule content in chondrites, further substantiating the role of chondrules as the carriers of K
361 depletion and low- $\delta^{41}\text{K}$ signature in chondrites (Hu et al., 2023). Therefore, the inter-group K
362 isotopic variations in carbonaceous chondrites primarily reflect variable mixing between the two
363 main K hosts, i.e., chondrules and the matrix, accreted by the parent bodies of carbonaceous
364 chondrites.

365

366 **3.4 Potassium isotopic evidence for widespread aqueous alteration in chondrite parent** 367 **bodies**

368 Although chondrule-matrix mixing could explain the inter-group $\delta^{41}\text{K}$ variation among
369 carbonaceous chondrites, the significant variability within individual chondrites remains
370 perplexing. In addition to the anomalously low $\delta^{41}\text{K}$ value reported for Ivuna, substantial
371 variability has also been documented for Allende and Murchison. Our in-house Allende sample
372 has a higher $\delta^{41}\text{K}$ value ($-0.169 \pm 0.030\text{‰}$) than the sample provided by the Smithsonian

373 Museum ($-0.292 \pm 0.044\text{‰}$), while they both fall within the previously reported range of
374 -0.620‰ to -0.080‰ for Allende (Bloom et al., 2020; Jiang et al., 2021a; Koefoed et al., 2023;
375 Ku and Jacobsen, 2020; Nie et al., 2021). Furthermore, the Allende sample provided by the
376 Smithsonian Museum is an aliquot from a large batch of homogenized powder, and its $\delta^{41}\text{K}$
377 value is in the middle of the reported range. Our Murchison sample also has a $\delta^{41}\text{K}$ value (-0.250
378 $\pm 0.043\text{‰}$) that is in the middle of the previously reported range for Murchison (-0.410‰ to
379 -0.101‰ , Jiang et al., 2021b; Koefoed et al., 2023; Ku and Jacobsen, 2020; Nie et al., 2021).

380 To identify the cause of intra-chondrite isotopic variability, we compare the two sets of $\delta^{41}\text{K}$
381 vs. $\delta^{87}\text{Rb}$ data obtained from IPGP and those reported in Nie et al. (2021), respectively, for four
382 carbonaceous chondrites. While the two datasets differ by varying extents for individual
383 chondrites, the $\delta^{41}\text{K}$ and $\delta^{87}\text{Rb}$ values correlate in both datasets (Fig. 3A). For example, the
384 Murchison specimen analyzed at IPGP has both lower $\delta^{41}\text{K}$ and $\delta^{87}\text{Rb}$ values than the specimen
385 reported by Nie et al. (2021). The correlation between $\delta^{41}\text{K}$ and $\delta^{87}\text{Rb}$ values suggests that the
386 variability reflects isotopic heterogeneity at the analyzed sample scale. This isotopic
387 heterogeneity could reflect different proportions of chondrules present in the two different
388 specimens of a given chondrite. In this case, the specimen containing more chondrules should
389 have a lower K abundance and a lower $\delta^{41}\text{K}$ value because chondrules are inferred to be depleted
390 in K and its lighter isotope compared with the matrix, as indicated by the mixing curve in Fig.
391 3B. However, the specimen with the lower K abundance is often found to have a higher $\delta^{41}\text{K}$
392 value than that with the higher K abundance of the same chondrite (Fig. 3B), which is opposite
393 to the expectation of mixing between varying proportions of chondrules with the matrix.
394 Therefore, the observed sample heterogeneity cannot be due to the chondrule nugget effect alone
395 but, to a greater extent, reflects the redistribution of K and its isotopes during aqueous alteration
396 in chondrite parent bodies, which significantly altered the original K abundances and isotopic
397 compositions in chondrules and matrix.

398 The effects of asteroidal aqueous alteration on the redistribution of alkalis in carbonaceous
399 chondrites have been well documented, including the formation of diverse K-bearing
400 phyllosilicate phases at $< 100^\circ\text{C}$ and/or anhydrous phases (e.g., nepheline and sodalite) due to
401 Fe-alkali-halogen metasomatism at $\sim 200^\circ\text{C}$ - 300°C (e.g., Brearley, 2014; Ikeda and Kimura, 1996;
402 Kimura and Ikeda, 1998; Krot et al., 1998; Krot et al., 1995). Ordinary chondrites also show
403 signs, albeit more subtle, of interaction with aqueous fluids at varying temperatures, ranging

404 from low-temperature alteration (< 200°C, in the least-equilibrated petrologic type 3) to fluid-
405 assisted metamorphism (200-400°C, in petrologic types of 4-6). Before or during the earliest
406 stage of thermal metamorphism, aqueous fluids were suggested to have flowed through the fine-
407 grained matrix, dissolving the mesostasis near the edge of chondrules and migrating alkalis into
408 chondrule cores (e.g., Grossman et al., 2000; Grossman and Brearley, 2005). The presence of
409 fluids is evidenced by the brine-bearing salt crystals found in the Monahans and Zag ordinary
410 chondrites, which are argued to represent fluid samples from the early Solar System (Rubin et
411 al., 2002; Zolensky et al., 1999). More recently, fluid activity on S-type asteroids is further
412 supported by nanometer-sized halite crystals detected in particles returned from the asteroid
413 Itokawa by JAXA's Hayabusa mission (Che and Zega, 2023). These observations collectively
414 suggest the widespread action of fluids on primitive asteroids in the early Solar System.

415 Fluid-rock interactions have also been invoked to explain K isotopic variations in chondritic
416 components of the Allende and Hamlet (LL4) chondrites (Jiang et al., 2021a; Koefoed et al.,
417 2020). While inter-group variations in $\delta^{41}\text{K}$ are attributable to mixing between a CI-like matrix
418 and K-poor, isotopically light chondrules, direct measurements on Allende and Hamlet show
419 highly variable $\delta^{41}\text{K}$ in chondrules with values both higher and lower than those of the matrices
420 (Fig. 3C and D). Furthermore, Ca-Al-inclusions (CAIs) formed well above the condensation
421 temperature of K contain anomalously high K concentrations and $\delta^{41}\text{K}$ values. These
422 observations are inconsistent with the low $\delta^{66}\text{Zn}$ and $\delta^{71}\text{Ga}$ values measured in CAIs interpreted
423 as reflecting incomplete condensation (Kato et al., 2017; Luck et al., 2005). Instead of pristine
424 nebular signatures, Jiang et al. (2021a) considered the high $\delta^{41}\text{K}$ values of CAIs as a secondary
425 feature inherited from aqueous fluids assuming that most K in the CAIs was secondary in origin
426 (Fig. 3C). Furthermore, they proposed that as the fluids interact with the porous matrix, the
427 heavier K isotope was preferentially leached from the matrix and entered chondrules or CAIs,
428 overprinting the original K abundances and isotopic signatures of the chondritic components.

429 The preference for the heavier K isotope by aqueous fluids is supported by the direction of K
430 isotope fractionation during chemical weathering on Earth. Global rivers (-0.59‰ to 0.12‰)
431 and oceans ($0.12 \pm 0.07\text{‰}$, 2SD) have heavier K isotopic compositions (Hille et al., 2019; Li et
432 al., 2019; Wang et al., 2020; Wang et al., 2021) than fresh basaltic rocks averaging at $-0.42 \pm$
433 0.08‰ (2SD) (Hu et al., 2021; Tuller-Ross et al., 2019a; Tuller-Ross et al., 2019b). In addition,
434 studies of weathering profiles and riverine sediments indicate that aqueous fluids preferentially

435 leach the heavy K isotope from the bedrock with apparent isotope fractionation factors of 0.08‰
436 to 0.55‰, leaving behind isotopically light weathered residues (down to $-0.94‰$, Chen et al.,
437 2020; Li et al., 2019; Teng et al., 2020). Consistent with field analyses, theoretical calculations
438 based on K-O bond strengths also suggest an enrichment of the heavier K isotope in aqueous
439 solutions relative to typical silicate minerals and a fractionation factor of 0.24‰ between
440 aqueous fluids and illite (Zeng et al., 2019).

441 At the onset of aqueous alteration, the K abundance and $\delta^{41}\text{K}$ value of matrix would have
442 decreased, while the chondrules or CAIs would have increased K abundances and $\delta^{41}\text{K}$ values.
443 With increasing degrees of alteration, chondrules, CAIs, and matrix would tend to approach
444 equilibrium. The significant $\delta^{41}\text{K}$ variability in Allende and Murchison (Fig. 2) and the fact that
445 the specimen with a lower K abundance often has a higher $\delta^{41}\text{K}$ value (Fig. 3B) suggest that
446 these chondrites record early stages of aqueous alteration. A broad negative correlation between
447 $\delta^{41}\text{K}$ values and K abundances is also observed in major groups of carbonaceous chondrites (Fig.
448 S1), suggesting that aqueous alteration and associated redistribution of K isotopes occurred
449 pervasively in the parent bodies of carbonaceous chondrites. Therefore, K isotopic variations in
450 carbonaceous chondrites record volatility-driven isotope fractionations during chondrule
451 formation, variable chondrule-matrix mixing at the accretion of chondrite parent bodies, and
452 post-accretionary redistribution of K in chondrite parent bodies assisted by aqueous fluids.

453

454 **4 Conclusions**

455 Previous estimates of the K isotopic composition of the bulk Solar System were based
456 heavily on CI chondrite Orgueil, whereas $\delta^{41}\text{K}$ values reported for two fragments of the CI-type
457 specimen Ivuna show a large discrepancy. This study reports the K isotopic composition of the
458 first samples returned from a carbonaceous asteroid (Ryugu) by JAXA's Hayabusa2 mission,
459 thereby allowing for a direct comparison of CI chondrites with pristine asteroidal samples not
460 contaminated by terrestrial environments. Analyses of three aliquots of Ryugu particles yield
461 similar $\delta^{41}\text{K}$ values between $-0.207‰$ and $-0.172‰$. This range is consistent with the K isotopic
462 compositions of most CI chondrites except for lower $\delta^{41}\text{K}$ values found in contaminated Orgueil
463 samples and an anomalously low value previously reported for Ivuna. Based on the Ryugu
464 samples and representative analyses of CI chondrites, we provide an updated estimate of the $\delta^{41}\text{K}$

465 value for the bulk Solar System ($-0.185 \pm 0.078\%$, 2SE). We suggest that samples of CI
466 chondrites with $\delta^{41}\text{K}$ values significantly deviating from this range may reflect compositional
467 heterogeneities due to aqueous alteration in CI parent bodies. Considerable K isotopic
468 heterogeneity is also observed in other carbonaceous chondrites and within individual chondritic
469 groups, whereby $\delta^{41}\text{K}$ values are broadly negatively correlated with K abundances. These
470 observations suggest the widespread redistribution of K and its isotopes during aqueous
471 alteration in chondrite parent bodies.

472

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481

482 All data referred to in this article can be found in the tables.

483

484 **Tables**485 **Table 1.** Potassium abundance (ppm) and isotopic composition (‰) of Ryugu samples and
486 carbonaceous chondrites analyzed in this study.

| Sample | Group | $\delta^{41}\text{K}$ (‰) | 2SD (‰) | 95% c.i. (‰) | N | [K] (ppm) | Source | Treatment |
|------------------------|--------|---------------------------|---------|--------------|---|-----------|------------|----------------------------|
| Ryugu particles | | | | | | | | |
| A0106 | | -0.150 | 0.065 | 0.060 | 4 | 436 | Tokyo Tech | SOM, Zn,Fe,Ti,Cr chemistry |
| duplicate | | -0.172 | 0.085 | 0.056 | 5 | | | |
| repeat | | -0.191 | 0.059 | 0.048 | 5 | | | |
| Wtd average | | -0.172 | 0.040 | 0.039 | | | | |
| C0107 | | -0.192 | 0.070 | 0.040 | 6 | 457 | Tokyo Tech | SOM, Zn,Fe,Ti,Cr chemistry |
| duplicate | | -0.214 | 0.066 | 0.056 | 5 | | | |
| Wtd average | | -0.204 | 0.030 | 0.048 | | | | |
| C0108 | | -0.188 | 0.086 | 0.059 | 5 | 604 | Tokyo Tech | Zn,Fe,Ti,Cr chemistry |
| repeat | | -0.220 | 0.072 | 0.069 | 4 | | | |
| Wtd average | | -0.207 | 0.045 | 0.055 | | | | |
| Ryugu average | | -0.194 | 0.038 | | | | | |
| Carbonaceous | | | | | | | | |
| Orgueil | CI1 | -0.081 | 0.070 | 0.066 | 5 | 568 | IPGP | Powder dissolution |
| Tagish lake-1 | C2-ung | -0.178 | 0.097 | 0.050 | 5 | 358 | Tokyo Tech | Zn,Fe,Ti,Cr chemistry |
| duplicate | | -0.181 | 0.057 | 0.048 | 5 | | | |
| replicate | | -0.215 | 0.057 | 0.043 | 6 | | | Powder dissolution |
| Wtd average | | -0.195 | 0.042 | 0.037 | | | | |
| Tagish lake-2 | | -0.201 | 0.042 | 0.023 | 8 | 360 | IPGP | Powder dissolution |
| repeat | | -0.236 | 0.051 | 0.030 | 7 | | | |
| Wtd average | | -0.215 | 0.049 | 0.032 | | | | |
| Tagish lake average | | -0.205 | 0.029 | | | | | |
| Tarda-1 | C2-ung | -0.217 | 0.067 | 0.054 | 6 | 416 | Tokyo Tech | Zn,Fe,Ti,Cr chemistry |
| duplicate | | -0.230 | 0.076 | 0.037 | 4 | | | |
| repeat | | -0.224 | 0.056 | 0.045 | 5 | | | |
| Wtd average | | -0.223 | 0.013 | 0.037 | | | | |
| Tarda-2 | | -0.166 | 0.063 | 0.050 | 5 | 416 | IPGP | Powder dissolution |
| repeat | | -0.222 | 0.054 | 0.047 | 5 | | | |
| repeat | | -0.218 | 0.058 | 0.039 | 4 | | | |
| replicate | | -0.204 | 0.056 | 0.038 | 7 | | | Powder dissolution |
| duplicate | | -0.189 | 0.060 | 0.047 | 6 | | | |
| repeat | | -0.222 | 0.060 | 0.050 | 5 | | | |
| Wtd average | | -0.205 | 0.044 | 0.024 | | | | |
| Tarda average | | -0.214 | 0.026 | | | | | |
| Murchison | CM2 | -0.250 | 0.049 | 0.043 | 7 | 411 | IPGP | Powder dissolution |
| Cold Bokkeveld | CM2 | -0.267 | 0.064 | 0.035 | 7 | 476 | IPGP | Powder dissolution |
| Lancé | CO3.5 | -0.382 | 0.087 | 0.040 | 7 | 430 | IPGP | Powder dissolution |
| Allende (A)-1 | CV3 | -0.293 | 0.056 | 0.052 | 5 | 296 | Tokyo Tech | Zn,Fe,Ti,Cr chemistry |
| repeat | | -0.292 | 0.072 | 0.048 | 5 | | | |
| Wtd average | | -0.292 | 0.001 | 0.044 | | | | |
| Allende-2 | | -0.151 | 0.062 | 0.059 | 5 | 267 | IPGP | Powder dissolution |
| repeat | | -0.181 | 0.062 | 0.055 | 4 | | | |
| repeat | | -0.168 | 0.052 | 0.045 | 6 | | | |
| replicate | | -0.178 | 0.065 | 0.031 | 4 | | | Powder dissolution |
| Wtd average | | -0.169 | 0.027 | 0.030 | | | | |

487 Note: 'repeat' indicates repeat instrumental analysis. 'duplicate' indicate repeat column chemistry and instrumental
488 analysis. 'replicate' indicates repeat sample dissolution, column chemistry, and instrumental analysis. The number '-
489 1' and '-2' attached to the same names indicate different pieces of a given chondrite. 'SOM' is short for 'Soluble
490 Organic Matter extraction.'

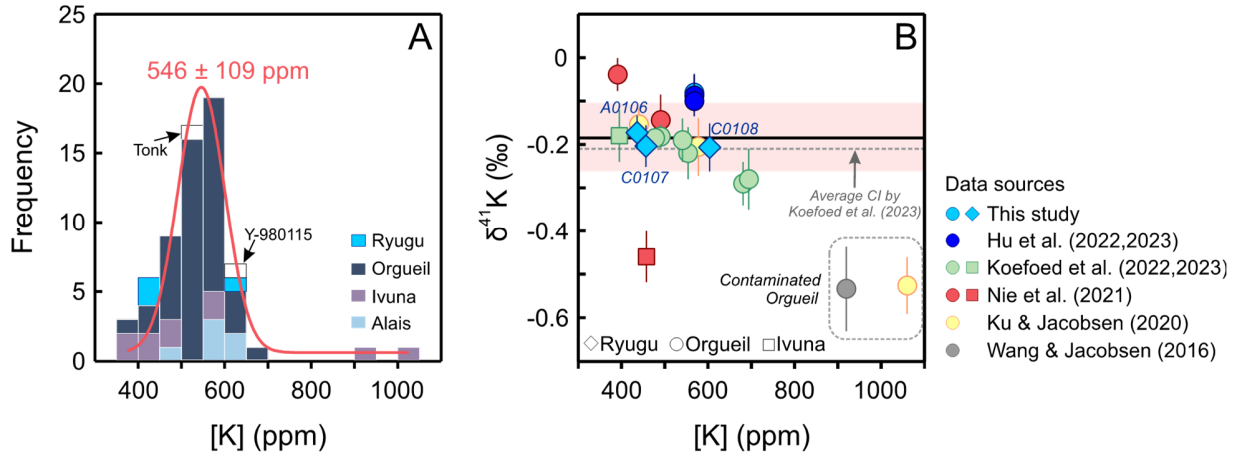
491 **Table S1.** Potassium isotopic compositions (‰) of terrestrial samples analyzed in this study.

| Geostandard | Description | $\delta^{41}\text{K}$ (‰) | 2SD (‰) | 95% c.i. (‰) | N |
|------------------|-------------|---------------------------|---------|--------------|---|
| BIR | Basalt | -0.448 | 0.059 | 0.036 | 7 |
| | replicate | -0.400 | 0.052 | 0.076 | 4 |
| | repeat | -0.414 | 0.050 | 0.062 | 4 |
| | Wtd average | -0.418 | 0.050 | 0.031 | |
| BCR-2 | Basalt | -0.404 | 0.051 | 0.019 | 8 |
| | repeat | -0.443 | 0.044 | 0.024 | 7 |
| | replicate | -0.425 | 0.066 | 0.034 | 5 |
| | Wtd average | -0.426 | 0.039 | 0.030 | |
| AGV-2 | Andesite | -0.455 | 0.042 | 0.030 | 7 |
| Pacific seawater | | 0.152 | 0.060 | 0.054 | 4 |
| | repeat | 0.128 | 0.078 | 0.050 | 6 |
| | repeat | 0.113 | 0.039 | 0.037 | 7 |
| | repeat | 0.133 | 0.084 | 0.056 | 5 |
| | Wtd average | 0.126 | 0.032 | 0.028 | |

492

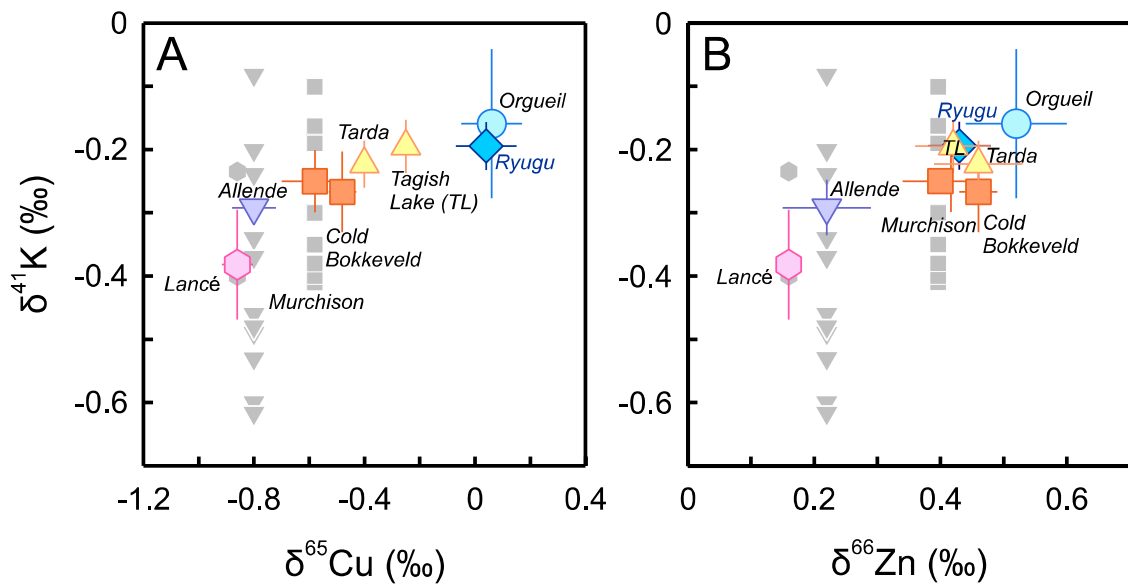
493 **Figures**

494



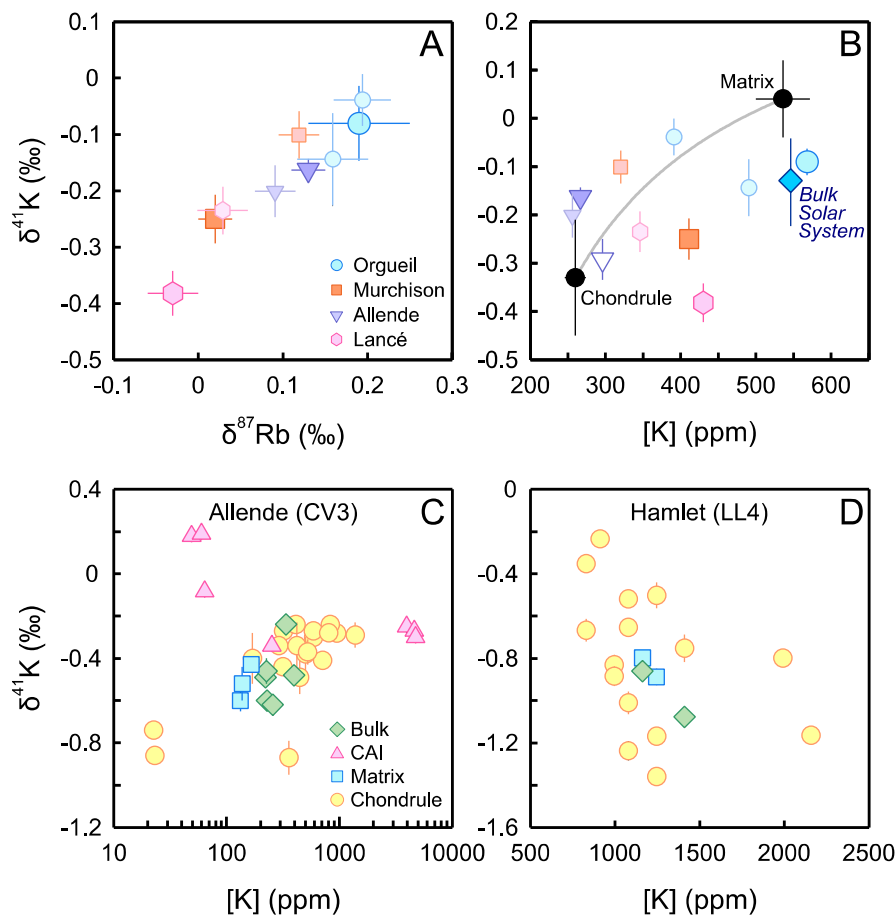
495

496 **Figure 1.** Potassium abundances ([K], ppm) and isotope compositions ($\delta^{41}\text{K}$, ‰) of CI
 497 chondrites and Ryugu samples. (A) A histogram of K abundances in CI chondrites reported in
 498 the literature (Ahrens et al., 1969; Anders and Grevesse, 1989; Barrat et al., 2012; Beer et al.,
 499 1984; Braukmüller et al., 2018; Edwards and Urey, 1955; Jarosewich, 1990; Kallemeyn and
 500 Wasson, 1981; Kaushal and Wetherill, 1970; Makishima and Nakamura, 2006; Mittlefehldt,
 501 2002; Nakamura, 1974; Nichiporuk and Moore, 1974; Palme and Zipfel, 2022; Von Michaelis et
 502 al., 1969; Wiik, 1956). (B) The three aliquots of Ryugu particles have indistinguishable $\delta^{41}\text{K}$
 503 values and fall within the range of CI chondrites except for an anomalously lower $\delta^{41}\text{K}$ value
 504 previously reported for Ivuna. The horizontal black line and pink field indicate the average $\delta^{41}\text{K}$
 505 value of Ryugu samples and CI chondrites ($-0.185 \pm 0.078\text{‰}$), representing the current best
 506 estimate of the bulk K isotopic composition of the Solar System. **The dashed grey line is the**
 507 **average CI chondrites reported by Koefoed et al. (2023) and is plotted for comparison.** Crustal
 508 contamination results in elevated K abundances and crust-like, lower $\delta^{41}\text{K}$ compared with
 509 pristine CI chondrite values (Ku and Jacobsen, 2020; Wang and Jacobsen, 2016). The K
 510 abundance of Ryugu C0108 is from Yokoyama et al. (2022), while those of A0106 and C0107
 511 are new data reported in this study (Table 1).



512

513 **Figure 2.** Positive correlations of $\delta^{41}\text{K}$ values with (A) $\delta^{65}\text{Cu}$ values and (B) $\delta^{66}\text{Zn}$ values in
 514 Ryugu samples and different groups of carbonaceous chondrites. These correlations suggest
 515 inter-group variations in $\delta^{41}\text{K}$ primarily reflect mixing of a CI-, or Ryugu-like matrix component
 516 with the heaviest isotopic compositions and a volatile-depleted refractory component such as
 517 chondrules carrying the light isotopic signatures. The colored symbols are data from this study
 518 and the grey symbols are literature values. In (A), $\delta^{41}\text{K}$ and $\delta^{65}\text{Cu}$ values (Paquet et al., 2022) for
 519 Ryugu, Tarda, Allende were measured from the same specimens. $\delta^{65}\text{Cu}$ values for Orgueil and
 520 Murchison are from Paquet et al. (2022), for Tagish Lake, Cold Bokkeveld, and Lancé are from
 521 Luck et al. (2003). In (B), $\delta^{41}\text{K}$ and $\delta^{66}\text{Zn}$ values (Paquet et al., 2022; Pringle et al., 2017) for
 522 Ryugu, Tagish Lake, Tarda, Murchison, Cold Bokkeveld, Lancé, and Allende were measured
 523 from the same specimens. $\delta^{66}\text{Zn}$ value for Orgueil is from Paquet et al. (2022). Literature $\delta^{41}\text{K}$
 524 values are from Bloom et al. (2020); Jiang et al. (2021a); Jiang et al. (2021b); Koefoed et al.
 525 (2023); Koefoed et al. (2022); Ku and Jacobsen (2020); Nie et al. (2021). Although literature
 526 $\delta^{41}\text{K}$ values were plotted with $\delta^{65}\text{Cu}$ and $\delta^{66}\text{Zn}$ values measured from different specimens of a
 527 given chondrite, $\delta^{65}\text{Cu}$ and $\delta^{66}\text{Zn}$ values reported for different specimens of Orgueil, Murchison,
 528 and Allende generally agree well (Barrat et al., 2012; Luck et al., 2005; Luck et al., 2003; Paquet
 529 et al., 2022; Pringle et al., 2017).



530

531 **Figure 3.** K isotopic evidence of aqueous alteration of chondrites in their parent bodies. (A)
 532 Positively correlated $\delta^{41}\text{K}$ and $\delta^{87}\text{Rb}$ values indicate sample heterogeneity for a given chondrite.
 533 The larger symbols are data from this study and the smaller symbols are from Nie et al. (2021).
 534 (B) The broad positive correlation between $\delta^{41}\text{K}$ values and K abundances across different
 535 carbonaceous chondrite groups indicates mixing of the CI-like, isotopically heavy matrix with K-
 536 depleted, isotopically light chondrules, as indicated by the mixing curve using endmember
 537 compositions extrapolated in Nie et al. (2021). However, for a given chondrite (coded with the
 538 same color), the specimen with the higher K abundance often has a lower $\delta^{41}\text{K}$ value than that
 539 with the lower K abundance. The large solid triangle down is an Allende specimen from IPGP,
 540 and the open triangle down is an aliquot of pulverized Allende provided by the Smithsonian
 541 Museum. **The blue diamond symbol indicates the new estimate of the composition of the bulk**
 542 **Solar System from this study.** (C) and (D) $\delta^{41}\text{K}$ values measured in chondrules, calcium-
 543 aluminum-rich inclusions (CAIs), and matrices in Allende and Hamlet chondrites show
 544 significant alteration of the original K abundances and $\delta^{41}\text{K}$ values of all chondritic components.
 545 Data source: Data presented in (A) and (B) are from this study and the literature (Nie et al., 2021;
 546 Pringle and Moynier, 2017), whereas those presented in (C) and (D) are from Jiang et al.
 547 (2021a); Ku et al. (2022), and Koefoed et al. (2020).

548 **References**

- 549 Ahrens, L. H., Von Michaelis, H., Fesq, H. W., 1969. The composition of the stony meteorites (IV) some
550 analytical data on Orgueil, Nogoya, Ornans and Ngawi. *Earth and Planetary Science Letters*. 6,
551 285-288.
- 552 Alexander, C. M. O. D., 2005. Re-examining the role of chondrules in producing the elemental
553 fractionations in chondrites. *Meteoritics & Planetary Science*. 40, 943-965.
- 554 Anders, E., Grevesse, N., 1989. Abundances of the elements: Meteoritic and solar. *Geochimica et*
555 *Cosmochimica Acta*. 53, 197-214.
- 556 Arakawa, M., et al., 2020. An artificial impact on the asteroid (162173) Ryugu formed a crater in the
557 gravity-dominated regime. *Science*. 368, 67-71.
- 558 Barrat, J. A., Zanda, B., Moynier, F., Bollinger, C., Liorzou, C., Bayon, G., 2012. Geochemistry of CI
559 chondrites: Major and trace elements, and Cu and Zn Isotopes. *Geochimica et Cosmochimica*
560 *Acta*. 83, 79-92.
- 561 Beer, H., Walter, G., Macklin, R. L., Patchett, P. J., 1984. Neutron capture cross sections and solar
562 abundances of ^{160,161}Dy, ^{170,171}Yb, ^{175,176}Lu, and ^{176,177}Hf for the s-process analysis of
563 the radionuclide ¹⁷⁶Lu. *Physical Review C*. 30, 464-478.
- 564 Bloom, H., et al., 2020. Potassium isotope compositions of carbonaceous and ordinary chondrites:
565 Implications on the origin of volatile depletion in the early solar system. *Geochimica et*
566 *Cosmochimica Acta*. 277, 111-131.
- 567 Braukmüller, N., Wombacher, F., Hezel, D. C., Escoube, R., Münker, C., 2018. The chemical
568 composition of carbonaceous chondrites: Implications for volatile element depletion,
569 complementarity and alteration. *Geochimica et Cosmochimica Acta*. 239, 17-48.
- 570 Brearley, A. J., 2006. The action of water. *Meteorites and the early solar system II*. 943, 587-624.
- 571 Brearley, A. J., 2014 1.9 - Nebular Versus Parent Body Processing. In: H. D. Holland, K. K. Turekian,
572 (Eds.), *Treatise on Geochemistry (Second Edition)*. Elsevier, Oxford, pp. 309-334.
- 573 Che, S., Zega, T. J., 2023. Hydrothermal fluid activity on asteroid Itokawa. *Nature Astronomy*.
- 574 Chen, H., Liu, X.-M., Wang, K., 2020. Potassium isotope fractionation during chemical weathering of
575 basalts. *Earth and Planetary Science Letters*. 539, 116192.
- 576 Chen, H., Tian, Z., Tuller-Ross, B., Korotev, Randy L., Wang, K., 2019. High-precision potassium
577 isotopic analysis by MC-ICP-MS: an inter-laboratory comparison and refined K atomic weight.
578 *Journal of Analytical Atomic Spectrometry*. 34, 160-171.
- 579 Clayton, R. N., 1993. Oxygen Isotopes in Meteorites. *Annual Review of Earth and Planetary Sciences*.
580 21, 115-149.
- 581 Edwards, G., Urey, H. C., 1955. Determination of alkali metals in meteorites by a distillation process.
582 *Geochimica et Cosmochimica Acta*. 7, 154-168.
- 583 Grossman, J. N., ALEXANDER, C. M. O. D., Wang, J., Brearley, A. J., 2000. Bleached chondrules:
584 Evidence for widespread aqueous processes on the parent asteroids of ordinary chondrites.
585 *Meteoritics & Planetary Science*. 35, 467-486.
- 586 Grossman, J. N., Brearley, A. J., 2005. The onset of metamorphism in ordinary and carbonaceous
587 chondrites. *Meteoritics & Planetary Science*. 40, 87-122.
- 588 Hille, M., Hu, Y., Huang, T.-Y., Teng, F.-Z., 2019. Homogeneous and heavy potassium isotopic
589 composition of global oceans. *Science Bulletin*. 64, 1740-1742.
- 590 Hopp, T., et al., 2022. Ryugu's nucleosynthetic heritage from the outskirts of the Solar System. *Science*
591 *Advances*. 8, eadd8141.
- 592 Hu, Y., Chen, X.-Y., Xu, Y.-K., Teng, F.-Z., 2018. High-precision analysis of potassium isotopes by HR-
593 MC-ICPMS. *Chemical Geology*. 493, 100-108.
- 594 Hu, Y., Moynier, F., Bizzarro, M., 2022. Potassium isotope heterogeneity in the early Solar System
595 controlled by extensive evaporation and partial recondensation. *Nature Communications*. 13,
596 7669.

597 Hu, Y., Moynier, F., Yang, X., 2023. Volatile-depletion processing of the building blocks of Earth and
598 Mars as recorded by potassium isotopes. *Earth and Planetary Science Letters*. 620, 118319.

599 Hu, Y., Teng, F.-Z., Helz, R. T., Chauvel, C., 2021. Potassium isotope fractionation during magmatic
600 differentiation and the composition of the mantle. *Journal of Geophysical Research: Solid Earth*.
601 126, e2020JB021543.

602 Ikeda, Y., Kimura, M., 1996. Anhydrous alteration of Allende chondrules in the solar nebula III: Alkali-
603 zoned chondrules and heating experiments for anhydrous alteration. *Antarctic Meteorite*
604 *Research*. 9, 51.

605 Jarosewich, E., 1990. Chemical analyses of meteorites: A compilation of stony and iron meteorite
606 analyses. *Meteoritics*. 25, 323-337.

607 Jiang, Y., et al., 2021a. Early solar system aqueous activity: K isotope evidence from Allende. *Meteoritics*
608 *& Planetary Science*. 56, 61-76.

609 Jiang, Y., Koefoed, P., Wang, K., Xu, W.-B., 2021b. High precision potassium isotopic study of Chinese
610 Antarctic chondrites. *Acta Geologica Sinica*. 95.

611 Kallemeyn, G. W., Wasson, J. T., 1981. The compositional classification of chondrites—I. The
612 carbonaceous chondrite groups. *Geochimica et Cosmochimica Acta*. 45, 1217-1230.

613 Kato, C., Moynier, F., Foriel, J., Teng, F.-Z., Puchtel, I. S., 2017. The gallium isotopic composition of the
614 bulk silicate Earth. *Chemical Geology*. 448, 164-172.

615 Kaushal, S. K., Wetherill, G. W., 1970. Rubidium 87-Strontium 87 age of carbonaceous chondrites.
616 *Journal of Geophysical Research (1896-1977)*. 75, 463-468.

617 Kimura, M., Ikeda, Y., 1998. Hydrous and anhydrous alterations of chondrules in Kaba and Mokoia CV
618 chondrites. *Meteoritics & Planetary Science*. 33, 1139-1146.

619 Kitazato, K., et al., 2019. The surface composition of asteroid 162173 Ryugu from Hayabusa2 near-
620 infrared spectroscopy. *Science*. 364, 272-275.

621 Koefoed, P., et al., 2023. The potassium isotopic composition of CI chondrites and the origin of isotopic
622 variations among primitive planetary bodies. *Geochimica et Cosmochimica Acta*.

623 Koefoed, P., Pravdivtseva, O., Chen, H., Gerritzen, C., Thiemens, M. M., Wang, K., 2020. Potassium
624 isotope systematics of the LL4 chondrite Hamlet: Implications for chondrule formation and
625 alteration. *Meteoritics & Planetary Science*. 55.

626 Koefoed, P., et al., 2022. The dynamic formation process of the CB chondrite Gujba. *Geochimica et*
627 *Cosmochimica Acta*. 332, 33-56.

628 Krot, A. N., Petaev, M. I., Scott, E. R. D., Choi, B.-G., Zolensky, M. E., Keil, K., 1998. Progressive
629 alteration in CV3 chondrites: More evidence for asteroidal alteration. *Meteoritics & Planetary*
630 *Science*. 33, 1065-1085.

631 Krot, A. N., Scott, E. R. D., Zolensky, M. E., 1995. Mineralogical and chemical modification of
632 components in CV3 chondrites: Nebular or asteroidal processing? *Meteoritics*. 30, 748-775.

633 Ku, Y., Jacobsen, S. B., 2020. Potassium isotope anomalies in meteorites inherited from the protosolar
634 molecular cloud. *Science Advances*. 6, eabd0511.

635 Ku, Y., Petaev, M. I., Jacobsen, S. B., 2022. The Timing of Potential Last Nucleosynthetic Injections into
636 the Protosolar Molecular Cloud Inferred from ⁴¹Ca–²⁶Al Systematics
637 of Bulk CAIs. *The Astrophysical Journal Letters*. 931, L13.

638 Li, S., et al., 2019. K isotopes as a tracer for continental weathering and geological K cycling.
639 *Proceedings of the National Academy of Sciences*. 116, 8740-8745.

640 Luck, J.-M., Othman, D. B., Albarède, F., 2005. Zn and Cu isotopic variations in chondrites and iron
641 meteorites: Early solar nebula reservoirs and parent-body processes. *Geochimica et*
642 *Cosmochimica Acta*. 69, 5351-5363.

643 Luck, J. M., Othman, D. B., Barrat, J. A., Albarède, F., 2003. Coupled ⁶³Cu and ¹⁶O excesses in
644 chondrites. *Geochimica et Cosmochimica Acta*. 67, 143-151.

645 Makishima, A., Nakamura, E., 2006. Determination of major/minor and trace elements in silicate samples
646 by ICP-QMS and ICP-SFMS applying isotope dilution-internal standardisation (ID-IS) and multi-
647 stage internal standardisation. *Geostandards and Geoanalytical Research*. 30, 245-271.

648 McSween, H. Y., Ghosh, A., Grimm, R. E., Wilson, L., Young, E. D., 2002. Thermal evolution models of
649 asteroids. *Asteroids III*. 559, 559-572.

650 Mittlefehldt, D. W., 2002. Geochemistry of the ungrouped carbonaceous chondrite Tagish Lake, the
651 anomalous CM chondrite Bells, and comparison with CI and CM chondrites. *Meteoritics &
652 Planetary Science*. 37, 703-712.

653 Morlok, A., Bischoff, A., Stephan, T., Floss, C., Zinner, E., Jessberger, E. K., 2006. Brecciation and
654 chemical heterogeneities of CI chondrites. *Geochimica et Cosmochimica Acta*. 70, 5371-5394.

655 Morota, T., et al., 2020. Sample collection from asteroid (162173) Ryugu by Hayabusa2: Implications for
656 surface evolution. *Science*. 368, 654-659.

657 Moynier, F., et al., 2022. The Solar System calcium isotopic composition inferred from Ryugu samples.
658 *Geochemical Perspectives Letters*. 24, 1-6.

659 Moynier, F., et al., 2021. Potassium isotopic composition of various samples using a dual-path collision
660 cell-capable multiple-collector inductively coupled plasma mass spectrometer, Nu instruments
661 Sapphire. *Chemical Geology*. 571, 120144.

662 Nakamura, E., et al., 2022. On the origin and evolution of the asteroid Ryugu: A comprehensive
663 geochemical perspective. *Proceedings of the Japan Academy, Series B*. 98, 227-282.

664 Nakamura, N., 1974. Determination of REE, Ba, Fe, Mg, Na and K in carbonaceous and ordinary
665 chondrites. *Geochimica et Cosmochimica Acta*. 38, 757-775.

666 Naraoka, H., et al., 2023. Soluble organic molecules in samples of the carbonaceous asteroid (162173)
667 Ryugu. *Science*. 379, eabn9033.

668 Nichiporuk, W., Moore, C. B., 1974. Lithium, sodium and potassium abundances in carbonaceous
669 chondrites. *Geochimica et Cosmochimica Acta*. 38, 1691-1701.

670 Nie, N. X., et al., 2021. Imprint of chondrule formation on the K and Rb isotopic compositions of
671 carbonaceous meteorites. *Science Advances*. 7, eabl3929.

672 Palme, H., Lodders, K., Jones, A., 2014 2.2 - Solar System abundances of the elements. In: H. D.
673 Holland, K. K. Turekian, (Eds.), *Treatise on Geochemistry (Second Edition)*. Elsevier, Oxford,
674 pp. 15-36.

675 Palme, H., Zipfel, J., 2022. The composition of CI chondrites and their contents of chlorine and bromine:
676 Results from instrumental neutron activation analysis. *Meteoritics & Planetary Science*. 57, 317-
677 333.

678 Paquet, M., et al., 2022. Contribution of Ryugu-like material to Earth's volatile inventory by Cu and Zn
679 isotopic analysis. *Nature Astronomy*.

680 Pilorget, C., et al., 2022. First compositional analysis of Ryugu samples by the MicrOmega hyperspectral
681 microscope. *Nature Astronomy*. 6, 221-225.

682 Pringle, E. A., Moynier, F., 2017. Rubidium isotopic composition of the Earth, meteorites, and the Moon:
683 Evidence for the origin of volatile loss during planetary accretion. *Earth and Planetary Science
684 Letters*. 473, 62-70.

685 Pringle, E. A., Moynier, F., Beck, P., Paniello, R., Hezel, D. C., 2017. The origin of volatile element
686 depletion in early solar system material: Clues from Zn isotopes in chondrules. *Earth and
687 Planetary Science Letters*. 468, 62-71.

688 Rubin, A. E., 2012. Collisional facilitation of aqueous alteration of CM and CV carbonaceous chondrites.
689 *Geochimica et Cosmochimica Acta*. 90, 181-194.

690 Rubin, A. E., Zolensky, M. E., Bodnar, R. J., 2002. The halite-bearing Zag and Monahans (1998)
691 meteorite breccias: Shock metamorphism, thermal metamorphism and aqueous alteration on the
692 H-chondrite parent body. *Meteoritics & Planetary Science*. 37, 125-141.

693 Sugita, S., et al., 2019. The geomorphology, color, and thermal properties of Ryugu: Implications for
694 parent-body processes. *Science*. 364, eaaw0422.
695 Tachibana, S., et al., 2022. Pebbles and sand on asteroid (162173) Ryugu: In situ observation and
696 particles returned to Earth. *Science*. 375, 1011-1016.
697 Teng, F.-Z., Hu, Y., Ma, J.-L., Wei, G.-J., Rudnick, R. L., 2020. Potassium isotope fractionation during
698 continental weathering and implications for global K isotopic balance. *Geochimica et*
699 *Cosmochimica Acta*. 278, 261-271.
700 Tomeoka, K., Buseck, P. R., 1988. Matrix mineralogy of the Orgueil CI carbonaceous chondrite.
701 *Geochimica et Cosmochimica Acta*. 52, 1627-1640.
702 Tuller-Ross, B., Marty, B., Chen, H., Kelley, K. A., Lee, H., Wang, K., 2019a. Potassium isotope
703 systematics of oceanic basalts. *Geochimica et Cosmochimica Acta*. 259, 144-154.
704 Tuller-Ross, B., Savage, P. S., Chen, H., Wang, K., 2019b. Potassium isotope fractionation during
705 magmatic differentiation of basalt to rhyolite. *Chemical Geology*. 525, 37-45.
706 Von Michaelis, H., Ahrens, L. H., Willis, J. P., 1969. The composition of stony meteorites. II. The
707 analytical data and an assessment of their quality. *Earth and Planetary Science Letters*. 5, 387-
708 394.
709 Wang, K., Close, H. G., Tuller-Ross, B., Chen, H., 2020. Global Average Potassium Isotope Composition
710 of Modern Seawater. *ACS Earth and Space Chemistry*. 4, 1010-1017.
711 Wang, K., Jacobsen, S. B., 2016. Potassium isotopic evidence for a high-energy giant impact origin of the
712 Moon. *Nature*. 538, 487-490.
713 Wang, K., Peucker-Ehrenbrink, B., Chen, H., Lee, H., Hasenmueller, E. A., 2021. Dissolved potassium
714 isotopic composition of major world rivers. *Geochimica et Cosmochimica Acta*. 294, 145-159.
715 Wiik, H. B., 1956. The chemical composition of some stony meteorites. *Geochimica et Cosmochimica*
716 *Acta*. 9, 279-289.
717 Xu, Y.-K., et al., 2019. Potassium isotopic compositions of international geological reference materials.
718 *Chemical Geology*. 513, 101-107.
719 Yada, T., et al., 2022. Preliminary analysis of the Hayabusa2 samples returned from C-type asteroid
720 Ryugu. *Nature Astronomy*. 6, 214-220.
721 Yokoyama, T., et al., 2022. Samples returned from the asteroid Ryugu are similar to Ivuna-type
722 carbonaceous meteorites. *Science*. 0, eabn7850.
723 Zeng, H., et al., 2019. Ab Initio Calculation of Equilibrium Isotopic Fractionations of Potassium and
724 Rubidium in Minerals and Water. *ACS Earth and Space Chemistry*. 3, 2601-2612.
725 Zolensky, M., McSween, H. Y., 1988 Aqueous alteration. In: J. F. Kerridge, M. S. Matthews, (Eds.),
726 *Meteorites and the Early Solar System*, pp. 114-143.
727 Zolensky, M. E., et al., 1999. Asteroidal water within fluid inclusion-bearing halite in an H5 chondrite,
728 Monahans (1998). *Science*. 285, 1377-1379.

Declaration of interests

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.